

BAB V

KESIMPULAN DAN SARAN

A. Kesimpulan

Berdasarkan hasil penelitian optimasi natrium karboksi metil selulosa (Na-CMC) dan karbopol 941 pada sediaan gel dispersi oadat ibuprofen dengan metode *simplex lattice design* dapat disimpulkan :

1. Gel dispersi padat ibuprofen dengan campuran CMC-Na dan karbopol 941 memberikan mutu fisik yang meliputi organoleptis, homogenitas, daya lekat, daya sebar, pH, dan viskositas serta stabilitas yang baik.
2. Formula optimum gel dispersi padat ibuprofen diperoleh proporsi 3% Na-CMC dan 0% karbopol 941.

B. Saran

Saran dari penelitian berdasarkan hasil penelitian optimasi natrium karboksi metil selulosa (Na-CMC) dan karbopol 941 pada sediaan gel dispersi oadat ibuprofen dengan metode *simplex lattice design* yaitu :

1. Perlu dilakukan uji laju penetrasi secara *in vitro* untuk mengetahui jumlah obat yang terpenetrasi.
2. Perlu dilakukan penelitian lebih lanjut dengan variasi *gelling agent* lainnya untuk mendapatkan formula gel yang baik.

DAFTAR PUSTAKA

- Afifi, S. 2015. Solid Dispersion Approach Improving Dissolution Rate of Stiripentol: A Novel Antiepileptic Drug. *Iranian Journal of Pharmaceutical Research*, 14 (4), 1001-1014.
- Agoes, G., dan Darijanto, S.T.1993. Teknologi Farmasi Likuida Dan Semi Solida. Bandung: Pusat Antar Universitas Bidang Ilmu Hayati ITB.
- Anderson, PO., James, EK, & William, GT., 2002. *Handbook of Clinical Drug Data*. (10thEd). New York: McGraw_Hill Companies.
- Ansel, H. C., Allen, L. V. dan Popovich, N. G. 1999. Pharmaceutical Dosage Forms and Drug Delivery Systems. Seventh Edition. Philadelphia: Lippincot Williams dan Wilkins.
- Aulton, M. E. 1991. *Pharmaceutical Practice*. 109-123. Longman Singapore Publisher Ptc Ltd. Singapore.
- Bajaj H, Bisht S, Yadav M, Singh V. Bioavailability enhancement: a review. *Int J Pharm Bio Sci*. 2011; 2(2): 202-216.
- Balasaheb P, Balaji T, Avinash B. 2014. Solid dispersion on overview on solubility enhancement of poorly water soluble drugs. *International Journal of Pharma and Bio Sciences* 5(3) : 7-25.
- Bley, H., Fussnegger, B., and Bodmeier, R. 2010. Characterization and Stability of Solid Dispersions Based on PEG/Polymer Blends. *International Journal of Pharmaceutics*, 390 (2), 165-173.
- Bolton. (1997). *Pharmaceutical Statistic*. 3rd Ed. 308-337. Marcel Dekker Inc. New York.
- Bushra, R., dan Aslam, N. 2010. An Overview of Clinical Pharmacology of Ibuprofen. *Oman Media Journal*. 25 (3): 155-166.
- Chiou, W. L. dan Riegeman, S. 1971. Pharmaceutical Application of Solid of Solid Dispersion System. *J. Pharm. Sci*. 60(9): 1281-1302.
- Chowdari, K.P.R. dan Srinivas, L. 2000. Physical Stability and Dissolution Rate of Ibuprofen Suspension Formulated Employing Solid Dispersion. *Indian J. Pharm. Sci.*, 62, 253-256.
- Clarke, E.G.C., Moffat, A.C., Osselton, M.D., dan Widdop, B. 2005. Clarke's Analysis of Drugs and Poisons. London : Pharmaceutical Press.
- Clegg. 1995. <http://simonbwidjanarko.files.wordpress.com/2008/06/bahan-pembentuk-gel-2.pdf> di akses pada tanggal 20 September 2019.
- Depkes RI. 1979. *Farmakope Indonesia*. Edisi 3. Jakarta: Departemen Kesehatan RI.

- Depkes RI. 1995. Farmakope Indonesia. Edisi IV. Jakarta: Departemen Kesehatan RI.
- Djajadisastra, J. 2004. *Cosmetic Stability*. Depok : Universitas Indonesia.
- Djajadisastra J, A. Mun'im, dan Dessy NP. Formulasi gel topikal dari ekstrak Nerii folium dalam sediaan anti jerawat. *Jurnal Farmasi Indonesia*. 2009; 4(4):2106.
- Draganoiu, 2009, Karbomer, Dalam Rowe R. C., Paul J. S. and Marian E. Q., eds., *Handbook of Pharmaceutical Exipients*, Pharmaceutical Press and American Pharmaceutical Assosiation, London, 110-114.
- Erawati, T., Rosita, N., Hendropasetyo, W., dan Juwita, D.R. 2005. Pengaruh Jenis Basis Gel dan Penambahan NaCl (0,5% b/b) terhadap Intensitas Echo Gelombang Ultrasonik Sediaan Gel untuk Pemeriksaan USG (Acoustic Coupling Agent). *Majalah Farmasi Airlangga*. Vol. 5 No. 2.
- Erlianti, R., Darusalam, F., & Herawati, D. (2015). Praperlakuan Bahan Baku Glimepirid Melalui Metode Kokristalisasi Untuk Meningkatkan Kelarutan dan Laju Disolusi. *Prosiding Penelitian SPeSIA Unisba*, 3, 671–680.
- Florence, A.T., dan Attwood, D. 1998. *Phisicochemical Princples of Pharmacy*, 3rd Edition, London : Mac Millan Press LTD.
- Gavrilin M. V., Fat'yanova E.A., Lukashova L.A., Kompantseva E. V. and Ziep C., 2000, Effect of Crystallization Conditions on the Solubility of Ibuprofen, *Pharmaceutical Chemistry Journal*, 34 (10), 555–557.
- Gandjar, I.G., Rohman, A. 2007. *Analisis Obat secara Kromatografi dan Spektroskopi*. Yogyakarta: Pustaka Pelajar. hlm 466-497.
- Gandjar IG, Rohman A. *Kimia Farmasi Analisis*. Edisi XI. Yogyakarta: Pustaka Pelajar; 2013.
- Harmita. 2004. *Petunjuk Pelaksanaan Validasi Metode dan Cara Perhitungannya*. *Majalah Ilmu Kefarmasian* 1 : 117-135.
- Hascicek, C., Bedis, A., dan Gonul, N. 2009. Preparation and Evaluation of Different Gel Formulations for Transdermal Delivery of Meloxicam. *Turk. J. Pharm Sci*, 6(3) : 177 – 186.
- Husnani, H., & Al Muazham, M. F. 2017. Optimasi Parameter Fisik Viskositas, Daya Sebar dan Daya Lekat pada Basis Natrium CMC dan Carbopol 940 pada Gel Madu dengan Metode *Simplex Lattice Design*. *JIFFK: Jurnal Ilmu Farmasi dan Farmasi Klinik*, 14(1), 11-18.
- Johari G.P., Kim S., Shanker R.M. Dielectric studies of molecular motions in amorphous solid and ultraviscous acetaminophen. *J Pharm Sci*. 2005; 94(10): 2207–2223.
- Karavas, E., Ktistis, G., Xenakis, A., dan Georgarakis, E. 2006. Effect of hydrogen bonding interactions on the release mechanism of felodipine from nanodispersions with polyvinylpyrrolidone. *Eur. J. Pharm. Biopharm.*, 63: 103–114.

