

BAB V

KESIMPULAN DAN SARAN

A. Kesimpulan

Berdasarkan dari hasil penelitian yang telah dilakukan dapat diperoleh kesimpulan bahwa:

Pertama, *Solid Lipid Nanoparticles* (SLN) fisetin yang dibuat kombinasi belum bisa stabil, namun SLN fisetin memiliki efektifitas antioksidan yang sangat kuat karena memiliki rata-rata replikasi IC_{50} sebesar 12.14 ppm.

Kedua, konsentrasi lipid padat setil alkohol 0,15; 0,25%; dan 0,50% berpengaruh terhadap ukuran partikel karena semakin kecil konsentrasi yang digunakan semakin kecil juga ukuran partikel yang dihasilkan. Berbanding terbalik dengan efisiensi penyerapan, semakin besar konsentrasi yang digunakan semakin banyak obat yang terjerap dalam SLN.

Ketiga, formula terbaik dari SLN fisetin yang dihasilkan yaitu pada formula 6 untuk ukuran partikel dan tingkat keseragaman ukuran yang baik, sedangkan untuk efisiensi penyerapan terbaik dihasilkan oleh formula 5.

B. Saran

Penelitian ini masih banyak kekurangan, maka perlu dilakukan penelitian lebih lanjut mengenai :

Pertama, perlu dilakukan analisis *screening lipid* dengan menggunakan jenis lipid yang lebih beragam.

Kedua, perlu dilakukan uji TEM untuk mengetahui morfologi atau bentuk dari SLN fisetin secara detail.

Ketiga, perlu dilakukan uji kelarutan kinetik dan uji disolusi untuk mengetahui kelarutan SLN zat aktif dan perlu dilakukan optimasi pada surfaktan untuk uji stabilitas penyimpanan dalam jangka waktu yang lebih lama minimal 4 minggu.

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Lampiran 1. Certificate of analysis (COA) fisetin



Certificate of Analysis

Print Date: Jan 14th 2016

www.tocris.com

Product Name: Fisetin

Catalog No.: 5016

Batch No.: 1

CAS Number: 528-48-3

IUPAC Name: 2-(3,4-Dihydroxyphenyl)-3,7-dihydroxy-4H-1-benzopyran-4-one

1. PHYSICAL AND CHEMICAL PROPERTIES

Batch Molecular Formula: C₁₅H₁₀O₆

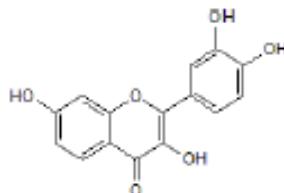
Batch Molecular Weight: 286.24

Physical Appearance: Yellow solid

Solubility: DMSO to 100 mM
ethanol to 10 mM

Storage: Store at -20°C

Batch Molecular Structure:



2. ANALYTICAL DATA

HPLC: Shows 98.1% purity

¹H NMR: Consistent with structure

Mass Spectrum: Consistent with structure

Microanalysis: Carbon Hydrogen Nitrogen

Theoretical 62.94 3.52

Found 62.81 3.58

Caution - Not Fully Tested • Research Use Only • Not For Human or Veterinary Use

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Product Name: Fisetin

Catalog No.: 5016

Batch No.: 1

CAS Number: 528-48-3

IUPAC Name: 2-(3,4-Dihydroxyphenyl)-3,7-dihydroxy-4H-1-benzopyran-4-one

Description:

Naturally occurring flavonoid and antioxidant. Inhibits PI 3-K, Akt, mTOR and Cdk5. Displays antiproliferative activity in prostate cancer cells. Shown to activate ERK; exhibits neuroprotective activity in Huntington's disease models. Also a DNMT1 inhibitor.

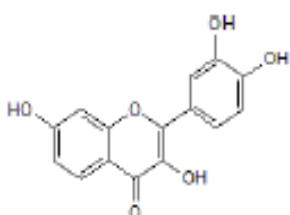
Physical and Chemical Properties:

Batch Molecular Formula: C₁₉H₁₈O₆
 Batch Molecular Weight: 386.24

Physical Appearance: Yellow solid

Minimum Purity: >98%

Batch Molecular Structure:



Storage: Store at -20°C

Solubility & Usage Info:

DMSO to 100 mM
 ethanol to 10 mM

Stability and Solubility Advice:

Some solutions can be difficult to obtain and can be encouraged by rapid stirring, sonication or gentle warming (in a 45–60°C water bath).

Information concerning product stability, particularly in solution, has rarely been reported and in most cases we can only offer a general guide. Our standard recommendations are:

SOLIDS: Provided storage is as stated on the product label and the vial is kept tightly sealed, the product can be stored for up to 6 months from date of receipt.

SOLUTIONS: We recommend that stock solutions, once prepared, are stored aliquoted in tightly sealed vials at -20°C or below and used within 1 month. Wherever possible solutions should be made up and used on the same day.

