

## **BAB V**

### **KESIMPULAN DAN SARAN**

#### **A. Kesimpulan**

Berdasarkan dari hasil penelitian yang telah dilakukan dapat diperoleh kesimpulan bahwa:

Pertama, *Solid Lipid Nanoparticles* (SLN) myrisetin dapat dibuat menggunakan kombinasi metode emulsifikasi pelarut dan sonikasi.

Kedua, myrisetin dalam bentuk SLN tidak stabil pada minggu kedua sudah memisah terbentuk endapan irreversibel.

Ketiga, karakterisasi myrisetin dalam SLN dengan lipid yang terpilih yaitu Apifil 2% dapat menghasilkan ukuran partikel terkecil  $105,5 \pm 0,70$ ; potensial zeta  $-20,52$  mV, dapat menghasilkan efisiensi penjerapan sebesar 73,56% dan memiliki aktifitas antioksidan dengan  $IC_{50}$  38,77 ppm.

#### **B. Saran**

Penelitian ini masih banyak kekurangan, maka perlu dilakukan penelitian lebih lanjut mengenai :

Pertama, perlu dilakukan optimasi surfaktan yang lebih beragam.

Kedua, perlu dilakukan uji morfologi untuk mengetahui bentuk sediaan dari SLN.

Ketiga, pengembangan dalam bentuk sediaan obat.

## DAFTAR PUSTAKA

- Annette ZMH, Schwarz C, Wolfgang M. 1998. Solid lipid nanoparticles (SLN) for controlled drug delivery-drug release and release mechanism. *Eur J Pharm Biopharm* 45(2): 149-155.
- Bae S, Suh HJ. 2007. Antioxidant Activities Of Five Different Mulberry Cultivars In Korea. *LWT Food Sci Technol.*;40:955–62. [[Ref list](#)]
- Bennett, C.J.; Caldwell, S.T.; McPhail, D.B.; Morrice, P.C.; Duthie, G.G.; Hartley, R.C. 2004, *Potential Therapeutic Antioxidants That Combine The Radical Scavenging Ability Of Myrisetin And The Lipophilic Chain Of Vitamin E To Effectively Inhibit Microsomal Lipid Peroxidation. Bioorg. Med. Chem.*, 12, 2079–2098. [[Google Scholar](#)] [[CrossRef](#)] [[PubMed](#)]
- Bhattacharjee A (2013) *Solid Lipid Nanoparticles Technology As A Novel Platform For Delivery Of Drugs*. *Indo Am j pharm Res* 3(5): 4079-4097.
- Bonifácio et al, 2014, Nanotechnology-based drug delivery systems and herbal medicines: a review, *International Journal of Nanomedicine*, Brazil.
- C. Shah, V. Shah, U. Upadhyay. 2011. Solid Lipid Nanoparticles: A Review *Current Pharma Research*, 1(4), 351-368.
- Chen XQ, Antman MD, Gesenberg C, Gudmundsson OS, 2006, *Discovery pharmaceutics—challenges and opportunities*, *AAPS J* 8: E402–408.
- Chen, C.; Zhang, B.; Fu, X.; You, L. J.; Abbasi, A. M.; Liu, R. H. 2016. *The digestibility of mulberry fruit polysaccharides and its impact on lipolysis under simulated saliva, gastric and intestinal conditions*. *Food Hydrocolloids*, 58, 171–178.
- Couvreur, P., Barratt, G., Fattal, E., Legrand, P., Vauthier, C., 2002. Nanocapsule technology: a review. *Crit. Rev. Ther. Drug Carrier Syst.* 19, 99–134.
- Graf A et al. 2008 Protein delivery using nanoparticles based on microemulsions with different structure types. *Eur J Pharm Sci*; 33(4–5):434–444.
- Hagenacker, T.; Hillebrand, I.; Wissmann, A.; Büsselberg, D.; Schäfers, M. 2010. Anti-Allodynic Effect Of The Flavonoid Myrisetin In A Rat Model Of Neuropathic Pain: Involvement Of P38 And Protein Kinase C Mediated Modulation Of Ca<sup>2+</sup> Channels. *Eur. J. Pain*, 14, 992–998. [[Google Scholar](#)] [[CrossRef](#)] [[PubMed](#)]
- Harmita. 2004. Petunjuk Pelaksanaan Validasi Metode Dan Cara Perhitungannya. *Majalah Ilmu Kefarmasian*, Vol. I, No.3, Desember, 117 – 135.

- Hu X, Fan W, Yu Z, Lu Y, Qi J, Zhang J, Dong X, Zhao W, Wu W. 2016. Evidence Does Not Support Absorption Of Intact Solid Lipid Nanoparticles Via Oral Delivery. *Nanoscale*. Apr 7; 8(13):7024-35. [[PubMed](#)] [[Ref list](#)]
- Iqbal, M.A., Md, S., Sahni, J.K., Baboota, S., Dang, S., Ali, J., 2012. Nanostructured Lipid Carriers System: Recent Advances In Drug Delivery. *J. Drug Target*. 20, 813–830.
- Kobus-Cisowska J, Gramza-Michalowska A, Kmiecik D, Flaczyk E, Korczak J. 2013, *Mulberry Fruit As An Antioxidant Component In Muesli*. *Agric Sci*.;4:130.
- Lin CH, Chen CH, Lin ZC, Fang JY. 2017. Recent Advances In Oral Delivery Of Drugs And Bioactive Natural Products Using Solid Lipid Nanoparticles As The Carriers. *J Food Drug Anal* 25(2): 219-234
- Lu Y, Wang L, Wei H, Yang ZQ, Wang W. Structure-Activity Relationship Of Flavonoids In Antioxidant Activity. *Food Sci*. 2006;27:233–7. [[Ref List](#)]
- Mohanraj VJ, Chen Y. 2006. Nanoparticles – A Review.. Evidence Does Not Support Absorption Of Intact Solid Lipid Nanoparticles Via Oral Delivery. *Trop J Pharm Res*.5:561–73.
- Neves, A.R.; Lucio, M.; Martins, S.; Lima, J.L.; Reis, S. 2013. Novel Resveratrol Nanodelivery Systems Based On Lipid Nanoparticles To Enhance Its Oral Bioavailability. *Int. J. Nanomed*. 2013, 8, 177–187. [[Google Scholar](#)]
- Pel, P.; Chae, H. S.; Nhoek, P.; Kim, Y. M.; Chin, Y. W. 2017, Chemical Constituents With Proprotein Convertase Subtilisin/Kexin Type 9 Mrna Expression Inhibitory Activity From Dried Immature Morus Alba Fruits. *J.Agric. Food Chem.*, 65, 5316–5321.
- Phytochemical And Antioxidant Properties Of Anthocyanin-Rich Zgen Mustafa, Sedat Serc, Cemal Kaya, 2008, *Morus Nigra And Morus Rubra Fruits*, *Scientia Horticulturae*, Turki: University Of Gaziosmanpas
- Qingxia Yuan And Longyan Zhao, 2017, The Mulberry (Morus Alba L.) Fruit • A Review Of Characteristic Components And Health Benefits
- Rahul N, Arunkumar KS, Priya KV (2011) Recent Advances In Solid Lipid Nanoparticle Based Drug Delivery Systems, *J Biomed Sci And Res* 3(2): 368-384.
- Raman Sivakumar Thasma, 2016. *In Vitro And In Vivo* Antioxidant Activity Of Flavonoid Extracted From Mulberry Fruit (*Morus Alba L.*) , *Pharmacogn Mag, China* 12(46): 128–133