- Lachman, L., Liebermann, H.A., dan Kanig, J.I. 1994. Teori and Praktek Farmasi Industri II. Edisi III. Jakarta: UI Press.
- Lachman. 1989. Pharmaceutical Dosage System. *Disperse System*. 2: 495-496.
- Laksmi,P.K., 2011. Formulation and evaluation of ibuprofen topical gel: a novel approach for penetration enhancement. *Int. J. Pharm*ISSN- 0975-7058 Vol 3, Issue 3, 2011.
- Lieberman, H.A., Rieger, M.M., dan Banker, G.S. 1996. Pharmaceutical Dosage Forms: Disperse Systems, Second Edition. New York: Marcell Dekker Inc.
- Malolepsza, K. Dan Jarmolowska. 2007. The Influence of Glycerol on the Release of Metronidazole From gels Containing Lactid Acid Complexed with Chitosan. Polish Chitin Society, Monograph XII.
- Mappa T, Edy HJ, Kojong N. Formulasi gel ekstrak daun sasaladahan (*Peperomia pellucida* (L.) H.B.K) dan uji efektivitasnya terhadap luka bakar pada kelinci (*Oryctolagus cuniculus*).*Jurnal Ilmiah Farmasi*. 2013;2(2):49-55.
- Marinda, Wenny Silvia. 2012. Formulasi dan Uji Stabilitas Fisik Gel Liposom yang Mengandung Fraksinasi Ekstrak Metanol Kulit Manggis (*Garcinia mangostana* L.) sebagai Antioksidan. Depok. Universitas Indonesia.
- Mulhaquddin. 2014. *Validation Method*. Dipresentasikan pada Diklat Validasi Metode, Baristand Industri Ambon 10 - 13 Juni 2014.
- Mulja, M., dan Suharman. 1995. Analisis Instrumental. Airlangga University Press. Surabaya. pp. 26-31.
- Niyaz, B., P. Kalyani & G. Divakar. (2011). Formulation and Evaluation of Gel Containing Fluconazole-Antifungal Agent. *International Journal Of Drug Development & Research*, 3 : 109-128.
- Paradkar, A., Ambike, A.A., Jadhav, B.K. dan Mahadik, K.K. 2004. Haracterization of curcumin-PVP solid dispersion obtained by spray drying. *Int. J. Pharm.*, 271, 281-286.
- Patel, A., Bell,M., O'Connorb, C., Inchleyb,A., Wibawab, J., dan Lanea, M.E. 2013. Delivery of ibuprofen to the skin. *Int. J. Pharm*. 457 (2013) 9– 13
- Prabhu., S. 2005. Novel lipid-based formulations enhancing the in vitro. *Eur. J. Pharm. Biopharm*.
- Rainsford KD, Stetsko PI, Sirko SP, Debski S. Gastrointestinal mucosal injury following repeated daily oral administration of conventional formulations of indometacin and other non-steroidal antiInflammatory drugs to pigs: a model for human gastrointestinal disease. *J Pharm Phramacol*. 2003; 55 (5): 661-668.
- Ramukutti, S., dan Ramachandran, E. 2014. Reaction Rate for the Thermal Decomposition of Ibuprofen Crystals. *J. Crys. Pro. Tech.*, 4: 71-78.

- Raton, F.L Boca and C.K Smoley. 1993. *Everything Added to Food in the United States*. Available at :<http://en.wikipedia.org/wiki/Gellingagent> Di akses pada tanggal 20 September 2019.
- Rawat, S. 2011. Release enhancement of meloxicam from transdermal gel through cyclodextrin complexation. *Int. J. Pharm. Sci. and Res.* 2 (2):357-365.
- Roni, M. A., Dipu, M. H., Kibria, G., Rahman, F., Rony, mD R., & Jalil, R. (2011). Dissolution Enhancement of Poorly Soluble Carbamazepine By Using. *Ijpsr*, 2(1), 49–57.
- Rowe, R. C., Sheskey, P. J., and Quinn, M. E. 2006. Handbook of Pharmaceutical Excipients. Sixth Edition. USA: Pharmaceutical Press and American Pharmacists Association: 110, 442, 592.
- Rowe, R. C., Sheskey, P. J., and Quinn, M. E. 2006. *Handbook of Pharmaceutical Excipients*. Sixth Edition. 549-553, 675-681. Pharmaceutical Press. London.
- Rowe, R. C., Sheskey, P. J., and Quinn, M. E., 2009, Handbook of Pharmaceutical Excipients, 6th edition, Pharmaceutical Press and American Pharmacists Association, Washington D. C., pp. 118-121, 592594.
- Sa'adah, Eva, dan Ari Surya Winata. 2010. Validasi metode pengujian logam tembaga pada produk air minum dalam kemasan secara spektrofotometri serapan atom nyala. *BIOPROPAL INDUSTRI* 01 (02) : 31–37.
- Saroja K, Singh S, Aggarwal A, Nanda S. Transdermal gels an alternative vehicle for drug delivery. *J Pharm Chem Bio Sci.* 2010; 3 (3): 495-503.
- Serajuddin, A. 1999. Solid dispersion technique. *Journal of Pharmaceutical Sciences*, 88(10), 891-900.
- Serajuddin A.T.M. Solid Dispersion of Poorly WaterSoluble Drugs: Early Promises, Subsequent Problems, and Recent Breakthroughs. *J Pharm Sci.* 1999; 88(10): 1058-1066.
- Shah, T. J., Amin, A. F., Parikh, J. R. dan Parikh, R. H. 2007. Process optimization and characterization of poloxamer solid dispersions of a poorly water-soluble drug. *AAPS PharmSciTech.*, vol. 8, issue 2, Article 29, pp. E18-E24.
- Simon, Patricia (2012). Formulasi dan Uji penetrasi mikroemulsi natrium diklofenak dengan metode sel difusi Franz dan metodeTape stripping [Skripsi], Depok : Prodi Farmasi FMIPA Universitas Indonesia.
- Sinatra, R.S, Hord, A.H., dan Grinsberg, B. 1992. *Acute Pain Mechanisms & Management*. Missouri: Mosby Year Book.
- Sinko, P. J. 2011. *Martin's Physical Pharmacy and Pharmaceutical Science: Physical Chemical and Biopharmaceutical Principle in the*

- Pharmaceutical Sciences*, 5th edition, Lippicott William and Wilkins, Philadelphia.
- Sinko, Patrick J. (2012). *Farmasi Fisik dan Ilmu Farmasetika*. Jakarta : EGC.
- Singh, J., dan Singh, R. 2009. Optimization and Formulation of Orodispersible Tablets of Meloxicam. *Trop. J.Pharm. Res.* Vol. 8 (2): 153-159.
- Stoelting, R.K., dan Hillier, S.C. 2006. *Pharmacology & Physiology in Anesthetic Practice*. Edisi ke-4. Lipincott: Williams & Wilkins.
- Sulaiman, T.N.S., Kurniawan. 2009. *Teknologi dan Formulasi Sediaan Tablet*. Laboratorium Teknologi Farmasi Fakultas Farmasi Universitas Gadjah Mada. Yogyakarta. 42-46.
- Swastika A., Mufrod dan Purwanto, 2013, Aktivitas Antioksidan Krim Ekstrak Sari Tomat (*Solanum lycopersicum L .*), *Trad. Med. J.*, 18 (September), 132–140.
- Sweetman SC. Martindale: The complete drug reference. 36th ed. London: Pharmaceutical Press; 2009.
- Taylor, L.S., and Zografi, G., 1997. Spectroscopic characterization of interactions between PVP and indomethacin in amorphous molecular dispersions. *Pharm. Res.*, 14: 1691–1698.
- Tranggono, R. I. dan Latifah, F. 2007. *Buku Pedoman Ilmu Pengetahuan Kosmetik*. Jakarta: Penerbit Gramedia Pustaka Utama.
- Tripathi, K.D. 2003. *Non steroidal anti inflammatory drugs and anti pyretic analgesics*. In: *Essentials of medical pharmacology*. New Delhi: Jaypee Brothers.
- Vasanthavada M., Tong W.Q., Joshi Y., Kislalioglu M.S. Phase behavior of amorphous molecular dispersions I: Determination of the degree and mechanism of solid solubility. *Pharm Res*. 2004; 21(9): 1598–1606.
- Vasconcelos, T., and Costa, P., 2007. Development of a rapid dissolving ibuprofen solid dispersion. In *PSWC – Pharmaceutical Sciences World Conference.*, 23:11-130.
- Voight, R.1995. *Buku Pelajaran Teknologi Farmasi Edisi V*, diterjemahkan Noerono, S. Yogyakarta : Gajah Mada University Press.
- Yanhendri dan Yenny S.W. 2012. Berbagai bentuk sediaan topical dalam dermatologi. *CDK-194*. 39(6).
- Zatz, J, L., & G.P. Kushla. (1996). *Pharmacheutical dosage forms: disperse system* vol ke-2. Gels. Lieberman, H, A., M.M. Rieger., & G.S. Banker. (Editor) Edisi II. Marcel Dekker Inc, New York.

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Lampiran 1. Sertifikat analisis ibuprofen



IOL CHEMICALS AND PHARMACEUTICALS LIMITED

CERTIFICATE OF ANALYSIS

Product Name : **IBUPROFEN BP** Batch No. : 4001/1201/18/A-4151
 Date of Mfg. : Nov. - 2018 Date of Analysis : 27/11/2018
 Date of Expiry : Oct. - 2023 A.R. No. : 4001/1151/1118/A-4151/10667
 Drug Lic No. : 1689-OSP Batch Qty : 1000 Kg
 Dispatch Qty : 1000 Kg Packing : 20 X 50 Kg

Sr. No	TEST	SPECIFICATIONS	RESULTS
1.	Appearance	White or almost White Crystalline Powder or Colourless Crystals	White crystalline powder.
2.	Solubility	Practically insoluble in water, freely soluble in Acetone, Methanol, and Methylene chloride. It dissolves in dilute solution of alkali hydroxide and carbonate	Complies
3.	Identification 1 st Identification A,C 2 nd Identification A,B,D	A) Melting Point 75-78°C B) By UV Exhibits 2 absorption maxima, at 264 nm and 272 nm. The ratio of the absorbance measured at the maximum at 264 nm to that measured at the shoulder at 258 nm is 1.20 to 1.30. The ratio of the absorbance measured at the maximum at 272 nm to that measured at the shoulder at 258 nm is 1.00 to 1.10. C) By IR The infra-red absorbance spectrum obtained from the sample should be concordant with spectrum obtained from the standard D) By TLC The principal spot in the chromatogram obtained with the test solution is similar in position, color and size to the principal spot in the chromatogram obtained with the reference solution	76°C Omitted as per pharmacopoeia Complies. Omitted as per pharmacopoeia