References:

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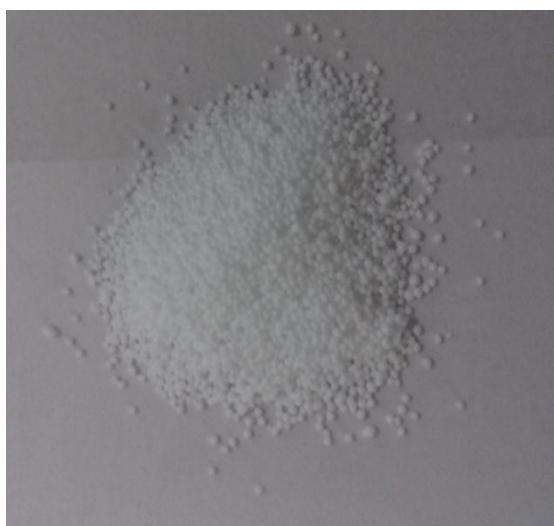
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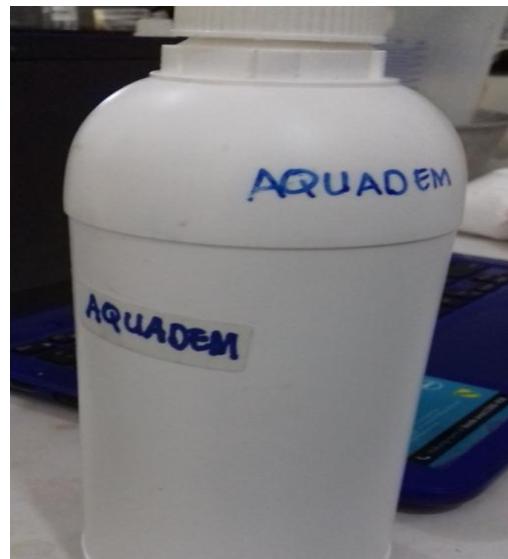
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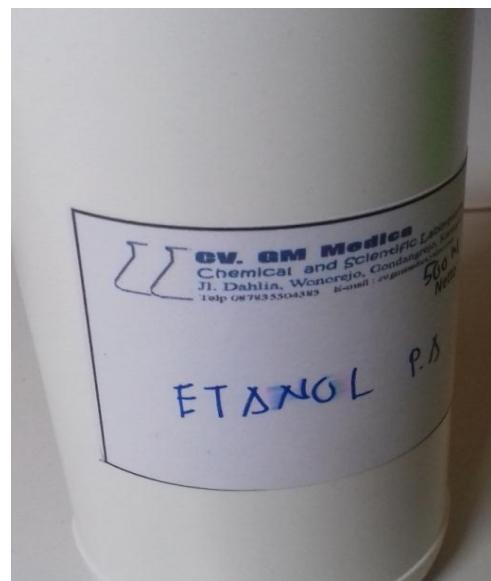
Lampiran 2. Foto serbuk fisetin**Lampiran 3. Foto Lipid Setil Alkohol**

Lampiran 4. Foto Lipid Stearil Alkohol**Lampiran 5. Foto lipid Setostearil Alkohol****Lampiran 6. Foto Surfaktan Tween 80**

