

BAB V

KESIMPULAN DAN SARAN

A. KESIMPULAN

Berdasarkan dari hasil penelitian yang telah dilakukan dapat diperoleh kesimpulan bahwa:

Pertama, myricetin dapat dibuat nanofitosom dengan menggunakan metode hidrasi lapis tipis-sonikasi.

Kedua, karakterisasi nanofitosom dilihat dari ukuran partikel semua formula memiliki ukuran di antara 10-1000 nm, penggunaan fosfatidilkolin pada konsentrasi tertinggi sebesar 118 mg mampu menghasilkan ukuran partikel paling kecil yaitu 139 nm dan indeks polidipersitas terendah yaitu 0,380 dengan efisiensi penjerapan tertinggi 91,54%.

Ketiga, nanofitosom myricetin formula 1, 2, 3 dan 4 tidak stabil selama penyimpanan lebih dari 3 minggu, formula 5 memiliki nilai potensial zeta -6,350 mV, sehingga kurang stabil setelah penyimpanan karena nilai potensial zeta kurang dari ± 30 mV.

B. SARAN

Penelitian ini masih banyak kekurangan, maka perlu dilakukan penelitian lebih lanjut mengenai:

Pertama, perlu dilakukan pembuatan nanofitosom dengan metode lainnya yang bisa memberikan ukuran partikel lebih kecil dan stabil.

Kedua, perlu dilakukan uji karakterisasi morfologi nanofitosom menggunakan alat *Transmission Electron Microscopy* (TEM).

DAFTAR PUSTAKA

- Adawiah, Sukandar D, Muawanah A. 2015. Aktivitas antioksidan dan kandungan komponen bioaktif sari buah Namnam. *Jurnal Kimia Valensi*. 1(2).
- Ajazuddin, Saraf S. 2010. Applications of novel drug delivery system for herbal formulations. *Fitoterapia*. 81(7):680–9.
- Aqil F, Munagala R, Jeyabalan J, Vadhanam MV. 2013. Bioavailability of phytochemicals and its enhancement by drug delivery systems. *Cancer Lett.* 334:133–41.
- Arora, Rajnish, Jain CP. 2007. Advances in niosome as a drug carrier. *Asian Journal of Pharmaceutics*. 1(1).
- Babazadeh, Afshin M, Zeinali H, Hamishehka. 2018. Nano-Phytosome: A developing platform for herbal anti-cancer agents in cancer therapy. *Current Drug Targets*. 19:170-180.
- Badarinath A, Rao K, Chetty CS, Ramkanth S, Rajan T, Gnanaprakash K. 2010. A Review on in-vitro antioxidant methods: comparisons, correlations, and considerations. *International Journal of PharmTech Research*. 1276-1285.
- Blazek-Welsh AI, DG Rhodes. 2001. Maltodextrin-based proniosome. *AAP Pharmaceutical Sciences*.
- Chan, Chung Chown, Lam Y.C, Lee, Xue Ming Zhang. 2004. *Analitical Method Validation and Instrument Performance Verification*. John Willey & Sons. New Jersey: Inc. Publication.
- Chandira RM, Pradeep, A. Pasupathi, D. Bhowmik, Chiranjib, B. Jayakar, KK Tripathi, KP Sampath Kumar, *et al.* 2010. Design, development, and formulation of antiacne dermatological gel. *Journal of Chemical and Pharmaceutical Research*.
- Colas JC, Shi W, Rao VS, Omri A, Mozafari MR, Singh H, *et al.* 2007. Microscopical investigations of nisin-loaded nanoliposomes prepared by Mozafari method and their bacterial targeting. *Micron*. 38:841–847.

- Comoglio A, Tomasi A, Malandrino S, Poli G, Albano E. 1995. Scavenging effect of silipide, a new silybin-phospholipid complex, on ethanol-derived free radicals. *Biochem Pharmacol.* 50:1313-1316.
- Dang Y, Xie Y, Duan JZ, Ma P, Li GW, Ji G, et al. 2014. Quantitative determination of myricetin in rat plasma by ultra performance liquid chromatography tandem mass spectrometry and its absolute bioavailability. *Drug Res.* 64: 516-522.
- Delmifiana B, Astuti. 2013. Pengaruh sonikasi terhadap struktur dan morfologi nanopartikel magnetik yang disintesis dengan metode kopresipitasi. *Jurnal Fisika Unand.* 2:3.
- Du X, AJ Brown, H Yang. 2015. Novel mechanisms of intracellular cholesterol transport: oxysterol-binding proteins and membrane contact sites. *Current Opinion in Cell Biology.* 35:37–42.
- Dua JS, Rana AC, Bhandari AK. 2012. Liposome: methods of preparation and applications. *Int J Pharm.* 3:14–20.
- Dwiastuti R, Noegrohati S, Istyastono EP, Marchaban. 2016. Metode pemanasan dan sonikasi menghasilkan nanoliposom dari fosfolipid lesein kedelai (*Soy Lecithin*). *Jurnal Farmasi Sains dan Komunitas.* 13(1):23-27.
- El-Gawad AH, Soliman OA, Shams MEE, Maria DN. 2014. Formulation and in vitro evaluation of loratadine gels for ophthalmic use. *RGUHS J Pharm Sci* 4.
- Gaber DM, Nafee N, Abdallah OY. 2017. Myricetin solid lipid nanoparticles: Stability assurance from system preparation to site of action. *European Journal of Pharmaceutical Sciences.*
- Gandjar IG, Rohman A. 2007. *Kimia Farmasi Analisis.* Yogyakarta: Pustaka Pelajar. Hal 419-425.
- Ghanbarzadeh B, A Babazadeh, H Hamishehkar. 2016. Nano phytosome as a potential food-grade delivery system. *Food Bio science.* 15:126–135.
- Gregoriadis, Gregory. 2007. Liposome technology: Liposome preparation and related techniques. Third Edition. Vol 1. *Informa Healthcare USA.* Inc 21-32.

- Gupta BP, Jain N, Thakur N, Ruchi J, Banweer J, Deepak KJ, Surendra J, *et al.* 2010. Phytosome: a novel drug delivery system for herbal medicine. *International Journal of Pharmaceutical Sciences and Drug Research.* 2(4):224-228.
- Gupta, Dixit. 2011. Bioavailability enhancement of curcumin by complexation with phosphatidylcholine. *J Pharm Sci.* 100:1987-1995.
- Hamada A, Kawaguchi T, Nakano M. 2002. Clinical pharmakokinetics of cytarabine formulations. *Clin Pharmacokinet.* 41(10): 705-71.
- Harmita. 2004. Petunjuk pelaksanaan validasi metode dan cara perhitungannya. *Majalah Ilmu Kefarmasian.* 1:117-135.
- Harvey, David. 2000. *Modern Analytical Chemistry.* USA: The McGraw-Hill Companies.
- Honary S, Zahir F. 2013. Effect of zeta potensial on the properties of nano drug delivery system—a review (part 1). *Trop. J. of Pharmaceutical Research.* 12 (2):255-264.
- Hong C, Dang Y, Lin G, Yao Y, Li G, Ji G, Shen H, Xie Y, *et al.* 2014. Effects of stabilizing agents on the development of myricetin nanosuspension and its characterization: an in vitro and in vivo evaluation. *International Journal of Pharmaceutics.* 477: 251–260.
- Huang L, Kirschke CP, Gitschier J. 2002. Functional characterization of a novel mammalian zinc transporter, ZnT6. *J Biol Chem.* 277(29): 26389-95.
- Husni P, Puspitaningrum K. 2017. Pengembangan formula nano-fitosom serbuk liofilisasi seduhan teh hitam (*Camellia sinensis* L. Kuntze). *Indonesian Journal of Pharmaceutical Science and Technology.* Vol. 4(3).
- ICH Q2A. 2005. *Validation of Analytical Procedures: Definitions and Terminology.* Geneva 1995, in 2005 incorporated in Q2(R1).
- ICH Q2B. 2005. *Validation of Analytical Procedures: Methodology,* adopted in 1996, Geneva Q2B in 2005 incorporated in Q2(R1).
- Jain S, Dhanotiya C, Malviya N. 2012. Physicochemical characterization and determination of free radical scavenging activity of rutin-phospholipid complex. *Int J Pharm Sci Res.* 3(3):909-913.

- Jonassen H. 2014. Polysaccharide based nanoparticle for drug delivery application [Thesis School of Pharmacy]. Faculty of Mathematics and Natural Science, University of Oslo.
- Juniarti D, Osmeli, Yuhernita. 2009. Kandungan Senyawa Kimia, Uji Toksisitas (Brine Shrimp Lethality Test) dan Antioksidan (1,1-diphenyl-2 pikrilhydrazyl) dari Ekstrak Daun Saga (*Abrus precatorius* L.). *Makara Sains*. 13(1): 50-54.
- Kalita B, Malay K.Das, Sharma AK. 2013. Novel phytosome formulations in making herbal extracts more effective. *Research J. Pharm. and Tech.* 6(11).
- Karatas A, Turhan F. 2015. Phyto-phospholipid complexes as drug delivery system for herbal extracts/molecules. *Turk J Pharm Sci.* 12(1):93-102.
- Karlik M. 2001. *Lattice Imaging In Transmission Electron Microscopy*, Department of Materials, Faculty of Nuclear Sciences and Physical Engineering, Czech Technical University in Prague, Trojanova 13, 12000 Prague 2, Czech Republic, Keil FJ. 2007. Modeling of process intensification. *AIDIC Conference Series*. 9:1-8.
- Khan N, Deeba N. Syed, Nihal A, Hasan M. 2013. Fisetin: a dietary for health promotion. *Antioxidants and Redox Signaling*. 19(2).
- Kidd PM, Head K. 2005. A review of the bioavailability and clinical efficacy of milk thistle phytosome: a silybin-phosphatidylcholine complex. *Alternative Med Rev.* 10(3):193-203.
- Kidd PM. 2009. Bioavailability and activity of phytosome complexes from botanical polyphenols: The silymarin, curcumin, green tea, and grape seed extracts. *Alternative Medicine Review*. 14(3): 226-246.
- Kumar K, Rai AK. 2011. Development and evaluation of proniosomes as a promising Drg carrier to improve transdermal drug delivery. *IRJP*. 2(11): 71-74.
- Kusumawardhani N, Sulistyarti H, Atikah. 2015. Penentuan Panjang Gelombang Maksimum dan pH Optimum dalam Pembuatan Tes Kit Sianida

- Berdasarkan Pembentukan Hidrindantin. Jurnal. Jurusan Kimia. Fakultas Matematika dan Ilmu Pengetahuan Alam. Universitas Brawijaya, Malang.
- Lalatendu Panigrahi, Snigdha P dan Saroj KG. 2004. Design and characterization of mucoadhesive buccal patches of salbutamol sulphate. *Acta Poloniae Pharmaceutica-Drug Research.* 61(5):351- 360.
- Leekumjorn, Sukit. 2004. Synthesis and characterization of potential drug delivery systems using nonionic surfactant “niosom”.
- Liu QM. 2010. Optimization of ultrasonic-assisted extraction of chlorogenic acid from *Folium eucommiae* and evaluation of its antioxidant activity. *Journal of Medicinal Plants Research.* 4(23):2503-2511.
- Maghraby GMME, Williams AC, Barry BW. 2001. Skin delivery of 5-fluorouracil from ultra deformable and standard liposomes in-vitro. *J Pharm Pharmacol.* 53:1069–1077.
- Makeshawar K, Wasankar, S. 2013. Niosome: a novel drug delivery system. *Asian Parmapres.* 3(1):16-20.
- Manglani N, Vaishnava S. 2012. Phytosomes: A novel herbal drug delivery. *Journal of Pharmaceutical and Scientific Innovation.* 35-40.
- Mason TJ, Lorimer JP. 2002. *Applied sonochemistry: The uses of power ultrasound in chemistry and processing.* Verlag: Whiley-VCH.
- Mayes AP. 2003. Lipid yang memiliki makna fisiologi. Dalam: Murray RK, Granner DK, Mayes PA, Rodwell VW. *Biokimia harper.* Jakarta: EGC. hlm 151-154.
- Medicago AB. 2010. *Phosphate Buffered Saline Specification Sheet.*
- Molyneux P. 2004. The use of the stable free radical diphenylpicrylhydrazyl (DPPH) for estimating antioxidant activity. *Songklanakarin J. Sci. Technol.* 26(2): 211-219.
- Muller RH, Jacobs C, Kayser O. 2000. Nanosuspensions for the formulation of poorly soluble drugs. *Pharmaceutical Emulsions and Suspensions.* 105:383-407.
- Nakahira A, Nakamura S, Horimoto M. 2007. Synthesis of Modified Hydroxyapatite (HAP) Substituted with Fe ion for DDS Application.

