

BAB V

KESIMPULAN DAN SARAN

A. Kesimpulan

Berdasarkan dari hasil penelitian yang telah dilakukan dapat diperoleh kesimpulan bahwa:

Pertama, *Nanostructured Lipid Carriers* (NLC) fisetin dapat dibuat menggunakan lipid golongan gliserida (Precirol) dengan metode emulsifikasi sonikasi.

Kedua, karakterisasi NLC Fisetin menghasilkan ukuran partikel yaitu F1 $651,1 \text{ nm} \pm 7,11$; F2 $246,2 \text{ nm} \pm 2,45$ dan F3 $119,6 \text{ nm} \pm 5,36$. Menghasilkan PDI yaitu pada F1 0,315; F2 0,266 dan F3 0,619. Menghasilkan zeta potensial yaitu pada F1 -13,13 mV; F2 -20,36 mV dan F3 -20,53 mV.

Ketiga, karakterisasi serum NLC Fisetin pada uji organoleptik menghasilkan warna kuning keruh, bau khas dan sedikit kental. Pada uji pH menghasilkan pH sebesar 6,27, pada uji homogenitas menghasilkan sediaan yang homogen dan pada uji viskositas menghasilkan viskositas sebesar 1000 cPs.

B. Saran

Penelitian ini masih banyak kekurangan, maka perlu dilakukan penelitian lebih lanjut mengenai :

Pertama, perlu dilakukan analisis *screening* lipid dengan menggunakan jenis lipid yang lebih beragam.

Kedua, perlu dilakukan uji *Transmission Electron Microscopy* (TEM) untuk mengetahui morfologi NLC (*Nanostructured Lipid Carriers*) fisetin.

Ketiga, perlu dilakukan uji pelepasan sediaan (Difusi Franz).

DAFTAR PUSTAKA

- Abdel-Salam, F., Mahmoud, A., Ammar, H., & Elkheshen, S. (2016). Nanostructured lipid carriers as semisolid topical delivery formulations for diflucortolone valerate. *Journal Of Liposome Research*, 27(1), 41-55. <https://doi.org/10.3109/08982104.2016.1149866>
- Abhijit et al. 2011. Lipid nanocarriers (GeluPearl) containing amphiphilic lipid Gelucire 50/13 as a novel stabilizer: fabrication, characterization and evaluation for oral drug delivery. *Nanotechnology* 22 (2011) 275102 (12pp).
- Afianti Hanum Pramuji, Mimiek .M. 2015. Pengaruh Variasi Kadar Gelling Agent HPMC Terhadap Sifat Fisik dan Aktivitas Antibakteri Sediaan Gel Ekstrak Etanolik Daun Kemangi. *Majalah Farmaseutik*. 11 : 308.
- Alkilani, Ahlam Zaid., NcCrudden Maeliosa T.C.,Donelly, & Ryan F. 2015. Transdermal Drug Delivery: Innovative Pharmaceutical Developments Based on Disruption of the Barriers Properties of the stratum corneum. *Pharmaceutics*;7(4):438-470
- Anggreini, Y., Hendradi, E., Purwanti, T., 2012. Karakteristik Sediaan dan Pelepasan Natrium Diclofenak dalam Sistem Niosom dengan Basis Gel Carbomer 940. *PharmaScientia* 1: 6
- Annisa R, Hendradi E, Melani D. 2016. Pengembangan Sistem Nanostructure Lipid Carriers (NLC) Meloxicam dengan Lipid Monostearin dan Miglycol 808 Menggunakan Metode Emulsifikasi. *J. Trop. Pharm. Chem* 3:3
- Apriyani 2019. Formulasi dan Karakterisasi NLC (*Nanostructured Lipid Carriers*) Fisetin Menggunakan Lipid Golongan Gliserida Dengan Metode Emulsifikasi [Skripsi]. Surakarta: Fakultas Farmasi, Universitas Setia Budi.
- Arai Y, Watanabe S, Kimira M, Shimoi K, Mochizuki R, Kinae N 2000. Dietary intakes of flavonols, flavones and isoflavones by Japanese women and the inverse correlation between quercetin intake and plasma LDL cholesterol concentration. *J Nutr* 130:2243-50.
- Astuti, D. P., Patihul Husni, dan Kusdi Hartono., 2017, formulasi dan uji stabilitas fisik sediaan gel Antiseptik tangan minyak atsiri bunga lavender(lavandula angustifolia miller), *Jurnal Farmaka Unpad*, 15(1)
- Aswad, M., Fatmawaty, A., Nursamsiar, Rahmawanti. 2011. Validasi Metode Spektrofotometri Sinar Tampak untuk Analisis Formalin dalam Tahu. Majalah Farmasi dan Farmakologi. 15:26-29.

- Aulifa, D., Noerfitri, R., Tristiyanti, D., & Budiman, A. (2020). FORMULATION OF SERUM GEL CONTAINING ANGELICA KEISKEI LEAF EXTRACT AS AN ANTIOXIDANT AND TYROSINASE ENZYME INHIBITOR. *International Journal Of Applied Pharmaceutics*, 108-111
- Badarinath A, Rao K, Chetty CS, Ramkanth S, Rajan T, & Gnanaprakash K. A. 2010. Review on In-vitro Antioxidant Methods : Comparisons, Correlations, and Considerations. *International Journal of PharmTech Research* 12761285.
- Bayas MMS, Kadam Rs, Nalbalwar NN, Jain VKP. 2015. UV Spectofotometric Estimation of Loratadine in Bulk and Tablet Dosage Form Using Area Under Curve Method. *WJPPS*.
- Beifuss W, Leschke M, Weber K. A new concept to boost the preservative efficacy of phenoxyethanol. *SOFW Journal*. 2005; 11(131):2–7.
- Bhaskar, K., Anbu, J., Ravichandiran, V., Venkateswarlu, V., and Rao, Y.M., 2009. Lipid Nanoparticles for Transdermal Delivery of Flurbiprofen: Formulation, in vitro, ex vivo and in vivo Studies. *BioMed Cent* 8: 254-258.
- Budiman, Muhammad Haqqi. 2008. *Uji Stabilitas dan Aktivitas Antioksidan Sediaan Krim yang Mengandung Ekstrak Kering Tomat (Solanum lycopersicum L.)*. Depok: Universitas Indonesia
- Buzea, C., Blandino, I.I.P., dan Robbie, K., 2007, Nanomaterial and nanoparticles: sources and toxicity, *Biointerphases*, 2: MR170– MR172
- Carton, F., Chevalier, Y., Nicoletti, L., Tarnowska, M., Stella, B., Arpicco, S., Lollo, G. 2019. Rationally designed hyaluronic acid-based nanocomplexes for pentamidine delivery. *International Journal of Pharmaceutics*, 118526. doi:10.1016/j.ijpharm.2019.118526
- Chan CC, Herman L, Lee YC, Zhang XM. 2004. Analytical Method Validation and Instrument Performance Verification. New Jersey : Inc Publication
- Cirri, M., Bragagni, M., Mennini, P., 2012. Development of a New Delivery System Consisting “Drug-in Cyclodextrin-in Nanostructured Lipid Carriers” for Ketoprofen Topical J. Trop. Pharm. Chem. 2016. Vol 3. No. 3 p-ISSN: 20877099; e-ISSN: 2407-6090 168 Delivery. Eur J of Pharm and Biopharm 80(1): 46-53.
- Cserhati, T.1995. Alkyl Ethoxylated and alkylphenol ethoxylated nonionic surfactants: Interaction with bioactive compounds and biological effects. *Environmental Health Perspect* (103): 358-364.

- Dang Y, Xie Y, Duan, JZ, Ma P, Li GW, Ji G. 2014. Quantitative Determination of Myricetin in Rat Plasma by Ultra Performance Liquid Chromatography Tandem Mass Spectrometry and its Absolute Bioavailability. *Drug Res* 64: 516–522.
- Dechert T.A., Ducale A.E., Ward S.I., Yager D.R. (2006): Hyaluronan in human acute and chronic dermal wounds. *Wound Repair and Regeneration*, 14, 252–258.
- Dehpour, .A., Ebrahimzadeh, M.A., Fazel, N.S., and Mohammad, N.S., 2009, Antioxidant Activity of Methanol Extract of Ferula Assafoetida and Its Essential Oil Composition, *Grasas Aceites*, 60(4), 405-412.
- Djajadisastra, J. (2004). Cosmetic Stability. Seminar Setengah Hari HIKI, Farmasi Fakultas Matematika dan Ilmu Pengetahuan Alam Universitas Indonesia, Depok.
- Draelos, Z.D. (2010). Cosmetic Dermatology Products and Procedures. USA: Blackwell Publishing, Ltd
- Gandjar, I. G. dan Rohman, A. 2012. Analisa Obat secara Kromatografi dan Spektroskopi. Yogyakarta: Pustaka Pelajar
- Gandjar, I. G. & Rohman, A. 2013. Kimia farmasi analisis. (Edisi XI).Yogyakarta: Pustaka Pelajar.
- Gaonkar T A, Geraldo I, Shintre M, Modak S M 2006. In vivo efficacy of an alcoholbased surgical hand disinfectant containing a synergistic combination of ethylhexylglycerin and preservatives. *J Hosp Infect*. 63: 412.
- Garg, A., Deepika, A., Sanjay, G., & Anil, K. S.,2002, Spreading of Semisolid Formulations: An Update, 178-180, *Pharmaceutical Technology*, USA.
- Godin, B., and Touitou, E., 2007. Transdermal Skin Delivery: Prediction for Human Fron in Vivo, Ex Vivo and Animals Model. *Adv Drug Deliv Rev* 59: 1152-1161.
- Hadi, A. 2007. Pemahaman dan Penerapan ISO/IEC 17025: Persyaratan Umum Kompetensi Laboratorium Pengujian dan Laboratorium Kalibrasi. PT. Gramedia Pustaka Utama.259-274.
- Harjanti, R., & Nilawati, A. (2020). Aktivitas Antioksidan dan Potensi Tabir Surya Serum Ekstrak Terpurifikasi Daun Wangon (*Olax psittacorum* (Willd.) Vahl.). *Jurnal Farmasi Indonesia*, 17(1), 18-28.
- Harmita. 2004. Petunjuk Pelaksanaan Validasi Metode dan Cara Perhitungannya. Majalah Ilmu Kefarmasian.I: 117-135.