Lampiran 7. Foto Aquademineralisata



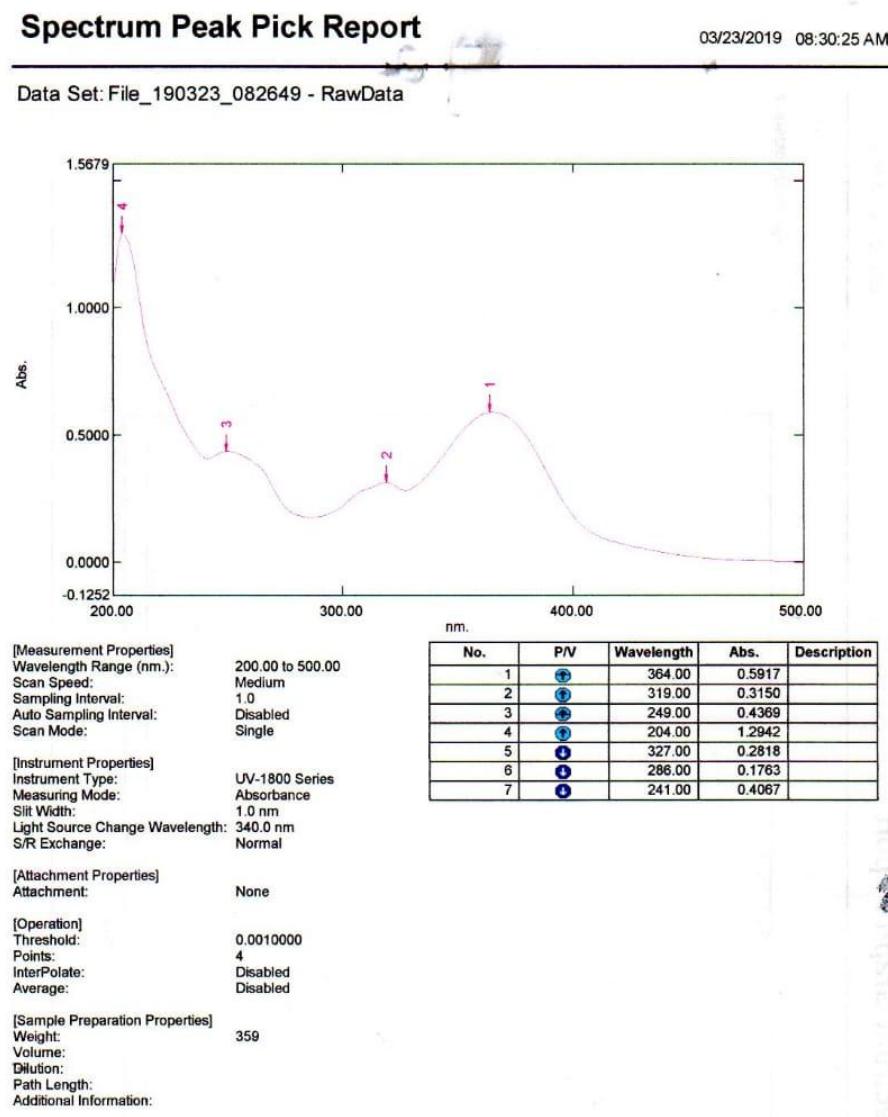
Lampiran 8. Foto etanol



Lampiran 9. Emulsi SLN Fisetin

Lampiran 10. Penentuan panjang gelombang dan pembuatan kurva baku

a. Penentuan panjang gelombang



b. Penentuan operating time

Kinetics Data Print Report

05/15/2019 03:46:21 PM

Time (Minute)	RawData ...
0.000	0.457
1.000	0.458
2.000	0.459
3.000	0.460
4.000	0.459
5.000	0.461
6.000	0.460
7.000	0.461
8.000	0.460
9.000	0.460
10.000	0.462
11.000	0.462
12.000	0.462
13.000	0.462
14.000	0.463
15.000	0.463
16.000	0.464
17.000	0.464
18.000	0.463
19.000	0.464
20.000	0.464
21.000	0.465
22.000	0.465
23.000	0.464
24.000	0.464
25.000	0.466
26.000	0.465
27.000	0.466
28.000	0.466
29.000	0.466
30.000	0.466

c. Kurva kalibrasi (Linieritas (*linearity*))

- Membuat larutan induk sebesar 46 ppm dengan menimbang 4,6mg fisetin ditambahkan etanol *p.a* sampai 100 ml.

Kertas + sampel : 0,2786g

Kertas + sisa : 0,2740g

Serbuk fisetin : 0,0046g

➤ Perhitungan seri konsentrasi

1. 3,68 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 3,68 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 0,8 \text{ ml}$$

2. 4,6 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 4,6 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 1 \text{ ml}$$

3. 5,52 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 5,52 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 1,2 \text{ ml}$$

4. 6,44 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 6,44 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 1,4 \text{ ml}$$

5. 7,36 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 7,36 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 1,6 \text{ ml}$$

6. 8,28 ppm

$$C_1 \times V_1 = C_2 \times V_2$$

$$46 \text{ ppm} \times V_1 = 8,28 \text{ ppm} \times 10 \text{ ml}$$

$$V_1 = 1,8 \text{ ml}$$

Konsentrasi (ppm)	Absorbansi
3,68	0,246
4,6	0,305
5,52	0,371
6,44	0,424
7,36	0,478
8,28	0,541

Persamaan regresi linier antara konsentrasi (ppm) dan serapan diperoleh nilai :

$$a = 0,01401$$

$$b = 0,063571$$

$$r = 0,99951$$

$$y = a + bx$$

$$y = 0,01401 + 0,063571 x$$

keterangan :

x = konsentrasi ($\mu\text{g}/\text{ml}$)

y = serapan

Hasil linearitas diperoleh $r = 0,99951$; sehingga dapat disimpulkan bahwa data linear

d. Akurasi (Accuracy)

Konsentrasi Sebenarnya	Absorbansi	Konsentrasi Terukur	%recovery	Rata-rata
4,6	0,308	4,624	100,52%	
4,6	0,308	4,624	100,52%	100,19%
4,6	0,305	4,578	99,52%	
5,52	0,366	5,508	99,78%	
5,52	0,366	5,508	99,78%	99,69%
5,52	0,365	5,493	99,51%	
6,44	0,429	6,469	100,45%	
6,44	0,429	6,469	100,45%	
6,44	0,425	6,408	99,50%	100,13%

$$\% \text{ Recovery} = \frac{\text{konsentrasi terukur}}{\text{konsentrasi sebenarnya}} \times 100\%$$

➤ **Konsentrasi 4,6 ppm**

- % Recovery $= \frac{4,624}{4,6} X 100\% = 100,52\%$
- % Recovery $= \frac{4,624}{4,6} X 100\% = 100,52\%$
- % Recovery $= \frac{4,578}{4,6} X 100\% = 99,52\%$

➤ **Konsentrasi 5,52 ppm**

- % Recovery $= \frac{5,508}{5,52} X 100\% = 99,78\%$
- % Recovery $= \frac{5,508}{5,52} X 100\% = 99,78\%$
- % Recovery $= \frac{5,493}{5,52} X 100\% = 99,51\%$

➤ **Konsentrasi 6,44 ppm**

- % Recovery $= \frac{6,469}{6,44} X 100\% = 100,45\%$
- % Recovery $= \frac{6,469}{6,44} X 100\% = 100,45\%$
- % Recovery $= \frac{6,408}{6,44} X 100\% = 99,50\%$