- Osaka: IEEE Transactions on Magnetic* 43 (6): 2465-2467. Dalam: Hapsari BW. 2009. Sintesis Nanosfer Berbasis *Ferrofluid* dan *Polylactic Acid* (PLA) dengan Metode Sonikasi [skripsi]. Bogor: Fakultas Matematika dan Ilmu Pengetahuan Alam, Institut Pertanian Bogor.
- Nandure HP, Puranik P, Giram P, Lone V. 2013. Ethosome: a novel drug carrier. *Int J Pharm Res Allied Sci.* 2(3):18–30.
- New RRC. 1990. Introduction. In: New RRC (ed). *Liposomes. A practical Approach.* IRL Press. hlm 1-31.
- Papakostas D, Rancan F, Sterry W, Peytavi UB, Vogt, A. 2011. Nanoparticles in dermatology. *Department of Dermatology and Allergy Research.* 303:533–550.
- Patihul H, Dara AI. 2017. Teknik meningkatkan kelarutan obat. *Farmaka.* 15(4).
- Pham, Thi Thuy. 2012. Colloid and surfaces B: Biointerfaces, liposome and niosome preparation using a membran contractor for scale-up. *France.* 94:15-21
- Pirrung MC. 2007. *The Synthetic Organic Chemist's Companion.* New Jersey: John Wiley & Sons Inc.
- Prochazkova D, Bousova I, Wilhelmova N. 2011. Antioxidant and prooxidant properties of flavonoids. *Fitoterapia.* 82:513–523.
- Purwaningsih EH. 2002. Inkorporasi Metilprednisolon Palmitat pada Membran Liposom yang Mengandung Tetraeter Lipid Berasal dari Archea serta Gambaran Distribusinya di Beberapa Organ Limfoid pada Mencit [Disertasi]. Depok: Program Studi Ilmu Biomedik FKUI, Program Doktor UI.
- Rachmawati H, Reker-Smit C, Lub-de Hooge MN, van Loenen-Weemaes A, Poelstra K, Beljaars L, *et al.* 2007. Chemical modification of interleukin-10 with mannose 6-phosphate groups yield a liver-selective cytokine. *DMD.* 35:814-821.
- Ramadon D, Mun'im A. 2016. Utilization of nanotechnology in drug delivery system for natural products. *Jurnal Ilmu Kefarmasian Indonesia.* 14(2):118-127.

- Rasaie S, Ghanbarzadeh S, Mohammadi M, HamishehkarH. 2014. Nano phytosomes of quercetin: a promising formulation for fortification of food products with antioxidants. *Pharmaceutical sciences.* 20:96-101.
- Redha A. 2010. Flavonoid: Struktur, sifat antioksidatif dan peranannya dalam sistem biologis. *Jurnal Belian.* 9(2): 196-202.
- Rohman A, Riyanto S. 2005. Daya antioksidan ekstrak etanol Daun Kemuning (*Murraya paniculata* (L) Jack) secara *in vitro*. *Majalah Farmasi Indonesia.* 16 (3): 136–140.
- Ronson. 2012. UV/Vis/IR Spectroscopy Analysis of Nanoparticles. *Nano Composit.* 1(1):1-6.
- Ross JA, Kasum CM. 2002. Dietary flavonoids: bioavailability, metabolic effects, and safety. *Annual Review of Nutrition.* 22: 19–34.
- Rowe RC, Sheskey PJ, Owen SC. 2006. *Handbook of Pharmaceutical Excipients Fifth Edition.* London: Pharmaceutical Press. hlm 182-184.
- Ruozi B, Belletti D, Tombesi A, Tosi G, Bondioli L, Forni F, Vandelli MA, *et al.* 2011. AFM, ESEM, TEM, and CLSM in liposomal characterization: a comparative study. *International Journal of Nanomedicine.* 6:557–563.
- Sharma S, Sikarwar M. 2005. Phytosome: a review. *Planta. Indica.* 1(2):1-3.
- Shivanand P, Kinjal P. 2010. Phytosomes: technical revolution in phytomedicine. *International Journal of PharmTech Research CODEN (USA): IJPRIF.* 2(1):627-631.
- Singh D, Rawat MS, Semalty A, Semalty M. 2012. Rutin-phospholipid complex: an innovative technique in novel drug delivery system-NDDS. *Curr Drug Deliv.* 9:305-314.
- Sinko, P. J. 2011. *Martin Farmasi Fisika dan Ilmu Farmasetika edisi 5,* diterjemahkan oleh Tim Alih Bahasa Sekolah Farmasi ITB, 706. Jakarta: Penerbit Buku Kedokteran EGC.
- Sjahbanar SZ, Setiadi E, Stryer L. 2000. *Biokimia Vol 1. Edisi 4.* Jakarta: EGC. hlm 264-77.

- Steven, Malcolm P. 2001. *Polymer Chemistry: An Introduction*. Oxford University Press. Diterjemahkan oleh Iis Sopyan. *Kimia Polimer*. Jakarta: PT Pradnya Paramita.
- Sultana B, Anwar F. 2008. Flavonols (kaempferol, quercetin, myricetin) contents of selected fruits, vegetables and medicinal plants. *Food Chemistry*. 879–884.
- Sunardi IK. 2007. Uji Aktivitas Antioksidan Ekstrak Belimbing Wuluh (*Averrhoa bilimbi* L) terhadap 1,1-diphenyl-2-picrilhidrazyl (DPPH), *Seminar Nasional Teknologi (SNT)*, D-III Teknologi Farmasi Fakultas Teknik USB, Yogyakarta.
- Suslick KS, Price GJ. 1999. Application of ultrasound to materials chemistry. *Annual Review of Materials Science*. 29: 295-326
- Tarekegn A, Joseph NM, Palani S, Zacharia A. 2010. Niosom in targeted drug delivery: some recent advances. *IJPSPR*. Vol 1(9):1-8.
- Tertasari Hermini. 2003. *Validasi Metode Analisis*. Pusat pengkajian Obat dan Makanan BPOM.
- Thurapati, P.R., Mettu S.R., Veerareddy, P.R. 2011. Phytosomes: A Novel Phyto-Phospholipid Carriers for Herbal Drug Delivery. *International Research Journal of Pharmacy*. 2(6): 28-33.
- Tippler PA. 1998. *Fisika untuk Sains dan Teknik*. Prasetyo L & Adi RW, penerjemah. Jakarta: Erlangga. Terjemahan dari: Physics for Scientists and Engineers.
- Tulandi P, Grace, Sudewi S, Lolo W, Astuty. 2015. Validasi metode analisis untuk penetapan kadar paracetamol dalam sediaan tablet secara spektrofotometri ultraviolet. FMIPA. UNSRAT, Manado.
- Utama RB, I Gusti. 2016. Korelasi linier dan berganda. Universitas Dhyana Pura, Bali.
- Verma S, SK Singh, Syan N, Mathur P, Valecha V. 2010. Nanoparticle vesicular system: A versatile tool for drug delivery. *J. Chem. Res.* Vol 2 (2):496-509.
- Wardiyati S. 2004. Pemanfaatan ultrasonik dalam bidang kimia. Dalam: *Penguasaan IPTEK Bahan untuk Meningkatkan Kualitas Produk*

- Nasional. Prosiding Pertemuan Ilmiah IPTEK Bahan; Serpong, 7 Sep 2004. Serpong: P3IB Batan. hlm 419-424.
- William D, O'brien JR. 1986. Biological effects of ultrasound: Rationale for the measurement of selected ultrasonic output quantities. *Echocardiography*. Vol.3 No.3.
- Wilson and Gisvold. 1982. *Buku Teks Wilson dan Gisvold: Kimia Farmasi dan Medisinal Organik, Edisi VIII, I.B.* Philadelphia-Toronto: Lippincott Company. hlm 351-353.
- Wu L, Zhang J, Watanabe W. 2010. Physical and chemical stability of drug nanoparticles. *Advances Drug Delivery Reviews*. 63: 456-469.
- Yao YS, Li GB, Xie Y, Ma P, Li GW, Meng QC, Wu T, et al. 2013. Preformulation studies of myricetin; a natural antioxidant flavonoid. *Pharmazie*. 69: 19-26.
- Yashu Y, et al. 2014. Preformulation studies of myricetin: a natural antioxidant flavonoid. *Pharmazie*. 69: 19-26.

L

A

M

P

I

R

A

N

Lampiran 1. Sertifikat analisis myricetin



Certificate of Analysis

Print Date: Jul 20th 2017www.tocris.com

Product Name: Myricetin

Catalog No.: 6189

Batch No.: 1

CAS Number: 529-44-2

IUPAC Name: 3,5,7-Trihydroxy-2-(3,4,5-trihydroxyphenyl)-4H-1-benzopyran-4-one

1. PHYSICAL AND CHEMICAL PROPERTIES

Batch Molecular Formula: C₁₅H₁₀O₈·H₂O

Batch Molecular Weight: 336.26

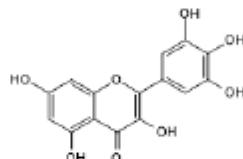
Physical Appearance: Yellow solid

Solubility: DMSO to 100 mM

ethanol to 50 mM

Storage: Store at -20°C

Batch Molecular Structure:



2. ANALYTICAL DATA

HPLC: Shows 97.7% purity

¹H NMR: Consistent with structure

Mass Spectrum: Consistent with structure

Microanalysis: Carbon Hydrogen Nitrogen

Theoretical 53.58 3.6

Found 53.6 3.57

Caution - Not Fully Tested • Research Use Only • Not For Human or Veterinary Use

bio-tecne.com
info@bio-tecne.com
techsupport@bio-tecne.com

North America
 Tel: (800) 343 7475

China
info.cn@bio-tecne.com
 Tel: +86 (21) 52380373

Europe Middle East Africa
 Tel: +44 (0)1235 529449
 Tel: +86 (21) 52380373

Rest of World
www.tocris.com/distributors
 Tel: +1 612 379 2956

Lampiran 2. Sertifikat analisis fosfatidilkolin

Lipoid

PHOSPHOLIPID GmbH - Member of the Lipoid Group

ANALYTICAL DATA

AN33102844

- 1 -

PHOSPHOLIPON 90 G

Batch	228154-3180044	Recommended storage Date of production	n.m.t., +8 °C 07/2018
-------	----------------	---	--------------------------

Sample for laboratory use only					
Parameter	Result	Specification min	Specification max	Unit	Method
Phosphatidylcholine	96,1	94,0	102,0	% (m/m)	05.P07.857
Identity (TLC)	conform to reference	conform to reference			05.P08.300
Lysophosphatidylcholine	1,4		4,0	% (m/m)	05.P07.857
Nonpolar Lipids	1,0		3,0	% (m/m)	05.P03.008
Tocopherol	0,21		0,30	% (m/m)	05.P07.142
Acid value	0,2		0,5		05.P03.002
Peroxide value	1,8		5,0		05.P06.120
Water	0,2		1,5	% (m/m)	05.P10.013
Toluene insolubles	0,00		0,05	% (m/m)	05.P06.001
Ethanol	0,1		0,2	% (m/m)	05.P05.049
Heavy metals	< 10		10	mg/kg	USP <231> method II
Arsenic	< 0,015		0,15	mg/kg	USP <232>/ USP <233>
Lead	< 0,015		0,10	mg/kg	USP <232>/ USP <233>
Appearance	yellowish, waxy	yellowish, waxy			05.P06.155