- Hatakeyama, H., Asano, Y., Hatakeyama, T., & Kennedy, J. F. 2002. Thermal Properties of Hyaluronic Acid-based Polyurethane Derivatives Associated With Water. *Hyaluronan*, 313–322.
- Jain, S.K., Jain, N. K. 2010. Multiparticulate carriers for sun-screening agents. *Int. J. Cosmet. Sci*, 32(2), 89–98
- Jiménez, M., Escribano-Cebrián, J., & García-Carmona, F. (1998). Oxidation of the flavonol fisetin by polyphenol oxidase. *Biochimica Et Biophysica Acta (BBA) - General Subjects*, 1425(3), 534-542. doi: 10.1016/s0304-4165(98)00108-1
- Jumaetri et al. 2016. Uji Aktifitas Antioksidan dari Beberapa Ekstrak Kulit Batang Jamblang (*Syzygium cumini*) Menggunakan Metode Peredaman Radikal 2,2-Diphenyl-1-Picrylhydrazyl (DPPH). *JF FIK UINAM* Vol.4 No.4
- Kawashima, Y., Yamamoto, H., Takeuchi, H., and Kuno, Y., 2000, Mucoadhesive DLlactide/glycolide copolymer nanospheres coated with chitosan to improve oral delivery of elcatonin, *Pharmaceutical Development and Technology*, 5(1): 77-85
- Kim, H. J., Kim, S. H., & Yun, J. M. (2012). Fisetin inhibits hyperglycemia-induced proinflammatory cytokine production by epigenetic mechanisms. *Evidence-Based Complementary and Alternative Medicine*, 2012.
- Kosasih, E.N., Tony S. dan Hendro H. 2006. Peran Antioksidan pada Lanjut Usia. Pusat Kajian Nasional Masalah Lanjut Usia. Jakarta
- Lai-Cheong, J. E., & McGrath, J. A. 2017. Structure and function of skin, hair and nails. *Medicine (United Kingdom)*, Elsevier Ltd.
- Li Q et al. 2017. A Review of the Structure, Preparation, and Application of NLCs, PNPs, and PLNs. *Nanomaterials*. MDPI AG.
- Lullung, A., & Suprapti, S. (2016). Pengaruh Surfaktan Terhadap Diameter Partikel, Viskositas dan Indeks Polidispersitas pada Pembuatan Sln dari Lemak Kakao. *Jurnal Riset Teknologi Industri*, 6(12), 1
- Lund, W. 1994. The Pharmaceutical Codex 12th Edition. *Pharmaceutica Acta Helvetiae*, 69(1), 53.
- Madani, S. Y., Mandel, A., & Seifalian, A. M. (2013). A concise review of carbon nanotube's toxicology. *Nano Reviews*, 4(1), 21521.
- Mardiyanti Siti, Effionora Anwar, Fadlina Chany Saputri. 2016. Formulasi Serum sebagai Penyembuh Luka Bakar Berbahan Baku Utama Serbuk Konsentrat Ikan Gabus (*Channa striatus*). *Jurnal Ilmu Kefarmasian Indonesia* 14(2) : 181-189

- Mardliyati Etik, Sjaikhurral El Muttaqien dan Damai Ria Setyawati. 2012. Sintesis Nanopartikel Kitosan – Trypoly Phosphate dengan Metode Gelasi Ionik : Pengaruh Konsentrasi dan Rasio Volum Terhadap Karakteristik Partikel. Pusat Teknologi Farmasi dan Medika. Jakarta.
- Marinda, Wenny Silvia. 2012. Formulasi dan Uji Stabilitas Fisik Gel Liposom yang Mengandung Fraksinasi Ekstrak Metanol Kulit Manggis (*Garcinia mangostana L.*) sebagai Antioksidan. Depok: Universitas Indonesia
- Martien et al. 2012. Perkembangan Teknologi Nanopartikel Sebagai Sistem Penghantaran Obat. *Majalah Farmaseutik*, Vol. 8 No. 1 Tahun 2012.
- Martin A, Swarbrick J, Cammarata A. 1993. Farmasi Fisik. Edisi III. Jakarta: UI Press.
- Mehnert, W. 2001. Solid lipid nanoparticles production, characterization and applications. *Advanced Drug Delivery Reviews*, 47(2-3), 165–196.
- Misnadiarly, A.S. 2006. Faktor-Faktor yang Berpengaruh terhadap Kerusakan Kulit. *Cermain Dunia Kedokteran*, 152, 43-45
- M.M , Akhmetov , G.G. Gumarov, V. Yu. Petukhov, M. Yu. Volkov. 2019. NMR study of sodium gluconate solutions. *Journal of Molecular Structure* (1193): s373-377
- Muller, R. H., Mader, K., & Gohla, S. 2000. Solid lipid nanoparticles (SLN) for controlled drug delivery - a review of the state of the art. *European Journal of Pharmaceutics and Biopharmaceutics*, 50, 161-17
- Müller, R. H., Radtke, M., Wissing, S. A. 2002. Solid lipid nanoparticles (SLN) and nanostructured lipid carriers (NLC) in cosmetic and dermatological preparations. *Advanced Drug Delivery Reviews* (Vol. 54).
- Naeimi, A. F., & Alizadeh, M. (2017). Antioxidant properties of the flavonoid fisetin: An updated review of in vivo and in vitro studies. *Trends in Food Science & Technology*, 70, 34–44.
- Nurdianti, L., Aryani, R., & Indra, I. (2017). Formulasi dan Karakterisasi SNE (Self Nanoemulsion) Astaxanthin dari Haematococcus pluvialis sebagai Super Antioksidan Alami. *Jurnal Sains Farmasi & Klinis*, 4(1), 36. doi: 10.29208/jsfk.2017.4.1.168
- Pardeike, J., Hommos, A., Muller, R.H., 2009. Lipid Nanoparticles (SLN and NLCs) in Cosmetic and Pharmaceutical Dermal Product. *Inter J of Pharm* 336: 170-184
- Pardeike, J., Weber, S., Haber, T., Wagner, J., Zarfl, H., Plank, H., & Zimmer, A. (2011). Development of an Itraconazole-loaded nanostructured lipid carrier

- (NLC) formulation for pulmonary application. *International Journal Of Pharmaceutics*, 419(1-2), 329-338.
- Pardeshi C et al. 2012. Solid lipid based nanocarriers. *Acta Pharm.* 62 (2012) 433– 472.
- Parhi, R., & Suresh, P. (2010). Production of Solid Lipid Nanoparticles-Drug Loading and Release Mechanism. *Journal of Chemical and Pharmacuetical Research*, 2(1), 211–227.
- Prayoga G. Fraksinasi, Uji Aktivitas Antioksidan dengan Metode DPPH dan Identifikasi Golongan Senyawa Kimia dari Ekstrak Teraktif Daun Sambang Darah (*Excoecaria cochinchinensis* Lour). Fakultas Farmasi Program Studi Sarjana Ekstensi Universitas Indonesia.2013.
- Rahmi A, Hendradi E, Melani D. 2016. Pengembangan Sistem Nanostructured Lipid Carriers (NLC) Meloxicam Dengan Lipid Monostearin Dan Miglyol 808 Menggunakan Metode Emulsifikasi. *J. Trop. Pharm. Chem.* 2016. Vol 3. No. 3 156 p-ISSN: 2087-7099; e-ISSN: 2407-6090.
- Rahmi D dan Suprobo G. 2015. Pengaruh Kecepatan Homogenisasi Terhadap Sifat Fisika dan Kimia Krim Nanopartikel dengan Metode High Speed Homogenization (HSH). Jurnal Litbang Industri Vol. 5 No. 1 : 1-12
- Rohman, A. 2007. Kimia Farmasi Analisis. Pustaka pelajar, Yogyakarta.
- Riyanto. (2014). Validasi Dan Verifikasi Sesuai dengan ISO/IEC 17025 Laboratorium Pengujian dan Kalibrasi. Yogyakarta: Deepublish.
- Rosli, N. A., Hasham, R., Aziz, A. A., & Aziz, R. (2015). Formulation and characterization of nanostructured lipid carrier encapsulated Zingiber zerumbet oil using ultrasonication technique. *Journal of Advanced Research in Applied Mechanic*, 11(1), 16–23.
- Rowe C, Sheskey P, Quinn M, editor. 2009. Handbook of Pharmaceutical Excipients. Ed ke-6. London. Hlm 155-156.
- Sasmiyandri, B., Samsul, E., & Indriyanti, N. (2019). Efektivitas Serum Lidah Buaya (*Aloe vera*) terhadap Peningkatan Laju Pertumbuhan Rambut dan Sun Protection. *Proceeding Of Mulawarman Pharmaceuticals Conferences*, 10, 81-85.
- Sechi, M., Syed, D., Pala, N., Mariani, A., Marceddu, S., & Brunetti, A. et al. (2016). Nanoencapsulation of dietary flavonoid fisetin: Formulation and in vitro antioxidant and α -glucosidase inhibition activities. *Materials Science And Engineering: C*, 68, 594-602.
- Sengupta B, Banerjee A, & Sengupta P.K. 2005. *J. Photochem. Photobiol. B*,Biol.80, 79-86

- Sinko, P.J. and Singh, H., (ed). (2011). *Martin's Physical Pharmacy and Pharmaceutical Sciences Physical Chemical and Biopharmaceutical Principles in the Pharmaceutical Sciences: Sixth Edition* (pp. 1–659). Wolters Kluwer Health Adis (ESP).
- Souto, E. B., & Müller, R. H. (2019). Lipid Nanoparticles (Solid Lipid Nanoparticles and Nanostructured Lipid Carriers) for Cosmetic, Dermal, and Transdermal Applications. In *Nanoparticulate Drug Delivery Systems* (pp. 213–233).
- Sriarumtias et al. 2017. Formulasi dan Uji Potensi Antioksidan Nanostructured Lipid Carrier (NLC) Retinil Palmitat. *Acta Pharmaceutica Indonesia* Vol. 42 No. 1 hlm. 25-31
- Sulastri, E., Yusriadi & Rahmiyati D. (2016). Pengaruh Pati Pragelatinasi Beras Hitam Sebagai Bahan Pembentuk Gel Terhadap Mutu Fisik Sediaan Masker Gel Peel Off. *Jurnal Pharmascience*. Vol. 03 No.02 hal: 69 - 79
- Swarbrick, J. dan Boylan, J. 1995, *Percutaneous Absorption, in Encyclopedia of Pharmaceutiacal Technology*, Volume 11,Marcel Dekker Inc., New York.
- Tamjidi, F., Shahedi, M., Varshosaz, J., & Nasirpour, A. 2013. Nanostructured Lipid Carriers (NLC): A potential delivery Sistem for Bioactive Food Molecules. *Innovative Food Science and Emerging Technologies* Vol.19, p.29-43
- Touitou, Elka. Barry W. 2007. *Enhancement In Drug Delivery*. New York:CRC Press, 220-221,237,246
- Wahyuningsih, I., & Putranti, W., (2015). Optimasi Perbandingan Tween 80 dan Polietilenglikol 400 Pada Fromula Self Nanoemulsifying Drug Delivery System (SNEDDS) Minyak Biji Jinten Hitam, *Pharmacy*, 12(02): 223-241
- Walters, K.A. 2004. Dermatological and Transdermal Formulation. Marcel Dekker. New York. p. 18, 25 -26, 33, 103 -105, 210, 337 -338.
- Williams B et al. 1994. Use of a Free Radical Method to Evaluate Antioxidant Activity. Departement Science 91305 Massy (France).
- Witt, Krista & Bucks, Daniel. 2003. Studying In Vitro Skin Penetration and Drug Release to Optimize Dermatological Formulations. *Pharmaceutical Technology*, Advanstar USA
- Yao Y, Lin G, Xie Y, Ma P, Li G, Meng Q, Wu T. 2013. Preformulation Studies of Myricetin: a Natural Antioxidant Flavonoid, *Pharmazie* 69:19-26.
- Yuan, H., Wang, L. L., Du, Y. Z., You, J., Hu, F. Q., & Zeng, S. (2007). Preparation and characteristics of nanostructured lipid carriers for control-

- releasing progesterone by melt-emulsification. *Colloids and Surfaces B: Biointerfaces*, 60(2), 174–179.
- Zhang et al. 2007. Preparation and characterization of solid lipid nanoparticles containing silibinin. *Drug Delivery*, 14(6), 381–387.
- Zhao X, Li XL, Liu X, Wang C, Zhou DS, Ma Q, Zhou WH, Hu ZY 2015 Antinociceptive effects of fisetin against diabetic neuropathic pain in mice: Engagement of antioxidant mechanisms and spinal GABAA receptors. *Pharmacol Res* 102:286-97.
- Zou, F., Tan, H., Guo, Y., Ma, B., He, X., & Zhou, Y. 2017. Effect of sodium gluconate on dispersion of polycarboxylate superplasticizer with different grafting density in side chain. *Journal of Industrial and Engineering Chemistry*, 55, 91–100.
- Zulfa, E., Novianto, D., & Setiawan, D. (2014). Formulasi Nanoemulsi Natrium Diklofenak Dengan Variasi Kombinasi Tween 80 Dan Span 80 : Kajian Karakteristik Fisik Sediaan. *Media Farmasi Indonesia*, 14(1), 1471–1477.

