

BAB V

KESIMPULAN DAN SARAN

A. Kesimpulan

Berdasarkan hasil penelitian yang telah dilakukan dapat diperoleh kesimpulan, bahwa :

Pertama, *retinyl palmitate* dapat dibuat sediaan serum pada semua formula sesuai dengan karakterisasi sediaan.

Kedua, serum *retinyl palmitate* dengan berbagai variasi konsentrasi sodium glukonat tidak stabil pada pengamatan visual selama proses penyimpanan.

B. Saran

Penelitian ini masih banyak kekurangan, maka perlu dilakukan penelitian lebih lanjut mengenai :

Pertama, perlu dilakukan uji difusi sel menggunakan membran biologis untuk mengetahui serum *retinyl palmitate* yang dapat berpenetrasi ke dalam kulit

Kedua, perlu dilakukan uji DPPH sediaan serum untuk mengetahui serum *retinyl palmitate* mempunyai aktivitas antioksidan.

Ketiga, pengembangan sediaan serum *retinyl palmitate* dalam bentuk nanopartikel dan enkapsulasi.

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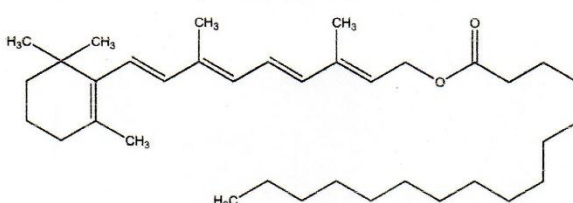
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Lampiran 1. Sertifikat analisis retinyl palmitate

Certificate of Analysis	
ISO GUIDE 34 ANAB Cert# AR-1470	RETINYL PALMITATE (Vitamin A Palmitate) CERTIFIED REFERENCE MATERIAL
ISO/IEC 17025 ANAB Cert# AT-1467	
CERTIFIED PURITY: 99.7%, $U_{\text{cm}} = \pm 0.1\%$ $k = 2$ (Mass Balance/as is basis)	
CERTIFIED ACTIVITY: 1756000IU/g, $U_{\text{cm}} = \pm 8297\text{IU/g}$ $k = 2.06$ (UV Assay/as is basis)	
NOMINAL PACKAGE SIZE: 1g	
CATALOG #: PHR1235	LOT #: P500235
CERTIFICATE VERSION: 500235.4	ISSUE DATE: 19 May 2016
<i>Note: Certificates may be updated due to Pharmacopeial Lot changes or the availability of new data. Check our website at: www.sigma-aldrich.com for the most current version.</i>	
CRM EXPIRATION: 12 Months from Receipt (Proper Storage and Handling Required).	
RECEIPT DATE: _____ Note: this space is provided for convenience only and its use is not required.	
STORAGE: Store in a Refrigerator/Protect from Light, keep container tightly closed. Discard unused portions.	
CHEMICAL FORMULA: $\text{C}_{36}\text{H}_{60}\text{O}_2$	MW: 524.9
SIGMA-ALDRICH®	
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PHYSICAL DESCRIPTION: Pale yellow viscous oil in amber ampule **CAS #:** 79-81-2

HAZARDS: Read Safety Data Sheet before using. All chemical reference materials should be considered potentially hazardous and should be used only by qualified laboratory personnel.

INSTRUCTIONS FOR USE: Do not dry, use on the as is basis. The internal pressure of the container may be slightly different from the atmospheric pressure at the user's location. Open slowly and carefully to avoid dispersion of the material. This material is intended for R&D use only. Not for drug, household or other uses.

TRACEABILITY ASSAY

Comparative assay demonstrates direct traceability to Pharmacopeial Standards
Specification: None

METHOD: HPLC (ref.: Vitamin A Tablets, USP34)

Column: Supelcosil LC-NH₂, 4.6 x 150mm, 3µm

Mobile Phase: Hexane, Ethyl Acetate (98:2)

Flow Rate: 1mL/min

Column Temperature: 30°C

Injection: 2.5µL

Detector: 325nm

ASSAY vs. USP REFERENCE STANDARD (as is basis)