./2

Lampiran 3. Alat yang digunakan dalam penelitian

Alat	Nama alat	Kegunaan
	Neraca analitik	Menimbang bahan baku dan eksipien
	Spektrofotometer UV-Vis	Membaca absorbansi, mencari lamda maksimum dan OT sampel
	<i>Magnetic stirrer</i>	Menghomogenkan formula dengan pengadukan
	<i>Rotary evaporator</i>	Menguapkan pelarut organik



pH meter

Mengukur pH



Sonikasi probe

Menghomogenkan dan memperkecil ukuran partikel



Particle size analyzer

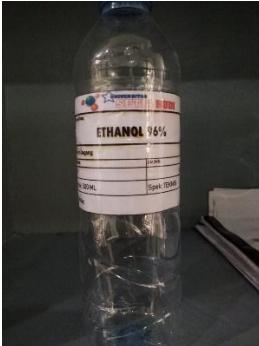
Mengukur ukuran partikel dan zeta potensial

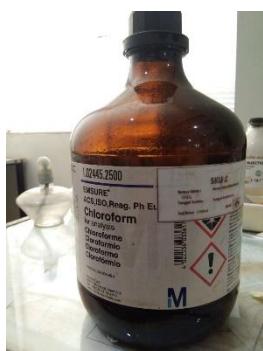


Sentrifugasi

Memisahkan obat bebas dengan obat yang terjerap dalam nanofitosom

Lampiran 4. Bahan yang digunakan dalam penelitian

Alat	Nama alat	Kegunaan
	Myricetin	Zat aktif
	Fosfatidilkolin	Zat pembawa
	Kolesterol	Penstabil
	Etanol 96%	Pelarut myricetin dan fosfatidilkolin



Kloroform

Pelarut kolesterol

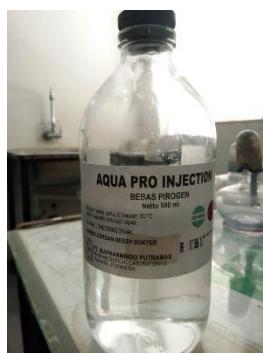
*Phosphat Buffer Saline*
(PBS) pH 7,4Untuk menghidrasi
lapisan tipis nanofitosom

DPPH

Radikal bebas untuk
mengukur aktivitas
antioksidan sampel

Etanol p.a

Pelarut DPPH



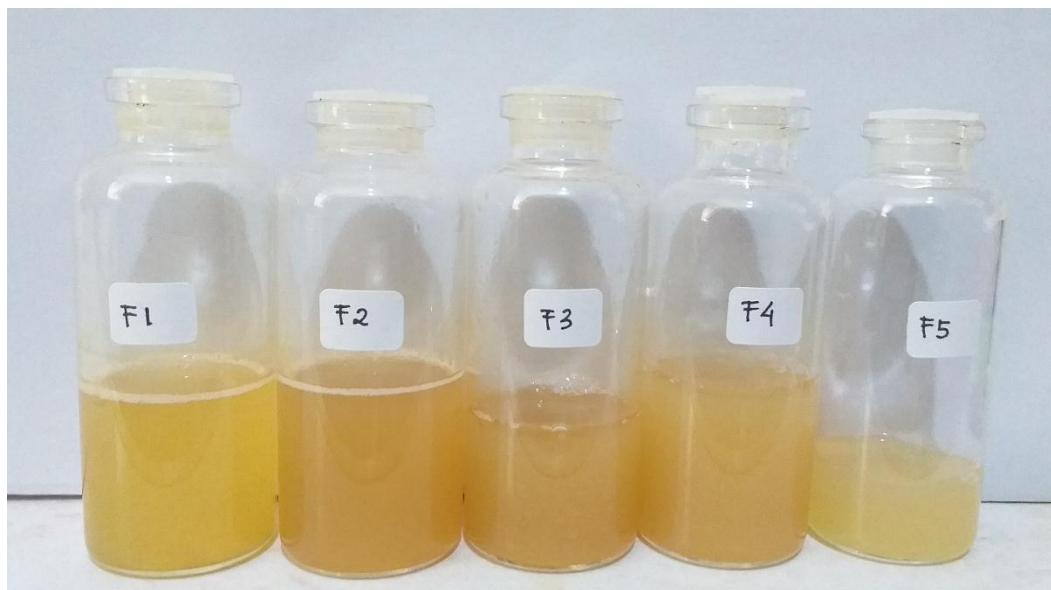
Aqua p.a

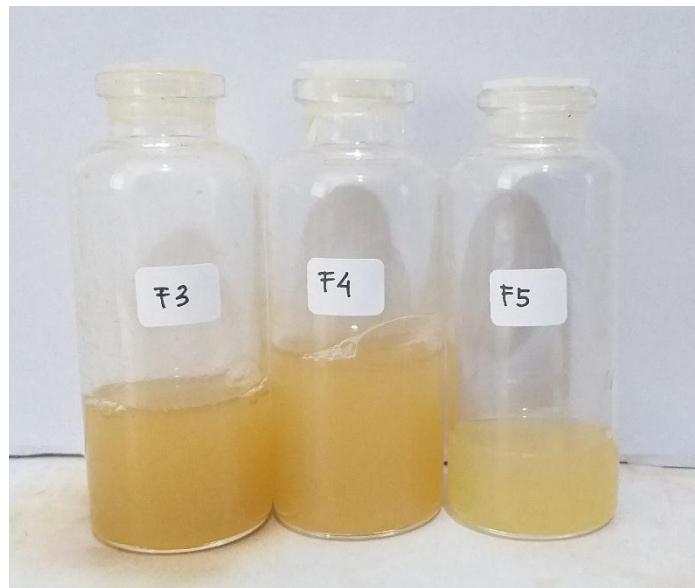
Untuk membuat larutan
Phosphat Buffer Saline
(PBS)



NaOH

Mengendalikan tingkat
keasaman atau pH pada
larutan PBS

Lampiran 5. Nanofitosom myricetin setelah disonikasi

Lampiran 6. Nanofitosom myricetin setelah uji stabilitas

Lampiran 7. Perhitungan formula

Kadar myricetin murni : 98%

Berat molekul myricetin : 318,2351 gram/mol

Berat molekul fosfatidilkolin : 768 gram/mol

Berat molekul kolesterol : 386,67 gram/mol

- Kandungan myricetin dalam 10 mg serbuk

$$\begin{aligned}\text{Myricetin (mg)} &= 10 \text{ mg} \times \frac{98}{100} \\ &= 9,8 \text{ mg}\end{aligned}$$

- Mol myricetin dalam 10 mg serbuk

$$\begin{aligned}\text{Mol myricetin (\mu mol)} &= \frac{9,8 \text{ mg}}{318235 \text{ mg} \times 10^{-6} \mu \text{mol}} \\ &= 30,7949 \mu \text{mol}\end{aligned}$$

- Fosfatidilkolin

- a. Formula 1:1

$$\begin{aligned}\text{Fosfatidilkolin yang ditimbang} &= 30,7949 \mu \text{mol} \times \frac{768.000 \text{ mg}}{10^6 \mu \text{mol}} \\ &= 23,6505 \text{ mg}\end{aligned}$$

- b. Formula 1:2

$$\begin{aligned}\text{Fosfatidilkolin yang ditimbang} &= 61,5898 \mu \text{mol} \times \frac{768.000 \text{ mg}}{10^6 \mu \text{mol}} \\ &= 47,3010 \text{ mg}\end{aligned}$$

- c. Formula 1:3

$$\begin{aligned}\text{Fosfatidilkolin yang ditimbang} &= 92,3847 \mu \text{mol} \times \frac{768.000 \text{ mg}}{10^6 \mu \text{mol}} \\ &= 70,9514 \text{ mg}\end{aligned}$$

- d. Formula 1:4

$$\begin{aligned}\text{Fosfatidilkolin yang ditimbang} &= 123,1796 \mu \text{mol} \times \frac{768.000 \text{ mg}}{10^6 \mu \text{mol}} \\ &= 94,6019 \text{ mg}\end{aligned}$$

- e. Formula 1:5

$$\begin{aligned}\text{Fosfatidilkolin yang ditimbang} &= 153,9745 \mu \text{mol} \times \frac{768.000 \text{ mg}}{10^6 \mu \text{mol}} \\ &= 118,2524 \text{ mg}\end{aligned}$$

- Kolesterol

Perbandingan myricetin:kolesterol (1:0,2), maka jumlah kolesterol yang dibutuhkan $30,7949 \mu\text{mol} \times 0,2 = 6,1590 \mu\text{mol}$

$$\begin{aligned}\text{Kolesterol yang ditimbang} &= 6,1590 \mu\text{mol} \times \frac{386.670 \text{ mg}}{10^6 \mu\text{mol}} \\ &= 2,3815 \text{ mg}\end{aligned}$$

Lampiran 8. Ukuran partikel

Ukuran partikel	Formula				
	1	2	3	4	5
Replikasi 1	905,2	688,5	163,5	211,2	139,1
Replikasi 2	912,2	471,8	155,9	197,9	139,2
Replikasi 3	930,5	619,1	192,9	211,1	140,1
Rata- rata±SD	915,967	539,133	170,767	206,733	139,467
	13,064	110,659	19,541	7,650	0,551

Lampiran 9. Indeks Polidispersitas

Indeks polidispersitas	Formula				
	1	2	3	4	5
Replikasi 1	0,572	1,000	0,474	0,385	0,375
Replikasi 2	0,484	1,000	0,417	0,405	0,387
Replikasi 3	0,575	0,924	0,431	0,454	0,377
Rata-rata±SD	0,544 0,052	0,975 0,044	0,441 0,030	0,415 0,036	0,380 0,006

Hasil Ukuran Partikel F1

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: Nanofitosom Myricetin F1 1

SOP Name: mansettings.nano

General Notes:

File Name: NanofotosomMyricetin_2...	Dispersant Name: Water
Record Number: 1	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: Jumat, 26 April 2019 09.51....

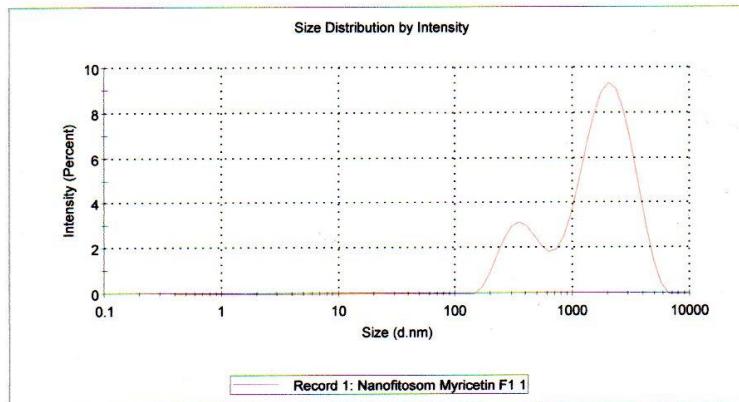
System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 343,8	Measurement Position (mm): 1,25
Cell Description: Disposable sizing cuvette	Attenuator: 4

Results

		Size (d.nm)	% Intensity:	St Dev (d.nm)
Z-Average (d.nm):	905,2	Peak 1:	2091	79,5
Pdl:	0,572	Peak 2:	372,4	20,5
Intercept:	0,860	Peak 3:	0,000	0,000

Result quality Good



Hasil Ukuran Partikel F2

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: Nanofitosom Myricetin F2 1

SOP Name: mansettings.nano

General Notes:

File Name: NanofotosomMyricetin_2...	Dispersant Name: Water
Record Number: 7	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absortion: 0,100	Measurement Date and Time: Jumat, 26 April 2019 10.21....