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Lampiran 1. Certificate of Analysis (CoA) Fisetin



Product Information

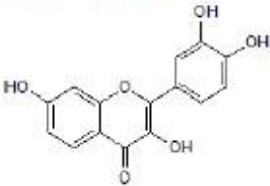
Print Date: Jan 14th 2016
www.tocris.com

Product Name:	Fisetin	Catalog No.:	5016	Batch No.:	1
CAS Number:	528-48-3				
IUPAC Name:	2-(3,4-Dihydroxyphenyl)-3,7-dihydroxy-4H-1-benzopyran-4-one				

Description:
Naturally occurring flavonoid and antioxidant. Inhibits PI 3-K, Akt, mTOR and Cdk5. Displays antiproliferative activity in prostate cancer cells. Shown to activate ERK; exhibits neuroprotective activity in Huntington's disease models. Also a DNMT1 inhibitor.

Physical and Chemical Properties:
Batch Molecular Formula: C₁₅H₁₀O₈
Batch Molecular Weight: 286.24
Physical Appearance: Yellow solid
Minimum Purity: >98%

Batch Molecular Structure:



Storage: Store at -20°C

Solubility & Usage Info:
DMSO to 100 mM
ethanol to 10 mM

Stability and Solubility Advice:
Some solutions can be difficult to obtain and can be encouraged by rapid stirring, sonication or gentle warming (in a 45-60°C water bath).
Information concerning product stability, particularly in solution, has rarely been reported and in most cases we can only offer a general guide. Our standard recommendations are:
SOLIDS: Provided storage is as stated on the product label and the vial is kept tightly sealed, the product can be stored for up to 6 months from date of receipt.
SOLUTIONS: We recommend that stock solutions, once prepared, are stored aliquoted in tightly sealed vials at -20°C or below and used within 1 month. Wherever possible solutions should be made up and used on the same day.

References:

- Lu et al (2005) Crystal structure of a human cyclin-dependent kinase 6 complex with a flavonol inhibitor, fisetin. *J.Med.Chem.* **48** 737. PMID: 15689157.
- Lee et al (2005) Mechanisms for the inhibition of DNA methyltransferases by tea catechins and bioflavonoids. *Mol.Pharmacol.* **68** 1018. PMID: 16037419.
- Haddad et al (2010) Antiproliferative mechanisms of the flavonoids 2,2'-dihydroxychalcone and fisetin in human prostate cancer cells. *Nutr.Cancer* **62** 668. PMID: 20574928.
- Maher et al (2011) ERK activation by the polyphenols fisetin and resveratrol provides neuroprotection in multiple models of Huntington's disease. *Hum.Mol.Genet.* **20** 261. PMID: 20952447.
- Adhami et al (2012) Dietary flavonoid fisetin: a novel dual inhibitor of PI3K/Akt and mTOR for prostate cancer management. *Biochem.Pharmacol.* **84** 1277. PMID: 22842629.

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Lampiran 2. Certificate of Analysis (CoA) Precirol


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Certificate of Analysis

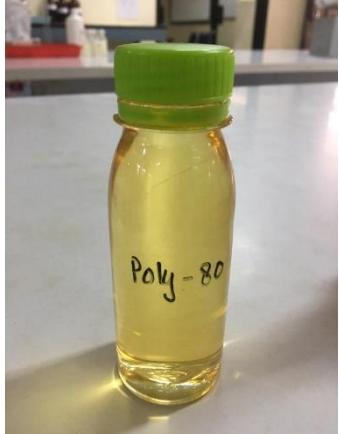
Product name	: PRECIROL ATO 5		
Item N°	: 3092PPD		
Batch N°	: 169604		
Manufacturing date	: 05.2018		
Expiry date	: 05.2020		
CHEMICAL DEFINITION : Glycerol distearate (type I) EP / Glyceryl distearate NF.			
Characteristic	Method	Specification	Result
MANUFACTURING PLANT		Gattefossé : 01 / Innova : 06	01/01
APPEARANCE	MA0193	Fine white powder	Conforms
ODOUR	MA0170	Faint	Conforms
COLOUR (Gardner Scale)	MA0214	< = 3.0	0.9
DROP POINT (METTLER)	MA0003	53.0 to 57.0 °C	56.3
ACID VALUE	MA0008	< = 6.00 mgKOH/g	1.76
SAPONIFICATION VALUE	MA0172	165 to 195 mgKOH/g	187
IODINE VALUE	MA0092	< = 3.0 gl2/100g	< = 3.0*
FREE GLYCEROL CONTENT	MA0006	< = 1.0 %	0.2
TOTAL MONOGLYCERIDES CONTENT	MA0006	8.0 to 22.0 %	17.4
TOTAL DIESTERS CONTENT	MA0006	40.0 to 60.0 %	50.7
TOTAL TRIESTERS CONTENT	MA0006	25.0 to 35.0 %	30.8
PALMITIC ACID (C16)	MA0002	40.0 to 60.0 %	44.5
STEARIC ACID (C18)	MA0002	40.0 to 60.0 %	53.2
SUM OF PALMITIC AND STEARIC ACIDS	MA0002	> = 90.0 %	97.7
WATER CONTENT	MA0007	< = 1.0 %	< 0.1
TOTAL ASHES CONTENT	MA0028	< = 0.1 %	< = 0.1*
NICKEL CONTENT	MA0256	< 1 ppm	Conforms
HEAVY METALS CONTENT (Pb)	MA0032	< 10 ppm	Conforms *
* Characteristic guaranteed but not tested.			
This batch was released by our authorized Quality Manager and was found to meet the specifications as stated above. The above results do not discharge the customer from the obligation to carry out an inspection of goods received. This report has been produced electronically and does not require a signature.			
Printing Date : 09.26.2018	Quality Manager Magali DRONNIER		

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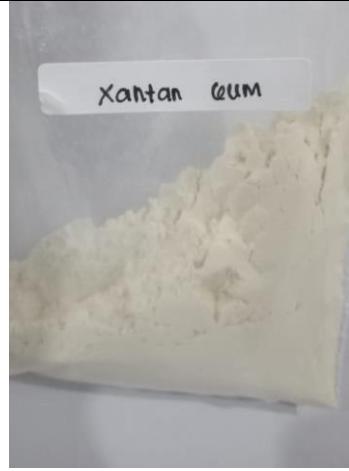
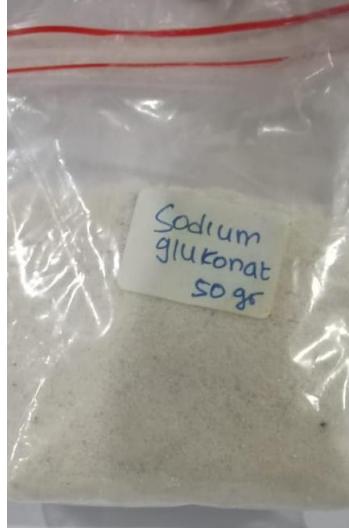
GATTEFOSSÉ SAS: 36 chemin de Genas - CS 70070 - 69804 SAINT-PRIEST Cedex - France - Tél: +33 4 72 22 98 00 - Télécopie: +33 4 78 90 45 67
 SAS au capital de 4 650 000 Euros - R.C.S. Lyon B 389 586 900 - N° d'identité CEE FR 49 389 586 900

Lampiran 3. Bahan yang digunakan untuk penelitian

Nama Bahan	Foto Bahan	Kegunaan
Serbuk Fisetin		Zat aktif
Precirol		Sebagai Lipid padat dalam komponen NLC.
Isopropil Miristat		Sebagai lipid cair dalam komponen NLC.

Tween80		Sebagai surfaktan dalam komponen NLC.
Aquade-mineralisata		Sebagai pelarut.
Ethanol PA		Sebagai pelarut.

Hyaluronic Acid		Berfungsi sebagai humektan
Propilen Glikol		Sebagai humektan
Distilled Water		Sebagai pelarut pada sediaan serum.

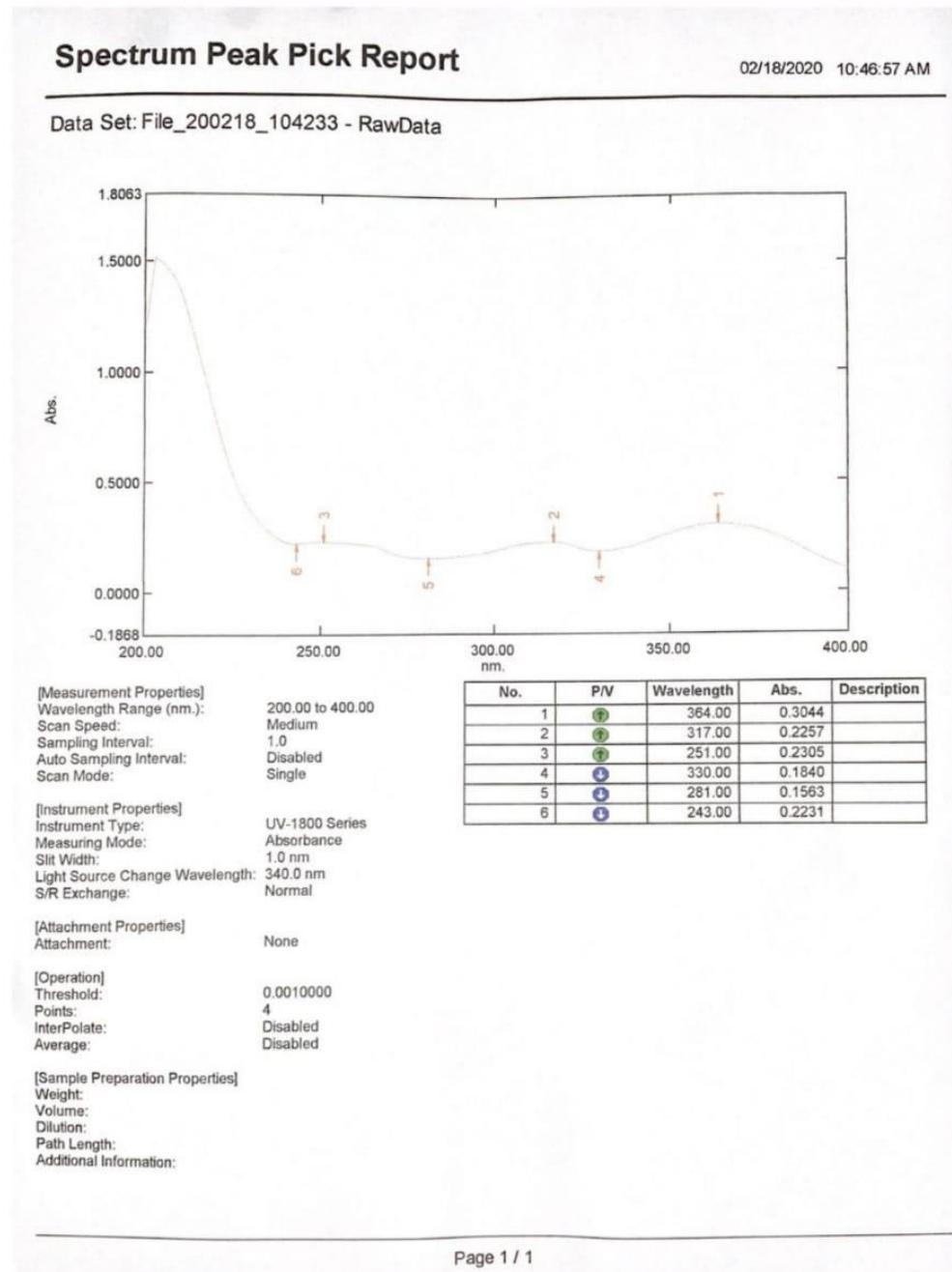
Xanthan Gum		Sebagai pelembut dalam serum.
Sodium Gluconate		Sebagai stabilizer dalam sediaan serum
Optiphen		Berfungsi sebagai pengawet

Lampiran 4. Alat yang digunakan dalam penelitian

Nama Alat	Foto Alat	Kegunaan
Spektrofotometri Uv-vis		Merupakan spektrofotometer yang digunakan untuk pengukuran serapan cahaya di daerah ultraviolet (200-350nm) dan sinar tampak (350-800nm) oleh suatu senyawa.
Particle Size Analyzer (PSA)		Untuk mengukur ukuran partikel.
Zeta nansizer		Analisa zeta potensial menggunakan hamburan cahaya elektroforesis.
Magnetic Stirrer		Berfungsi untuk menghomogenkan suatu larutan dengan pengadukan serta suhu.