ASSAY VALUE

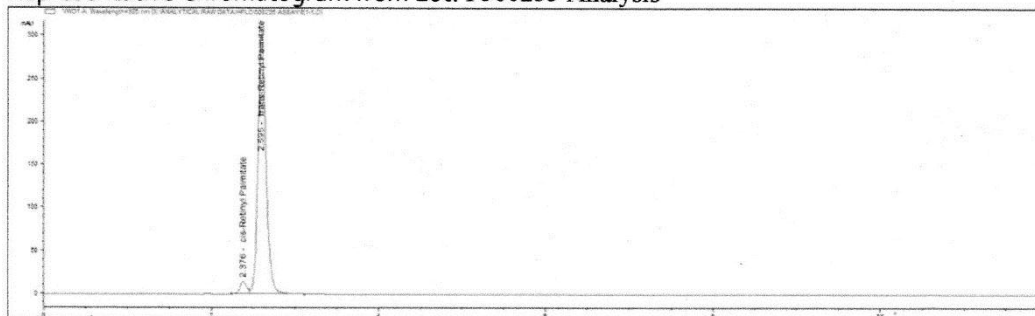
98.7%

vs. USP LOT

F0J106

Labeled Content = 0.957mg/mg

Representative Chromatogram from Lot: P500235 Analysis



ACTIVITY DETERMINATION

METHOD: UV (ref.: Vitamin A Concentrate, EP7)

Solvent: Dissolve in pentane, dilute with isopropanol to ~15IU/g

Cell Pathlength: 1cm

Wavelength: 326nm

CERTIFIED ACTIVITY

1756000IU/g, $U_{cm} = \pm 8297IU/g$ $k = 2.06$
(as is basis)

PURITY DETERMINATION BY MASS BALANCE

CHROMATOGRAPHIC IMPURITY ANALYSIS

METHOD: HPLC

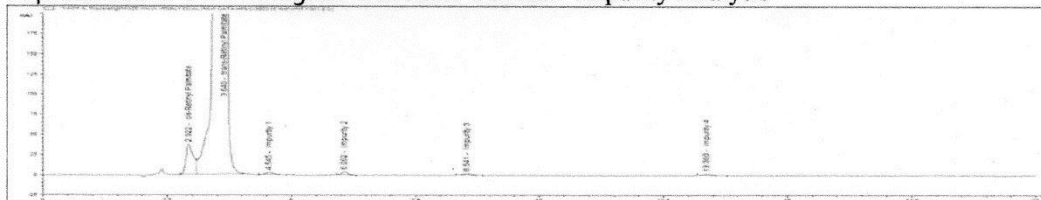
See Assay

Impurities Detected:

Impurity 1:	0.1%
Impurity 2:	0.1%
Impurity 3:	0.07%
Impurity 4:	0.07%

Total Impurities: **0.3%**

Representative Chromatogram from Lot: P500235 Impurity Analysis



RESIDUAL SOLVENTS

Method: GC-MS Headspace (ref.: Residual Solvents <467>, USP34)

Column: DB-1301

Carrier gas: He

Flow: 1.2mL/min

Split Ratio: 1:5

Injection/Temperature: 1µl/250°C

Temperature Program: 40°C for 20min, 10°C/min to 240°C, hold 20min

Solvents Detected: None

WATER DETERMINATION

Method: Karl Fisher titration

Mean of three measurements, Water Content = 0.002%

RESIDUE ANALYSIS

Method: Sulfated Ash

Sample Size: ~1g

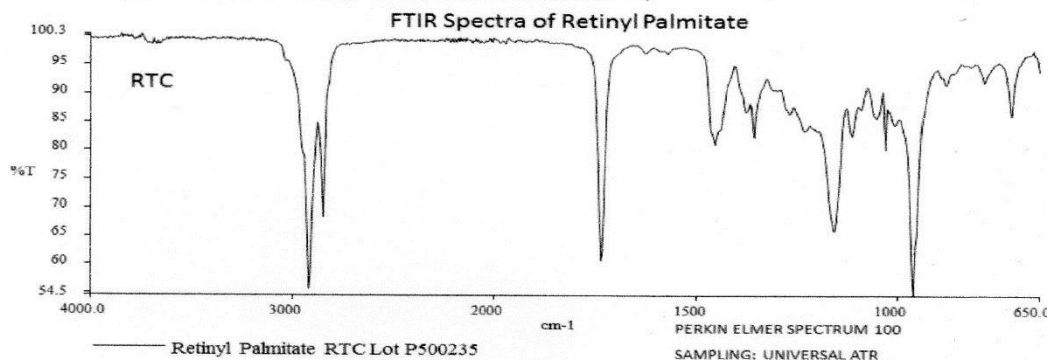
Mean of three measurements, Residue = 0.02%