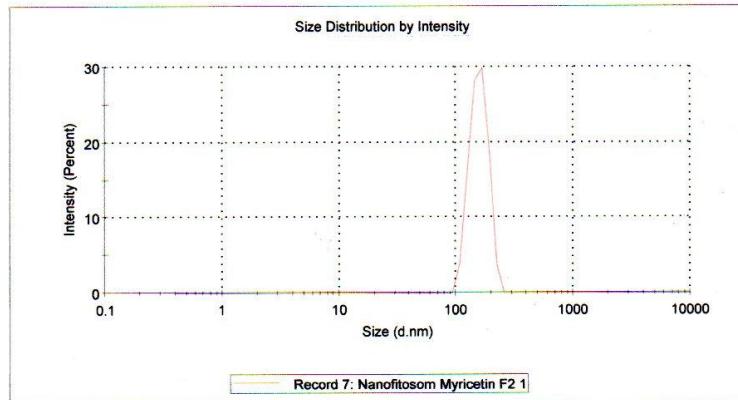
System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 323,3	Measurement Position (mm): 1,05
Cell Description: Disposable sizing cuvette	Attenuator: 3

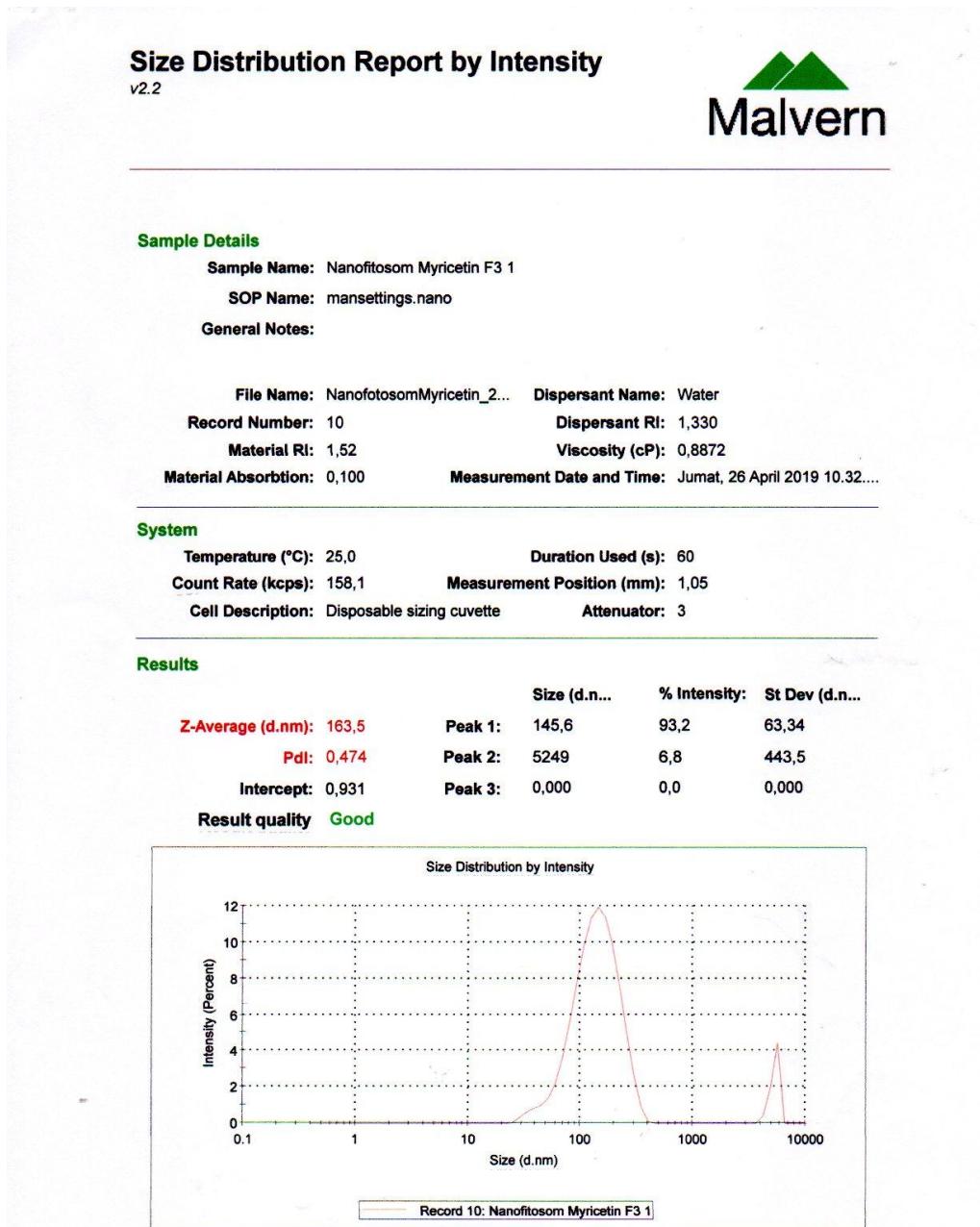
Results

		Size (d.n...)	% Intensity:	St Dev (d.n...
Z-Average (d.nm):	688,5	Peak 1:	155,8	100,0
Pdi:	1,000	Peak 2:	0,000	0,000
Intercept:	0,795	Peak 3:	0,000	0,000