Nilai rata- rata % Recovery diatas adalah 100%, hal ini menunjukkan nilai persen perolehan kembali yang baik.

e. Presisi

Konsentrasi	Absorbansi	Konsentrasi
4,6	0,367	6,131
4,6	0,365	6,098
4,6	0,369	6,164
4,6	0,370	6,181
4,6	0,366	6,114
4,6	0,368	6,148
4,6	0,369	6,164
4,6	0,371	6,197
4,6	0,370	6,181
4,6	0,366	6,114
RATA-RATA		6,149
	SD	0,034
	CV	1%

Nilai CV dilihat dari data diatas adalah 1%, hasil ini sesuai dengan pernyataan bahwa syarat presisi adalah kurang dari 2%.

Lampiran 11. Hasil Uji Ukuran Partikel Formula Setil Alkohol 0,5%

Instrument

Serial Number: 3214-DMP
 Model: DelsaMax Pro
 Pals Firmware Version: 1.1.0.6
 DLS Firmware Version: 2.3.1.0
 Assist Firmware Version: 1.0.0.9
 Instrument Name: BCI-3214-DMP
 Laser Wavelength (nm): 532.0
 Has DLS: Yes
 DLS Detector Angle (degrees): 163.5
 Minimum Temperature (C): 3.5
 Minimum Temperature without N2 (C): 20
 Maximum Temperature (C): 70
 Minimum Ramp Rate (C/min): 0
 Maximum Ramp Rate (C/min): 1.5

Instrument Parameters: Measurements

Collect Data: DLS and Pals (Simultaneous)
 Aq Time (s): 20
 Read Interval (s): 1
 Number Aqc: 3
 Electric Field Frequency (Hz): 10.0
 Voltage Amplitude (V): 2.5
 Collection Period (s): 15.0
 Auto-attenuation: Yes
 Attenuation Level (%): 0
 Auto-attenuation Time Limit(s): 0
 Laser Mode: Normal
 Set Temp On Connection: No
 Set Temp (C): 20
 Temp Ramp Enabled: Yes
 Temp Ramp Rate (C/min): 1

Datalog Table: Measurements

	Item	Zeta Potential	Radius	Diameter
		(mV)	(nm)	(nm)
1	Meas 1	-19.22	109.2	218.3
2	Meas 2	-21.00	138.6	277.1
3	Meas 3	-20.51	138.7	277.5
	Mean	-20.25	128.8	257.6
	S	0.92	17.0	34.1
	%S	4.55	13.2	13.2
	S ²	0.85	289.9	1159.5
	Min	-21.00	109.2	218.3
	Max	-19.22	138.7	277.5

Lampiran 12. Hasil Uji Ukuran Partikel Formula Setil Alkohol 0,25%

Instrument

Serial Number: 3214-DMP
 Model: DelsaMax Pro
 Pals Firmware Version: 1.1.0.6
 DLS Firmware Version: 2.3.1.0
 Assist Firmware Version: 1.0.0.9
 Instrument Name: BCI-3214-DMP
 Laser Wavelength (nm): 532.0
 Has DLS: Yes
 DLS Detector Angle (degrees): 163.5
 Minimum Temperature (C): 3.5
 Minimum Temperature without N2 (C): 20
 Maximum Temperature (C): 70
 Minimum Ramp Rate (C/min): 0
 Maximum Ramp Rate (C/min): 1.5

Instrument Parameters: Measurements

Collect Data: DLS and Pals (Simultaneous)
 Acq Time (s): 20
 Read Interval (s): 1
 Number Acq: 3
 Electric Field Frequency (Hz): 10.0
 Voltage Amplitude (V): 2.5
 Collection Period (s): 15.0
 Auto-attenuation: Yes
 Attenuation Level (%): 0
 Auto-attenuation Time Limit(s): 0
 Laser Mode: Normal
 Set Temp On Connection: No
 Set Temp (C): 20
 Temp Ramp Enabled: Yes
 Temp Ramp Rate (C/min): 1

Datalog Table: Measurements

Item	Zeta Potential	Diameter
	(mV)	(nm)
1 Meas 1	-19.88	26.1
2 Meas 2	-15.91	27.9
3 Meas 3	-19.41	31.9
Mean	-18.40	28.7
S	2.17	3.0
%S	11.80	10.3
S ²	4.71	8.8
Min	-19.88	26.1
Max	-15.91	31.9

Lampiran 13. Hasil Uji Ukuran Partikel Formula Setil Alkohol 0,15%

Instrument

Serial Number: 3214-DMP
 Model: DelsaMax Pro
 Pals Firmware Version: 1.1.0.6
 DLS Firmware Version: 2.3.1.0
 Asslet Firmware Version: 1.0.0.9
 Instrument Name: BCI-3214-DMP
 Laser Wavelength (nm): 532.0
 Has DLS: Yes
 DLS Detector Angle (degrees): 163.5
 Minimum Temperature (C): 3.5
 Minimum Temperature without N2 (C): 20
 Maximum Temperature (C): 70
 Minimum Ramp Rate (C/min): 0
 Maximum Ramp Rate (C/min): 1.5