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 Ph. : +91-1679-285285-86, Fax : +91-1679-285292

Lampiran 2. Gambar alat dan bahan penelitian

Alat	
Spektrofotometer UV-Vis 	Spektroskopi Inframerah 
Water Bath 	Ice Bath 
Mortir-Stamper 	Cawan Porselin 

Alat uji daya lekat



Alat uji daya sebar



Alat uji viskositas



pH meter



Alat uji homogenitas



Timbangan analitik digital



Ayakan mesh no 80



Pompa Hidrolik



Desikator

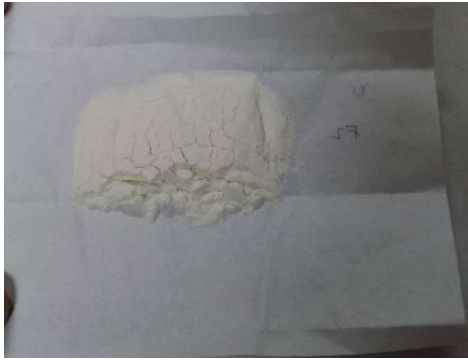


Alat-alat gelas

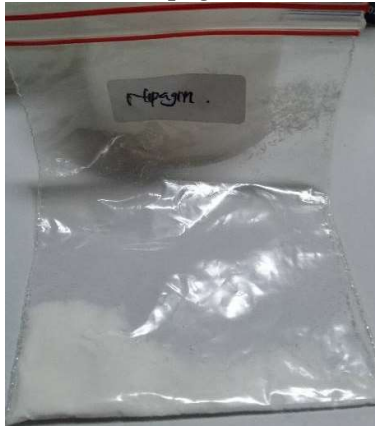


Oven

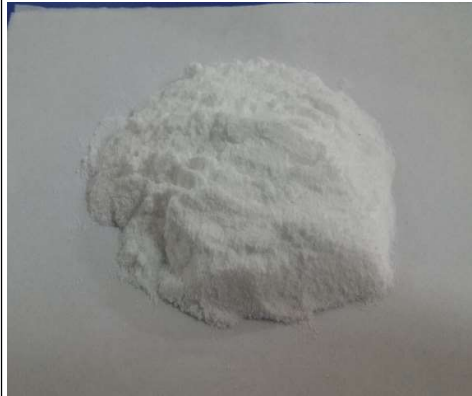


Bahan	
Ibuprofen 	PEG 6000 
KBr 	Karbopol 
CMC-Na 	TEA 

Nipagin



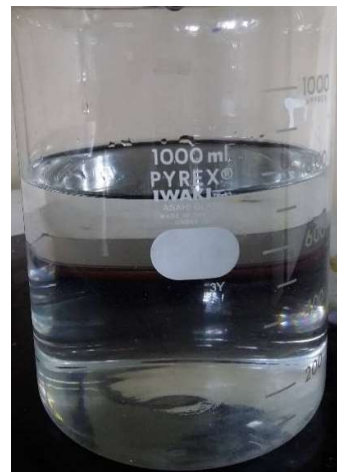
Dispersi padat ibuprofen-PEG 6000



Larutan NaOH



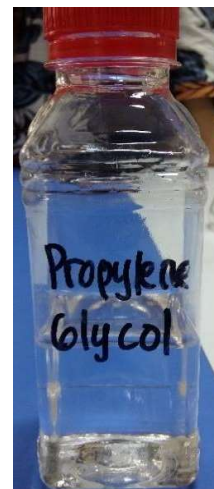
Aquadest



Tween 80



Propilen-glikol



Lampiran 3. Penimbangan pembuatan dispersi padat dan penetapan kadar dispersi padat

A. Penimbangan pembuatan dispersi padat

Kertas timbang + ibuprofen = 0,042 mg

Kertas timbang + sisa = 0,002 mg –

Ibuprofen = 0,04 mg = 40 g

PEG 6000 = 0,06 mg = 60 g +

Total = 100 g

B. Penimbangan kurva baku

Kertas timbang + ibuprofen = 0,3675 g

Kertas timbang + sisa = 0,2641 g –

Ibuprofen = 0,1034 g = 103,4 mg/100 ml = 1034 ppm

Pengenceran kurva baku

$$V_1 \times N_1 = V_2 \times N_2$$

$$10 \text{ ml} \times \mathbf{150 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 1,5 \text{ ml}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$10 \text{ ml} \times \mathbf{200 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 2 \text{ ml}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$10 \text{ ml} \times \mathbf{250 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 2,5 \text{ ml}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$10 \text{ ml} \times \mathbf{300 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 3 \text{ ml}$$

$$V1 \times N1 = V2 \times N2$$

$$10 \text{ ml} \times \mathbf{350 \text{ ppm}} = V2 \times 1000 \text{ ppm}$$

$$V2 = 3,5 \text{ ml}$$

$$V1 \times N1 = V2 \times N2$$

$$10 \text{ ml} \times \mathbf{400 \text{ ppm}} = V2 \times 1000 \text{ ppm}$$

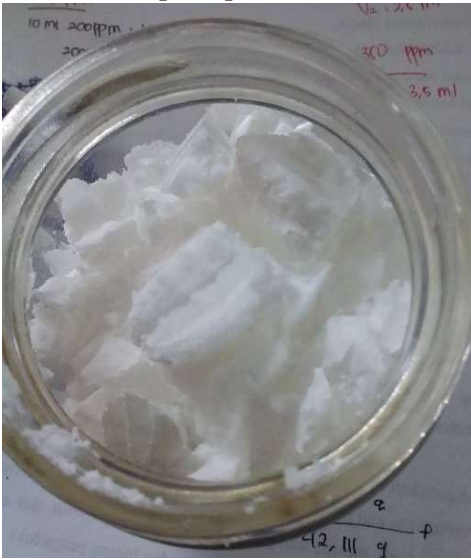
$$V2 = 4 \text{ ml}$$

$$V1 \times N1 = V2 \times N2$$

$$10 \text{ ml} \times \mathbf{450 \text{ ppm}} = V2 \times 1000 \text{ ppm}$$

$$V2 = 4,5 \text{ ml}$$

Lampiran 4. Foto hasil peleburan, pembekuan, dispersi padat kasar dan halus

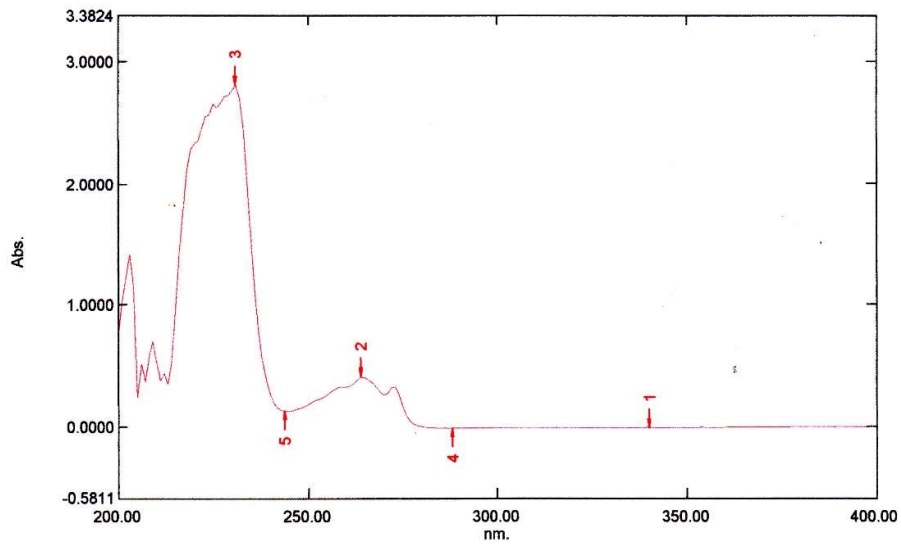
Peleburan PEG 6000 dan Ibuprofen 	Pembekuan leburan dispersi padat ibuprofen 
Dispersi padat kasar 	Dispersi padat halus 

Lampiran 5. Hasil penentuan panjang gelombang maksimum ibuprofen

Spectrum Peak Pick Report

02/01/2020 10:30:15 AM

Data Set: File_200201_102832 - RawData



[Measurement Properties]
Wavelength Range (nm.): 200.00 to 400.00
Scan Speed: Medium
Sampling Interval: 1.0
Auto Sampling Interval: Disabled
Scan Mode: Single

[Instrument Properties]
Instrument Type: UV-1800 Series
Measuring Mode: Absorbance
Slit Width: 1.0 nm
Light Source Change Wavelength: 340.0 nm
S/R Exchange: Normal

[Attachment Properties]
Attachment: None

[Operation]
Threshold: 0.0010000
Points: 4
InterPolate: Disabled
Average: Disabled