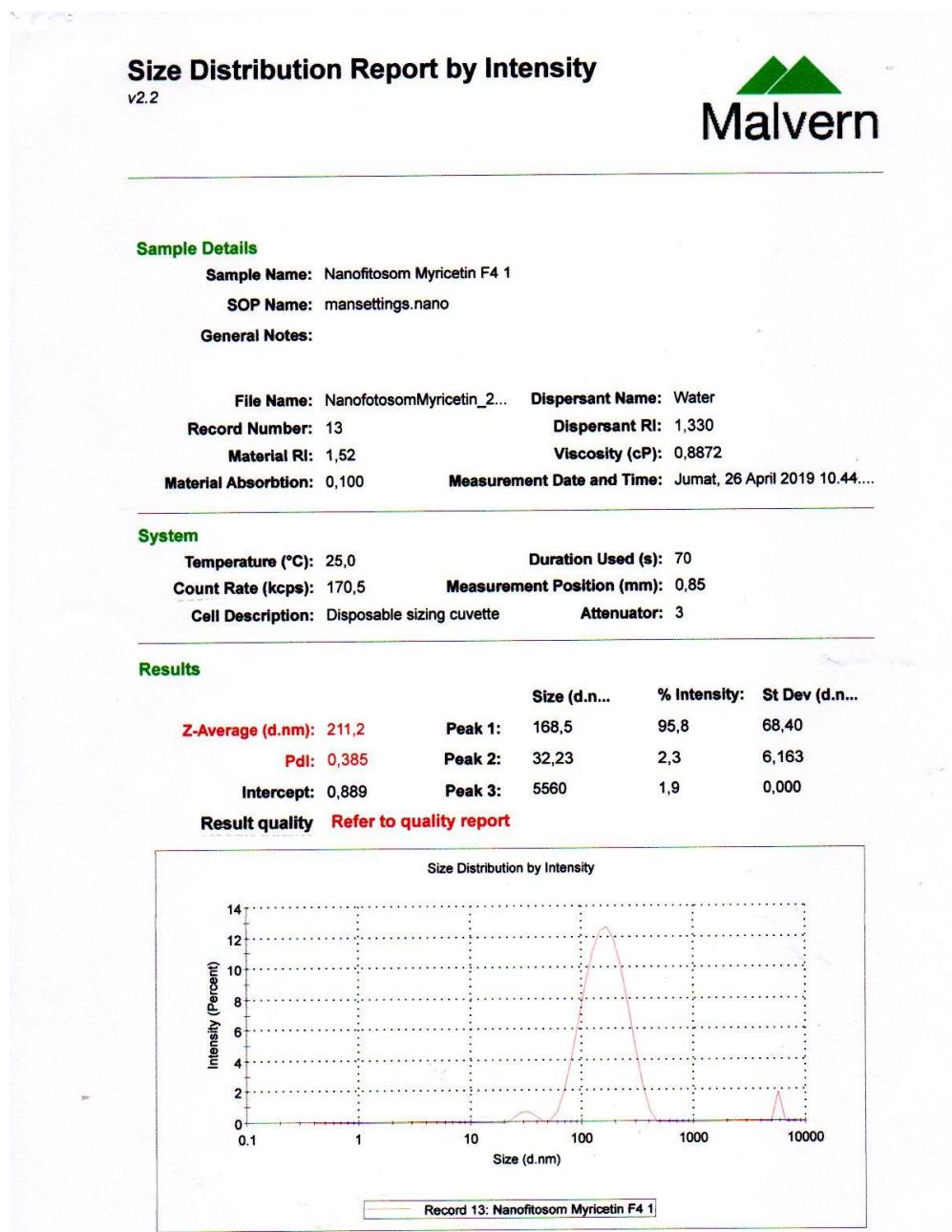
Result quality Refer to quality report



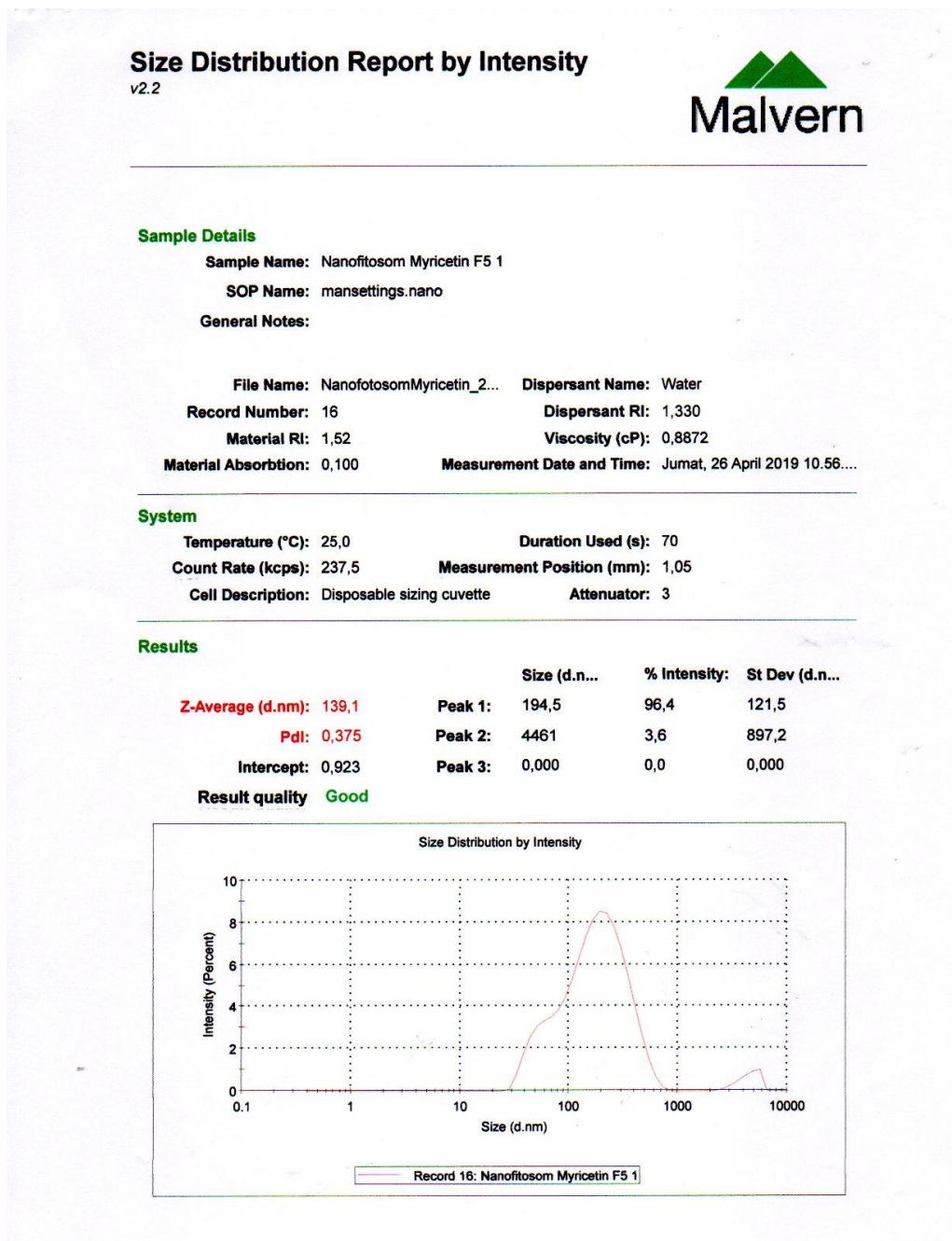
Hasil Ukuran Partikel F3



Hasil Ukuran Partikel F4

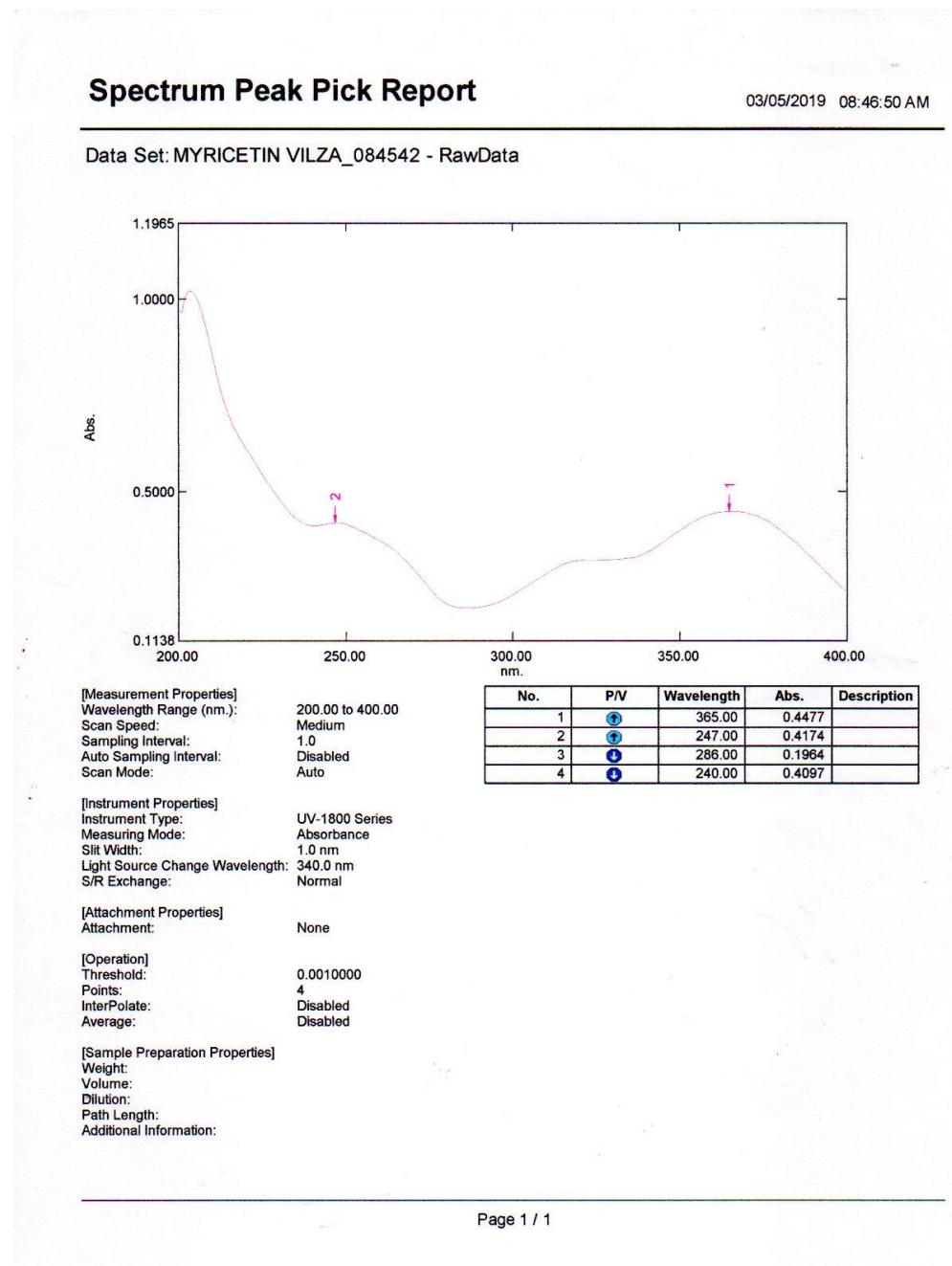


Hasil Ukuran Partikel F5



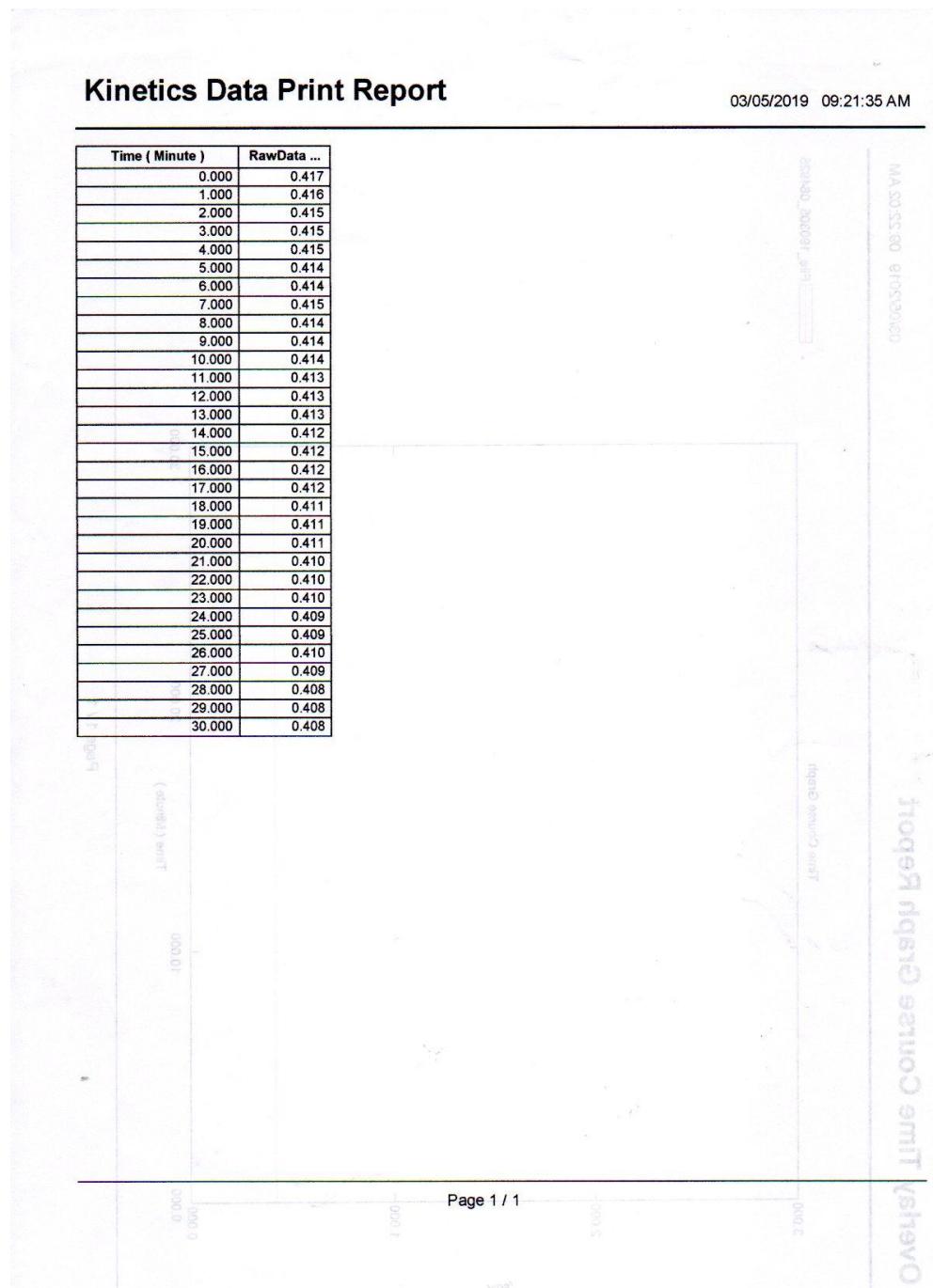
Lampiran 10. Kurva kalibrasi

a. Penetapan panjang gelombang myricetin dalam medium dapar pH 7,4



b. Penentuan *Operating Time* (OT)

Pembacaan *operating time* didapat nilai serapan yang stabil pada menit ke 14-17.



c. Pembuatan larutan induk myricetin dalam medium dapar pH 7,4

$$\text{Berat kertas kosong} = 276,9 \text{ mg}$$

$$\text{Berat kertas + bahan} = 287,2 \text{ mg}$$

$$\text{Berat bahan} = 287,2 \text{ mg} - 276,9 \text{ mg}$$

$$= 10,3 \text{ mg}$$

$$\text{Berat kertas + sisa} = 277,2 \text{ mg}$$

$$\text{Berat sisa} = 277,2 \text{ mg} - 276,9 \text{ mg}$$

$$= 0,3 \text{ mg}$$

$$\text{Berat myricetin} = 10,3 \text{ mg} - 0,3 \text{ mg}$$

$$= 10 \text{ mg}$$

$$\text{Volume dapar pH 7,4} = 100 \text{ mL}$$

$$\text{Larutan stok} = 10 \text{ mg}/100 \text{ mL}$$

$$= 100 \text{ mg}/1000 \text{ mL}$$

$$= 100 \text{ ppm}$$

d. Kurva baku myricetin dalam medium dapar pH 7,4

Larutan induk myricetin dibuat seri konsentrasi 6 ppm, 8 ppm, 10 ppm, 12 ppm dan 14 ppm dalam 25 mL.

1. 6 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 6 \text{ ppm}$$

$$V_1 = 0,5 \text{ mL}$$

2. 8 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 8 \text{ ppm}$$

$$V_1 = 0,8 \text{ mL}$$

3. 10 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 10 \text{ ppm}$$

$$V_1 = 1 \text{ mL}$$

4. 12 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 12 \text{ ppm}$$

$$V_1 = 1,2 \text{ mL}$$

5. 14 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 14 \text{ ppm}$$

$$V_1 = 1,4 \text{ mL}$$

6. 16 ppm

$$V_1 \times C_1 = V_2 \times C_2$$

$$V_1 \times 100 \text{ ppm} = 10 \text{ mL} \times 16 \text{ ppm}$$

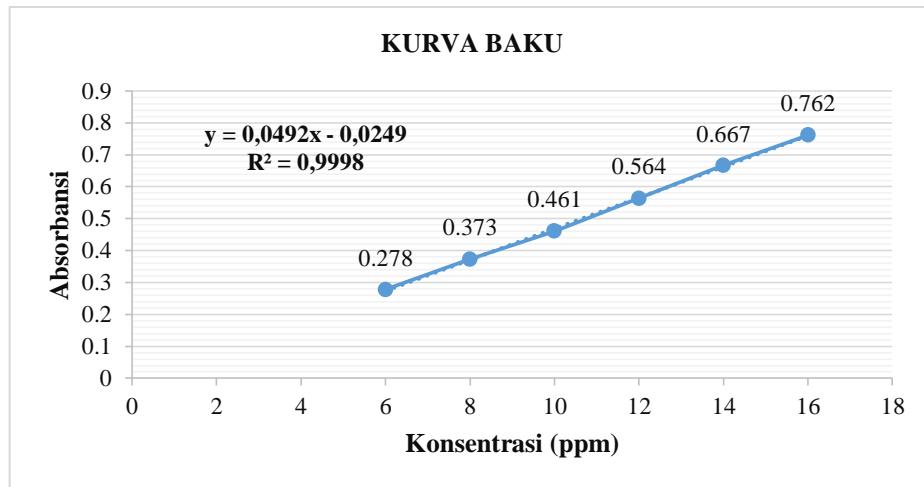
$$V_1 = 1,6 \text{ mL}$$

e. Tabel kurva baku medium dapar pH

Larutan stok 100 ppm \rightarrow 10 mg myricetin + 5 ml etanol p.a + ad 100 ml
dapar fosfat pH 7,4