Sonikator		Berfungsi memecah partikel dengan energi suara.
Viskometer		Untuk mengukur viskositas (kekentalan) suatu sediaan.
pH meter		Untuk mengukur pH (kadar keasaman) suatu sediaan.

Lampiran 5. Penentuan panjang gelombang dan pembuatan kurva baku



Hasil lamda max

Panjang gelombang maksimum yang diperoleh dari scaning larutan fisetin dalam etanol pro analisis sebesar 364 nm dengan serapan 0,3044 nm.

a. Penentuan *operating time*

Kinetics Data Print Report

02/18/2020 11:50:21 AM

Time (Minute)	RawData ...
0.000	0.301
1.000	0.301
2.000	0.301
3.000	0.301
4.000	0.300
5.000	0.300
6.000	0.300
7.000	0.301
8.000	0.301
9.000	0.301
10.000	0.300
11.000	0.301
12.000	0.300
13.000	0.300
14.000	0.300
15.000	0.300
16.000	0.300
17.000	0.300
18.000	0.300
19.000	0.300
20.000	0.300
21.000	0.300
22.000	0.300
23.000	0.300
24.000	0.300
25.000	0.300
26.000	0.300
27.000	0.300
28.000	0.300
29.000	0.300
30.000	0.300
31.000	
32.000	
33.000	
34.000	
35.000	
36.000	
37.000	
38.000	
39.000	
40.000	
41.000	
42.000	
43.000	
44.000	
45.000	
46.000	
47.000	
48.000	
49.000	
50.000	

Page 1 / 2

b. Kurva kalibrasi (linieritas)

Penimbangan bahan :

Kertas + fisetin = 0,1053

Kertas kosong = 0,0807 _

$$0,0246 \text{ gr} / 500 \text{ ml} = 49,2 \text{ ppm}$$

Perhitungan konsentrasi kurva kalibrasi :

- 1) $V_1 \times N_1 = V_2 \times N_2$
 $1 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 4,92 \text{ ppm}$
- 2) $V_1 \times N_1 = V_2 \times N_2$
 $1,2 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 5,904 \text{ ppm}$
- 3) $V_1 \times N_1 = V_2 \times N_2$
 $1,4 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 6,888 \text{ ppm}$
- 4) $V_1 \times N_1 = V_2 \times N_2$
 $1,6 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 7,872 \text{ ppm}$
- 5) $V_1 \times N_1 = V_2 \times N_2$
 $1,8 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 8,856 \text{ ppm}$
- 6) $V_1 \times N_1 = V_2 \times N_2$
 $2 \text{ ml} \times 49,2 \text{ ppm} = 10 \text{ ml} \times N_2$
 $N_2 = 9,84 \text{ ppm}$

Konsentrasi (ppm)	Absorbansi
4,920	0,315
5,904	0,427
6,888	0,527
7,872	0,646
8,856	0,741
9,840	0,860

Persamaan regresi linier antara konsentrasi (ppm) dan serapan diperoleh nilai :

$$a = -0,225285714$$

$$b = 0,109930314$$

$$r = 0,999669338$$

$$y = a + bx$$

$$y = -0,225285714 + 0,109930314x$$

keterangan :

x = konsentrasi ($\mu\text{g/ml}$)

y = serapan

Hasil linearitas diperoleh $R = 0.999669338$; sehingga dapat disimpulkan bahwa data tersebut linier.

c. Akurasi

X (ppm)	Y (abs)	Konsentrasi	Replikasi	Abs	Konsentrasi	Konsentrasi Sebenarnya	%	Rata-rata	Hasil
4,920	0,315	80%	1	0,326	5,0149	4,92	102%	99,89%	99,95%
5,904	0,427		2	0,304	4,8147	4,92	98%		
6,888	0,527		3	0,315	4,9148	4,92	100%		
7,872	0,646	100%	1	0,431	5,9700	5,904	101%	100,86%	99,95%
8,856	0,741		2	0,425	5,9154	5,904	100%		
9,840	0,860		3	0,432	5,9791	5,904	101%		
		120%	1	0,525	6,8251	6,888	99%	99,09%	
			2	0,527	6,8433	6,888	99%		
			3	0,523	6,8069	6,888	99%		

d. Presisi

Konsentrasi (ppm)	Absorbansi
6,888	0,552
6,888	0,537
6,888	0,545
6,888	0,557
6,888	0,558
6,888	0,525
6,888	0,543
6,888	0,547
6,888	0,552
6,888	0,529
Rata-rata	7,0025
SD	0,102222
CV	0,014598 %

Nilai CV dilihat dari data diatas adalah 0,014598%, hasil ini sesuai dengan persyaratan presisi yaitu $\leq 2\%$.

Lampiran 6. Foto NLC Fisetin**Lampiran 7. Foto NLC fisetin setelah disimpan selama 2 minggu**

Lampiran 8. Uji ukuran partikel formula 1 (Precirol 2%)

a. Replikasi I

Size Distribution Report by Intensity v2.2



Sample Details

Sample Name: NLC 2% RIZKY 1

SOP Name: mansettings.nano

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 1

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 09:57:42

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 227,4

Measurement Position (mm): 4,65

Cell Description: Disposable sizing cuvette

Attenuator: 5

Results

	Size (d.n...	% Intensity:	St Dev (d.n...
--	--------------	--------------	----------------

Z-Average (d.nm): 658,8

Peak 1: 634,9

96,9

219,3

Pdl: 0,315

Peak 2: 5376

3,1

324,8

Intercept: 0,949

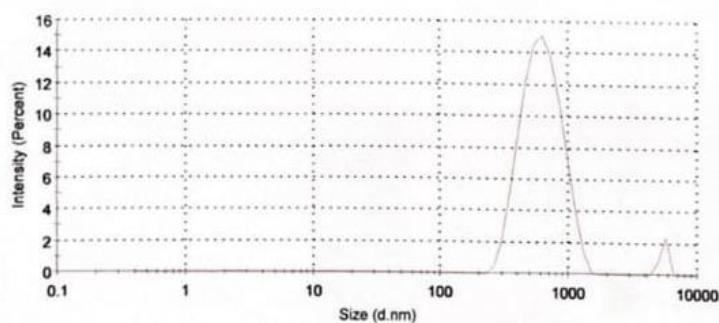
Peak 3: 0,000

0,0

0,000

Result quality Good

Size Distribution by Intensity



Record 1: NLC 2% RIZKY 1

b. Replikasi II

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: NLC 2% RIZKY 2

SOP Name: mansettings.nano

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 2

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 09:59:55

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 257,3

Measurement Position (mm): 4,65

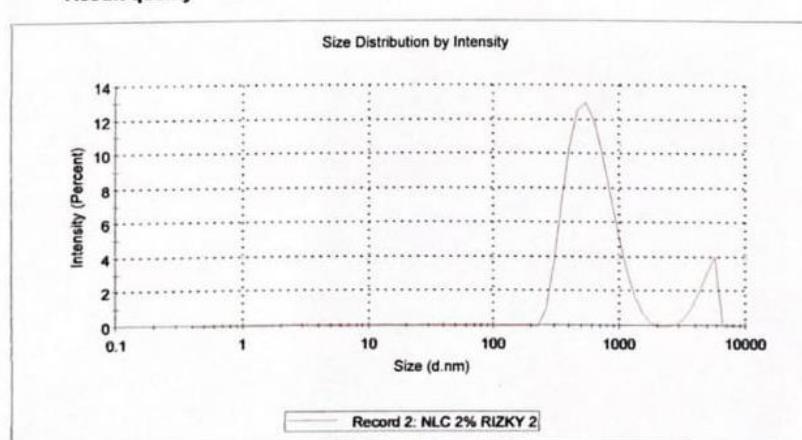
Cell Description: Disposable sizing cuvette

Attenuator: 5

Results

	Size (d.n...	% Intensity:	St Dev (d.n...
Z-Average (d.nm): 649,6	Peak 1: 606,2	89,6	246,8
Pdl: 0,350	Peak 2: 4788	10,4	755,6
Intercept: 0,942	Peak 3: 0,000	0,0	0,000

Result quality **Good**



c. Replikasi III

Size Distribution Report by Intensity v2.2



Sample Details

Sample Name: NLC 2% RIZKY 3

SOP Name: mansettings.nano

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts	Dispersant Name: Water
Record Number: 3	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorbtion: 0,100	Measurement Date and Time: 12 March 2020 10:02:09

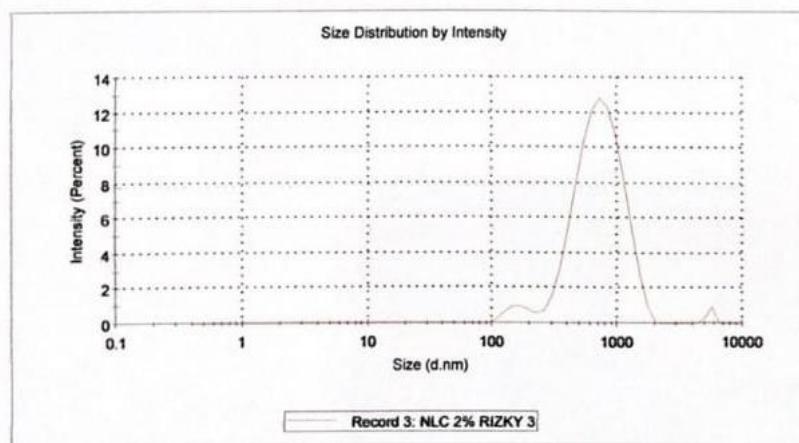
System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 216,1	Measurement Position (mm): 4,65
Cell Description: Disposable sizing cuvette	Attenuator: 5

Results

		Size (d.nm)	% Intensity:	St Dev (d.nm)
Z-Average (d.nm):	644,8	Peak 1: 756,1	94,5	305,6
Pdl:	0,279	Peak 2: 161,7	4,3	33,65
Intercept:	0,945	Peak 3: 5407	1,2	304,5