[Sample Preparation Properties]
Weight:
Volume:
Dilution:
Path Length:
Additional Information:

No.	P/V	Wavelength	Abs.	Description
1	●	340.00	-0.0081	
2	●	264.00	0.4073	
3	●	231.00	2.8152	
4	●	288.00	-0.0140	
5	●	244.00	0.1251	

Lampiran 6. Hasil penetapan *operating time*

Kinetics Data Print Report

Time (Minute)	RawData ...
0.000	0.568
1.000	0.568
2.000	0.569
3.000	0.569
4.000	0.569
5.000	0.570
6.000	0.570
7.000	0.569
8.000	0.569
9.000	0.570
10.000	0.569
11.000	0.569
12.000	0.569
13.000	0.568
14.000	0.569
15.000	0.569
16.000	0.569
17.000	0.568
18.000	0.568
19.000	0.568
20.000	0.568
21.000	0.569
22.000	0.569
23.000	0.569
24.000	0.569
25.000	0.569
26.000	0.569
27.000	0.569
28.000	0.569
29.000	0.569
30.000	0.569

Lampiran 7. Perhitungan % *recovery* ibuprofen dalam dispersi padat

$$a = 0,0324$$

$$b = 0,0018$$

$$r = 0,9993$$

$$Y = a + bx$$

$$= 0,0324 + 0,0018x$$

A. Replikasi 1

Dispersi padat ibuprofen yang ditimbang:

$$\text{Kertas timbang} + \text{dispersi} = 307,8 \text{ mg}$$

$$\text{Kertas timbang} + \text{sisir} = 276,7 \text{ mg}$$

$$\text{Dispersi} = 31,1 \text{ mg}$$

$$\text{Abs} = 0,238$$

Dalam 31,1 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{31,1 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 12,44 \text{ mg ibuprofen}$$

Kadar 12,44 mg ibuprofen yaitu:

$$\frac{12,44 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 124,4 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,238 = 0,0324 + 0,0018x$$

$$x = \frac{0,238 - 0,0324}{0,0018}$$

$$= 114,22 \text{ ppm}$$

$$\% \text{Recovery} = \frac{114,222 \text{ ppm}}{124,4 \text{ ppm}} \times 100\% = 91,82\%$$

B. Replikasi 2

Dispersi padat ibuprofen yang ditimbang:

Kertas timbang + dispersi = 298 mg

Kertas timbang + sisa = 266 mg –

Dispersi = 32 mg

Abs = 0,246

Dalam 32 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{32 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 12,8 \text{ mg ibuprofen}$$

Kadar 12,8 mg ibuprofen yaitu:

$$\frac{12,8 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 128 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,246 = 0,0324 + 0,0018x$$

$$x = \frac{0,246 - 0,0324}{0,0018}$$

$$= 118,67 \text{ ppm}$$

$$\% \text{Recovery} = \frac{118,67 \text{ ppm}}{128 \text{ ppm}} \times 100\% = 92,72\%$$

C. Replikasi 3

Dispersi padat ibuprofen yang ditimbang:

Kertas timbang + dispersi = 306,0 mg

Kertas timbang + sisa = 275,5 mg –

Dispersi = 30,5 mg

Abs = 0,231

Dalam 30,5 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{30,5 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 12,2 \text{ mg ibuprofen}$$

Kadar 12,2 mg ibuprofen yaitu:

$$\frac{12,2 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 122 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,231 = 0,0324 + 0,0018x$$

$$x = \frac{0,231 - 0,0324}{0,0018}$$

$$= 110,33 \text{ ppm}$$

$$\% \text{Recovery} = \frac{110,33 \text{ ppm}}{122 \text{ ppm}} \times 100\% = 90,43\%$$

$$\text{Rata-rata } \% \text{ recovery} = \frac{91,82 + 92,72 + 90,43}{3} = 91,66\%$$

$$\begin{aligned} \text{SD} &= \sqrt{\frac{(91,66 - 91,82)^2 + (91,66 - 92,71)^2 + (91,66 - 90,43)^2}{3 - 1}} \\ &= \sqrt{\frac{0,0256 + 1,236 + 1,5129}{2}} = 1,1537 \end{aligned}$$

$$\text{CV} = \frac{1,1537}{91,66} \times 100\% = 1,26\%$$

Lampiran 8. Perhitungan Penggunaan Ibuprofen dalam Formula Gel

Perbandingan ibuprofen : PEG 6000 = 1 : 1,5

Dalam 100 gram sediaan gel mengandung 2,5 gram dispersi padat ibuprofen-PEG 6000 terdiri atas 1 gram ibuprofen dan 1,5 gram PEG 6000. %*Recovery* ibuprofen dalam dispersi padat 91,66% sehingga dalam 2,5 gram dispersi padat mengandung 0,9166 gram ibuprofen.

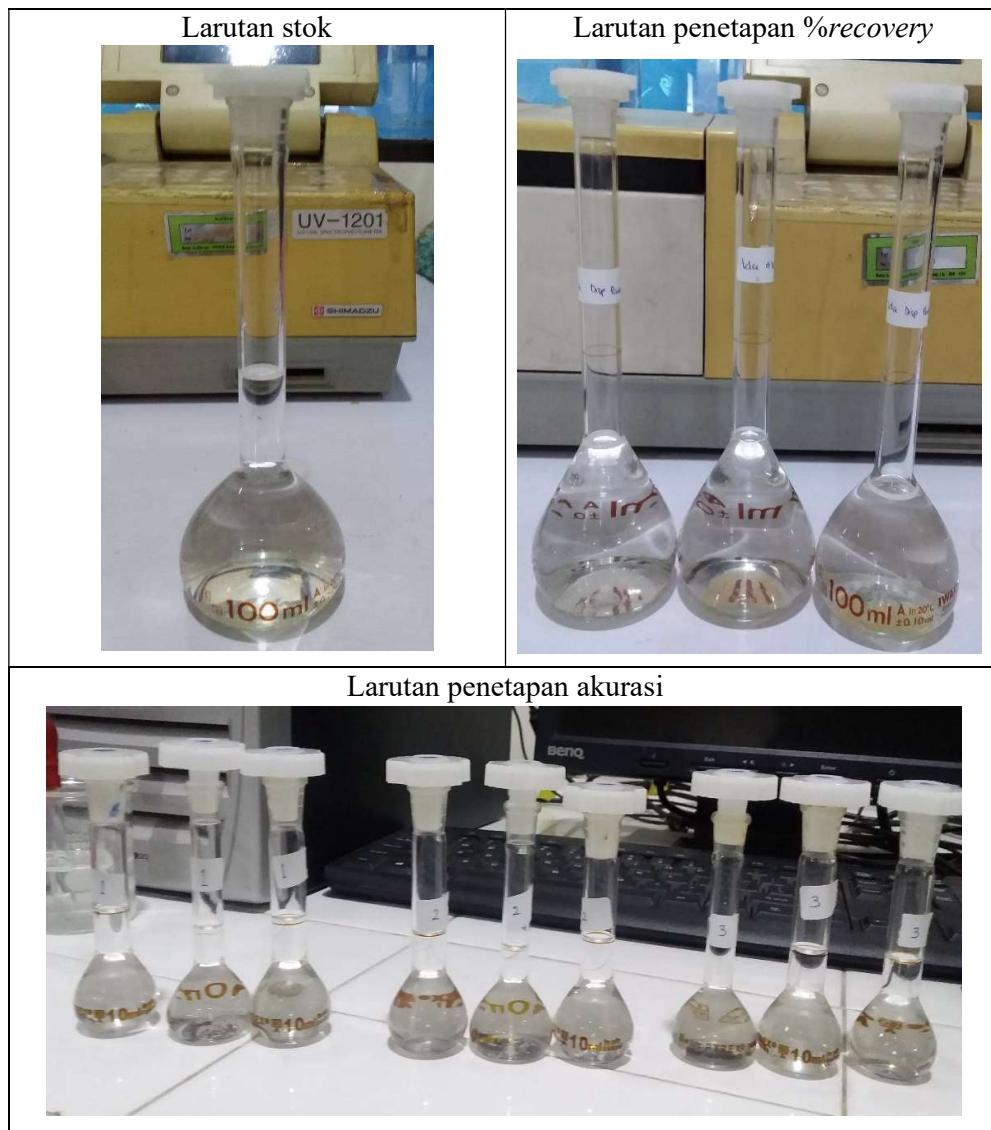
Pada formulasi 100 gram sediaan gel diinginkan mengandung 1 gram ibuprofen.

$$\frac{2,5 \text{ gram dispersi padat}}{x \text{ gram dispersi padat}} = \frac{0,9166 \text{ gram ibuprofen}}{1 \text{ gram ibuprofen}}$$

$$x = 2,73 \text{ gram dispersi padat}$$

Sehingga dapat disimpulkan bahwa dalam formulasi 100 gram sediaan gel mengandung 2,73% dispersi padat ibuprofen yang setara dengan 1% ibuprofen.