Konsentrasi (ppm)	Absorbansi
6	0,278
8	0,373
10	0,461
12	0,564
14	0,667
16	0,762

Persamaan regresi linier antara konsentrasi (ppm) dan serapan diperoleh:



$$a = -0,0249$$

$$b = 0,0492$$

$$r = 0,9998$$

$$y = a + b \cdot x$$

$$y = -0,0249 + 0,0492 \cdot x$$

Keterangan:

x = konsentrasi

y = serapan

Lampiran 11. Verifikasi metode analisis

a. Linieritas

Konsentrasi (ppm)	Absorbansi
6	0,278
8	0,373
10	0,461
12	0,564
14	0,667
16	0,762

$$a = -0,0249$$

$$b = 0,0492$$

$$r = 0,9998$$

Hasil linearitas diperoleh R sebesar 0,9998, sehingga dapat disimpulkan bahwa data linier.

b. Penentuan LOD dan LOQ

KONSENTRASI (PPM)	ABSORBANSI (y)	\hat{y}	$ y - \hat{y} $	$ y - \hat{y} ^2$
6	0,278	0,2703	0,0077	0,00005929
8	0,373	0,3687	0,0043	0,00001849
10	0,461	0,4671	-0,0061	0,00003721
12	0,564	0,5655	-0,0015	0,00000225
14	0,667	0,6639	0,0031	0,00000961
16	0,762	0,7623	-0,0003	0,00000009

$$\text{Jumlah total } (\sum |y - \hat{y}|)^2 = 0,00012694$$

Nilai \hat{y} diperoleh dari substitusi konsentrasi dalam persamaan $\hat{y} = -0,0249 + 0,0492x$ dengan x adalah konsentrasi (ppm) dan \hat{y} adalah serapan.

$$1. \quad \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 6$$

$$\hat{y} = 0,2703$$

$$2. \quad \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 8$$

$$\hat{y} = 0,3687$$

$$3. \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 10$$

$$\hat{y} = 0,4671$$

$$4. \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 12$$

$$\hat{y} = 0,5655$$

$$5. \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 14$$

$$\hat{y} = 0,6639$$

$$6. \hat{y} = -0,0249 + 0,0492x$$

$$\hat{y} = -0,0249 + 0,0492 \times 16$$

$$\hat{y} = 0,7623$$

- $S_{x/y} = \sqrt{\frac{(\sum|y-\hat{y}|)^2}{n-2}}$

$S_{x/y}$ = simpangan baku residual
 $(\sum|y-\hat{y}|)^2$ = jumlah kuadrat total residual
 n = jumlah data
 $S_{x/y} = \sqrt{\frac{0,00012694}{5-2}} = 0,00650487$

- $LOD = 3,3 \times \frac{Sx/y}{b}$
 $= 3,3 \times \frac{0,00650487}{0,0492}$
 $= 0,4363 \text{ ppm}$
 $y = -0,0249 + 0,0492x$
 $= -0,0249 + 0,0492 (0,4363)$
 $= -0,0249 + 0,0215$

Serapan LOD = -0,0034

- LOQ
$$= 10 \times \frac{Sx/y}{b}$$

$$= 10 \times \frac{0,00650487}{0,0492}$$

$$= 1,3221 \text{ ppm}$$

$$y = -0,0249 + 0,0492x$$

$$= -0,0249 + 0,0492 (1,3221)$$

$$= -0,0249 + 0,0650$$

Serapan LOQ = 0,0401

c. Akurasi

Konsentrasi	Absorbansi	Konsentrasi terukur (ppm)	Konsentrasi sebenarnya (ppm)	Konsentrasi (%)	% recovery	Rata-rata % recovery
80%	0,497	10,6112	10	106		
	0,499	10,6518	10	107	106%	
	0,492	10,5095	10	105		
100%	0,598	12,6646	12	106		
	0,596	12,6240	12	105	106%	105%
	0,601	12,7256	12	106		
120%	0,684	14,4132	14	103		
	0,687	14,4742	14	103	103%	
	0,689	14,5148	14	104		

a = -0,0249

b = 0,0492

r = 0,9998

Perhitungan konsentrasi (ppm)

- Konsentrasi 80%

$$\text{Replikasi 1} = \frac{\text{Absorbansi} - a}{b} = \frac{0,497 - (-0,0249)}{0,0492} = 10,6112$$

$$\text{Replikasi 2} = \frac{\text{Absorbansi} - a}{b} = \frac{0,499 - (-0,0249)}{0,0492} = 10,6518$$

$$\text{Replikasi 3} = \frac{\text{Absorbansi} - a}{b} = \frac{0,492 - (-0,0249)}{0,0492} = 10,5095$$

- Konsentrasi 100%

$$\text{Replikasi 1} = \frac{\text{Absorbansi} - a}{b} = \frac{0,598 - (-0,0249)}{0,0492} = 12,6646$$

$$\text{Replikasi 2} = \frac{\text{Absorbansi} - a}{b} = \frac{0,596 - (-0,0249)}{0,0492} = 12,6240$$

$$\text{Replikasi 3} = \frac{\text{Absorbansi} - a}{b} = \frac{0,601 - (-0,0249)}{0,0492} = 12,7256$$

- Konsentrasi 120%

$$\text{Replikasi 1} = \frac{\text{Absorbansi} - a}{b} = \frac{0,684 - (-0,0249)}{0,0492} = 14,4132$$

$$\text{Replikasi 2} = \frac{\text{Absorbansi} - a}{b} = \frac{0,687 - (-0,0249)}{0,0492} = 14,4742$$

$$\text{Replikasi 3} = \frac{\text{Absorbansi} - a}{b} = \frac{0,689 - (-0,0249)}{0,0492} = 14,5148$$

d. Presisi

Replikasi	Absorbansi	Konsentrasi terukur (ppm)	Konsentrasi sebenarnya (ppm)
1	0,475	10,1639	10
2	0,480	10,2655	10
3	0,479	10,2452	10
4	0,489	10,4485	10
5	0,484	10,3468	10
6	0,488	10,4282	10
7	0,492	10,5095	10
8	0,480	10,2655	10
9	0,473	10,1232	10
10	0,484	10,3468	10

$$a = -0,0249$$

$$b = 0,0492$$

$$r = 0,9998$$

- Rata-rata konsentrasi = 10,3143 ppm
- SD = 0,124671

- CV $= \frac{SD}{Rata-rata} \times 100\%$
 $= \frac{0,124671}{10,3143} \times 100\%$
 $= 0,012087$
- RSD $= 1\% < 2\%$
Keterangan:
 - SD = Simpangan baku
 - RSD = Simpangan baku relatif
 - CV = Koefisien variasi

Lampiran 12. Efisiensi penjerapan

$$y = -0,0249 + 0,0492.x$$

Efisiensi penjerapan	Formula		
	3	4	5
Replikasi 1	0,534	0,410	0,396
Replikasi 2	0,537	0,413	0,391
Replikasi 3	0,541	0,412	0,386
Rata-rata Absorbansi	0,537	0,412	0,391
%EE	88,57%	91,12%	91,54%

Formula 3

- a. Perhitungan kadar myricetin terjerap menggunakan persamaan regresi linier yang menggunakan pelarut PBS:

$$\begin{aligned} y &= a + b.x \\ 0,5373 &= -0,0249 + 0,0492.x \\ 0,0492.x &= 0,5622 \\ x &= 11,4303 \text{ ppm} \end{aligned}$$

$$b. \Sigma \text{myricetin tidak terjerap} = \frac{11,4303 \text{ ppm}}{100 \text{ ppm}} \times 10 \text{ mg}$$

$$= 1,14303 \text{ mg}$$

$$\begin{aligned} c. \% \text{ Efisiensi penjerapan} &= \frac{TD-FD}{TD} \times 100\% \\ &= \frac{10 \text{ mg} - 1,14303 \text{ mg}}{10 \text{ mg}} \times 100\% \\ &= 88,57\% \end{aligned}$$

Formula 4

- a. Perhitungan kadar myricetin terjerap menggunakan persamaan regresi linier yang menggunakan pelarut PBS:

$$\begin{aligned} y &= a + b.x \\ 0,4117 &= -0,0249 + 0,0492.x \\ 0,0492.x &= 0,4366 \end{aligned}$$

$$x = 8,8767 \text{ ppm}$$

b. $\Sigma \text{ myricetin tidak terjerap} = \frac{8,8767 \text{ ppm}}{100 \text{ ppm}} \times 10 \text{ mg}$
 $= 0,88767 \text{ mg}$

c. % Efisiensi penjerapan $= \frac{TD - FD}{TD} \times 100\%$
 $= \frac{10 \text{ mg} - 0,88767 \text{ mg}}{10 \text{ mg}} \times 100\%$
 $= 91,12\%$

Formula 5

- a. Perhitungan kadar myricetin terjerap menggunakan persamaan regresi linier yang menggunakan pelarut PBS:

$$\begin{aligned} y &= a + b.x \\ 0,391 &= -0,0249 + 0,0492.x \\ 0,0492.x &= 0,4159 \\ x &= 8,4558 \text{ ppm} \end{aligned}$$

b. $\Sigma \text{ myricetin tidak terjerap} = \frac{8,4558 \text{ ppm}}{100 \text{ ppm}} \times 10 \text{ mg}$
 $= 0,84558 \text{ mg}$

c. % Efisiensi penjerapan $= \frac{TD - FD}{TD} \times 100\%$
 $= \frac{10 \text{ mg} - 0,84558 \text{ mg}}{10 \text{ mg}} \times 100\%$
 $= 91,54\%$

Keterangan:

TD = total jumlah myricetin yang terdapat dalam formula

FD = jumlah myricetin yang terdeteksi pada supernatan (tidak terjerap)

Lampiran 13. Uji stabilitas fisik selama 3 minggu

Formula	Minggu ke-1	Minggu ke-2	Minggu ke-3
1	Tidak ada endapan	Ada endapan	Ada endapan
2	Tidak ada endapan	Ada endapan	Ada endapan
3	Tidak ada endapan	Ada endapan	Ada endapan
4	Tidak ada endapan	Ada endapan	Ada endapan
5	Tidak ada endapan	Tidak ada endapan	Tidak ada endapan

Ket: Formula 1 menggunakan perbandingan mol Myricetin:Fosfatidilkolin:Kolesterol (1:1:0,2)

Formula 2 menggunakan perbandingan mol Myricetin:Fosfatidilkolin:Kolesterol (1:2:0,2)

Formula 3 menggunakan perbandingan mol Myricetin:Fosfatidilkolin:Kolesterol (1:3:0,2)