- Review Solid Lipid Nanoparticles: Production, Characterization And Applications.
- Review. Mehnert W, Mäder K, 2001, *Solid Lipid Nanoparticles: Production, Characterization And Applications*. *Adv Drug Deliv Rev*. Apr 25; 47(2-3):165-96. [[Pubmed](#)] [[Ref List](#)]
- S. Nema, R.J. Brendel. 2011. Excipients And Their Role In Approved Injectable Products: Current Usage And Future Directions. *PDA J Pharm Sci And Tech*, **65**, 287–332
- Schroeder A, Kost J, Barenholz Y. 2009. Ultrasound, liposomes, and drug delivery: principles for using ultrasound to control the release of drugs from liposomes. *Chemistry and Physics of Lipids* 162:1–16.
- Sivakumar TR, Ajaykrishna PG, Fang Y, Ren ZX, Chen C, Jin C, Et Al. 2015. *Comparative Analysis Of The Chemical Composition Of Different Mulberry Fruit Varieties*. *Sericologia*.;55:221–8.
- Strambeanu, N., Demetrovici, L., & Dragos, D. 2015. *Natural Sources Of Nanoparticles. Nanoparticles' Promises And Risks* (Pp. 9-19). Springer International Publishing.
- Uner M, Yener G, 2007, Importance Of Solid Lipid Nanoparticles (SLN) In Various Administration Routes And Future Perspectives. *Int J Nanomedicine*.; 2(3):289-300. [[Pubmed](#)] [[Ref List](#)]
- Varshosaz, J., Hassanzadeh, F., Sadeghi, H., & Khadem, M. 2012, Galactosylated Nanostructured Lipid Carriers For Delivery Of 5-FU To Hepatocellular Carcinoma. *Journal Of Liposome Research*, 22(3), 224–236
- Weng C-J, Yen G-C. 2012, *Flavonoids, A Ubiquitous Dietary Phenolic Subclass, Exert Extensive In Vitro Anti-Invasive And In Vivo Anti-Metastatic Activities*. *Cancer And Metastasis Reviews*;31(1–2):323–51.
- Wu P, Ma G, Li N, Deng Q, Yin Y, Huang R, 2015. *Investigation Of In Vitro And In Vivo Antioxidant Activities Of Flavonoids Rich Extract From The Berries Of Rhodomyrtus Tomentosa(Ait.) Hassk*. *Food Chem*. Apr 15; 173():194-202.
- Xue M, Zhao Y, Li X-J, Jiang Z-Z, Zhang L, Liu S-H, Et Al. 2012, Comparison Of Toxicokinetic And Tissue Distribution Of Triptolide-Loaded Solid Lipid Nanoparticles Vs Free Triptolide In Rats. *Eur J Pharm Sci*.47(4):713–7

- Yang X, Yang L, Zheng H, 2010, *Hypolipidemic And Antioxidant Effects Of Mulberry (Morus Alba L.) Fruit In Hyperlipidaemia Rats*. Food Chem Toxicol.;48:2374–9. [[Pubmed](#)]
- Yuan Y, Xing J, Wang L, Yao J, Wang X. 2014. Study On The Pharmacokinetic Behavior Of *Dracocephalum Moldavica L.* Flavonoids In Rat. *J Shehezi Univ.*;32:69–72. Chinese.
- Yuan, Q. X.; Xie, Y. F.; Wang, W.; Yan, Y. H.; Ye, H.; Jabbar, S.; Zeng, X. X, 2015, *Extraction Optimization, Characterization And Antioxidant Activity In Vitro Of Polysaccharides From Mulberry (Morus Alba L.) Leaves*. Carbohydr. Polym., 128, 52–62.
- Yulia 2010 “Validasi Metode” Diktat Validasi Metode, Pusat Penelitian Kimia-LIPI, Bandung.
- Zhou, M.; Chen, Q. Q.; Bi, J. F.; Wang, Y. X.; Wu, X. Y, 2017, *Degradation Kinetics Of Cyanidin 3-O-Glucoside And Cyanidin 3-Orutinoside During Hot Air And Vacuum Drying In Mulberry (Morus Alba L.) Fruit: A Comparative Study Based On Solid Food System*. Food Chem., 229, 574–579.

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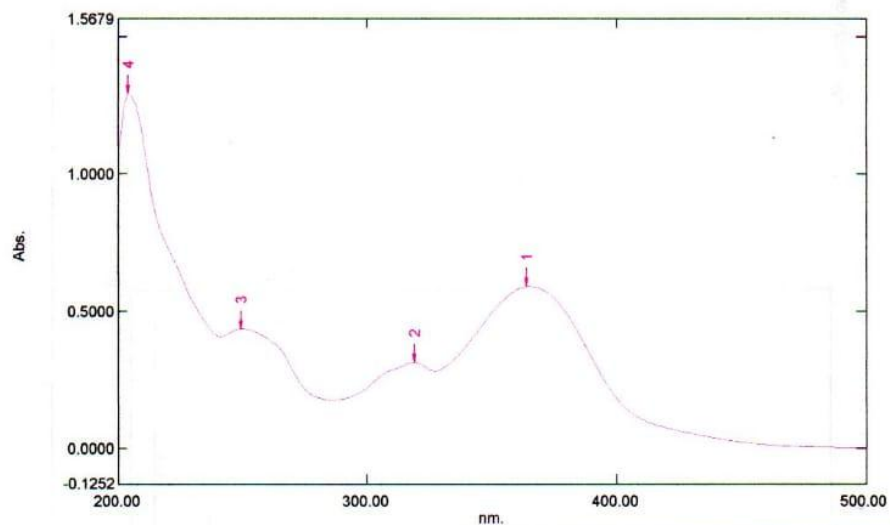
## Lampiran 1. Penentuan Panjang gelombang dan Pembuatan kurva baku

### a. Penentuan panjang gelombang maksimum

#### Spectrum Peak Pick Report

03/23/2019 08:30:25 AM

Data Set: File\_190323\_082649 - RawData



[Measurement Properties]  
 Wavelength Range (nm.): 200.00 to 500.00  
 Scan Speed: Medium  
 Sampling Interval: 1.0  
 Auto Sampling Interval: Disabled  
 Scan Mode: Single

[Instrument Properties]  
 Instrument Type: UV-1800 Series  
 Measuring Mode: Absorbance  
 Slit Width: 1.0 nm  
 Light Source Change Wavelength: 340.0 nm  
 S/R Exchange: Normal

[Attachment Properties]  
 Attachment: None

[Operation]  
 Threshold: 0.0010000  
 Points: 4  
 InterPolate: Disabled  
 Average: Disabled

[Sample Preparation Properties]  
 Weight: 359  
 Volume:  
 Dilution:  
 Path Length:  
 Additional Information:

No.	P/V	Wavelength	Abs.	Description
1	⊕	364.00	0.5917	
2	⊕	319.00	0.3150	
3	⊕	249.00	0.4369	
4	⊕	204.00	1.2942	
5	⊕	327.00	0.2818	
6	⊕	286.00	0.1763	
7	⊕	241.00	0.4067	

**b. Operating Time****Kinetics Data Print Report**

03/06/2019 08:28:48 AM

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Time ( Minute )	RawData ...
0.000	0.557
1.000	0.558
2.000	0.557
3.000	0.558
4.000	0.557
5.000	0.557
6.000	0.557
7.000	0.557
8.000	0.557
9.000	0.557
10.000	0.557
11.000	0.557
12.000	0.556
13.000	0.556
14.000	0.556
15.000	0.557
16.000	0.556
17.000	0.556
18.000	0.556
19.000	0.556
20.000	0.555
21.000	0.555
22.000	0.556
23.000	0.556
24.000	0.555
25.000	0.556
26.000	0.556
27.000	0.556
28.000	0.556
29.000	0.555
30.000	0.555



**c. Linearitas**

- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 0,5 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 2,35 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 1 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 4,7 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 2 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 9,4 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 2,5 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 11,75 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 3 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 14,1 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 3,5 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 16,45 \text{ ppm}$
- $c_1 \times v_1 = c_2 \times v_2$   
 $47 \text{ ppm} \times 4 \text{ ml} = c_2 \times 10 \text{ ml}$   
 $c_2 = 18,8 \text{ ppm}$

Konsentrasi(ppm)	Absorbansi
2,35	0,145
4,7	0,285
9,4	0,541
11,75	0,72
14,1	0,843
16,45	0,981
18,8	1,143

Persamaan linear anatar konsnetrasi (ppm) dan serapan diperoleh nilai:

$$a = -0,00280$$

$$b = 0,06032$$

$$r = 0,99935$$

$$y = a + bx$$

$$y = -0,00280 + 0,06032x$$

Keterangan:

x= konsntrasi (ppm)

y= serapan

hasil linearitas yang diperoleh  $r = 0,99935$ ; sehingga dapat disimpulkan bahwa data linear.