Instrument Parameters: Measurements

Collect Data: DLS and Pals (Simultaneous)
 Acq Time (s): 20
 Read Interval (s): 1
 Number Acq: 3
 Electric Field Frequency (Hz): 10.0
 Voltage Amplitude (V): 2.5
 Collection Period (s): 15.0
 Auto-attenuation: Yes
 Attenuation Level (%): 0
 Auto-attenuation Time Limit(s): 0
 Laser Mode: Normal
 Set Temp On Connection: No
 Set Temp (C): 20
 Temp Ramp Enabled: Yes
 Temp Ramp Rate (C/min): 1

Datalog Table: Measurements

	Item	Zeta Potential (mV)	Diameter (nm)
1	Meas 1	-17.25	15.8
2	Meas 2	-17.45	14.8
3	Meas 3	-19.52	14.9
Mean		-18.08	15.2
S		1.26	0.5
%S		6.95	3.4
S ²		1.58	0.3
Min		-19.52	14.8
Max		-17.25	15.8

Lampiran 14. Hasil Perhitungan Efisiensi Penjerapan SLN Fisetin

a. Formula 1 (Setil Alkohol 0,5%)

- Larutan induk \rightarrow 200 mg SLN fisetin / 10 ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\begin{aligned} \text{Fisetin} &= 10 \text{ mg} \\ \text{Eksipien (tween 80 + setil alcohol)} &= 10250 \text{ mg} \\ \% \text{ kadar fisetin} &= \frac{10}{10250+10} \times 100\% = 0,0975\% \\ \bullet \quad \text{Kadar dalam 200 mg SLN fisetin} &= 0,0975\% \times 200 \text{ mg} = 9,75 \text{ mg} \\ \bullet \quad \text{Perhitungan kadar fisetin terjerap menggunakan persamaan regresi linier :} \\ y &= a + bx \\ 0,946 &= 0,014 + 0,0636x \\ 0,0636x &= 0,932 \\ X &= 14,654 \text{ ppm} \\ \bullet \quad \% \text{ kadar} &= \frac{14,654}{20000} \times 100\% = 0,07327\% \\ \bullet \quad \text{Kadar dalam 200 mg SLN fisetin} &= 0,07327\% \times 200 \text{ mg} = 7,327 \text{ mg} \\ \bullet \quad \% \text{ efisiensi penjerapan} &= \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\% \\ &= \frac{7,327}{9,75} \times 100\% \\ &= 75,15\% \end{aligned}$$

b. Formula 2 (Setil Alkohol 0,25%)

- Larutan induk \rightarrow 200 mg SLN fisetin / 10 ml etanol p.a = 20.000 ppm
- Perhitungan teoritis

$$\begin{aligned} \text{Fisetin} &= 10 \text{ mg} \\ \text{Eksipien (tween 80 + setil alcohol)} &= 11000 \text{ mg} \\ \% \text{ kadar fisetin} &= \frac{10}{11000+10} \times 100\% = 0.0908\% \\ \text{Kadar dalam 200 mg SLN fisetin} &= 0,0908\% \times 200 \text{ mg} = 9,08 \text{ mg} \\ \bullet \quad \text{Perhitungan kadar fisetin terjerap menggunakan persamaan regresi linier :} \\ y &= a + bx \\ 0,548 &= 0,014 + 0,0636x \\ 0,0636x &= 0,534 \end{aligned}$$

$$\begin{aligned}
 X &= 8,396 \text{ ppm} \\
 \bullet \% \text{ kadar} &= \frac{8,396}{20000} \times 100\% = 0,04198\% \\
 \bullet \text{ Kadar dalam } 200 \text{ mg SLN fisetin} &= 0,04198\% \times 200 \text{ mg} = 4,198 \text{ mg} \\
 \bullet \% \text{ efisiensi penjerapan} &= \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\% \\
 &= \frac{4,198 \text{ mg}}{9,08 \text{ mg}} \times 100\% \\
 &= 46,23\%
 \end{aligned}$$

c. Formula 3 (Setil Alkohol 0,15%)

- Larutan induk \rightarrow 200 mg SLN fisetin / 10 ml etanol p.a = 20.000 ppm

- Perhitungan teoritis

$$\begin{aligned}
 \text{Fisetin} &= 10 \text{ mg} \\
 \text{Eksipien (tween 80 + setil alcohol)} &= 10750 \text{ mg} \\
 \% \text{ kadar fisetin} &= \frac{10}{10750+10} \times 100\% = 0,09294\%
 \end{aligned}$$

- Kadar dalam 200 mg SLN fisetin = $0,09294\% \times 200 \text{ mg} = 9,294 \text{ mg}$
- Perhitungan kadar fisetin terjerap menggunakan persamaan regresi linier :

$$\begin{aligned}
 y &= a + bx \\
 0,448 &= 0,014 + 0,0636x \\
 0,0636x &= 0,434 \\
 X &= 6,824 \text{ ppm} \\
 \bullet \% \text{ kadar} &= \frac{6,824}{20000} \times 100\% = 0,03412\% \\
 \bullet \text{ Kadar dalam } 200 \text{ mg SLN fisetin} &= 0,03412\% \times 200 \text{ mg} = 3,412 \text{ mg} \\
 \bullet \% \text{ efisiensi penjerapan} &= \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\% \\
 &= \frac{3,412 \text{ mg}}{9,294 \text{ mg}} \times 100\% \\
 &= 36,71\%
 \end{aligned}$$