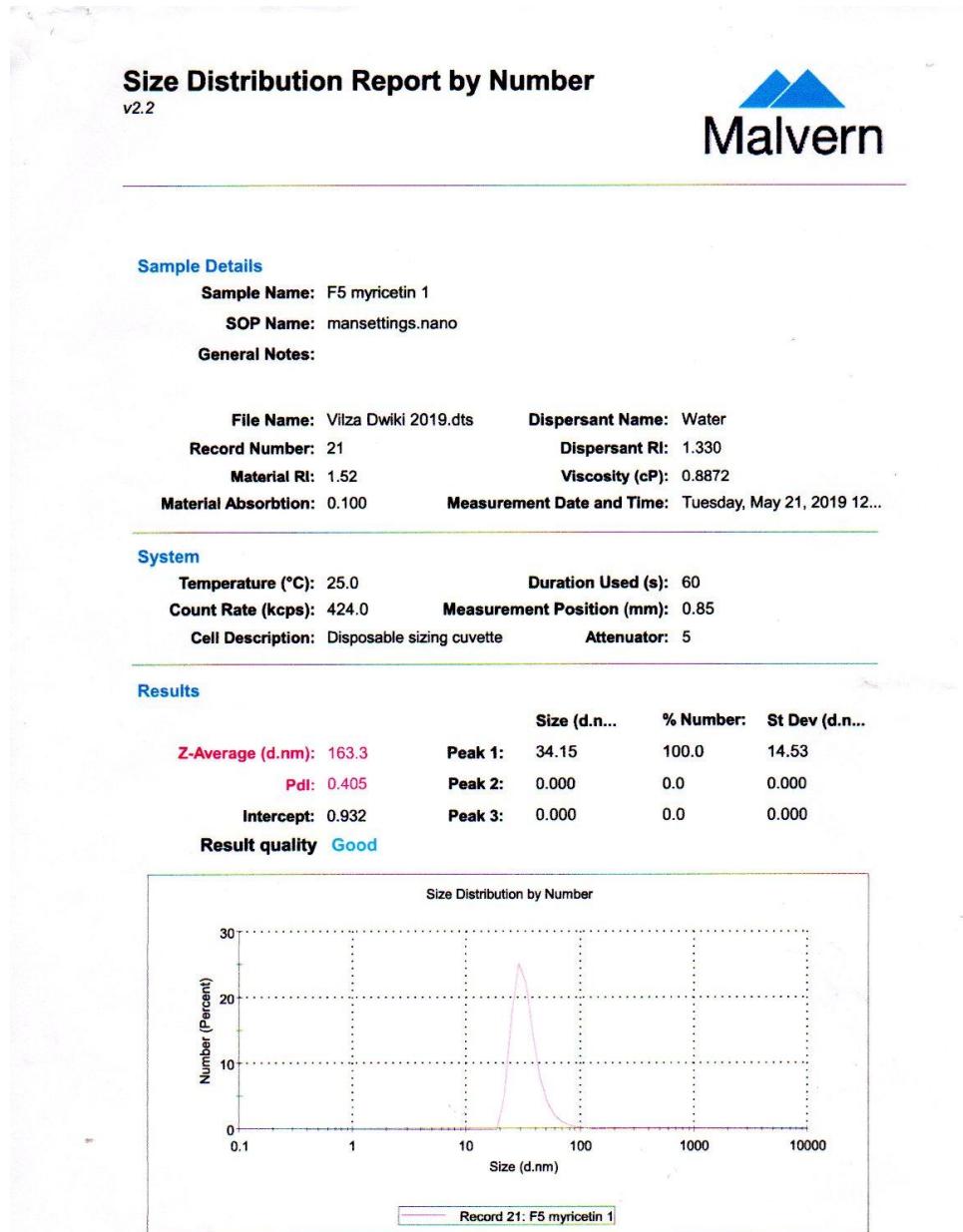
Formula 4 menggunakan perbandingan mol Myricetin:Fosfatidilkolin:Kolesterol (1:4:0,2)

Formula 5 menggunakan perbandingan mol Myricetin:Fosfatidilkolin:Kolesterol (1:5:0,2)

Lampiran 14. Ukuran dan distribusi partikel setelah penyimpanan

Formula	Ukuran Partikel (nm)		Indeks Polidispersitas	
	Sebelum	Sesudah	Sebelum	Sesudah
F5	139,467 nm	164,960 nm	0,380	0,393

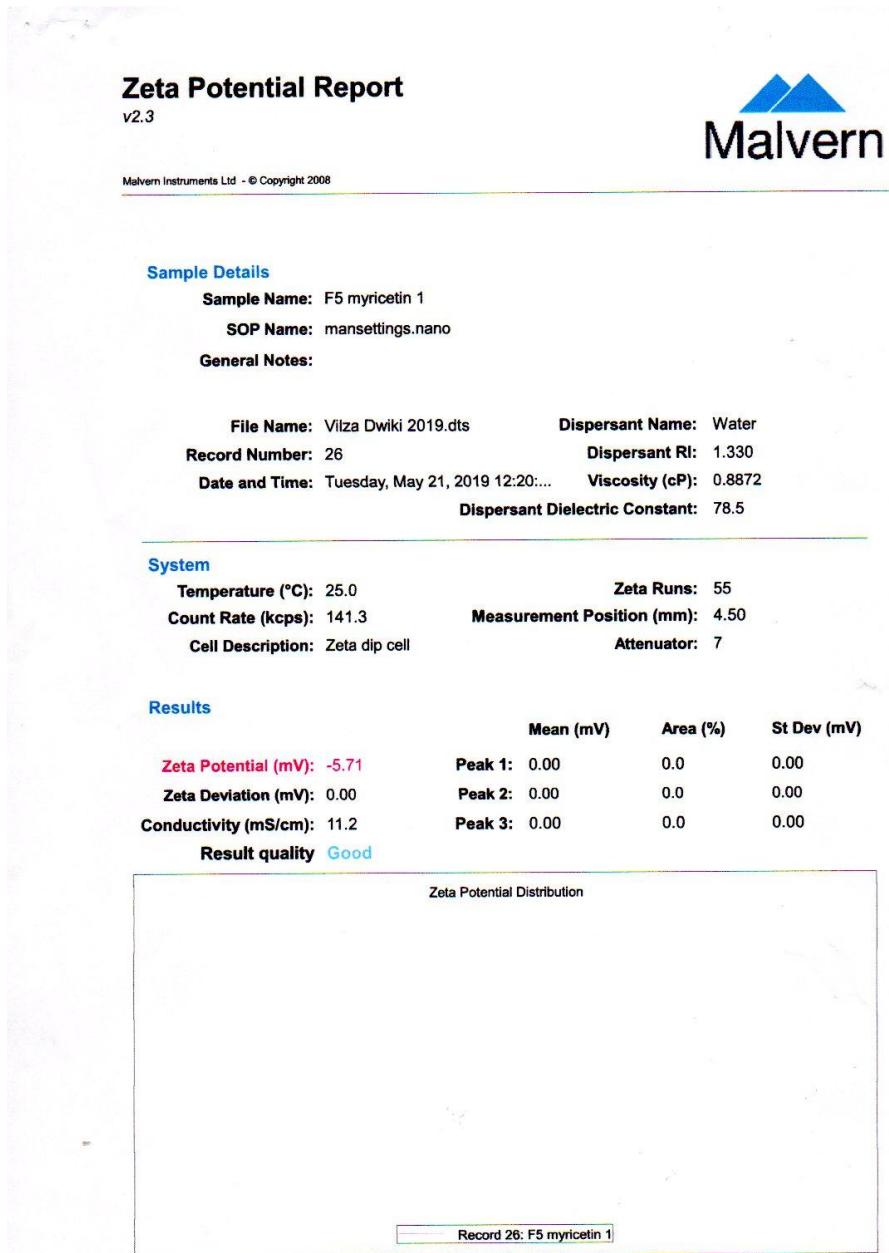
Hasil Ukuran Partikel F5 Setelah Penyimpanan



Lampiran 15. Zeta potensial nanofitosom myricetin setelah penyimpanan

Formula 5	Potensial zeta (Mv)				
	Repitasi 1	Repitasi 2	Repitasi 3	Rata-rata	SD
	-4,570	-5,930	-5,680	-5,393	0,724

Hasil Zeta Potensial F5 Setelah Penyimpanan



Lampiran 16. Hasil analisis statistik terhadap stabilitas ukuran partikel, indeks polidispersitas dan potensial zeta

1. Ukuran partikel

NPar Tests

One-Sample Kolmogorov-Smirnov Test

		Hasil ukuran partikel
N		8
Normal Parameters ^{a,b}	Mean	155.40000
	Std. Deviation	13.546744
Most Extreme Differences	Absolute	.299
	Positive	.246
	Negative	-.299
Kolmogorov-Smirnov Z		.845
Asymp. Sig. (2-tailed)		.473

a. Test distribution is Normal.

b. Calculated from data.

T-Test

Group Statistics

Waktu	N	Mean	Std. Deviation	Std. Error Mean
Hasil ukuran partikel sebelum	3	139.46667	.550757	.317980
sesudah	5	164.96000	4.043884	1.808480

Independent Samples Test

	Levene's Test for Equality of Variances		t-test for Equality of Means							95% Confidence Interval of the Difference			
	F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference	Lower	Upper				
Hasil ukuran partikel	Equal variances assumed	2.574	.160	-10.524	6	.000	-25.493333	2.422463	-31.420886	-19.565780			
	Equal variances not assumed			-13.884	4.243	.000	-25.493333	1.836222	-30.478378	-20.508289			

2. Indeks polidispersitas

NPar Tests

One-Sample Kolmogorov-Smirnov Test

		Hasil indeks polidispersitas
N		8
Normal Parameters ^{a,b}	Mean	.38813
	Std. Deviation	.016313
Most Extreme Differences	Absolute	.127
	Positive	.127
	Negative	-.100
Kolmogorov-Smirnov Z		.360
Asymp. Sig. (2-tailed)		.999

a. Test distribution is Normal.

b. Calculated from data.

T-Test

Group Statistics

	Waktu	N	Mean	Std. Deviation	Std. Error Mean
Hasil indeks polidispersitas	Sebelum	3	.37967	.006429	.003712
	Sesudah	5	.39320	.018953	.008476

Independent Samples Test

	Levene's Test for Equality of Variances		t-test for Equality of Means						95% Confidence Interval of the Difference	
	F	Sig.	t	df	Sig. (2- tailed)		Mean Difference	Std. Error Difference	Lower	Upper
Hasil indeks polidispersitas	Equal variances assumed	1.767	.232	-1.164	6	.288	-.013533	.011622	-.041971	.014904
				-1.463	5.292	.200	-.013533	.009253	-.036929	.009863

3. Potensial zeta

NPar Tests

One-Sample Kolmogorov-Smirnov Test

		Zeta potensial
N		8
Normal Parameters ^{a,b}	Mean	-5.9913
	Std. Deviation	.70548
Most Extreme Differences	Absolute	.205
	Positive	.205
	Negative	-.148
Kolmogorov-Smirnov Z		.579
Asymp. Sig. (2-tailed)		.891

a. Test distribution is Normal.

b. Calculated from data.

T-Test

Group Statistics

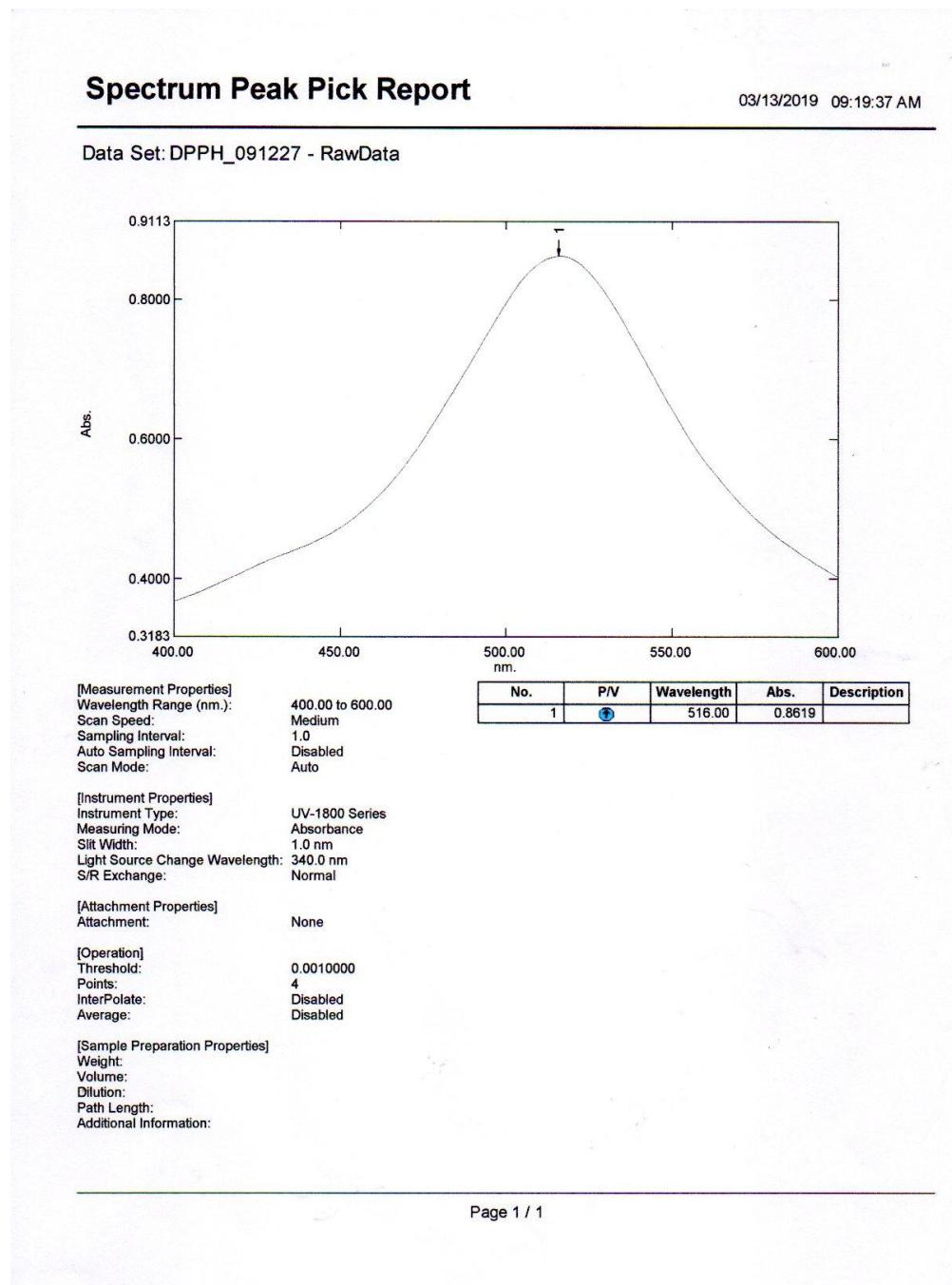
	Waktu	N	Mean	Std. Deviation	Std. Error Mean
Zetapotensial	Sebelum	3	-5.3933	.72390	.41794
	Sesudah	5	-6.3500	.42421	.18971