Lampiran 9. Foto larutan stok, penetapan %*recovery*, dan akurasi



Lampiran 10. Verifikasi metode analisis**A. Linearitas**

$$V1 \times N1 = V2 \times N2$$

$$1,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 155,1 \text{ ppm}$$

$$V1 \times N1 = V2 \times N2$$

$$2 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 206,8 \text{ ppm}$$

$$V1 \times N1 = V2 \times N2$$

$$2,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 258,5 \text{ ppm}$$

$$V1 \times N1 = V2 \times N2$$

$$3 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 310,2 \text{ ppm}$$

$$V1 \times N1 = V2 \times N2$$

$$3,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 361,9 \text{ ppm}$$

$$V1 \times N1 = V2 \times N2$$

$$4 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 413,6 \text{ ppm}$$

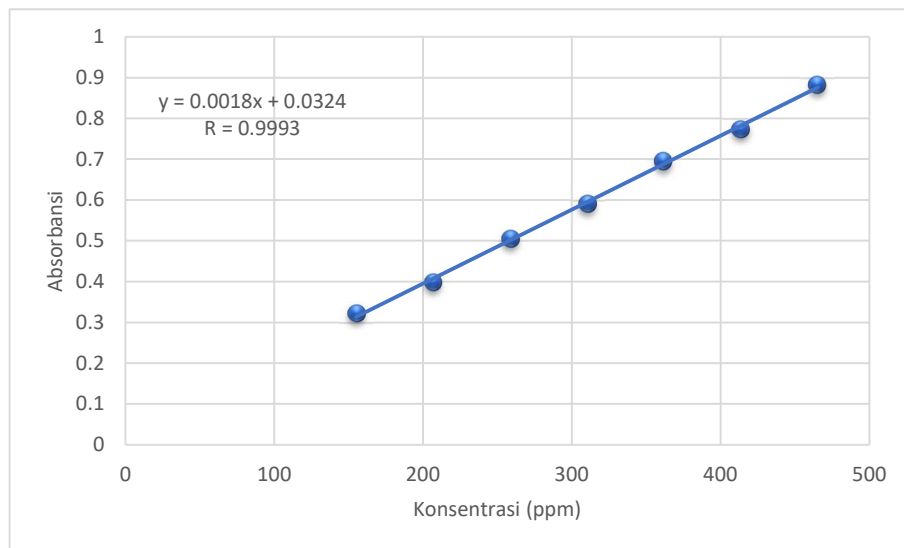
$$V1 \times N1 = V2 \times N2$$

$$4,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N2$$

$$N2 = 465,3 \text{ ppm}$$

Konsentrasi	Absorbansi
155.1	0.322
206.8	0.397
258.5	0.504
310.2	0.591
361.9	0.694
413.6	0.773
465.3	0.883

a	0.0324
b	0.0018
r	0.9993



A. Akurasi

Pengambilan larutan:

$$80\% \rightarrow \frac{80}{100} \times 310,2 = 248,16 \text{ ppm} \rightarrow \frac{248,16 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 2,48 \text{ ml} \approx 2,5 \text{ ml}$$

$$100\% \rightarrow \frac{100}{100} \times 310,2 = 310,2 \text{ ppm} \rightarrow \frac{310,2 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 3 \text{ ml}$$

$$120\% \rightarrow \frac{120}{100} \times 310,2 = 372,24 \text{ ppm} \rightarrow \frac{372,24 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 3,6 \text{ ml} \approx 3,5 \text{ ml}$$

KONSENTRASI	REPLIKASI	ABS	KONSENTRASI	PPM SEBENARNYA	%	RATA- RATA	RECOVERY
80%	1	0.491	252.9262	258.5	98%	99.34%	99.09%
	2	0.497	256.2350	258.5	99%		
	3	0.506	261.1982	258.5	101%		
100%	1	0.589	306.9700	310.2	99%	98.84%	
	2	0.588	306.4185	310.2	99%		
	3	0.588	306.4185	310.2	99%		
120%	1	0.676	354.9476	361.9	98%	99.35%	
	2	0.69	362.6681	361.9	100%		
	3	0.687	361.0137	361.9	100%		

B. Presisi

REPLIKASI	ABSORBANSI	KONSENTRASI
1	0.267	129.3977
2	0.272	132.1550
3	0.266	128.8462
4	0.271	131.6036
5	0.273	132.7065
6	0.268	129.9492
7	0.27	131.0521
8	0.274	133.2580
9	0.267	129.3977
10	0.272	132.1550