#### d. Presisi

Konsemtrasi	Absorbansi	Y
9,4	0,585	9,74515
9,4	0,582	9,69542
9,4	0,582	9,69542
9,4	0,582	9,69542
9,4	0,585	9,74515
9,4	0,586	9,76173
9,4	0,587	9,77831
9,4	0,588	9,79489
9,4	0,58	9,66226
9,4	0,579	9,64568
	RATA-RATA	9,72194
	SD	0,05016
	CV	0,52%

$$SD = \sqrt{\frac{\sum(x - \bar{x})^2}{n - 1}}$$

$$CV = \frac{SD}{X}$$

## e. Akurasi

konsentrasi sebenarnya	Absorbansi	Konsentrasi terukur	recovery	Rata-rata
4,7	0,276	4,622	98%	98%
4,7	0,272	4,556	97%	
4,7	0,274	4,589	98%	
9,4	0,561	9,347	99%	100%
9,4	0,562	9,364	100%	
9,4	0,564	9,397	100%	
11,75	0,725	12,066	103%	103%
11,75	0,729	12,133	103%	
11,75	0,73	12,149	103%	

$$\% \text{ recovery} = \frac{\text{konsentrasi terukur}}{\text{konsentrasi sebenarnya}} \times 100\%$$

## f. Penentuan LOD dan LOQ

Konsentrasi	Absorbansi (y)	$\hat{Y}$	$ Y-\hat{Y} $	$ Y-\hat{Y} ^2$
2,35	0,145	0,13894	0,00606	0,00004
4,7	0,285	0,28069	0,00431	0,00002
9,4	0,541	0,56418	-0,02318	0,00054
11,75	0,72	0,70593	0,01407	0,00020
14,1	0,843	0,84767	-0,00467	0,00002
16,45	0,981	0,98942	-0,00842	0,00007
18,8	1,143	1,13117	0,01183	0,00014
			jumlah	0,00102

$$S_{x/y} = \sqrt{\frac{\Sigma(Y-\hat{Y})^2}{n-2}}$$

$$S_{x/y} = \sqrt{\frac{0,00102}{5}}$$

$$S_{x/y} = \sqrt{0,000204}$$

$$S_{x/y} = 0,0143$$

$$LOD = 3,3 \times \frac{S_{y/x}}{sl}$$

$$LOD = 3,3 \times \frac{0,0143}{0,06032}$$

$$LOD = 0,782$$

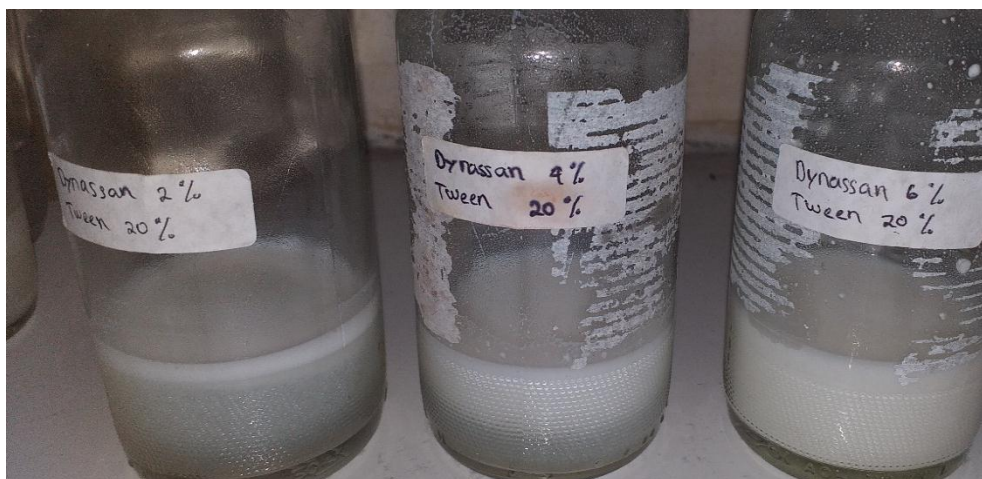
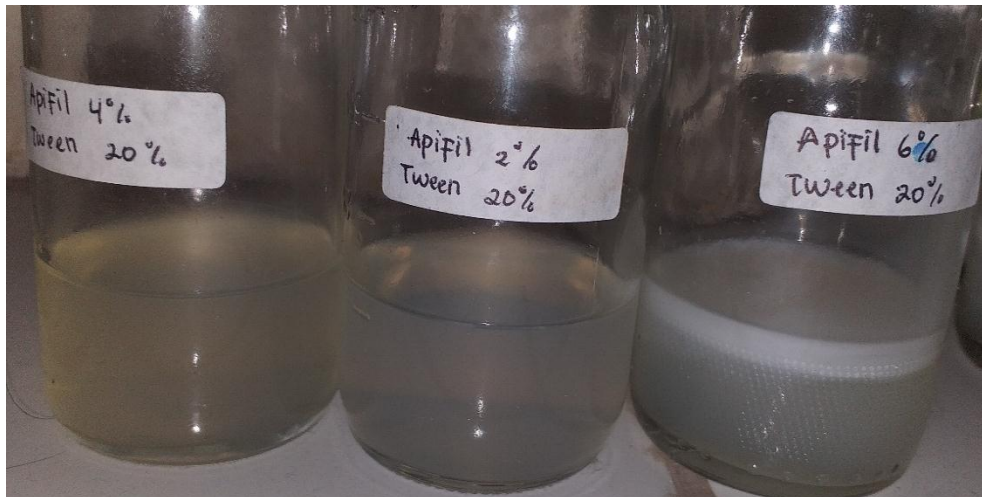
$$LOQ = 10 \times \frac{s_{y/x}}{s_l}$$

$$LOQ = 10 \times \frac{0,0143}{0,06032}$$

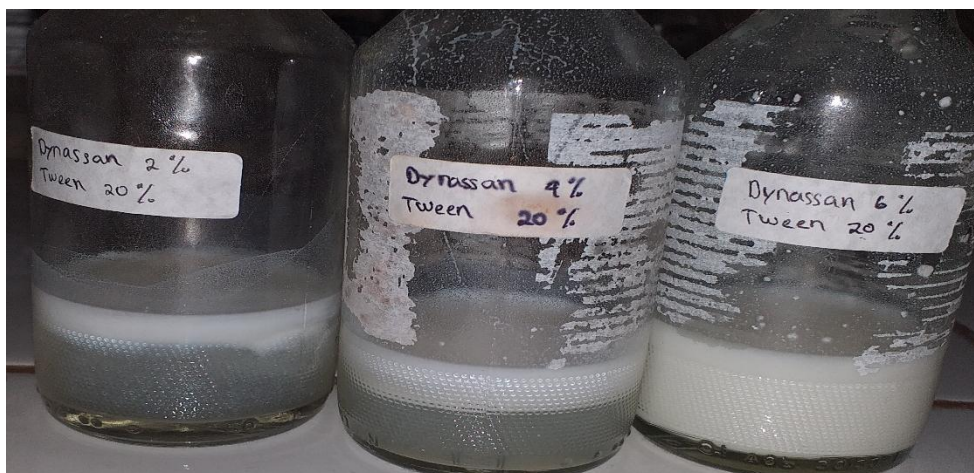
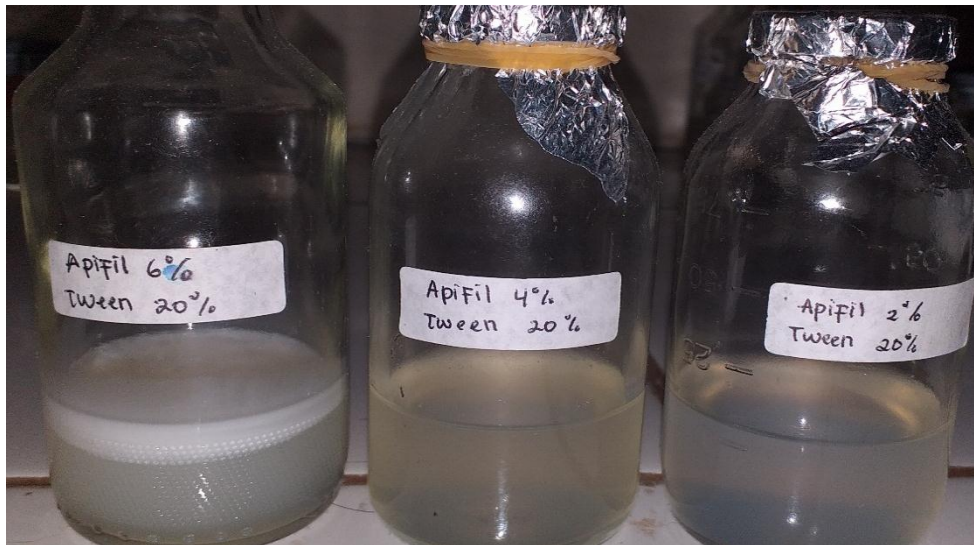
$$LOQ = 2,371$$