Independent Samples Test

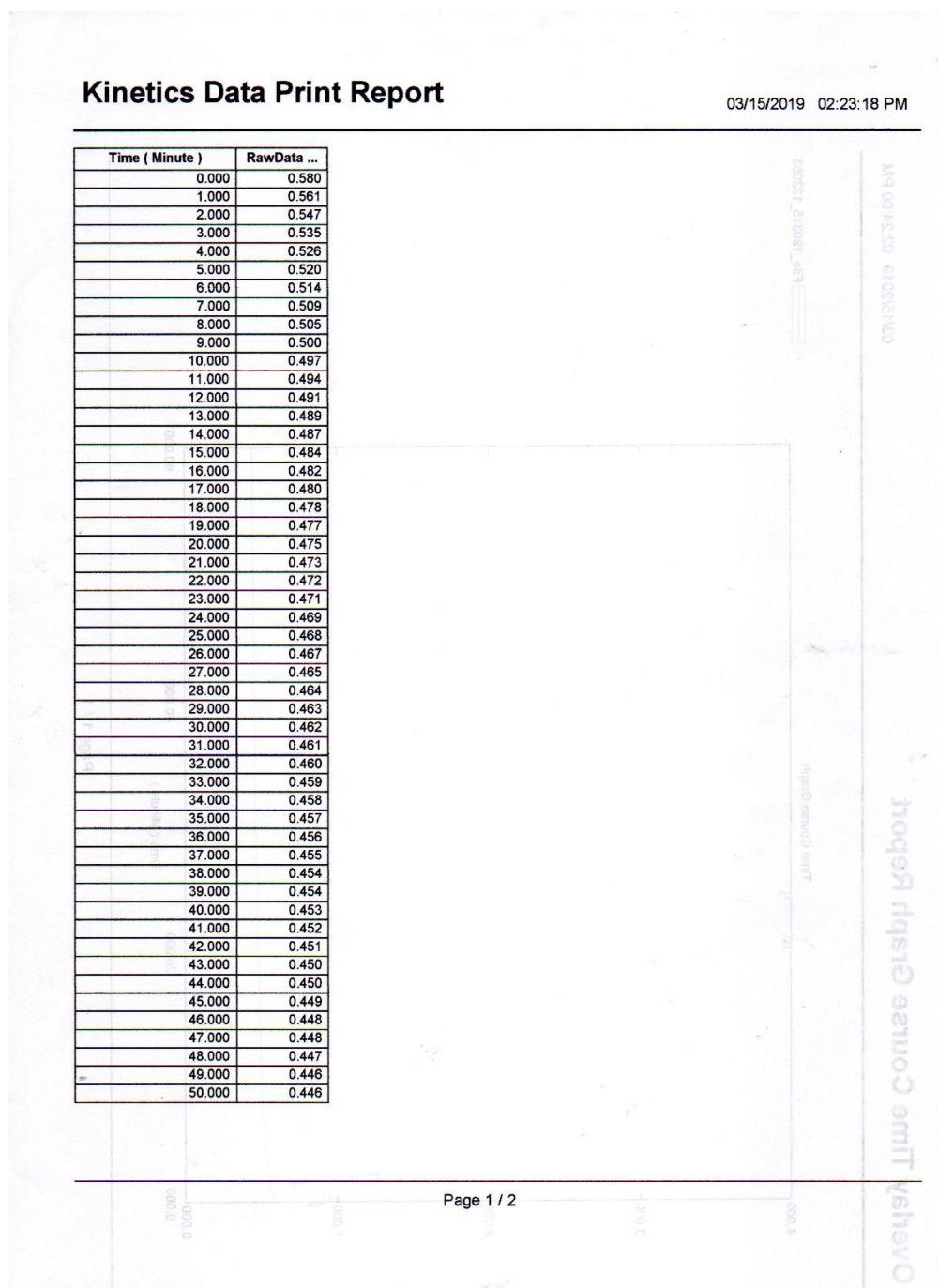
		Levene's Test for Equality of Variances		t-test for Equality of Means						95% Confidence Interval of the Difference	
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference	Lower	Upper	
Zeta potensial	Equal variances assumed	1.716	.238	2.413	6	.052	.95667	.39641	-.01332	1.92666	
	Equal variances not assumed			2.084	2.849	.133	.95667	.45899	-.54895	2.46229	

Lampiran 17. Uji DPPH

a. Penetapan panjang gelombang DPPH



b. Penetapan *Operating Time* (OT)



Kinetics Data Print Report

03/15/2019 02:23:18 PM

Time (Minute)	RawData ...
51.000	0.445
52.000	0.445
53.000	0.444
54.000	0.443
55.000	0.443
56.000	0.442
57.000	0.442
58.000	0.441
59.000	0.441
60.000	0.440

c. Uji DPPH myricetin

Absorbansi DPPH = 0,802

Lamda maksimal = 516 nm

Operating Time = 38-39 menit

Larutan stok 100 ppm → 10 mg myricetin + ad 100 ml etanol p.a

KONSENTRASI (PPM)	SERAPAN 1	SERAPAN 2	SERAPAN 3	RATA-RATA
2	0,700	0,701	0,705	0,702
4	0,625	0,624	0,624	0,624
6	0,560	0,563	0,570	0,564
8	0,525	0,535	0,566	0,542
10	0,478	0,481	0,484	0,481

Nilai IC₅₀

KONSENTRASI (PPM) x	ABSORBANSI	% PEREDAMAN y
2	0,702	18,56 %
4	0,624	27,61 %
6	0,564	34,57 %
8	0,542	37,12 %
10	0,481	44,20 %

$$a = 14,1750 \quad b = 3,0395 \quad r = 0,9852$$

- % Peredaman = $\frac{\text{Absorbansi blangko} - \text{Absorbansi sampel}}{\text{Absorbansi blangko}} \times 100\%$

1. 2 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,702}{0,862} \times 100\%$$

$$= 18,56 \%$$

2. 4 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,624}{0,862} \times 100\% \\ = 27,61 \%$$

3. 6 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,564}{0,862} \times 100\% \\ = 34,57 \%$$

4. 8 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,542}{0,862} \times 100\% \\ = 37,12\%$$

5. 10 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,481}{0,862} \times 100\% \\ = 44,20\%$$

- IC_{50}

$$Y = a + b.x$$

$$50 = 14,1750 + 3,0395.x$$

$$X = \frac{50 - 14,1750}{3,0395}$$

$$IC_{50} = 11,7865 \text{ ppm}$$

Nilai IC_{50} Myricetin sebesar 11,7865 ppm sehingga memiliki aktivitas antioksidan yang sangat kuat karena suatu senyawa dikatakan sebagai antioksidan sangat kuat jika nilai IC_{50} kurang dari 50 ppm ($IC_{50} < 50 \text{ ppm}$).

d. Uji DPPH sampel myricetin nanofitosom (Formula 5)

Absorbansi DPPH = 0,802
 Lamda maksimal = 512 nm
 Operating Time = 38-39 menit
 Konsentrasi = 10 mg/20 ml = 500 mg/1000 ml = 500 ppm
 Larutan stok 250 ppm → 5 ml sampel myricetin nanofitosom F5 + ad 10 ml etanol p.a

KONSENTRASI (PPM)	REPLIKASI 1	REPLIKASI 2	REPLIKASI 3	RATA-RATA
250	0,205	0,201	0,201	0,202
125	0,394	0,395	0,396	0,395
62,5	0,417	0,413	0,412	0,414
31,25	0,599	0,599	0,598	0,599
15,625	0,708	0,701	0,701	0,703
7,812	0,739	0,749	0,749	0,746

Perhitungan:

Sampel myricetin nanofitosom Formula 5 (konsentrasi 500 ppm) dibuat seri konsentrasi 250 ppm, 125 ppm, 62,5 ppm, 31,25 ppm, 15,625 ppm dan 7,812 ppm dalam 10 mL etanol p.a.

1. 250 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 250 \text{ ppm} \\ V_1 &= 5 \text{ mL} \end{aligned}$$

2. 125 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 125 \text{ ppm} \\ V_1 &= 2,5 \text{ mL} \end{aligned}$$

3. 62,5 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 62,5 \text{ ppm} \\ V_1 &= 1,25 \text{ mL} \end{aligned}$$

4. 31,25 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 31,25 \text{ ppm} \\ V_1 &= 0,625 \text{ mL} \end{aligned}$$

5. 15,625 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 15,625 \text{ ppm} \\ V_1 &= 0,312 \text{ mL} \end{aligned}$$

6. 7,812 ppm

$$\begin{aligned} V_1 \times C_1 &= V_2 \times C_2 \\ V_1 \times 500 \text{ ppm} &= 10 \text{ mL} \times 7,812 \text{ ppm} \\ V_1 &= 0,156 \text{ mL} \end{aligned}$$

Nilai IC₅₀

KONSENTRASI (PPM)	ABSORBANSI	% PEREDAMAN
		x
250	0,202	76,57 %
125	0,395	54,18 %
62,5	0,414	51,97 %
31,25	0,599	30,51 %
15,625	0,703	18,45 %
7,812	0,746	13,46 %

$$a = 20,9864$$

$$b = 0,2422$$

$$r = 0,9263$$

- % Peredaman = $\frac{\text{Absorbansi blangko} - \text{Absorbansi sampel}}{\text{Absorbansi blangko}} \times 100\%$

1. 250 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,202}{0,862} \times 100\% \\ = 76,57 \%$$

2. 125 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,395}{0,862} \times 100\% \\ = 54,18 \%$$

3. 62,5 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,414}{0,862} \times 100\% \\ = 51,97 \%$$

4. 31,25 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,599}{0,862} \times 100\% \\ = 30,51 \%$$

5. 15,625 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,703}{0,862} \times 100\% \\ = 18,45 \%$$

6. 7,812 ppm

$$\% \text{ peredaman} = \frac{0,862 - 0,746}{0,862} \times 100\% \\ = 13,46 \%$$

- IC₅₀

$$Y = a + b.x$$

$$50 = 20,9864 + 0,2422.x$$

$$X = \frac{50 - 20,9864}{0,2422}$$

$$IC_{50} = 119,7920 \text{ ppm}$$

Nilai IC₅₀ myricetin nanofitosom Formula 5 sebesar 119,7920 ppm sehingga memiliki aktivitas antioksidan yang sedang karena nilai IC₅₀ di atas 100 ppm (100 ppm < IC₅₀ < 150 ppm).