Result quality Good



Lampiran 9. Uji ukuran partikel formula 2 (Precirol 4%)

a. Replikasi I

Size Distribution Report by Intensity
v2.2



Sample Details

Sample Name: NLC 4% RIZKY 1

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 7

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 10:32:04

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 333,1

Measurement Position (mm): 1,25

Cell Description: Disposable sizing cuvette

Attenuator: 4

Results

	Size (d.nm)	% Intensity	St Dev (d.nm)
--	-------------	-------------	---------------

Z-Average (d.nm): 243,8

Peak 1:

100,0

223,1

Pdl: 0,269

Peak 2:

0,0

0,000

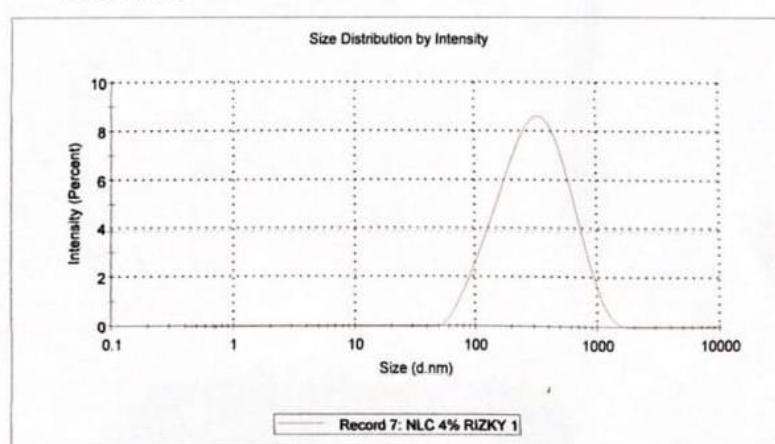
Intercept: 0,925

Peak 3:

0,0

0,000

Result quality Good



b. Replikasi II

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: NLC 4% RIZKY 2

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 8

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 10:34:07

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 331,9

Measurement Position (mm): 1,25

Cell Description: Disposable sizing cuvette

Attenuator: 4

Results

Size (d.nm) % Intensity St Dev (d.nm)

Z-Average (d.nm): 248,7

Peak 1: 309,0 97,3 162,5

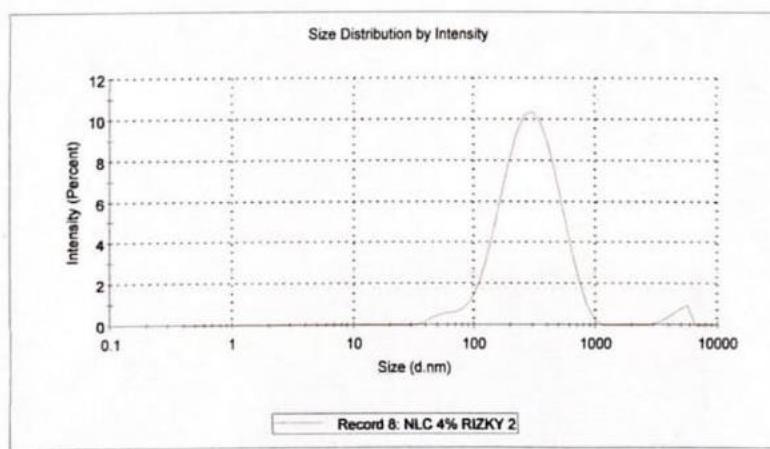
Pdl: 0,265

Peak 2: 4747 2,7 753,0

Intercept: 0,931

Peak 3: 0,000 0,0 0,000

Result quality Good



c. Replikasi III

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: NLC 4% RIZKY 3

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 9

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 10:36:11

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 331,2

Measurement Position (mm): 1,25

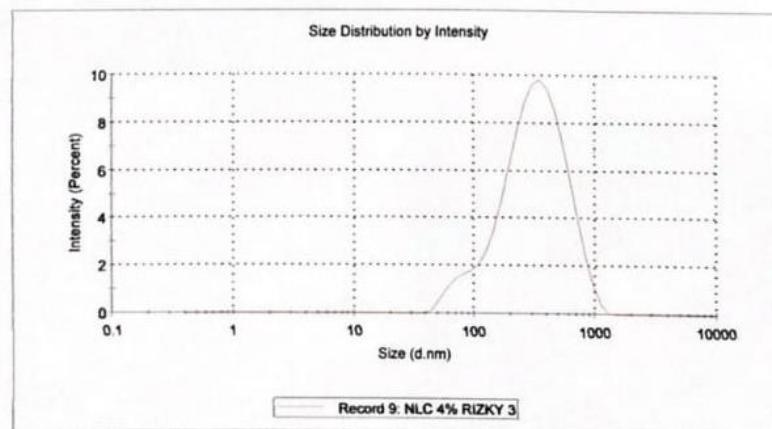
Cell Description: Disposable sizing cuvette

Attenuator: 4

Results

		Size (d.nm)	% Intensity	St Dev (d.nm)
Z-Average (d.nm):	246,1	Peak 1: 346,4	100,0	199,0
Pdl:	0,269	Peak 2: 0,000	0,0	0,000
Intercept:	0,929	Peak 3: 0,000	0,0	0,000

Result quality **Good**



Lampiran 10. Uji ukuran partikel formula 3 (Precirol 6%)

a. Replikasi I

Size Distribution Report by Intensity v2.2



Sample Details

Sample Name: NLC 6% RIZKY 1

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts	Dispersant Name: Water
Record Number: 13	Dispersant RI: 1,330
Material RI: 1,52	Viscosity (cP): 0,8872
Material Absorption: 0,100	Measurement Date and Time: 12 March 2020 11:00:24

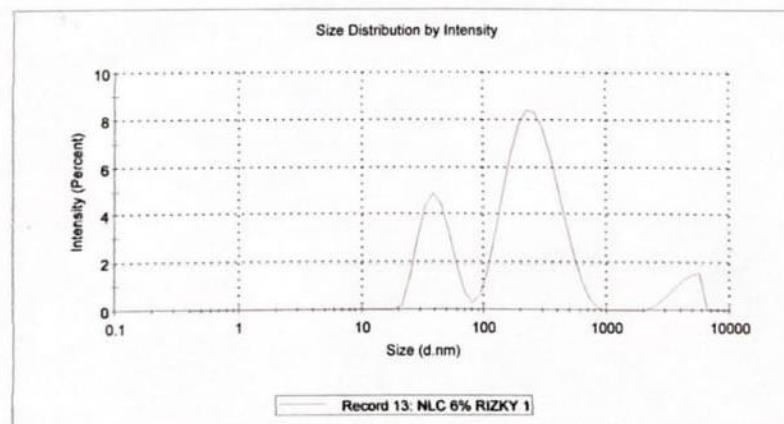
System

Temperature (°C): 25,0	Duration Used (s): 60
Count Rate (kcps): 266,1	Measurement Position (mm): 4,65
Cell Description: Disposable sizing cuvette	Attenuator: 5

Results

	Size (d.nm)	% Intensity	St Dev (d.nm)
Z-Average (d.nm): 121,5	Peak 1: 267,6	69,1	124,1
Pdi: 0,575	Peak 2: 40,66	24,4	11,50
Intercept: 0,948	Peak 3: 4309	6,5	965,2

Result quality Refer to quality report



b. Replikasi II

Size Distribution Report by Intensity

v2.2



Sample Details

Sample Name: NLC 6% RIZKY 2

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 14

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absortion: 0,100

Measurement Date and Time: 12 March 2020 11:02:37

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 263,4

Measurement Position (mm): 4,65

Cell Description: Disposable sizing cuvette

Attenuator: 5

Results

	Size (d.n...	% Intensity:	St Dev (d.n...
--	---------------------	---------------------	-----------------------

Z-Average (d.nm): 123,7

Peak 1: 282,8

69,7

167,1

Pdi: 0,515

Peak 2: 47,52

26,5

18,87

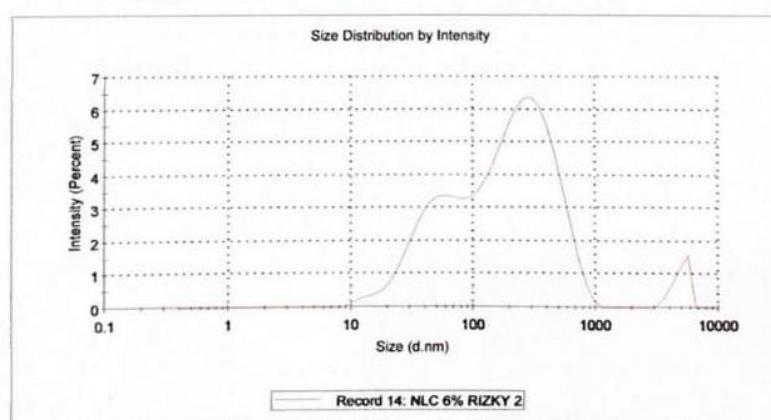
Intercept: 0,947

Peak 3: 4853

3,8

700,8

Result quality Refer to quality report



c. Replikasi III

Size Distribution Report by Intensity v2.2



Sample Details

Sample Name: NLC 6% RIZKY 3

SOP Name: nlc kiki.sop

General Notes: 2 tetes ke 5 ml

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 15

Dispersant RI: 1,330

Material RI: 1,52

Viscosity (cP): 0,8872

Material Absorbtion: 0,100

Measurement Date and Time: 12 March 2020 11:04:51

System

Temperature (°C): 25,0

Duration Used (s): 60

Count Rate (kcps): 261,9

Measurement Position (mm): 4,65

Cell Description: Disposable sizing cuvette

Attenuator: 5

Results

	Size (d.nm)	% Intensity	St Dev (d.nm)
--	--------------------	--------------------	----------------------

Z-Average (d.nm): 113,5

Peak 1: 291,6

66,4

135,8

Pdl: 0,768

Peak 2: 50,06

28,8

17,79

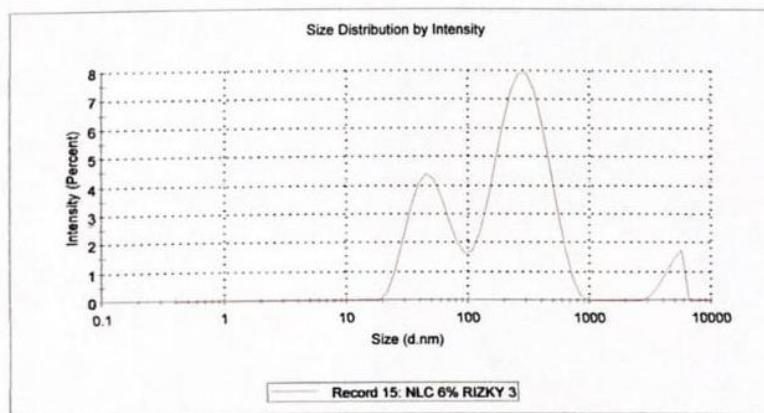
Intercept: 0,948

Peak 3: 4749

4,8

753,0

Result quality Refer to quality report



Lampiran 11. Uji zeta potensial formula I (Precirol 2%)

a. Replikasi I

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 2% RIZKY 1

SOP Name: mansettings.nano

General Notes: 2 TETES 5 ML

File Name:	Rizky_120320.dts	Dispersant Name:	Water
Record Number:	4	Dispersant RI:	1,330
Date and Time:	12 March 2020 10:11:06	Viscosity (cP):	0,8872
		Dispersant Dielectric Constant:	78,5