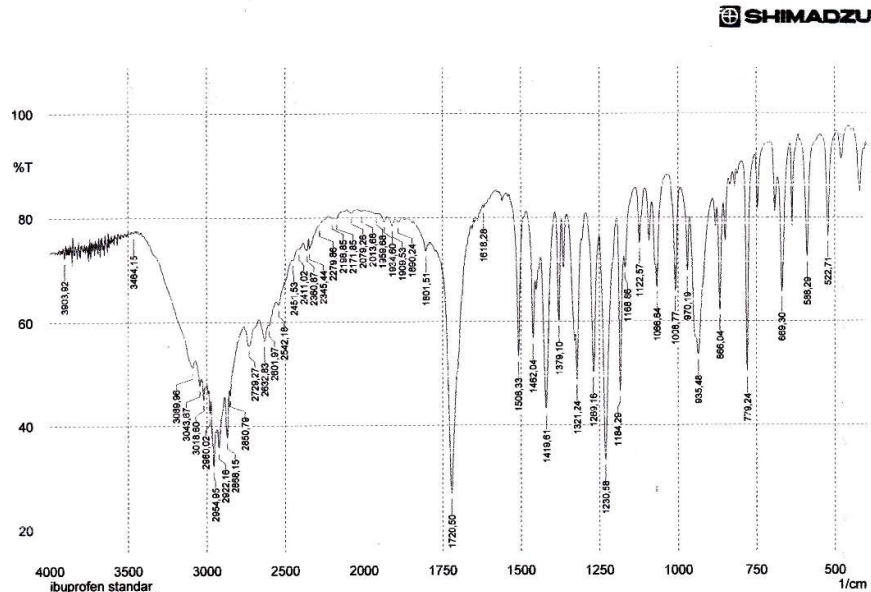
SD 1.559783

RATA-RATA 131.0521

CV 1.190201

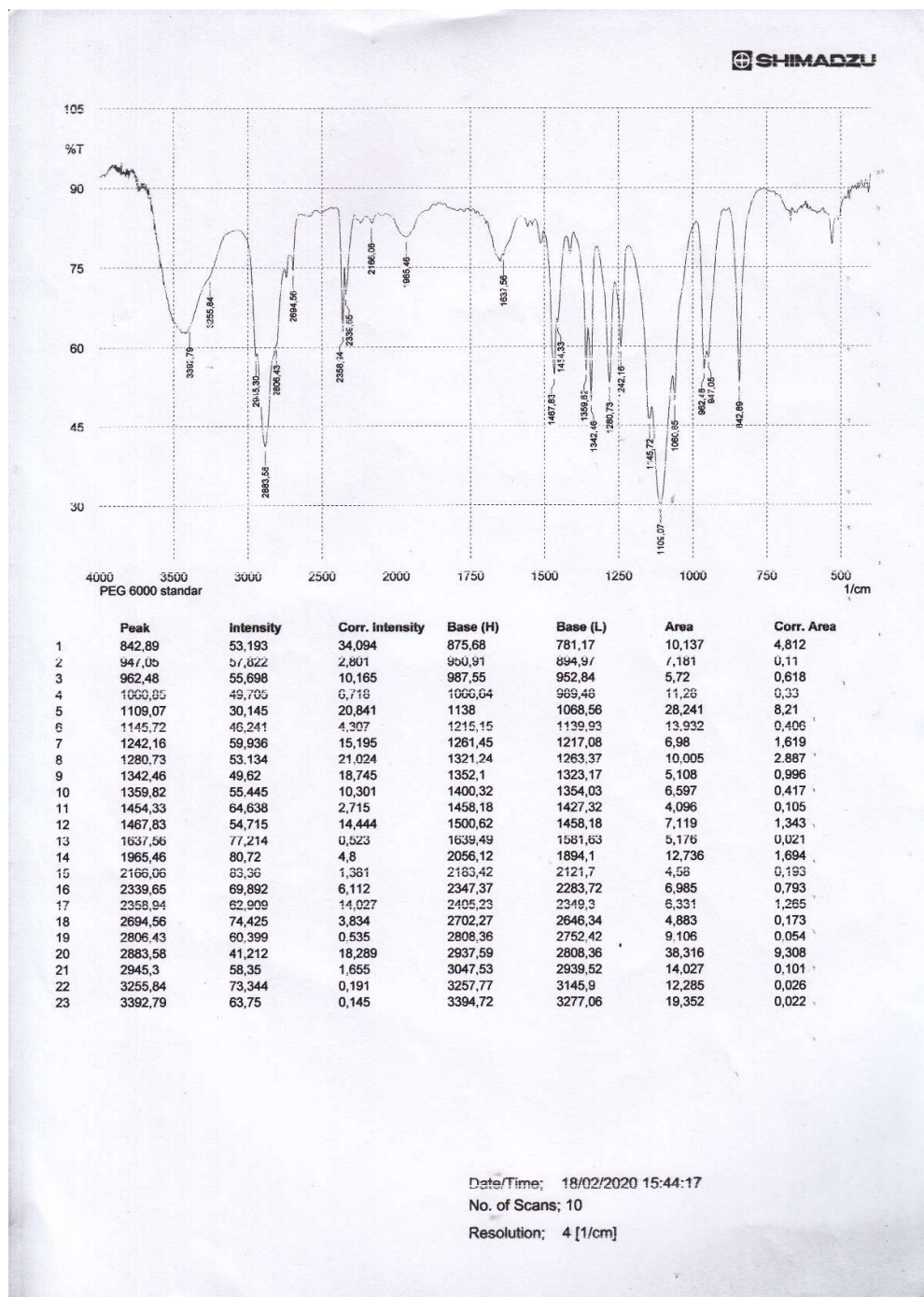
Lampiran 11. Hasil spektra IR

a. Ibuprofen

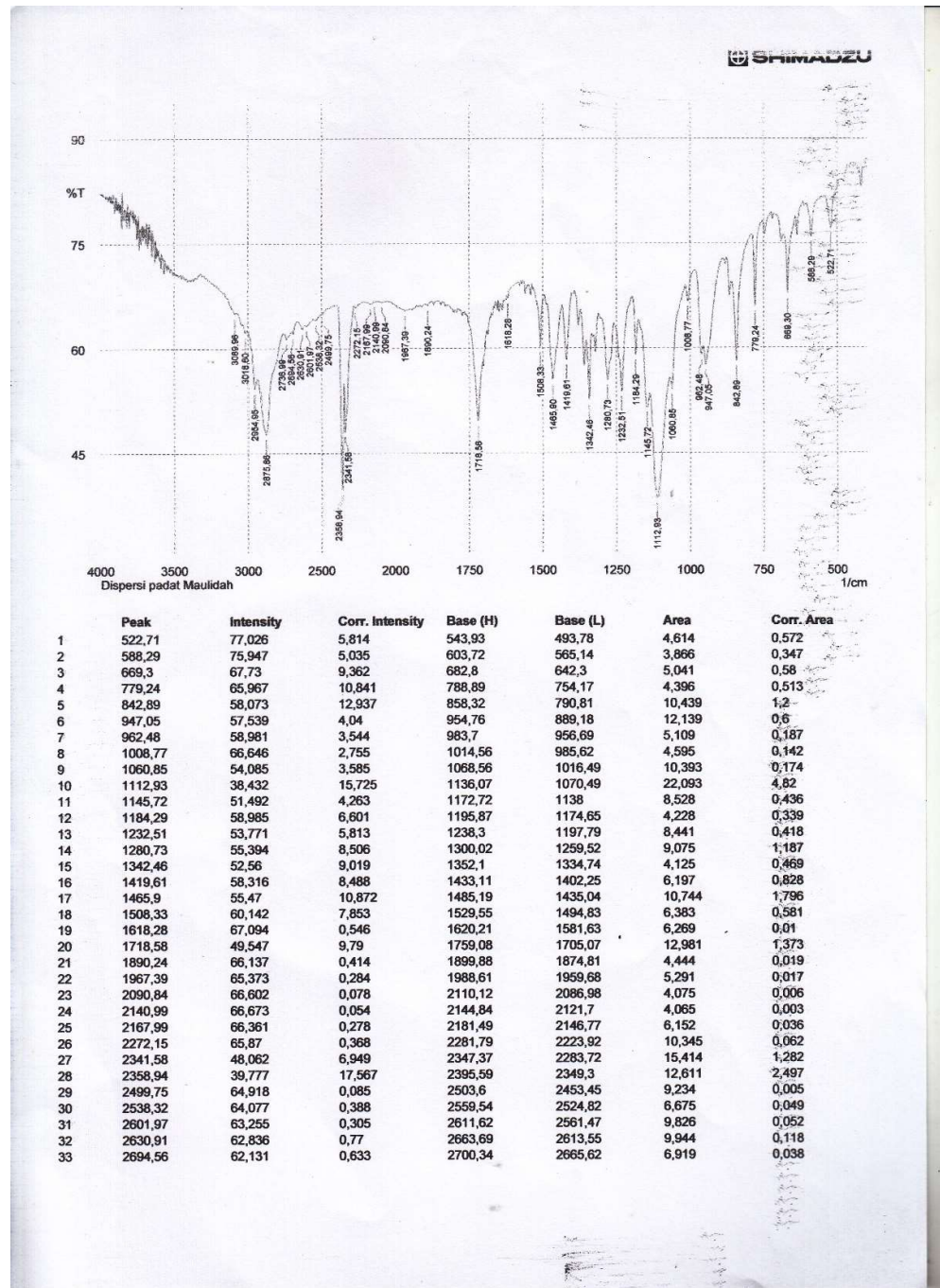


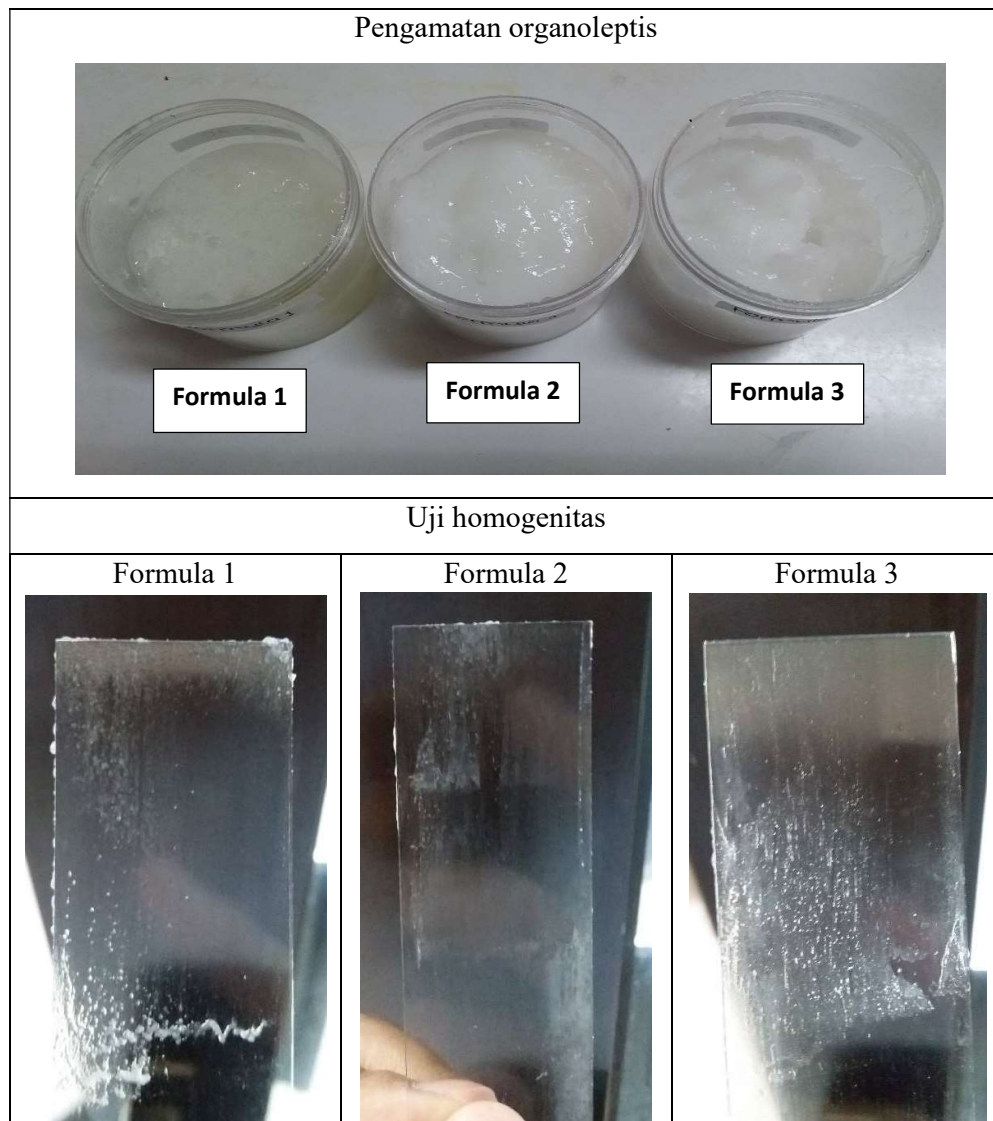
	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	522.71	76,352	19,952	545,85	493,78	2,072	1,231
2	588,29	72,492	23,266	615,29	547,78	3,011	1,757
3	669,3	65,632	24,135	680,87	644,22	3,168	1,685
4	779,24	50,13	41,196	796,6	756,1	3,829	2,257
5	866,04	62,106	19,629	873,75	856,39	2,447	0,94
6	935,48	53,583	5,99	941,26	889,18	8,15	0,41
7	970,19	69,662	11,443	991,41	964,41	2,698	0,45
8	1008,77	65,887	20,865	1031,92	993,34	3,665	1,321
9	1066,64	66,635	5,788	1070,49	1033,85	3,129	0,145
10	1122,57	75,089	10,604	1134,14	1105,21	2,529	0,588
11	1168,86	70,298	2,202	1170,79	1136,07	3,216	0,063
12	1184,29	46,441	29,934	1193,94	1172,72	4,282	1,712
13	1230,58	33,377	44,276	1249,87	1195,87	12,387	6,613
14	1269,16	50,184	27,89	1292,31	1251,8	6,815	2,504
15	1321,24	48,719	12,755	1325,1	1309,67	3,332	0,613
16	1379,1	60,118	20,15	1390,68	1371,39	2,816	0,981
17	1419,61	43,251	32,481	1436,97	1392,61	9,099	4
18	1462,04	56,709	13,513	1481,33	1456,26	3,936	0,612
19	1508,33	53,193	28	1537,27	1494,83	5,739	2,063
20	1618,28	81,9	0,643	1620,21	1581,63	2,987	0,044
21	1720,5	26,896	48,728	1768,72	1658,78	26,391	13,24
22	1801,51	73,554	2,616	1840,09	1789,94	5,899	0,232
23	1890,24	79,113	0,708	1899,88	1878,67	2,113	0,039
24	1909,53	78,414	1,443	1924,96	1901,81	2,329	0,075
25	1934,6	79,203	1,171	1953,89	1926,89	2,6	0,055
26	1959,68	80,784	0,086	1984,75	1955,82	2,651	0,01
27	2013,68	81,181	0,359	2038,76	2000,18	3,451	0,036
28	2079,26	80,872	0,65	2106,27	2056,12	4,549	0,099
29	2171,85	79,891	0,102	2173,78	2144,84	2,714	0,008
30	2198,85	79,907	0,156	2223,92	2193,06	2,976	0,011
31	2279,86	78,544	0,154	2283,72	2225,85	5,766	0,009
32	2345,44	74,07	2,047	2349,3	2285,65	7,403	0,289
33	2360,87	73,768	2,033	2385,95	2351,23	4,441	0,231