Lampiran 15. Uji stabilitas SLN fisetin dalam penyimpanan secara visual**a. Minggu pertama penyimpanan**

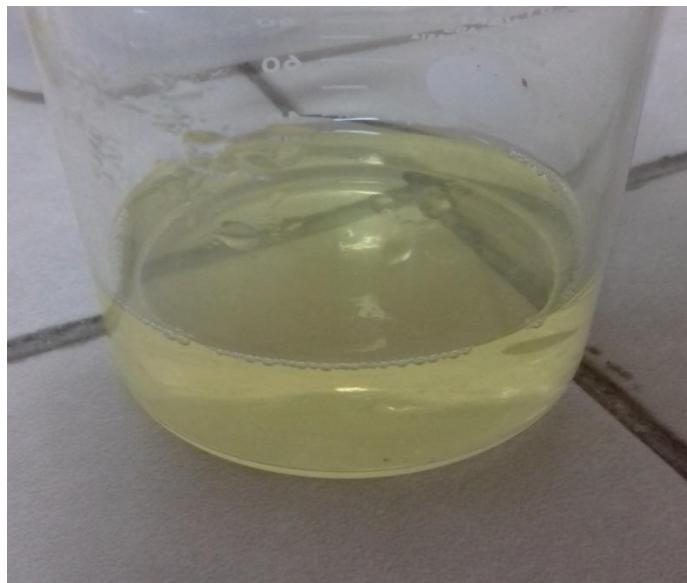
- ✓ Setil Alkohol 0,5%



- ✓ Setil Alkohol 0,25%



✓ Setil Alkohol 0,15%



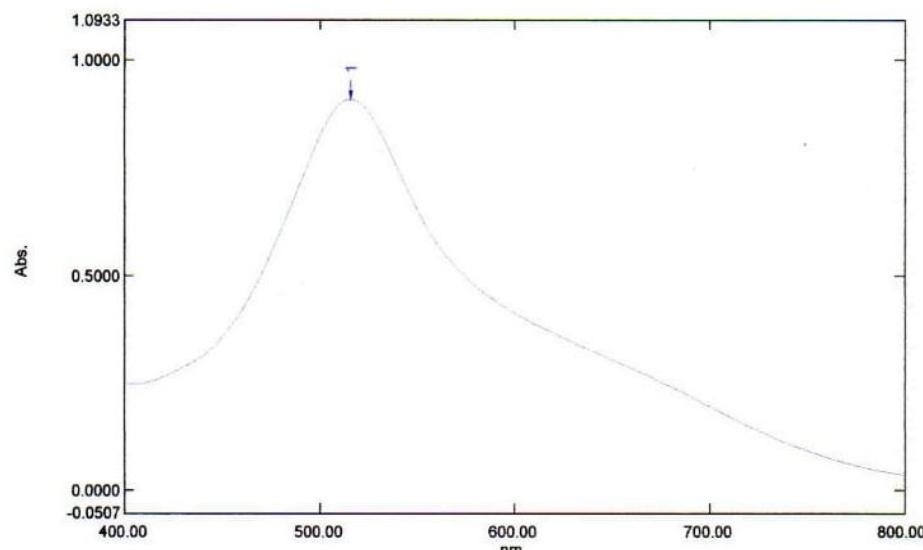
Lampiran 16. Uji aktifitas antioksidan

a. Penentuan panjang gelombang maks dpph

Spectrum Peak Pick Report

05/10/2019 10:03:42 AM

Data Set: lamda maks dpph fix 1 - RawData



[Measurement Properties]
Wavelength Range (nm.): 400.00 to 800.00
Scan Speed: Medium
Sampling Interval: 1.0
Auto Sampling Interval: Disabled
Scan Mode: Auto

No.	P/V	Wavelength	Abs.	Description
1	●	516.00	0.9096	
2	●	405.00	0.2479	

[Instrument Properties]
Instrument Type: UV-1800 Series
Measuring Mode: Absorbance
Slit Width: 1.0 nm
Light Source Change Wavelength: 340.0 nm
S/R Exchange: Normal

[Attachment Properties]
Attachment: None

[Operation]
Threshold: 0.0010000
Points: 4
Interpolate: Disabled
Average: Disabled

[Sample Preparation Properties]
Weight:
Volume:
Dilution:
Path Length:
Additional Information:

b. Penentuan operating time

OT Fisetur DPPH 4/5/19
Kinetics Data Print Report
 60 MEHIT
 05/04/2019 12:24:22 PM