CERTIFIED PURITY BY MASS BALANCE [100% - Impurities (normalized)]

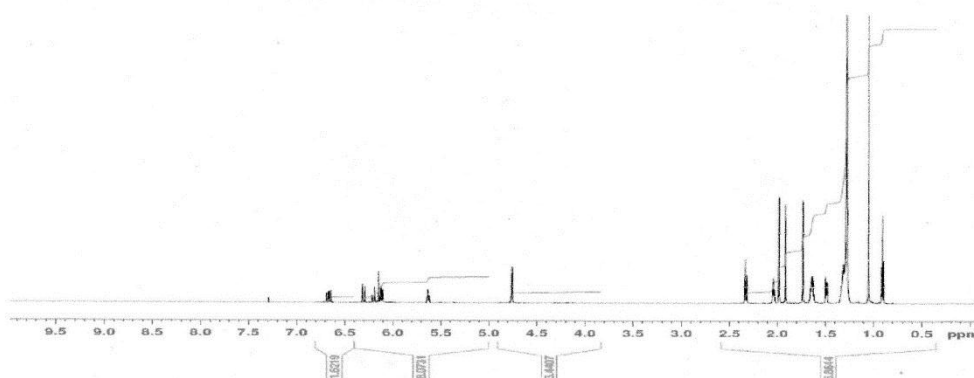
99.7% $U_{\text{crm}} = \pm 0.1\%$, $k = 2$
(as is basis)

IDENTIFICATION TESTS

INFRARED SPECTROPHOTOMETRY (This IR is for informational purposes only and shows good correlation to spectrum found in literature.)



¹H NMR (Data provided by an external laboratory; not in scope of accreditation)
P500235 Vitamin A Palmitate in CDCl₃



Consistent with structure

ELEMENTAL ANALYSIS (Data provided by an external laboratory; not in scope of accreditation)

Exeter Analytical 440 Elemental Analyzer

Combustion method

%	Theoretical	Result 1	Result 2	Mean
C	82.38	82.36	80.47	81.42
H	11.52	11.49	11.15	11.32

HOMOGENEITY ASSESSMENT

Homogeneity was assessed in accordance with ISO Guide 35. Completed units were sampled using a random stratified sampling protocol. The results of chemical analysis were then compared by Single Factor Analysis of Variance (ANOVA). The uncertainty due to homogeneity was derived from the ANOVA. Heterogeneity was not detected under the conditions of the ANOVA.

Analytical Method: HPLC

Sample size: ~30mg

UNCERTAINTY STATEMENT

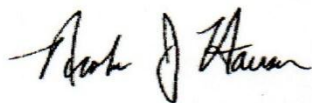
Uncertainty values in this document are expressed as Expanded Uncertainty (U_{cm}) corresponding to the 95% confidence interval. U_{cm} is derived from the combined standard uncertainty multiplied by the coverage factor k , which is obtained from a t -distribution and degrees of freedom. The components of combined standard uncertainty include the uncertainties due to characterization, homogeneity, long term stability, and short term stability (transport). The components due to stability are generally considered to be negligible unless otherwise indicated by stability studies.

STABILITY ASSESSMENT

Significance of the stability assessment will be demonstrated if the analytical result of the study and the range of values represented by the Expanded Uncertainty do not overlap the result of the original assay and the range of its values represented by the Expanded Uncertainty. The method employed will usually be the same method used to characterize the assay value in the initial evaluation.

Long Term Stability Evaluation - An assessment, or re-test, versus a Compendial Reference Standard may be scheduled, within the 3 year anniversary date of a release of a Secondary Standard. The re-test interval will be determined on a case-by-case basis.

Short Term Stability Study - It is useful to assess stability under reasonably anticipated, short term transport conditions by simulating exposure of the product to humidity and temperature stress. This type of study is conducted under controlled conditions of elevated temperature and humidity.