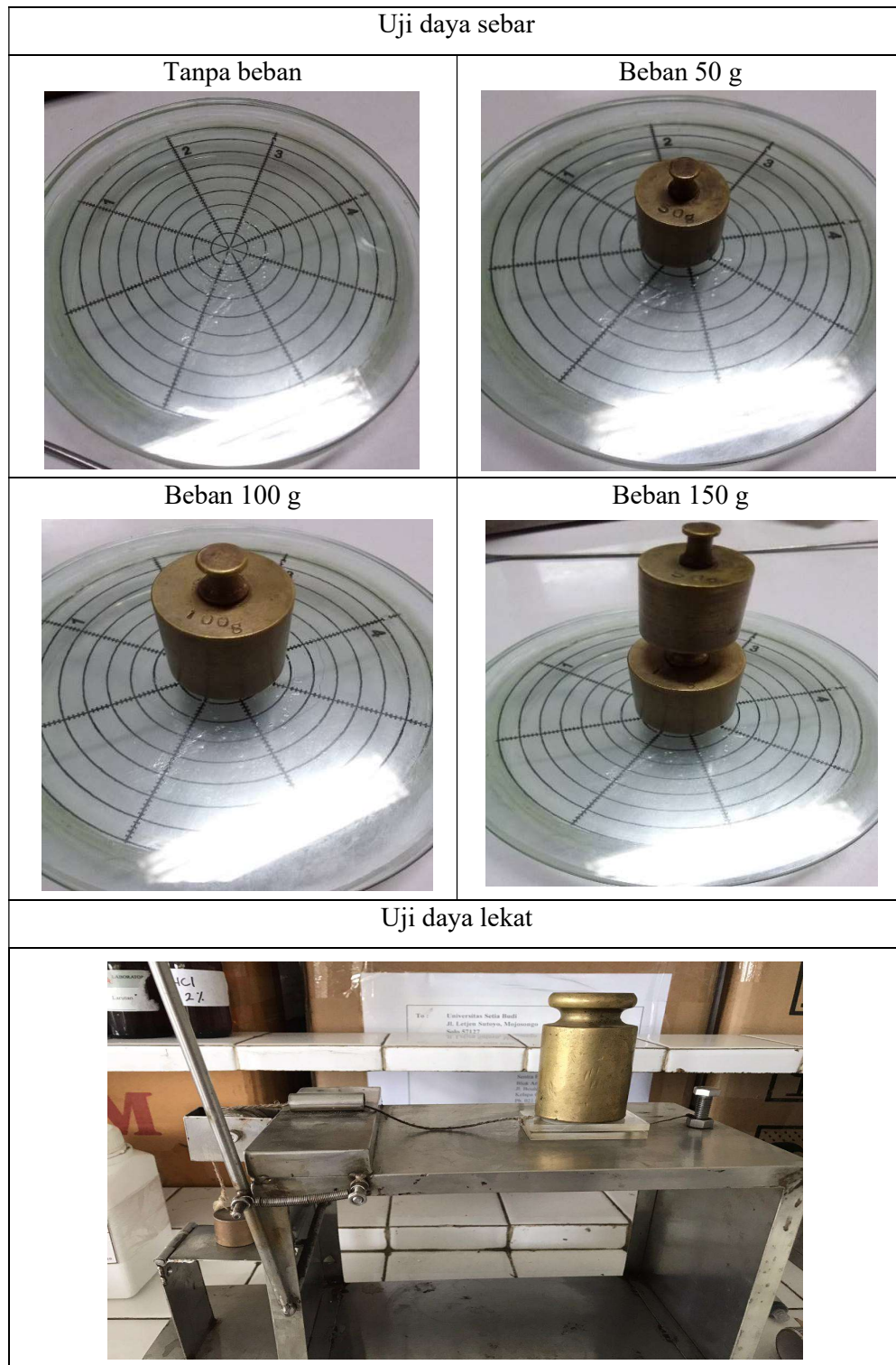
b. PEG 6000



c. Dispersi padat



Lampiran 12. Foto hasil uji mutu fisik gel dispersi padat ibuprofen

Lampiran 13. Foto uji daya sebar dan daya lekat

Lampiran 14. Data uji viskositas, daya lekat, dan daya sebar

Uji	Formula	Replikasi	Hasil	Rata-rata
Viskositas (dpa's)	1	1	250	255
		2	260	
		3	255	
	2	1	210	203,33
		2	200	
		3	200	
	3	1	300	303,33
		2	310	
		3	300	
Daya Lekat (detik)	1	1	1	1,03
		2	1,1	
		3	1	
	2	1	1	1,04
		2	1,05	
		3	1,08	
	3	1	1,3	1,3
		2	1,4	
		3	1,2	

Daya Sebar

Replikasi	Tanpa beban			50 g			100 g			150 g		
1	F1	F2	F3	F1	F2	F3	F1	F2	F3	F1	F2	F3
	2,2	2,3	2,2	2,8	2,9	2,5	3	3,3	2,8	3,6	3,6	2,9
	2,4	2,4	2,3	2,9	3	2,5	3,2	3,5	2,8	3,6	3,7	2,9
	2,4	2,6	2,2	2,9	3	2,4	3,2	3,4	2,6	3,5	3,7	2,9
	2,3	2,7	2,1	2,7	3,1	2,4	3,1	3,6	2,6	3,4	3,8	2,8
Rata-rata	2,3	2,5	2,2	2,8	3	2,6	3,1	3,5	2,7	3,5	3,7	2,9

Replikasi	Tanpa beban			50 g			100 g			150 g		
2	F1	F2	F3	F1	F2	F3	F1	F2	F3	F1	F2	F3
	2,3	2,4	2,3	2,9	2,9	2,7	3,1	3,2	3	3,4	3,3	3,2
	2,5	2,5	2,3	2,8	2,8	2,8	3,3	3,3	3	3,5	3,7	3,3
	2,4	2,6	2,4	2,7	3	2,8	3,3	3,4	3,1	3,6	3,8	3,4
	2,4	2,8	2,3	2,6	3,2	2,7	3,2	3,5	3	3,5	3,8	3,3
Rata-rata	2,4	2,6	2,3	2,8	3	2,7	3,2	3,6	3	3,5	3,7	3,3

Replikasi	Tanpa beban			50 g			100 g			150 g		
3	F1	F2	F3	F1	F2	F3	F1	F2	F3	F1	F2	F3
	2,5	2,5	2,2	2,7	2,9	2,4	3	3,2	2,7	3,3	3,5	3
	2,4	2,4	2,3	2,6	3	2,5	2,9	3,3	2,8	3,2	3,7	3,1
	2,3	2,7	2,2	2,7	3,1	2,6	3,1	3,4	2,9	3,4	3,7	3,2
	2,4	2,6	2,1	2,8	3,1	2,5	3,3	3,4	2,9	3,7	3,8	3,4
Rata-rata	2,4	2,6	2,2	2,7	3	2,5	3,1	3,3	2,8	3,4	3,7	3,2