Time (Minute)	RawData ...
0.000	0.199
1.000	0.199
2.000	0.198
3.000	0.198
4.000	0.199
5.000	0.198
6.000	0.198
7.000	0.198
8.000	0.198
9.000	0.199
10.000	0.198
11.000	0.199
12.000	0.198
13.000	0.199
14.000	0.198
15.000	0.199
16.000	0.199
17.000	0.199
18.000	0.199
19.000	0.199
20.000	0.199
21.000	0.199
22.000	0.199
23.000	0.199
24.000	0.199
25.000	0.199
26.000	0.199
27.000	0.199
28.000	0.199
29.000	0.199
30.000	0.199
31.000	0.199
32.000	0.199
33.000	0.199
34.000	0.200
35.000	0.200
36.000	0.199
37.000	0.200
38.000	0.200
39.000	0.200
40.000	0.200
41.000	0.200
42.000	0.200
43.000	0.200
44.000	0.200
45.000	0.200
46.000	0.200
47.000	0.200
48.000	0.200
49.000	0.200
50.000	0.200

Kinetics Data Print Report

05/04/2019 12:24:22 PM

Time (Minute)	RawData ...
51.000	0.201
52.000	0.200
53.000	0.200
54.000	0.201
55.000	0.201
56.000	0.201
57.000	0.201
58.000	0.201
59.000	0.201
60.000	0.201

c. DPPH Fisetin murni

➤ Perhitungan bahan fisetin :

$$50 \text{ mg/ 100 ml} = 500 \text{ mg / 1000 ml} = 500 \text{ ppm}$$

Yang ditimbang 49,8 mg sehingga didapat konsentrasi 498 ppm

➤ Perhitungan konsentrasi (ppm)

$$\diamond \quad V_1 \times N_1 = V_2 \times N_2$$

$$500 \text{ ppm} \times 0,3112 \text{ ml} = V_2 \times 10 \text{ ml}$$

$$V_2 = 15,56 \text{ ppm}$$

$$\diamond \quad V_1 \times N_1 = V_2 \times N_2$$

$$500 \text{ ppm} \times 0,1556 \text{ ml} = V_2 \times 10 \text{ ml}$$

$$V_2 = 7,78 \text{ ppm}$$

$$\diamond \quad V_1 \times N_1 = V_2 \times N_2$$

$$500 \text{ ppm} \times 0,0778 \text{ ml} = V_2 \times 10 \text{ ml}$$

$$V_2 = 3,89 \text{ ppm}$$

$$\diamond \quad V_1 \times N_1 = V_2 \times N_2$$

$$500 \text{ ppm} \times 0,039 \text{ ml} = V_2 \times 10 \text{ ml}$$

$$V_2 = 1,95 \text{ ppm}$$

$$\diamond \quad V_1 \times N_1 = V_2 \times N_2$$

$$500 \text{ ppm} \times 0,0194 \text{ ml} = V_2 \times 10 \text{ ml}$$

$$V_2 = 0,97 \text{ ppm}$$

DPPH	Konsentrasi (ppm)	Volume (ml)	Etanol (ml)	Absorbansi replikasi 1	Absorbansi replikasi 2	Absorbansi replikasi 3
1 ml	160		4	0,909		
1 ml	15,56	1	3	0,212	0,221	0,224
1ml	7,78	1	3	0,405	0,416	0,418
1ml	3,89	1	3	0,507	0,509	0,511
1ml	1,95	1	3	0,555	0,557	0,559
1ml	0,97	1	3	0,578	0,578	0,588

$$\begin{aligned} \text{\%inhibisi} \\ = \frac{\text{absorbansi DPPH} - \text{absorbansi sampel}}{\text{absorbansi DPPH}} \times 100\% \end{aligned}$$

➤ Konsentrasi 15,56 ppm

$$\text{Replikasi 1 \%inhibisi} = \frac{0,909-0,212}{0,909} \times 100\% = 76,68\%$$

$$\text{Replikasi 2 \%inhibisi} = \frac{0,909-0,221}{0,909} \times 100\% = 75,69\%$$

$$\text{Replikasi 3 \%inhibisi} = \frac{0,909-0,224}{0,909} \times 100\% = 75,36\%$$

➤ Konsentrasi 7,78 ppm

$$\text{Replikasi 1 \%inhibisi} = \frac{0,909-0,405}{0,909} \times 100\% = 55,45\%$$

$$\text{Replikasi 2 \%inhibisi} = \frac{0,909-0,416}{0,909} \times 100\% = 54,24\%$$

$$\text{Replikasi 3 \%inhibisi} = \frac{0,909-0,418}{0,909} \times 100\% = 54,02\%$$

➤ Konsentrasi 3,89 ppm

$$\text{Replikasi 1 \%inhibisi} = \frac{0,909-0,507}{0,909} \times 100\% = 44,22\%$$

$$\text{Replikasi 2 \%inhibisi} = \frac{0,909-0,509}{0,909} \times 100\% = 44,00\%$$

$$\text{Replikasi 3 \%inhibisi} = \frac{0,909-0,511}{0,909} \times 100\% = 43,78\%$$

➤ Konsentrasi 1,95 ppm

$$\text{Replikasi 1 \%inhibisi} = \frac{0,909-0,555}{0,909} \times 100\% = 38,94\%$$

$$\text{Replikasi 2 \%inhibisi} = \frac{0,909-0,557}{0,909} \times 100\% = 38,72\%$$

$$\text{Replikasi 3 \%inhibisi} = \frac{0,909-0,559}{0,909} \times 100\% = 38,50\%$$