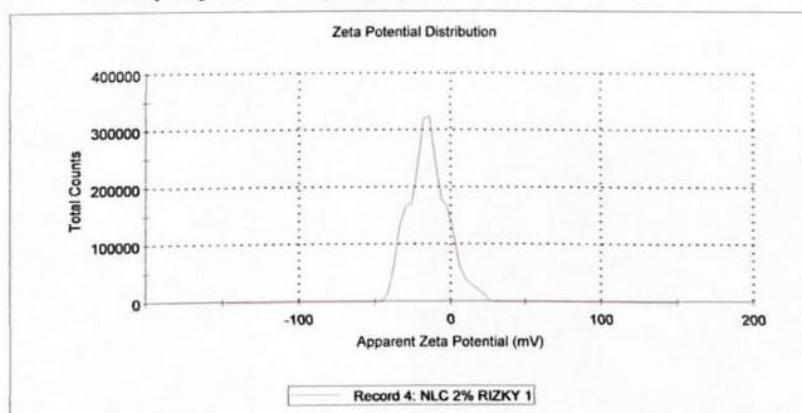
System

Temperature (°C):	25,0	Zeta Runs:	12
Count Rate (kcps):	1403,6	Measurement Position (mm):	4,50
Cell Description:	Zeta dip cell	Attenuator:	4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV):	-15,0	100,0	12,6
Zeta Deviation (mV):	12,6	0,0	0,00
Conductivity (mS/cm):	0,0166	0,0	0,00

Result quality See result quality report



b. Replikasi II

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 2% RIZKY 2

SOP Name: mansettings.nano

General Notes: 2 TETES 5 ML

File Name: Rizky_120320.dts	Dispersant Name: Water
Record Number: 5	Dispersant RI: 1,330
Date and Time: 12 March 2020 10:13:36	Viscosity (cP): 0,8872
	Dispersant Dielectric Constant: 78,5

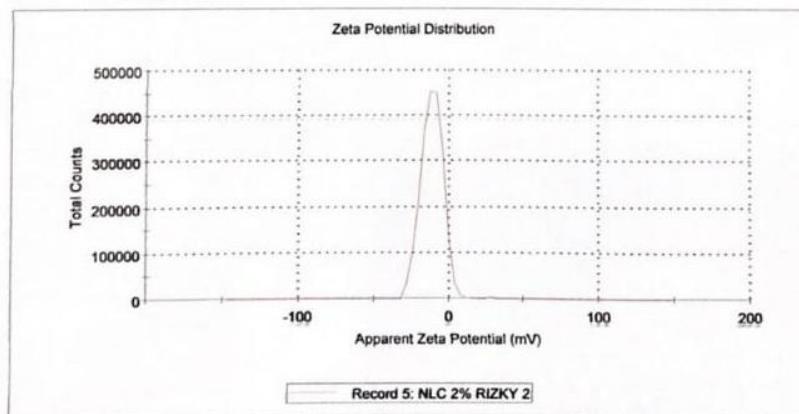
System

Temperature (°C): 25,0	Zeta Runs: 12
Count Rate (kcps): 825,0	Measurement Position (mm): 4,50
Cell Description: Zeta dip cell	Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -11,5	Peak 1: -11,6	99,8	6,99
Zeta Deviation (mV): 7,22	Peak 2: 27,4	0,2	0,00
Conductivity (mS/cm): 0,0170	Peak 3: 0,00	0,0	0,00

Result quality Good



c. Replikasi III

Zeta Potential Report v2.3



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Sample Details

Sample Name: NLC 2% RIZKY 3

SOP Name: mansettings.nano

General Notes: 2 TETES 5 ML

File Name:	Rizky_120320.dts	Dispersant Name:	Water
Record Number:	6	Dispersant RI:	1,330
Date and Time:	12 March 2020 10:14:16	Viscosity (cP):	0,8872
		Dispersant Dielectric Constant:	78,5

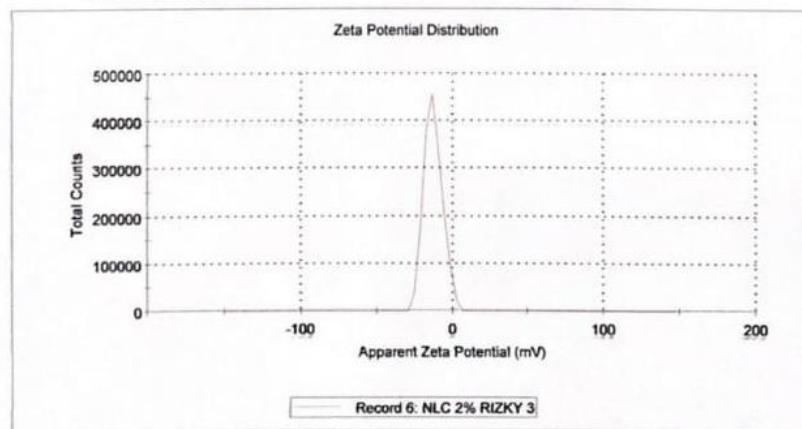
System

Temperature (°C):	25,0	Zeta Runs:	12
Count Rate (kcps):	794,9	Measurement Position (mm):	4,50
Cell Description:	Zeta dip cell	Attenuator:	4

Results

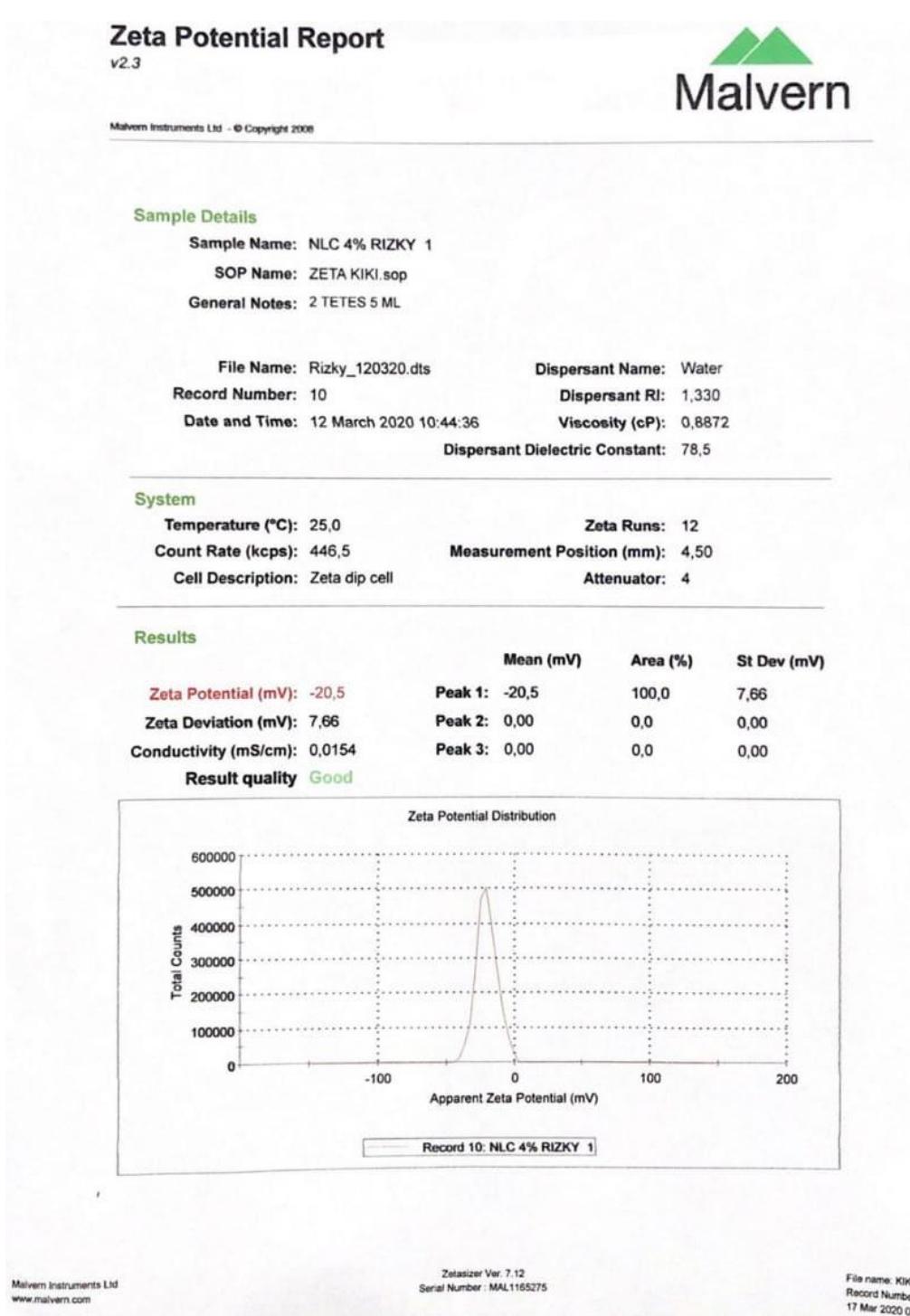
	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -12,9	Peak 1: -12,9	100,0	5,90
Zeta Deviation (mV): 5,90	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0172	Peak 3: 0,00	0,0	0,00

Result quality Good



Lampiran 12. Uji zeta potensial formula 2 (Precirol 4%)

a. Replikasi I



b. Replikasi II

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 4% RIZKY 2

SOP Name: ZETA KIKI.sop

General Notes: 2 TETES 5 ML

File Name: Rizky_120320.dts	Dispersant Name: Water
Record Number: 11	Dispersant RI: 1,330
Date and Time: 12 March 2020 10:47:09	Viscosity (cP): 0,8872
	Dispersant Dielectric Constant: 78,5

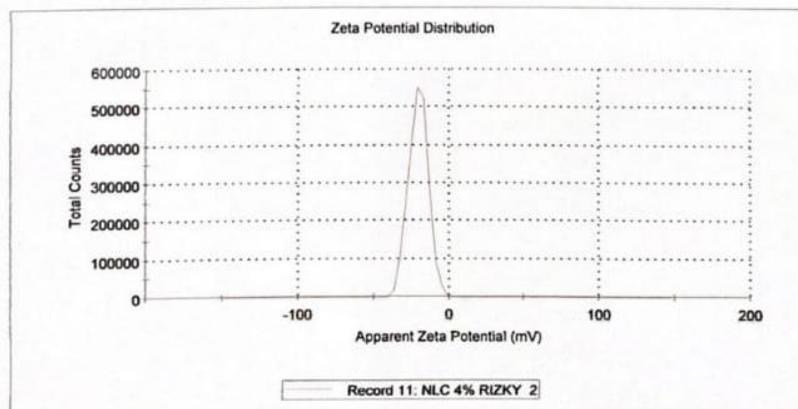
System

Temperature (°C): 25,0	Zeta Runs: 12
Count Rate (kcps): 534,5	Measurement Position (mm): 4,50
Cell Description: Zeta dip cell	Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -20,6	Peak 1: -20,6	100,0	6,31
Zeta Deviation (mV): 6,31	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0156	Peak 3: 0,00	0,0	0,00