## Lampiran 2. Skrining lipid

Minggu pertama



Minggu kedua

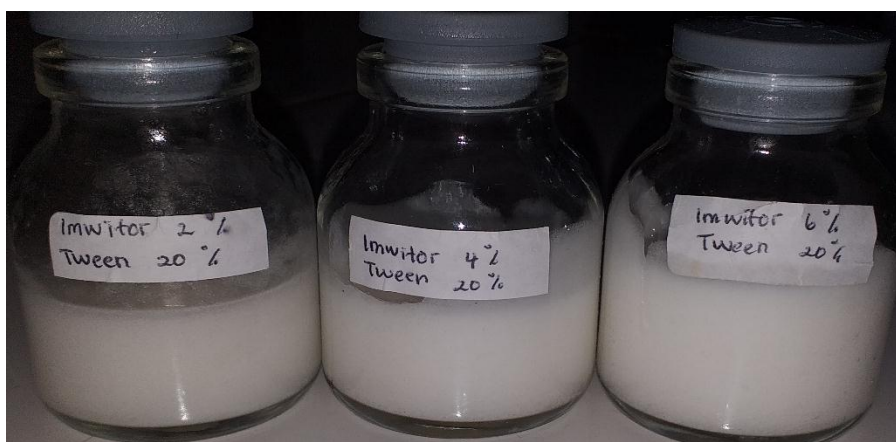




Minggu ketiga



## Minggu keempat





### Lampiran 3. Ukuran Partikel dan distribusi

#### Formula 7

#### Replikasi 1

## Size Distribution Report by Intensity

v2.2



### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 2% 1

SOP Name: mansettings.nano

General Notes:

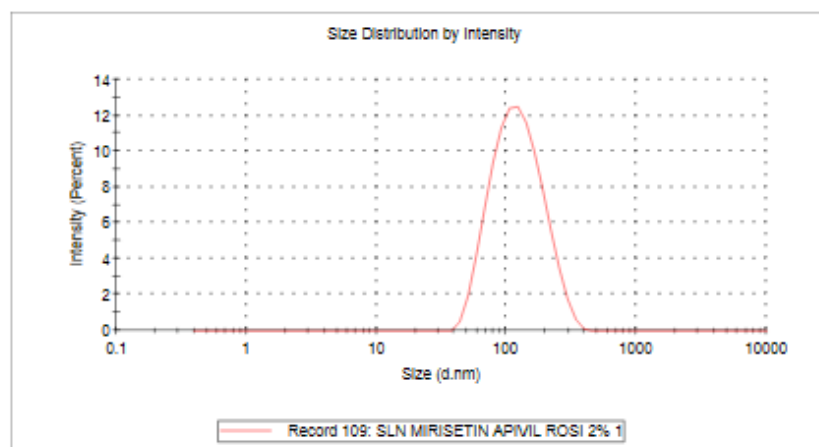
File Name:	NANOFITOSOM.dts	Dispersant Name:	Water
Record Number:	109	Dispersant RI:	1,330
Material RI:	1,52	Viscosity (cP):	0,8872
Material Absorbtion:	0,100	Measurement Date and Time:	26 April 2019 16:52:35

### System

Temperature (°C):	25,0	Duration Used (s):	60
Count Rate (kcps):	393,0	Measurement Position (mm):	4,65
Cell Description:	Disposable sizing cuvette	Attenuator:	4

### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm):</b> 105,5	<b>Peak 1:</b> 131,1	100,0	58,25
<b>Pdl:</b> 0,224	<b>Peak 2:</b> 0,000	0,0	0,000
<b>Intercept:</b> 0,906	<b>Peak 3:</b> 0,000	0,0	0,000

Result quality **Good**

## Replikasi 2

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 2% 2

SOP Name: mansettings.nano

General Notes:

File Name: NANOFITOSOM.dts

Dispersant Name: Water

Record Number: 110

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 26 April 2019 16:54:38

#### System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kops): 380,2

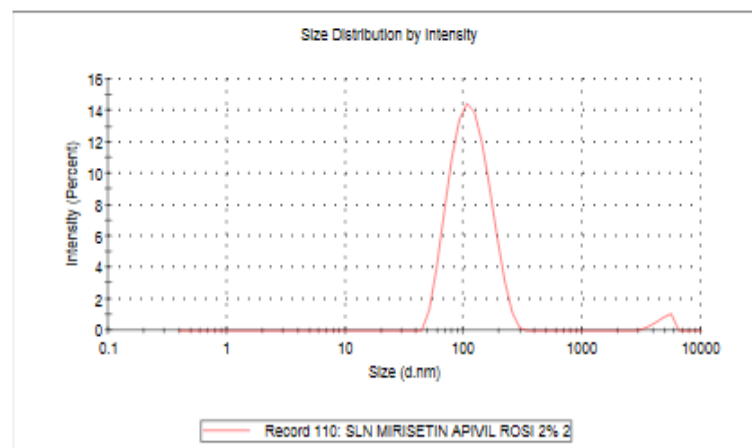
Measurement Position (mm): 4,65

Cell Description: Disposable sizing cuvette

Attenuator: 4

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm):</b> 104,4	<b>Peak 1:</b> 118,3	97,1	43,67
<b>PdI:</b> 0,239	<b>Peak 2:</b> 4795	2,9	726,1
<b>Intercept:</b> 0,905	<b>Peak 3:</b> 0,000	0,0	0,000

Result quality **Good**

## Replikasi 3

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 2% 3

SOP Name: mansettings.nano

General Notes:

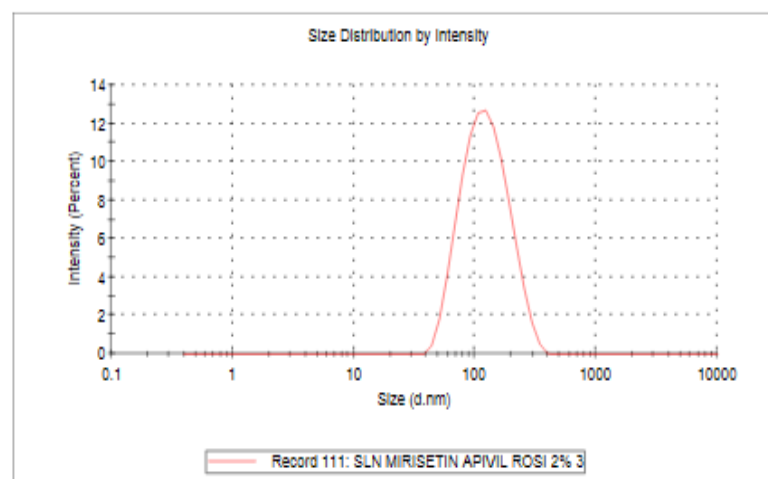
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 111	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 16:58:42

#### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 389,2	Measurement Position (mm): 4,65
Cell Description: Disposable sizing cuvette	Attenuator: 4

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 105,7</b>	<b>Peak 1: 130,9</b>	<b>100,0</b>	<b>56,90</b>
<b>Pdl: 0,234</b>	<b>Peak 2: 0,000</b>	<b>0,0</b>	<b>0,000</b>
<b>Intercept: 0,910</b>	<b>Peak 3: 0,000</b>	<b>0,0</b>	<b>0,000</b>

Result quality **Good**

## Formulasi 8

### Replikasi 1

## Size Distribution Report by Intensity

v2.2



### Sample Details

Sample Name: SLN MIRISETIN APIVL ROSI 3% 1

SOP Name: mansettings.nano

General Notes:

File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 106	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 16:42:39