➤ **Konsentrasi 0,97 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,578}{0,909} \times 100\% = 36,41\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,578}{0,909} \times 100\% = 36,41\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,588}{0,909} \times 100\% = 35,31\%$$

➤ **Konsentrasi dan % inhibisi**

Konsentrasi (ppm)	Replikasi 1 % inhibisi	Replikasi 2 % inhibisi	Replikasi 3 % inhibisi
15,56	76,68	75,69	75,36
7,78	55,45	54,24	54,02
3,89	44,22	44,00	43,78
1,95	38,94	38,72	38,50
0,97	36,41	36,41	35,31
A	33,625	33,533	32,969
B	2,772	2,700	2,724
R	0,9999	0,9999	0,9999

$$➤ IC_{50} = \frac{50-a}{b}$$

$$\text{Replikasi 1 } IC_{50} = \frac{(50-33,625)}{2,772} = 5,91 ppm$$

$$\text{Replikasi 2 } IC_{50} = \frac{(50-33,533)}{2,700} = 6,09 ppm$$

$$\text{Replikasi 3 } IC_{50} = \frac{(50-32,969)}{2,724} = 6,25 ppm$$

d. DPPH formula (Setil alkohol 0,5%)

DPPH	Konsentrasi (ppm)	Volume (ml)	Etanol (ml)	Absorbansi replikasi 1	Absorbansi replikasi 2	Absorbansi replikasi 3
1 ml	100		4	0,909		
1 ml	25	1	3	0,174	0,180	0,178
1ml	12,25	1	3	0,442	0,448	0,450
1ml	6,25	1	3	0,584	0,580	0,577
1ml	1,56	1	3	0,687	0,682	0,685
1ml	0,78	1	3	0,698	0,710	0,697

$$\begin{aligned} & \text{\%inhibisi} \\ &= \frac{\text{absorbansi DPPH} - \text{absorbansi sampel}}{\text{absorbansi DPPH}} \times 100\% \end{aligned}$$

➤ **Konsentrasi 25 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,174}{0,909} \times 100\% = 80,86\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,180}{0,909} \times 100\% = 80,20\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,178}{0,909} \times 100\% = 80,42\%$$

➤ **Konsentrasi 12,25 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,442}{0,909} \times 100\% = 51,38\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,448}{0,909} \times 100\% = 50,72\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,450}{0,909} \times 100\% = 50,50\%$$

➤ **Konsentrasi 6,25 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,584}{0,909} \times 100\% = 35,75\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,580}{0,909} \times 100\% = 36,19\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,577}{0,909} \times 100\% = 36,52\%$$

➤ **Konsentrasi 1,56 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,687}{0,909} \times 100\% = 24,42\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,682}{0,909} \times 100\% = 24,97\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,685}{0,909} \times 100\% = 24,64\%$$

➤ **Konsentrasi 0,78 ppm**

$$\text{Replikasi 1 \% inhibisi} = \frac{0,909-0,698}{0,909} \times 100\% = 23,21\%$$

$$\text{Replikasi 2 \% inhibisi} = \frac{0,909-0,710}{0,909} \times 100\% = 21,89\%$$

$$\text{Replikasi 3 \% inhibisi} = \frac{0,909-0,697}{0,909} \times 100\% = 23,32\%$$

➤ Konsentrasi dan % inhibisi

Konsentrasi (ppm)	Replikasi 1 %inhibisi	Replikasi 2 %inhibisi	Replikasi 3 %inhibisi
25	80,86	80,20	80,42
12,25	51,38	50,72	50,50
6,25	35,75	36,19	36,52
1,56	24,42	24,97	24,64
0,78	23,21	21,89	23,32
A	20,228	20,923	21,385
B	2,449	2,386	2,366
R	0,9993	0,9997	0,9999

➤ $IC_{50} = \frac{50-a}{b}$

Replikasi 1 $IC_{50} = \frac{(50-20,228)}{2,449} = 12,15 ppm$

Replikasi 2 $IC_{50} = \frac{(50-20,923)}{2,386} = 12,18 ppm$

Replikasi 3 $IC_{50} = \frac{(50-21,385)}{2,366} = 12,09 ppm$

e. Uji T-test Fisetin Murni

➔ T-Test

[DataSet0]

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
IC50	3	6,0700	,17521	,10116

One-Sample Test

	Test Value = 9					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
IC50	-28,964	2	,001	-2,93000	-3,3653	-2,4947

Kesimpulan : Nilai sig. (2-tailed) yaitu $0,001 < 0,05$ yang berarti terdapat perbedaan yang bermakna antara nilai IC50 hasil praktikum dengan nilai IC50 teoritis (IC50 fisetin 9ppm).

f. Uji T-test Formula Setil Alkohol 0,5%

➔ T-Test

[DataSet0]

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
IC50	3	10,7633	,27465	,15857

One-Sample Test

	Test Value = 9					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
IC50	11,120	2	,008	1,76333	1,0811	2,4456

Kesimpulan : Nilai sig. (2-tailed) yaitu $0,008 < 0,05$ yang berarti terdapat perbedaan yang bermakna antara nilai IC50 formula fisetin hasil praktikum dengan nilai IC50 teoritis (IC50 fisetin 9ppm).