Result quality Good



c. Replikasi III

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 4% RIZKY 3

SOP Name: ZETA KIKI.sop

General Notes: 2 TETES 5 ML

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 12

Dispersant RI: 1,330

Date and Time: 12 March 2020 10:47:48

Viscosity (cP): 0,8872

Dispersant Dielectric Constant: 78,5

System

Temperature (°C): 25,0

Zeta Runs: 12

Count Rate (kcps): 586,7

Measurement Position (mm): 4,50

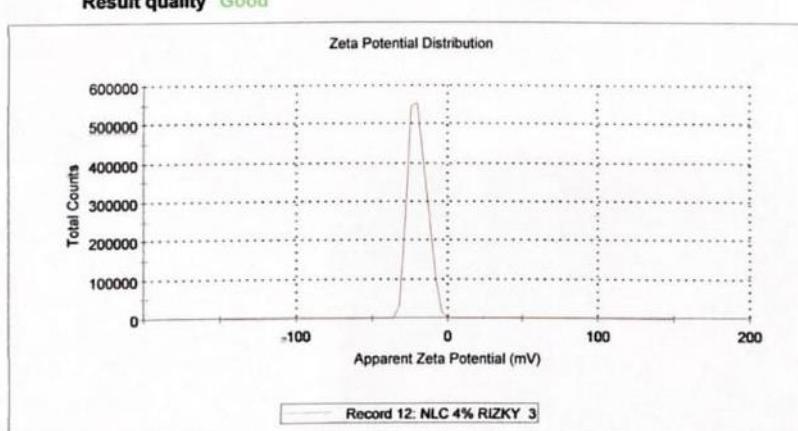
Cell Description: Zeta dip cell

Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -20,0	Peak 1: -20,0	100,0	5,81
Zeta Deviation (mV): 5,81	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0158	Peak 3: 0,00	0,0	0,00

Result quality **Good**



Lampiran 13. Uji zeta potensial formula 3 (Precirol 6%)

a. Replikasi I

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 2% RIZKY 1

SOP Name: mansettings.nano

General Notes: 2 TETES 5 ML

File Name: Rizky_120320.dts	Dispersant Name: Water
Record Number: 4	Dispersant RI: 1,330
Date and Time: 12 March 2020 10:11:06	Viscosity (cP): 0,8872
	Dispersant Dielectric Constant: 78,5

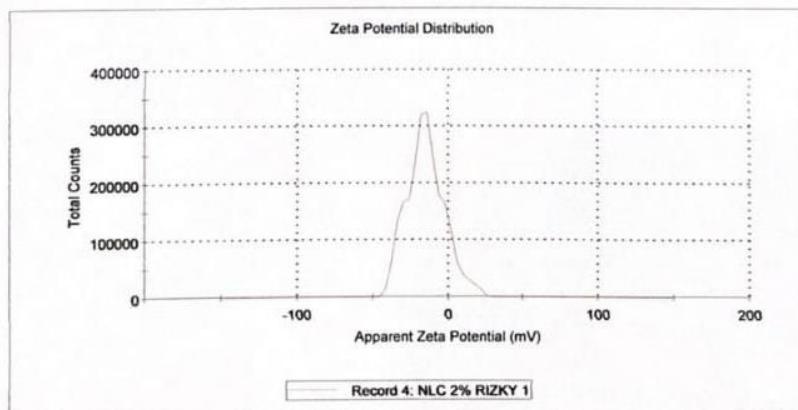
System

Temperature (°C): 25,0	Zeta Runs: 12
Count Rate (kcps): 1403,6	Measurement Position (mm): 4,50
Cell Description: Zeta dip cell	Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -15,0	Peak 1: -15,0	100,0	12,6
Zeta Deviation (mV): 12,6	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0166	Peak 3: 0,00	0,0	0,00

Result quality See result quality report



b. Replikasi II

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 6% RIZKY 2

SOP Name: ZETA KIKI.sop

General Notes: 2 TETES 5 ML

File Name: Rizky_120320.dts

Dispersant Name: Water

Record Number: 17

Dispersant RI: 1,330

Date and Time: 12 March 2020 11:14:08

Viscosity (cP): 0,8872

Dispersant Dielectric Constant: 78,5

System

Temperature (°C): 25,0

Zeta Runs: 12

Count Rate (kcps): 271,6

Measurement Position (mm): 4,50

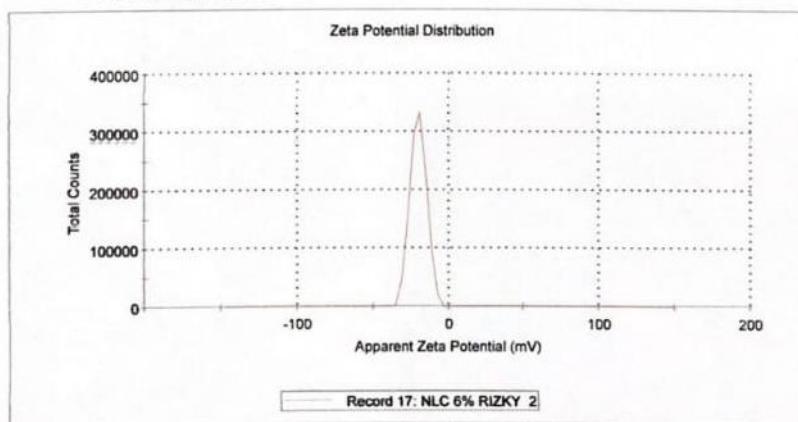
Cell Description: Zeta dip cell

Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -20,2	Peak 1: -20,2	100,0	5,33
Zeta Deviation (mV): 5,33	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0213	Peak 3: 0,00	0,0	0,00

Result quality Good



c. Replikasi III

Zeta Potential Report

v2.3



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Sample Details

Sample Name: NLC 6% RIZKY 3

SOP Name: ZETA KIKI.sop

General Notes: 2 TETES 5 ML

File Name:	Rizky_120320.dts	Dispersant Name:	Water
Record Number:	18	Dispersant RI:	1,330
Date and Time:	12 March 2020 11:14:49	Viscosity (cP):	0,8872
		Dispersant Dielectric Constant:	78,5

System

Temperature (°C): 25,0

Zeta Runs: 12

Count Rate (kcps): 388,3

Measurement Position (mm): 4,50

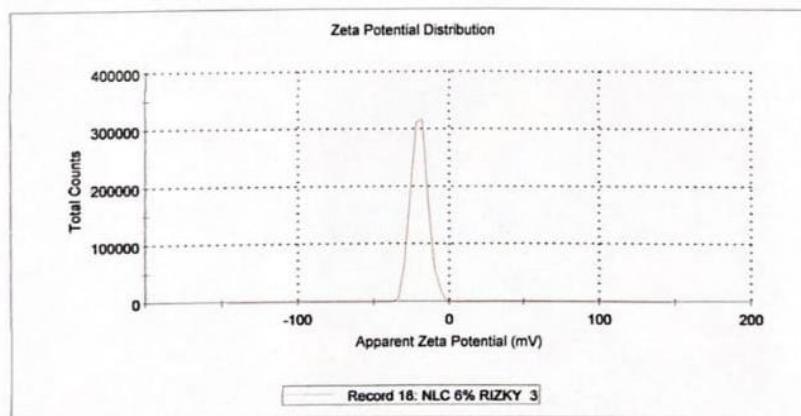
Cell Description: Zeta dip cell

Attenuator: 4

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -20,0	Peak 1: -20,0	100,0	5,44
Zeta Deviation (mV): 5,44	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0215	Peak 3: 0,00	0,0	0,00

Result quality Good



Lampiran 14. Foto serum NLC fisetin baru dibuat.



Lampiran 15. Foto serum NLC fisetin setelah 30 hari



Lampiran 16. Data skunder efisiensi penjerapan.

a. Rancangan Formula NLC

Table 3. Composition of modified diflucortolone valerate in Precirol®-based nanostructured lipid carriers using lipid-based surfactants.

Formula ^a	Type	Total lipid (20% w/w/formulation)		Poloxamer [®] 407 concentration (% w/w/formulation)	Lipid based surfactant		
		Solid lipid			Liquid lipid		
		Concentration (% w/w/base)	Type		Concentration (% w/w/base)	Type	
NLCM1	Precirol®	70	Capryol™	30	10	Labrasol® 5	
NLCM2		70		30	10	10	
NLCM3		70		30	10	5	
NLCM4		70		30	10	10	
NLCM5	Precirol®	50	Capryol™	50	10	Labrasol® 5	
NLCM6		50		50	10	10	
NLCM7		50		50	10	5	
NLCM8		50		50	10	10	
NLCM9	Precirol®	70	IPM	30	10	Labrasol® 5	
NLCM10		70		30	10	10	
NLCM11		70		30	10	5	
NLCM12		70		30	10	10	
NLCM13	Precirol®	50	IPM	50	10	Labrasol® 5	
NLCM14		50		50	10	10	
NLCM15		50		50	10	5	
NLCM16		50		50	10	10	
NLCM17	Tristearin®	70	Capryol™	30	10	Labrasol® 5	
NLCM18		70		30	10	10	
NLCM19		70		30	10	5	
NLCM20		70		30	10	10	
NLCM21	Tristearin®	50	Capryol™	50	10	Labrasol® 5	
NLCM22		50		50	10	10	
NLCM23		50		50	10	5	
NLCM24		50		50	10	10	
NLCM25	Tristearin®	70	IPM	30	10	Labrasol® 5	
NLCM26		70		30	10	10	
NLCM27		70		30	10	5	
NLCM28		70		30	10	10	
NLCM29	Tristearin®	50	IPM	50	10	Labrasol® 5	
NLCM30		50		50	10	10	
NLCM31		50		50	10	5	
NLCM32		50		50	10	10	

^aFormulation weight was made to 100% with distilled water and contained 0.1% w/w of diflucortolone valerate.

(Abdel-Salam, Mahmoud, Ammar & Elkheshen 2016)

b. Hasil uji ukuran partikel, PDI dan efisiensi penjerapan

Table 4. Mean particle size, polydispersity index, and entrapment efficiency values of modified nanostructured lipid carriers.

Formulation	Mean particle size (nm) \pm SD	PDI \pm SD	EE% \pm SD
NLCM1	439.1 \pm 24.9	0.44 \pm 0.09	41.12 \pm 4.90
NLCM2	490.1 \pm 21.5	0.44 \pm 0.06	43.06 \pm 4.50
NLCM3	229.0 \pm 25.4	0.23 \pm 0.02	44.82 \pm 3.52
NLCM4	245.1 \pm 14.5	0.28 \pm 0.08	68.00 \pm 7.40
NLCM5	206.1 \pm 20.5	0.24 \pm 0.12	42.06 \pm 5.80
NLCM6	743.7 \pm 20.4	0.52 \pm 0.03	42.02 \pm 4.78
NLCM7	246.3 \pm 17.2	0.14 \pm 0.05	50.20 \pm 5.65
NLCM8	197.7 \pm 7.6	0.15 \pm 0.06	59.59 \pm 5.20
NLCM9	230.9 \pm 15.2	0.39 \pm 0.13	37.38 \pm 5.62
NLCM10	589.2 \pm 22.6	0.52 \pm 0.11	38.38 \pm 4.66
NLCM11	264.1 \pm 33.4	0.24 \pm 0.01	44.56 \pm 5.70
NLCM12	244.5 \pm 30.1	0.25 \pm 0.03	53.18 \pm 2.94
NLCM13	187.5 \pm 11.5	0.22 \pm 0.02	40.28 \pm 3.86
NLCM14	337.2 \pm 21.5	0.53 \pm 0.09	41.77 \pm 3.96
NLCM15	183.5 \pm 14.3	0.23 \pm 0.04	36.18 \pm 5.02
NLCM16	180.8 \pm 18.2	0.37 \pm 0.14	55.61 \pm 8.65
NLCM17	ND	ND	ND
NLCM18	ND	ND	ND
NLCM19	295.0 \pm 13.7	0.23 \pm 0.02	51.34 \pm 4.00
NLCM20	255.1 \pm 11.0	0.23 \pm 0.01	56.95 \pm 6.60
NLCM21	ND	ND	ND
NLCM22	ND	ND	ND
NLCM23	276.9 \pm 19.8	0.27 \pm 0.03	42.29 \pm 3.70
NLCM24	219.7 \pm 6.5	0.15 \pm 0.01	63.90 \pm 4.30
NLCM25	252.6 \pm 25.7	0.24 \pm 0.05	40.75 \pm 5.40
NLCM26	231.3 \pm 20.5	0.25 \pm 0.01	48.25 \pm 1.93
NLCM27	229.5 \pm 9.3	0.25 \pm 0.02	ND
NLCM28	217.7 \pm 8.0	0.28 \pm 0.02	64.59 \pm 1.09
NLCM29	ND	ND	ND
NLCM30	ND	ND	ND
NLCM31	ND	ND	ND

ND, not determined.