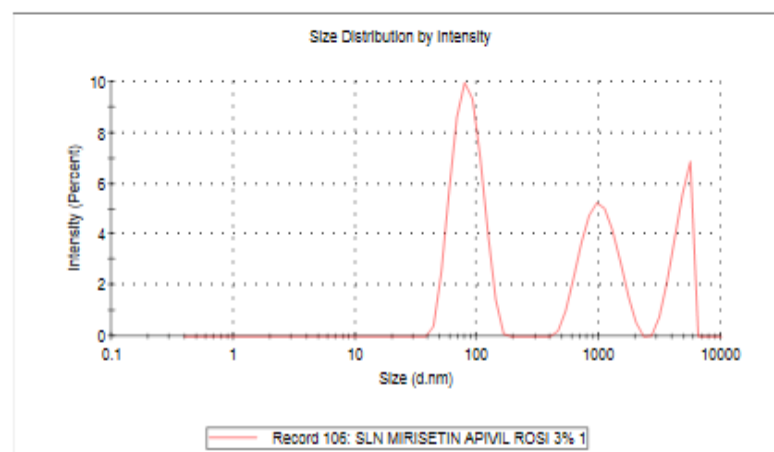
### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 270,6	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 196,9</b>	<b>Peak 1:</b> 84,71	49,2	22,50
<b>Pdl: 0,781</b>	<b>Peak 2:</b> 1040	31,5	333,3
<b>Intercept: 0,796</b>	<b>Peak 3:</b> 4732	19,3	758,1

Result quality **Refer to quality report**



## Replikasi 2

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIML ROSI 3% 2

SOP Name: mansettings.nano

General Notes:

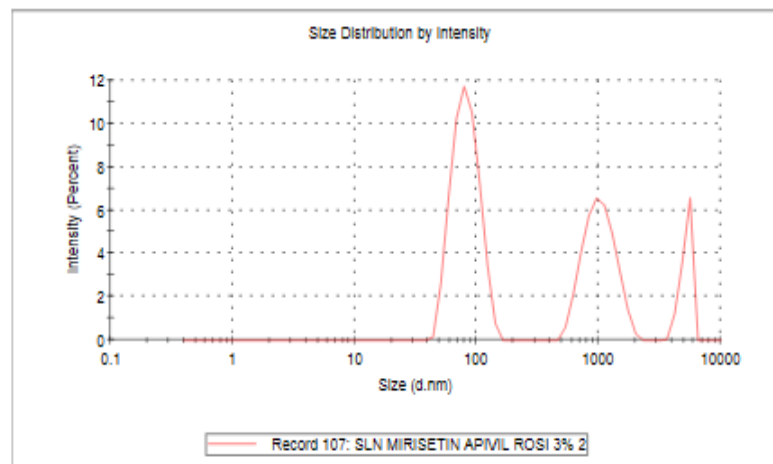
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 107	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 16:44:52

#### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 279,9	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 212,8</b>	<b>Peak 1: 82,65</b>	53,3	20,28
<b>Pdl: 0,720</b>	<b>Peak 2: 1039</b>	35,2	300,2
<b>Intercept: 0,786</b>	<b>Peak 3: 5160</b>	11,5	503,5

Result quality **Refer to quality report**

## Replikasi 3

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 3% 3

SOP Name: mansettings.nano

General Notes:

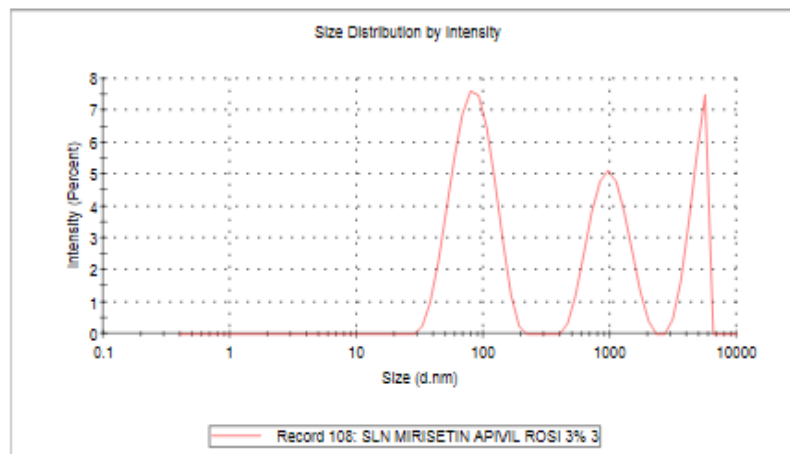
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 108	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 16:47:06

#### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 293,2	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 217,8</b>	<b>Peak 1: 88,06</b>	50,5	30,47
<b>Pdl: 0,732</b>	<b>Peak 2: 1009</b>	30,7	326,0
<b>Intercept: 0,776</b>	<b>Peak 3: 4828</b>	18,8	717,3

Result quality **Refer to quality report**

## Formulasi 10

### Replikasi 1

#### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIML ROSI 4% 1

SOP Name: mansettings.nano

General Notes:

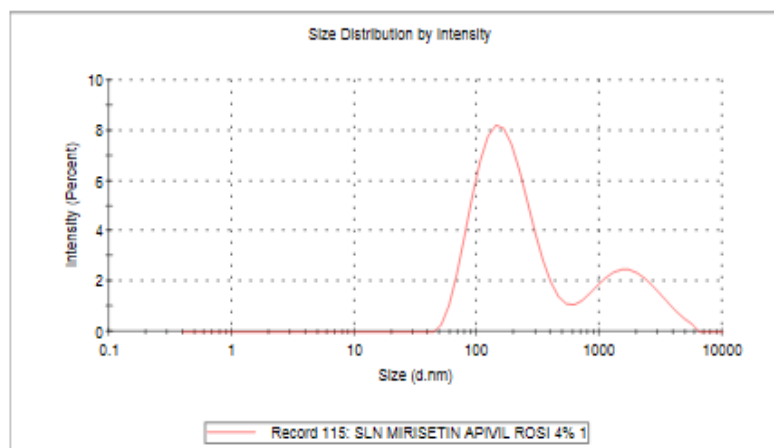
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 115	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 17:15:16

#### System

Temperature (°C): 25,0	Duration Used (s): 100
Count Rate (kcps): 99,4	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 2

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 187,4</b>	<b>Peak 1: 188,7</b>	74,4	110,5
<b>Pdl: 0,418</b>	<b>Peak 2: 1898</b>	25,6	1073
<b>Intercept: 0,933</b>	<b>Peak 3: 0,000</b>	0,0	0,000

Result quality **Good**

## Replikasi 2

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 4% 2

SOP Name: mansettings.nano

General Notes:

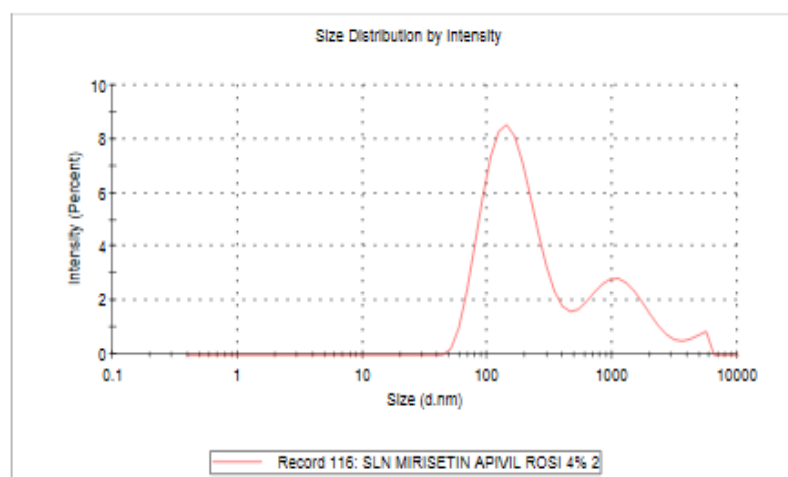
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 116	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 17:18:41

#### System

Temperature (°C): 25,0	Duration Used (s): 100
Count Rate (kcps): 101,1	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 2

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 187,6</b>	Peak 1: 172,0	70,7	88,24
<b>Pdl: 0,425</b>	Peak 2: 1242	26,6	689,2
<b>Intercept: 0,933</b>	Peak 3: 4871	2,6	739,5

Result quality **Good**



## Replikasi 3

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVL ROSI 4% 3

SOP Name: mansettings.nano

General Notes:

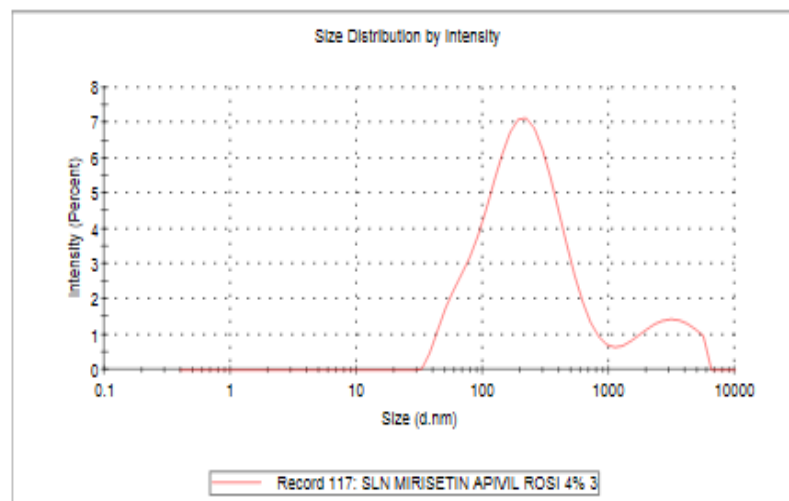
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 117	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorption: 0,100	Measurement Date and Time: 28 April 2019 17:22:08

#### System

Temperature (°C): 25,0	Duration Used (s): 100
Count Rate (kops): 101,1	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 2

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 188,0</b>	<b>Peak 1: 249,4</b>	<b>86,7</b>	<b>187,4</b>
<b>Pdl: 0,446</b>	<b>Peak 2: 2965</b>	<b>13,3</b>	<b>1291</b>
<b>Intercept: 0,929</b>	<b>Peak 3: 0,000</b>	<b>0,0</b>	<b>0,000</b>

Result quality **Good**

## Formulasi 11

### Replikasi 1

#### Size Distribution Report by Intensity

v2.2



##### Sample Details

Sample Name: SLN MIRISETIN APVIL ROSI 6% 1

SOP Name: mansettings.nano

General Notes:

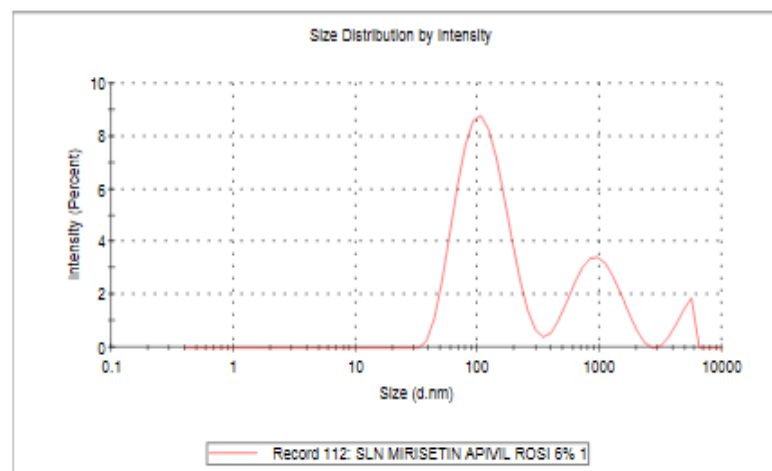
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 112	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 17:03:02

##### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kops): 302,2	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

##### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm):</b> 114,2	<b>Peak 1:</b> 117,5	69,3	53,97
<b>Pdl:</b> 0,756	<b>Peak 2:</b> 975,1	26,0	400,1
<b>Intercept:</b> 0,922	<b>Peak 3:</b> 4857	4,7	691,7

Result quality **Refer to quality report**

## Replikasi 2

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVL ROSI 6% 1

SOP Name: mansettings.nano

General Notes:

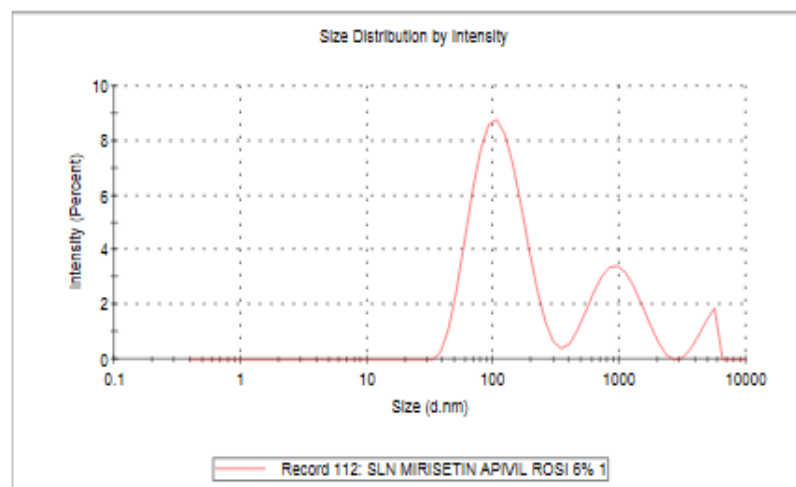
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 112	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 17:03:02

#### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 302,2	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 114,2</b>	Peak 1: 117,5	69,3	53,97
<b>Pdl: 0,756</b>	Peak 2: 975,1	26,0	400,1
<b>Intercept: 0,922</b>	Peak 3: 4857	4,7	691,7

Result quality **Refer to quality report**

## Replikasi 3

### Size Distribution Report by Intensity

v2.2



#### Sample Details

Sample Name: SLN MIRISETIN APIVIL ROSI 6% 3

SOP Name: mansettings.nano

General Notes:

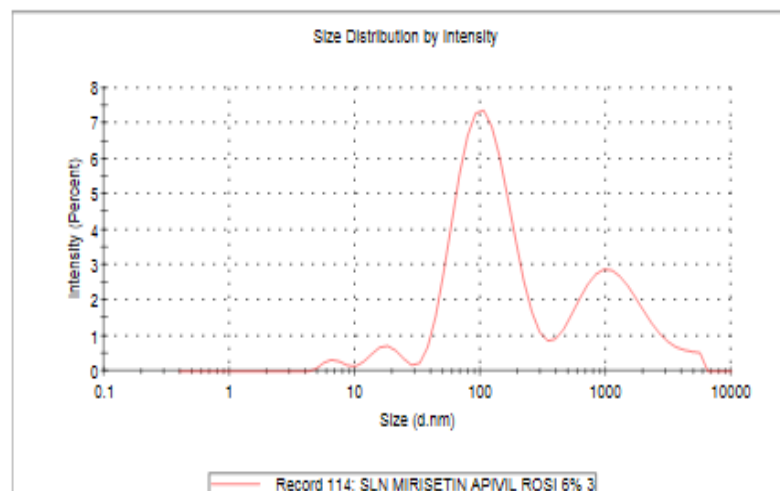
File Name: NANOFITOSOM.dts	Dispersant Name: Water
Record Number: 114	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 26 April 2019 17:07:29

#### System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kops): 299,3	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

#### Results

	Size (d.n...	% Intensity:	St Dev (d.n...
<b>Z-Average (d.nm): 112,8</b>	<b>Peak 1: 119,3</b>	63,5	61,65
<b>Pdl: 0,741</b>	<b>Peak 2: 1447</b>	31,8	1088
<b>Intercept: 0,921</b>	<b>Peak 3: 17,85</b>	3,4	4,607

Result quality **Refer to quality report**

**Lampiran 4. Potensial zeta**

## Lampiran 5. Perhitungan Efisiensi Penjerapan

### FORMULA 7 APIFIL 2%

- Formulasi induk → 200 mg SLN myrisetin/10ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\text{Myrisetin} = 19,2 \text{ mg}$$

$$\text{Eksipien (tween 80+ Apifil 4\%)} = 11000 \text{ mg}$$

$$\% \text{ kadar myrisetin} = \frac{19,2}{11000+19,2} \times 100\% = 0,174\%$$

$$\text{Kadar dalam 200 mg SLN} = 0,174\% \times 200 \text{ mg} = 34,8 \text{ mg}$$

- Perhitungan kadar Myrisetin terjerap menggunakan persamaan linear

$$y = a + bx$$

$$1,539 = -0,00280 + 0,06032x$$

$$1,539 + 0,00280 = 0,06032x$$

$$x = 25,560 \text{ ppm}$$

- $\% \text{ kadar} = \frac{25,560 \text{ ppm}}{20.000 \text{ ppm}} \times 100\% = 0,1278\%$

- Kadar dalam 200 mg SLN myrisetin =  $0,128\% \times 200 \text{ mg} = 25,6 \text{ mg}$