Lampiran 15. Data analysis *One Way* ANOVA

a. pH

Descriptive Statistics

	N	Mean	Std. Deviation	Minimum	Maximum
pH	9	5.2744	.99008	4.50	6.60

One-Sample Kolmogorov-Smirnov Test

		pH
N		9
Normal Parameters ^{a,b}	Mean	5.2744
	Std. Deviation	.99008
	Absolute	.379
Most Extreme Differences	Positive	.379
	Negative	-.240
Kolmogorov-Smirnov Z		1.137
Asymp. Sig. (2-tailed)		.151

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

pH

Levene Statistic	df1	df2	Sig.
1.639	2	6	.270

ANOVA

pH

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	7.840	2	3.920	14112.640	.000
Within Groups	.002	6	.000		
Total	7.842	8			

b. Daya sebar

Descriptive Statistics

	N	Mean	Std. Deviation	Minimum	Maximum
dayasebar	9	3.433	.2693	2.9	3.7

One-Sample Kolmogorov-Smirnov Test

		Dayasebar
N		9
Normal Parameters ^{a,b}	Mean	3.433
	Std. Deviation	.2693
	Absolute	.172
Most Extreme Differences	Positive	.161
	Negative	-.172
Kolmogorov-Smirnov Z		.517
Asymp. Sig. (2-tailed)		.952

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

dayasebar

Levene Statistic	df1	df2	Sig.
7.800	2	6	.021

ANOVA

dayasebar

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	.487	2	.243	15.643	.004
Within Groups	.093	6	.016		
Total	.580	8			

c. Daya lekat

Descriptive Statistics

	N	Mean	Std. Deviation	Minimum	Maximum
dayalekat	9	1.1256	.14449	1.00	1.40

One-Sample Kolmogorov-Smirnov Test

		Dayalekat
N		9
Normal Parameters ^{a,b}	Mean	1.1256
	Std. Deviation	.14449
	Absolute	.237
Most Extreme Differences	Positive	.237
	Negative	-.192
Kolmogorov-Smirnov Z		.711
Asymp. Sig. (2-tailed)		.694

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

dayalekat

Levene Statistic	df1	df2	Sig.
.794	2	6	.494

ANOVA

dayalekat

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	.137	2	.069	13.739	.006
Within Groups	.030	6	.005		
Total	.167	8			

d. Viskositas**Descriptive Statistics**

	N	Mean	Std. Deviation	Minimum	Maximum
viskositas	9	253.89	43.573	200	310

One-Sample Kolmogorov-Smirnov Test

		viskositas
N		9
Normal Parameters ^{a,b}	Mean	253.89
	Std. Deviation	43.573
	Absolute	.188
Most Extreme Differences	Positive	.176
	Negative	-.188
Kolmogorov-Smirnov Z		.565
Asymp. Sig. (2-tailed)		.907

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

viskositas

Levene Statistic	df1	df2	Sig.
.235	2	6	.797

ANOVA

viskositas

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	15005.556	2	7502.778	245.545	.000
Within Groups	183.333	6	30.556		
Total	15188.889	8			

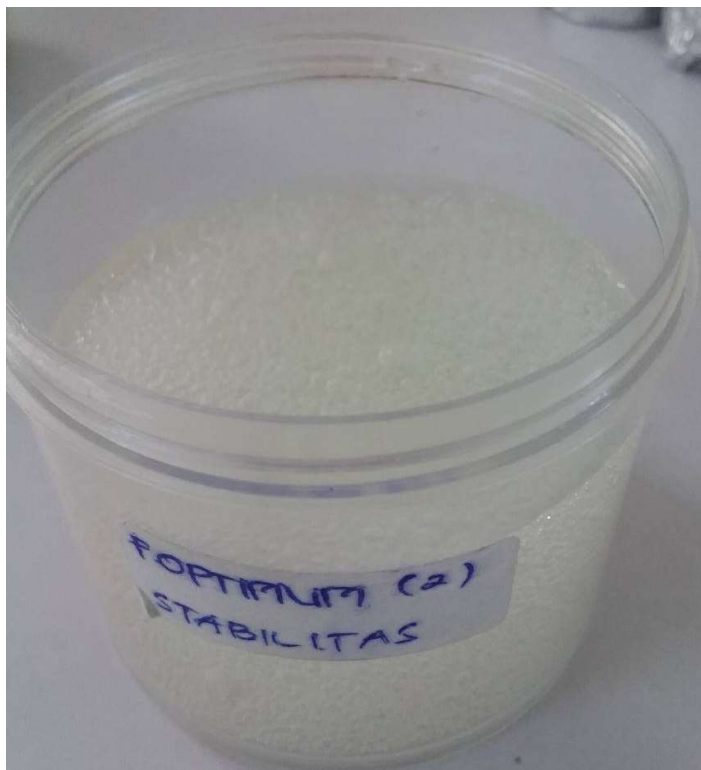
Lampiran 16. Penentuan formula optimum

Select	Std	Run ▼	Component 1 A:Na-CMC %	Component 2 B:Karbopol 9 %	Response 1 Daya sebar cm	Response 2 Daya lekat detik	Response 3 Viskositas dpa's
2		1	0	3	3.5	1	250
	5	2	0	3	3.5	1.1	260
	4	3	0	3	3.4	1	255
	3	4	1.5	1.5	3.7	1	210
	7	5	1.5	1.5	3.7	1.05	200
	6	6	1.5	1.5	3.7	1.08	200
	1	7	3	0	2.9	1.3	300
	8	8	3	0	3.3	1.4	310
	9	9	3	0	3.2	1.2	300

C. Viskositas

Transform Fit Summary Model ANOVA						
Use your mouse to right click on individual cells for definitions.						
Response 3 Viskositas						
ANOVA for Quadratic Mixture model						
*** Mixture Component Coding is L_Pseudo. ***						
Analysis of variance table [Partial sum of squares - Type III]						
	Sum of		Mean	F	p-value	
Source	Squares	df	Square	Value	Prob > F	
Model	15005.56	2	7502.78	245.55	< 0.0001	significant
1 Linear Mixtu	3504.17	1	3504.17	114.68	< 0.0001	
AB	11501.39	1	11501.39	376.41	< 0.0001	
Pure Error	183.33	6	30.56			
Cor Total	15188.89	8				
1 Inference for linear mixtures uses Type I sums of squares.						
The Model F-value of 245.55 implies the model is significant. There is only a 0.01% chance that an F-value this large could occur due to noise.						
Std. Dev.	5.53	R-Squared	0.9879			
Mean	253.89	Adj R-Squared	0.9839			
C.V. %	2.18	Pred R-Square	0.9728			
PRESS	412.50	Adeq Precisor	31.334			
-2 Log Likeliho	52.67	BIC	57.06			
		AICc	58.67			
The "Pred R-Squared" of 0.9728 is in reasonable agreement with the "Adj R-Squared" of 0.9839; i.e. the difference is less than 0.2.						
"Adeq Precision" measures the signal to noise ratio. A ratio greater than 4 is desirable. Your ratio of 31.334 indicates an adequate signal. This model can be used to navigate the design space.						
	Coefficient		Standard	95% CI	95% CI	
Component	Estimate	df	Error	Low	High	VIF
A-Na-CMC	303.33	1	3.19	295.52	311.14	1.25
B-Karbopol 94	255.00	1	3.19	247.19	262.81	1.25
AB	-303.33	1	15.63	-341.59	-265.08	1.50

Lampiran 18. Foto hasil gel dispersi padat dari formula optimum



Lampiran 19. Data uji viskositas, daya lekat, dan daya sebar formula optimum

Uji	Replikasi	Hasil	Rata-rata
Viskositas (dpa's)	1	300	301,67
	2	305	
	3	300	
Daya Lekat (detik)	1	1,46	1,316
	2	1,35	
	3	1,14	

Daya sebar

Beban	Replikasi 1 (cm)	Replikasi 2 (cm)	Replikasi 3 (cm)	Rata-rata
Tanpa beban	2,3	2,4	2,4	2,4
50 g	2,5	2,5	2,7	2,6
100 g	2,9	2,8	3	2,9
150 g	3,2	3,3	3,2	3,2

Lampiran 20. Data analisis *one sample t-test*

a. Daya sebar

One-Sample Statistics				
	N	Mean	Std. Deviation	Std. Error Mean
dayasebar	3	3.233	.0577	.0333

One-Sample Test						
	Test Value = 3.2					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
dayasebar	1.000	2	.423	.0333	-.110	.177

b. Daya lekat

One-Sample Statistics				
	N	Mean	Std. Deviation	Std. Error Mean
dayalekat	3	1.3167	.16258	.09387

One-Sample Test						
	Test Value = 1.3					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
dayalekat	.178	2	.875	.01667	-.3872	.4205

c. Viskositas

One-Sample Statistics				
	N	Mean	Std. Deviation	Std. Error Mean
viskositas	3	301.67	2.887	1.667

One-Sample Test						
	Test Value = 303.33					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
viskositas	-.998	2	.423	-1.663	-8.83	5.51

Lampiran 21. Foto hasil stabilitas dari formula optimum