(Abdel-Salam, Mahmoud, Ammar & Elkheshen 2016)

c. Hasil uji efisiensi penjerapan (Apriyani 2019).

Larutan induk → 200 mg NLC Fisetin/10 ml etanol p.a = 20.000 ppm

$$\text{Perhitungan teoritis Fisetin} = 10 \text{ mg}$$

$$\text{Eksipien (tween 80 + IPM + GMS)} = 11.500 \text{ mg %}$$

$$\text{Kadar fisetin} = \frac{10}{11.500+10} \times 100\% = 0,0869\%$$

$$\text{Kadar dalam 200 mg NLC} = 0,0869\% \times 200\text{mg} = 0,1738 \text{ mg}$$

Perhitungan kadar fisetin terjerap menggunakan persamaan regresi linier

$$: y = a + bx$$

$$0,449 = 0,014 + ,00636x$$

$$0,0636x = 0,435$$

$$x = 6,840 \text{ ppm}$$

$$\% \text{ kadar} = \frac{6840}{20000 \text{ ppm}} \times 100\% = 0,0324\%$$

$$\text{Kadar dalam 200mg NLC fisetin} = 0,0324\% \times 200 \text{ mg} = 0,0648 \text{ mg}$$

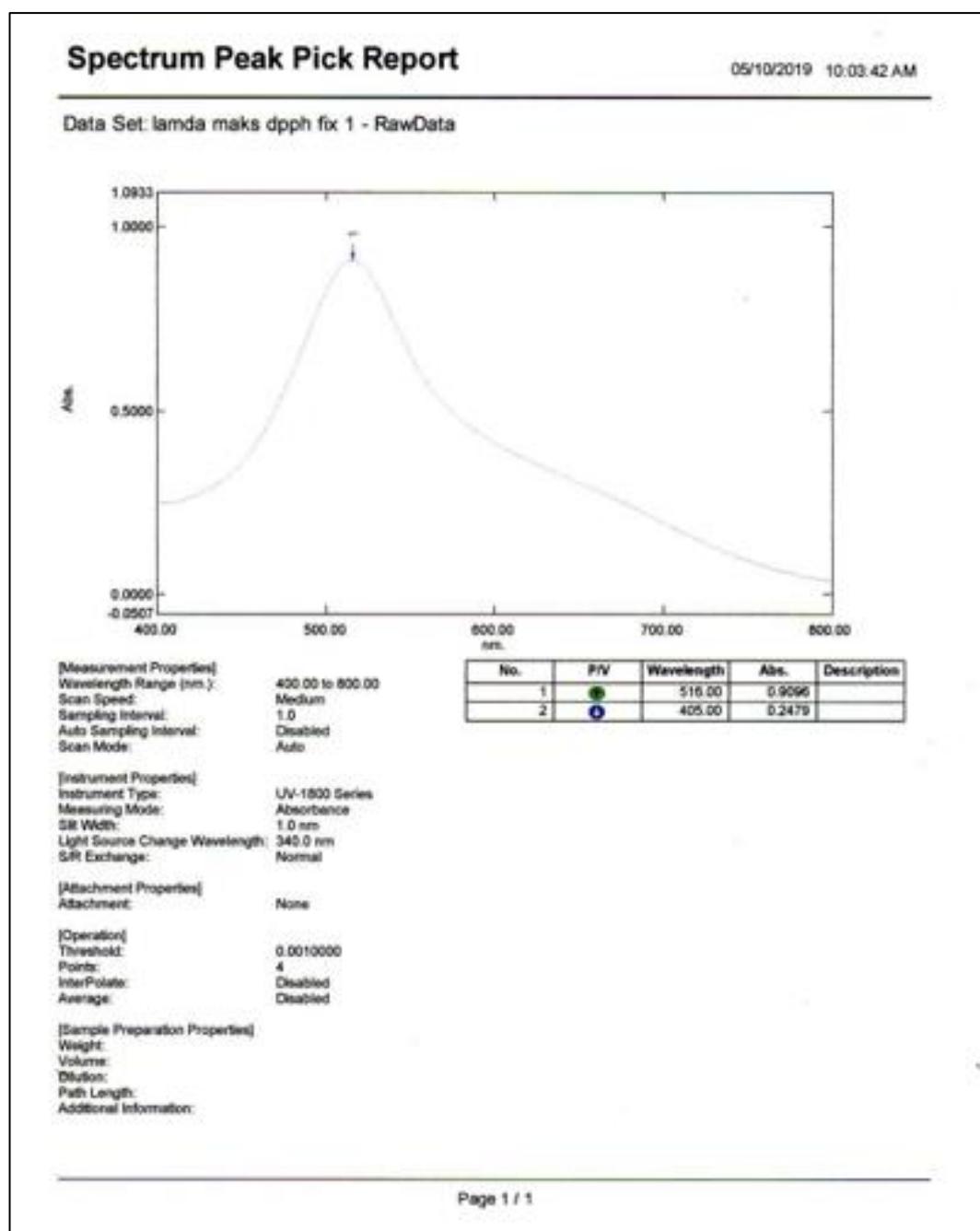
$$\% \text{ Efisiensi penjerapan (EE)} = \frac{\text{kadar terjerap}}{\text{kadar teoritis}} \times 100\%$$

$$= \frac{0,0648}{0,174} \times 100\%$$

$$= 59,20\%$$

Lampiran 17. Uji Aktivitas Antioksidan (Apriyani D 2019).

- a. Penentuan Lamda max.



b. Penentua *Operating Time*

Kinetics Data Print Report

OT Firekin DPPH 4/5/19
60 MEHIT

05/04/2019 12:24:22 PM

Time (Minute)	RawData ...
0.000	0.199
1.000	0.199
2.000	0.198
3.000	0.198
4.000	0.199
5.000	0.198
6.000	0.198
7.000	0.198
8.000	0.198
9.000	0.199
10.000	0.198
11.000	0.199
12.000	0.198
13.000	0.199
14.000	0.198
15.000	0.199
16.000	0.199
17.000	0.199
18.000	0.199
19.000	0.199
20.000	0.199
21.000	0.199
22.000	0.199
23.000	0.199
24.000	0.199
25.000	0.199
26.000	0.199
27.000	0.199
28.000	0.199
29.000	0.199
30.000	0.199
31.000	0.199
32.000	0.199
33.000	0.199
34.000	0.200
35.000	0.200
36.000	0.199
37.000	0.200
38.000	0.200
39.000	0.200
40.000	0.200
41.000	0.200
42.000	0.200
43.000	0.200
44.000	0.200
45.000	0.200
46.000	0.200
47.000	0.200
48.000	0.200
49.000	0.200
50.000	0.200

Page 1 / 2

Kinetics Data Print Report

05/04/2019 12:24:22 PM

Time (Minute)	RawData ...
51.000	0.201
52.000	0.200
53.000	0.200
54.000	0.201
55.000	0.201
56.000	0.201
57.000	0.201
58.000	0.201
59.000	0.201
60.000	0.201

Lampiran 18. DPPH Fisetin (Apriyani 2019).

Perhitungan bahan fisetin : $50\text{mg}/100\text{ml} = 500\text{mg}/1000\text{ml} = 500 \text{ ppm}$

Perhitungan konsentrasi (ppm) :

- 1) $V_1 \times N_1 = V_2 \times N_2$
 $500 \text{ ppm} \times 0,3112 \text{ ml} = 10\text{ml} \times N_2$
 $N_2 = 15,56 \text{ ppm}$
- 2) $V_1 \times N_1 = V_2 \times N_2$
 $500 \text{ ppm} \times 0,1556 \text{ ml} = 10 \text{ ml} \times N_2$
 $N_2 = 7,78 \text{ ppm}$
- 3) $V_1 \times N_1 = V_2 \times N_2$
 $500 \text{ ppm} \times 0,0778 \text{ ml} = 10\text{ml} \times N_2$
 $N_2 = 3,89 \text{ ppm}$
- 4) $V_1 \times N_1 = V_2 \times N_2$
 $500 \text{ ppm} \times 0,039 \text{ ml} = 10\text{ml} \times N_2$
 $N_2 = 1,95 \text{ ppm}$
- 5) $V_1 \times N_1 = V_2 \times N_2$
 $500 \text{ ppm} \times 0,0194 \text{ ml} = 10\text{ml} \times N_2$
 $N_2 = 0,97 \text{ ppm}$

a. Replikasi I

Konsentrasi (ppm)	Absorbansi	% Inhibisi
15,56	0,212	76,7
7,78	0,405	55,5
3,89	0,507	44,3
1,95	0,555	39,0
0,97	0,578	36,5

Persamaan regresi linier antara absorbansi dan % inhibisi diperoleh

nilai :

$$a = 33,6727$$

$$b = 2,7696$$

$$r = 0,99994$$

$$y = a + bx \quad y = 33,6727 + 2,7696x$$

keterangan :

x = konsentrasi (ppm)

$$y = 5$$

Perhitungan IC50 fisetin

$$50 = a + bx$$

$$50 = 33,6727 + 2,7696x$$

$$33,6272x = 16,3273 x$$

$$= 5,89 \text{ ppm}$$

b. Replikasi II

Konsentrasi (ppm)	Absorbansi	% Inhibisi
15,56	0,221	75,7
7,78	0,416	54,3
3,89	0,509	44,0
1,95	0,557	38,8
0,97	0,578	36,5

Persamaan regresi linier antara absorbansi dan % inhibisi diperoleh

nilai :

$$a = 33,5812$$

$$b = 2,6973$$

$$r = 0,9999$$

$$y = a + bx$$

$$y = 33,58124 + 2,6973x$$

keterangan :

x = konsentrasi (ppm)

y = 50

Perhitungan IC₅₀ fisetin

$$50 = a + bx$$

$$50 = 33,5812 + 2,6973x$$

$$33,5812x = 16,4188$$

$$x = 6,09 \text{ ppm}$$

c. Replikasi III

Konsentrasi (ppm)	Absorbansi	% Inhibisi
15,56	0,224	75,4
7,78	0,418	54,0
3,89	0,511	43,8
1,95	0,559	38,5
0,97	0,588	35,4

Persamaan regresi linier antara absorbansi dan % inhibisi diperoleh

nilai :

$$a = 33,0177$$

$$b = 2,7215$$

$$r = 0,9999$$

$$y = a + bx$$

$$y = 33,0177 + 2,7215x$$

keterangan :

x = konsentrasi (ppm)

$$y = 50$$

Perhitungan IC₅₀ fisetin

$$50 = a + bx$$

$$50 = 33,0177 + 2,7215x$$

$$33,0177x = 16,9823$$

$$x = 6,24 \text{ ppm}$$

Rata-rata IC50 Fisetin :

$$\left. \begin{array}{l} \text{Replikasi I} = 5,89 \text{ ppm} \\ \text{Replikasi II} = 6,09 \text{ ppm} \\ \text{Replikasi III} = 6,24 \text{ ppm} \end{array} \right\} 6,07 \text{ ppm}$$