- $\% \text{ Efisiensi Penjerapan} = \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\%$

$$= \frac{25,6 \text{ mg}}{34,8 \text{ mg}} \times 100\% = 73,56\%$$

### FORMULA 8 APIFIL 3%

- Formulasi induk → 200 mg SLN myrisetin/10ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\text{Myrisetin} = 20,8 \text{ mg}$$

$$\text{Eksipien (tween 80+ Apifil 4\%)} = 11500 \text{ mg}$$

$$\% \text{ kadar myrisetin} = \frac{20,8}{11500+20,8} \times 100\% = 0,180\%$$

$$\text{Kadar dalam 200 mg SLN} = 0,180\% \times 200 \text{ mg} = 36 \text{ mg}$$

- Perhitungan kadar Myrisetin terjerap menggunakan persamaan linear

$$y = a + bx$$

$$0,630 = -0,00280 + 0,06032x$$

$$0,630 + 0,00280 = 0,06032x$$

$$x = 10,4907 \text{ ppm}$$

- % kadar =  $\frac{10,4907 \text{ ppm}}{20.000 \text{ ppm}} \times 100\% = 0,052\%$
- Kadar dalam 200 mg SLN myrisetin =  $0,052\% \times 200 \text{ mg} = 10,4 \text{ mg}$
- % Efisiensi Penjerapan =  $\frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\%$   

$$= \frac{10,4 \text{ mg}}{36 \text{ mg}} \times 100\% = 28,89\%$$

#### FORMULA 10 APIFIL 4%

- Formulasi induk → 200 mg SLN myrisetin/10ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\text{Myrisetin} = 19,8 \text{ mg}$$

$$\text{Eksipien (tween 80+ Apifil 5\%)} = 12500 \text{ mg}$$

$$\% \text{ kadar myrisetin} = \frac{19,8}{12500+19,8} \times 100\% = 0,158\%$$

$$\text{Kadar dalam 200 mg SLN} = 0,158\% \times 200 \text{ mg} = 31,6 \text{ mg}$$

- Perhitungan kadar Myrisetin terjerap menggunakan persamaan linear

$$y = a + bx$$

$$0,763 = -0,00280 + 0,06032x$$

$$0,763 + 0,00280 = 0,06032x$$

$$x = 12,6956 \text{ ppm}$$

- % kadar =  $\frac{12,6956 \text{ ppm}}{20.000 \text{ ppm}} \times 100\% = 0,0635\%$
- Kadar dalam 200 mg SLN myrisetin =  $0,0635\% \times 200 \text{ mg} = 12,7 \text{ mg}$
- % Efisiensi Penjerapan  $\frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\%$

$$\frac{12,7 \text{ mg}}{31,6 \text{ mg}} \times 100\% = 40,19\%$$

#### FORMULA 11 APIFIL 5%

- Formulasi induk → 200 mg SLN myrisetin/10 ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\text{Myrisetin} = 20 \text{ mg}$$

$$\text{Eksipien (tween 80+ Apifil 6\%)} = 13000 \text{ mg}$$

$$\% \text{ kadar myrisetin} = \frac{20}{13000+20} \times 100 \% = 0.154 \%$$

$$\text{Kadar dalam 200 mg SLN} = 0.154\% \times 200 \text{ mg} = 30,72 \text{ mg}$$

- Perhitungan kadar Myrisetin terjerap menggunakan persamaan linear

$$y = a + bx$$

$$0,784 = -0,00280 + 0,06032x$$

$$0,763 + 0,00280 = 0,06032x$$

$$x = 13,044 \text{ ppm}$$

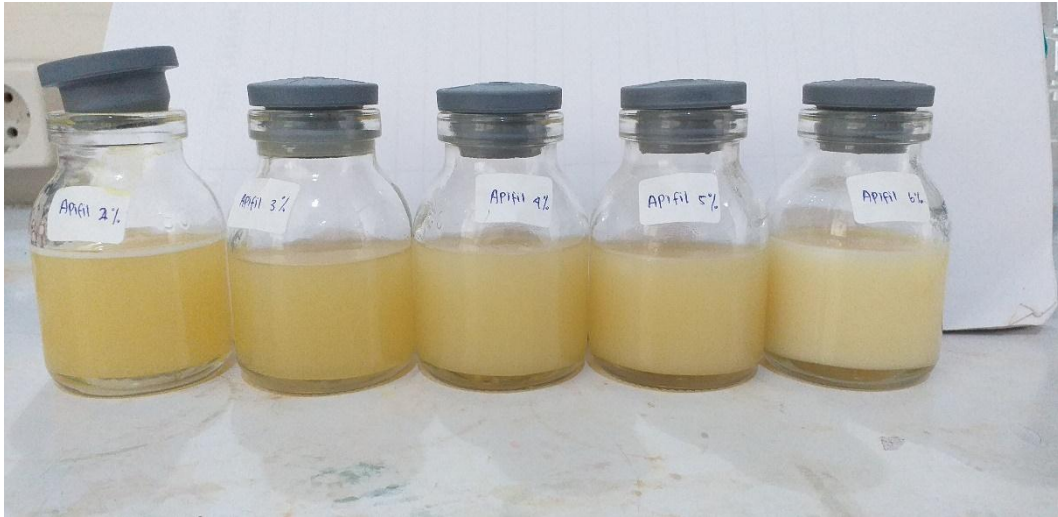
- $\% \text{ kadar} = \frac{13,044 \text{ ppm}}{20.000 \text{ ppm}} \times 100\% = 0,0652\%$
- $\text{Kadar dalam 200 mg SLN myrisetin} = 0,0652\% \times 200 \text{ mg} = 13,044 \text{ mg}$
- $\% \text{ Efisiensi Penjerapan} = \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\%$