Operations Manager

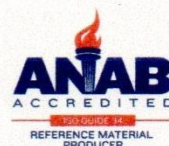


QA Supervisor

APPENDIX

Original Release Date: 31 May 2012
Stability Test Date: 18 May 2015

Manufactured and certified by Sigma-Aldrich RTC, Inc.
2931 Soldier Springs Rd, Laramie WY, USA 82070
(Phone): 1-307-742-5452 (Fax): 1-855-831-9212
email: RTCTechGroup@sial.com

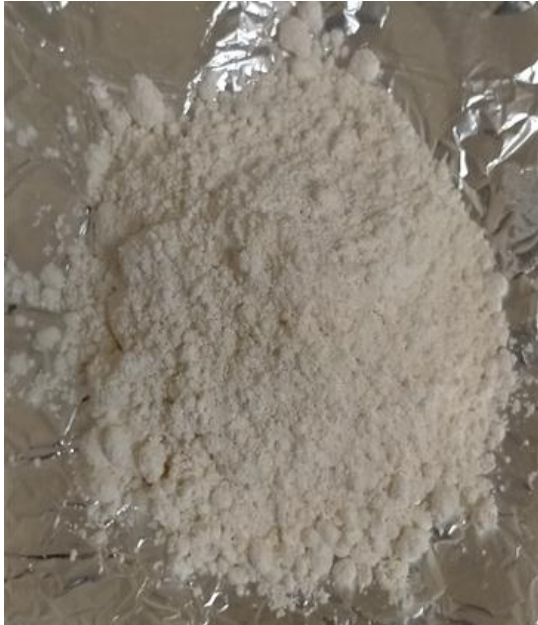
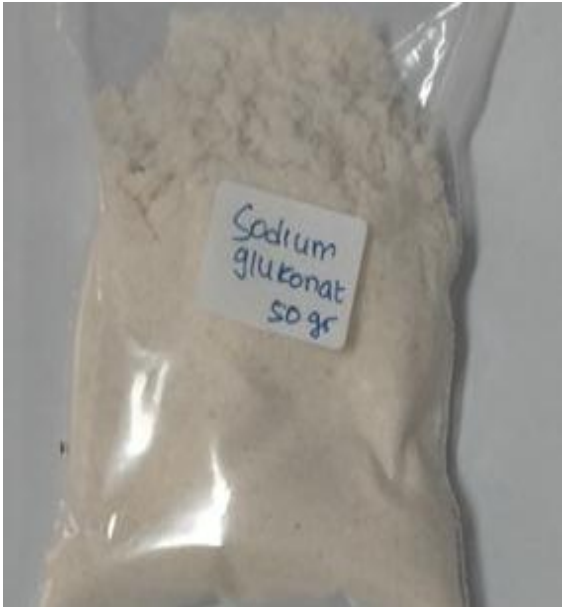
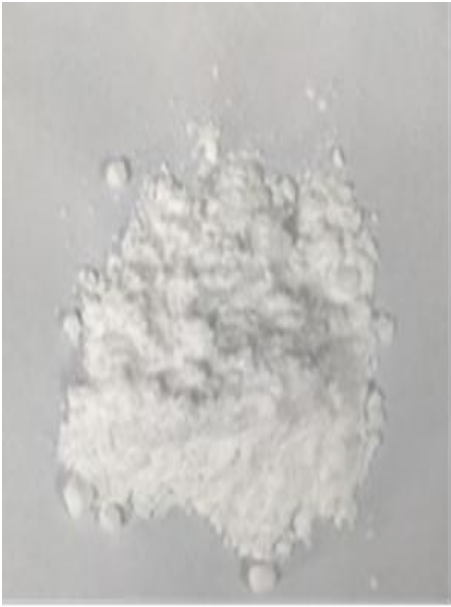


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Lampiran 2. Alat dalam praktikum

Alat praktikum	
 Neraca analitik	 Alat setirer
 Spektrofotometri Uv-Vis	 Pipet mikro
 Pipet volume	 Viskositas