$$\frac{13,044 \text{ mg}}{30,72 \text{ mg}} \times 100\% = 42,46\%$$



## Lampiran 6. Stabilitas penyimpanan

Minggu pertama

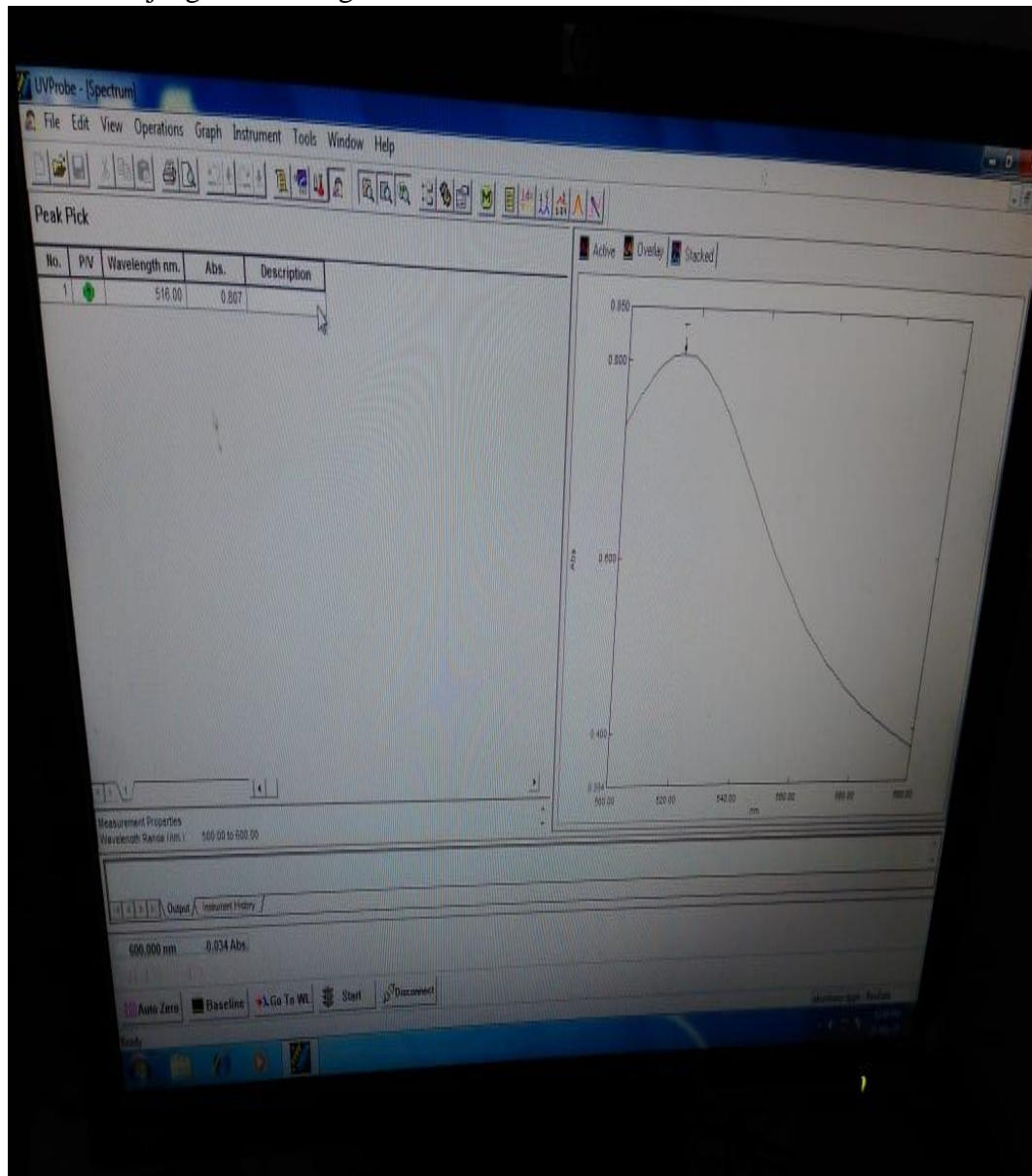


Minggu kedua



## Lampiran 7. Hasil DPPH

### a. Panjang Gelombang maksimal



## b. Operating time

OT Mincehn DPH 45 menit  
4-5-19.

**Kinetics Data Print Report**

05/04/2019 01:13:10 PM

Time ( Minute )	RawData ...
0.000	0.229
1.000	0.229
2.000	0.229
3.000	0.229
4.000	0.229
5.000	0.230
6.000	0.230
7.000	0.230
8.000	0.229
9.000	0.230
10.000	0.229
11.000	0.229
12.000	0.230
13.000	0.230
14.000	0.230
15.000	0.230
16.000	0.230
17.000	0.230
18.000	0.230
19.000	0.230
20.000	0.230
21.000	0.230
22.000	0.230
23.000	0.230
24.000	0.230
25.000	0.230
26.000	0.230
27.000	0.230
28.000	0.230
29.000	0.230
30.000	0.230
31.000	0.230
32.000	0.230
33.000	0.231
34.000	0.231
35.000	0.231
36.000	0.231
37.000	0.230
38.000	0.231
39.000	0.231
40.000	0.231
41.000	0.231
42.000	0.231
43.000	0.231
44.000	0.231
45.000	0.231
46.000	
47.000	
48.000	
49.000	
50.000	

c. Menghitung  $IC_{50}$  zat aktif

konsentrasi	ABS REP 1	ABS REP 2	ABS REP 3
15,53	0,249	0,237	0,24
7,77	0,445	0,442	0,438
3,88	0,526	0,528	0,53
1,94	0,578	0,578	0,577
0,97	0,595	0,593	0,597
<b>a</b>	0,6218	0,6237	0,6245
<b>b</b>	-0,0238	-0,0246	-0,0246
<b>r</b>	-0,9993	-0,9991	-0,9998

$$\%inhibisi = \frac{\text{absorbansi dpvh} - \text{absorbansi sampel}}{\text{absorbansi dpvh}} \times 100\%$$

- Konsentrasi 74,4 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,206}{0,807} \times 100\% = 74,47\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,214}{0,807} \times 100\% = 73,48\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,208}{0,807} \times 100\% = 74,23\%$$

- Konsentrasi 31 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,226}{0,807} \times 100\% = 72,00\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,229}{0,807} \times 100\% = 71,62\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,231}{0,807} \times 100\% = 71,38\%$$

- Konsentrasi 15,5 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,367}{0,807} \times 100\% = 54,52\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,356}{0,807} \times 100\% = 55,89\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,354}{0,807} \times 100\% = 56,13\%$$

- Konsentrasi 7,75 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,484}{0,807} \times 100\% = 40,02\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,487}{0,807} \times 100\% = 39,65\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,482}{0,807} \times 100\% = 35,81\%$$

- Konsentrasi 3,88 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,522}{0,807} \times 100\% = 35,32\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,522}{0,807} \times 100\% = 35,32\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,518}{0,807} \times 100\% = 35,81\%$$

Konsentrasi dan \%inhibisi

Konsentrasi (ppm)	replikasi 1 \%inhibisi	Replikasi 2 \%inhibisi	Replikasi 3 \%inhibisi
31	72,00	71,62	71,38
15,5	54,52	55,89	56,13
7,75	40,02	39,65	40,27
3,88	35,32	35,32	35,81
1,94	30,48	29,99	29,99
a	29,43	29,36	29,81
b	1,4181	1,4258	1,4069
r	0,9935	0,9876	0,9860

$$IC_{50} = \frac{50 - a}{b}$$

$$\text{replikasi 1 } IC_{50} = \frac{(50 - 29,43)}{1,4181} = 14,50 \text{ ppm}$$

$$\text{replikasi 2 } IC_{50} = \frac{(50 - 29,36)}{1,4258} = 14,47 \text{ ppm}$$

$$\text{replikasi 3 } IC_{50} = \frac{(50 - 29,81)}{1,4069} = 14,34 \text{ ppm}$$

Rata-ra  $IC_{50} = 14,44$  ppm

**d. Menghitung  $IC_{50}$  Formula**

konsentrasi	ABS REP 1	ABS REP 2	ABS REP 3
100	<b>0,152</b>	0,146	0,149
50	<b>0,538</b>	0,533	0,534
12,5	<b>0,798</b>	0,789	0,798
6,25	<b>0,825</b>	0,822	0,825
3,13	<b>0,873</b>	0,873	0,875
<b>A</b>	0,8879	0,8842	0,8885
<b>B</b>	-0,0073	-0,0073	-0,0073
<b>R</b>	-0,9991	-0,9991	-0,9992

$$\%inhibisi = \frac{\text{absorbansi dpph} - \text{absorbansi sampel}}{\text{absorbansi dpph}} \times 100\%$$

- Konsentrasi 80 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,274}{0,807} \times 100\% = 66,05\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,271}{0,807} \times 100\% = 66,42\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,268}{0,807} \times 100\% = 66,79\%$$

- Konsentrasi 60 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,322}{0,807} \times 100\% = 60,10\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,320}{0,807} \times 100\% = 60,35\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,332}{0,807} \times 100\% = 60,10\%$$

- Konsentrasi 40 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,387}{0,807} \times 100\% = 52,04\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,385}{0,807} \times 100\% = 52,29\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,382}{0,807} \times 100\% = 52,66\%$$

- Konsentrasi 25 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,426}{0,807} \times 100\% = 47,21\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,419}{0,807} \times 100\% = 48,08\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,425}{0,807} \times 100\% = 47,34\%$$

- Konsentrasi 12,5 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,503}{0,807} \times 100\% = 36,67\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,511}{0,807} \times 100\% = 36,68\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,509}{0,807} \times 100\% = 36,93\%$$

- Konsentrasi 6,25 ppm

$$\text{replikasi 1 \%inhibisi} = \frac{0,807 - 0,546}{0,807} \times 100\% = 32,34\%$$

$$\text{replikai 2 \%inhibisi} = \frac{0,807 - 0,544}{0,807} \times 100\% = 32,59\%$$

$$\text{replikasi 3 \%inhibisi} = \frac{0,807 - 0,546}{0,807} \times 100\% = 32,34\%$$

Konsentrasi dan % Inhibisi

Konsentrasi (ppm)	Replikasi 1 %Inhibisi	Replikasi 2 %inhibisi	Replikasi 3 %inhibisi
80	66,05	66,42	66,79
60	60,10	60,35	60,10
40	52,04	52,29	52,66
25	47,21	48,08	47,34
12,5	37,67	36,68	36,93
6,24	32,34	32,59	32,34
a	32,6946	32,6001	32,3580
b	0,4436	0,4505	0,4559
r	0,9821	0,9787	0,9823

$$\text{replikasi 1 } IC_{50} = \frac{(50 - 32,6946)}{0,4436} = 39,01 \text{ ppm}$$

$$\text{replikasi 1 } IC_{50} = \frac{(50 - 32,6001)}{0,4505} = 38,62 \text{ ppm}$$

$$\text{replikasi 1 } IC_{50} = \frac{(50 - 31,3580)}{0,4559} = 38,69 \text{ ppm}$$

$$\text{Rata-ra } IC_{50} = 38,77 \text{ ppm}$$