Lampiran 3. Bahan serum *retinyl palmitate*

Bahan serum	
	
<i>Retinyl palmitate</i>	Gliserin
	
Sodium glukonat	<i>Hyaluronic acid</i>



Asam glikolat



Optiphen

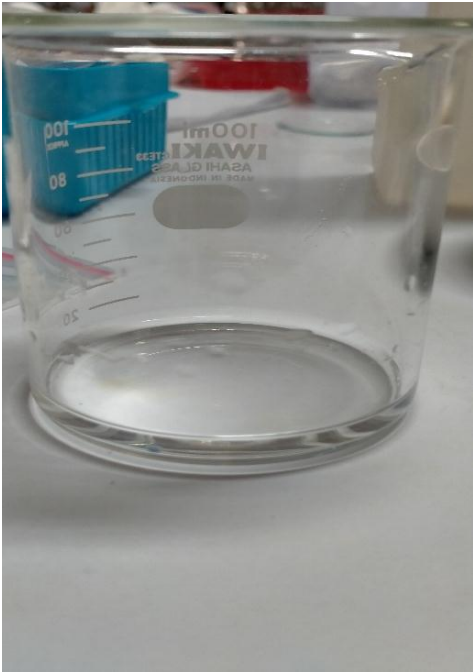
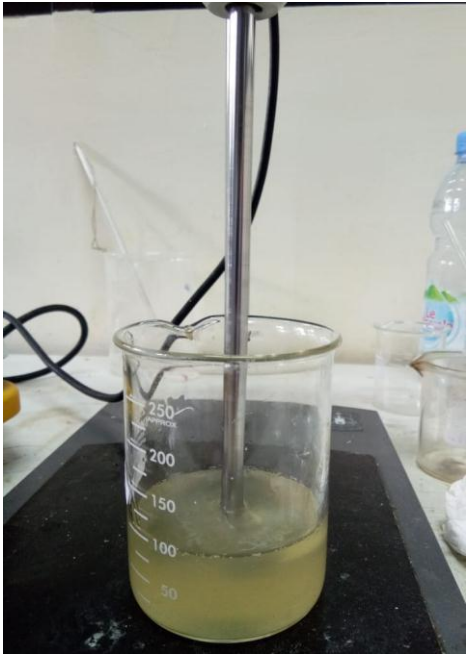


Xanthan gum



Aquadest

Lampiran 4. Pembuatan serum *retinyl palmitate*

Pengembangan <i>hyaluronic acid</i>	Pengembangan xanthan gum
	
Pelarutan sodium glukonat dengan aquadest	Pencampuran menggunakan alat setirer
	

Three glass vials containing yellow serum samples, labeled F1, F2, and F3. Each vial has a label that reads "Serum R. Palmitat".

Lampiran 6. Data uji viskositas dan pH

a. Viskositas

Formula	Replikasi	Hari ke-1	Hari ke-7	Hari ke-14	Hari ke-21
1	1	1000	750	650	600
	2	1100	850	700	550
	3	1000	800	750	650
	Rata-rata	1033	800	700	600
	SD	57,74	50	50	50
2	1	1000	750	650	600
	2	1000	800	600	500
	3	1100	750	550	550
	Rata-rata	1033	767	600	550
	SD	57,74	28,87	50	50
3	1	1000	700	550	450
	2	900	650	500	500
	3	800	600	500	400
	Rata-rata	900	650	517	450
	SD	100	50	28,87	50

b. pH

Formula	Replikasi	Hari ke-1	Hari ke-7	Hari ke-14	Hari ke-21
1	1	5,96	5,97	6,03	6,1
	2	5,96	5,98	6,05	6,12
	3	5,99	5,97	6,04	6,1
	Rata-rata	5,97	5,97	6,04	6,11
	SD	0,02	0,01	0,01	0,01
2	1	6,16	5,99	6,23	6,28
	2	6,14	6,07	6,2	6,22
	3	6,17	6,09	6,26	6,24
	Rata-rata	6,16	6,05	6,23	6,25
	SD	0,02	0,05	0,03	0,03
3	1	6,24	6,36	6,32	6,33
	2	6,27	6,31	6,28	6,36
	3	6,29	6,22	6,24	6,35
	Rata-rata	6,27	6,3	6,28	6,35
	SD	0,03	0,07	0,04	0,02

Lampiran 7. Data analisis *one way* ANOVA

a. Viskositas

Descriptive Statistics

	N	Mean	Std. Deviation	Minimum	Maximum
Hari	36	2,5000	1,13389	1,00	4,00
Formula	36	2,0000	,82808	1,00	3,00
Viskositas	36	713,8889	196,98119	400,00	1100,00

One-Sample Kolmogorov-Smirnov Test

		Hari	Formula	Viskositas
N		36	36	36
Normal Parameters ^{a,b}	Mean	2,5000	2,0000	713,8889
	Std. Deviation	1,13389	,82808	196,98119
	Absolute	,170	,220	,127
Most Extreme Differences	Positive	,170	,220	,127
	Negative	-,170	-,220	-,121
Kolmogorov-Smirnov Z		1,022	1,318	,763
Asymp. Sig. (2-tailed)		,247	,062	,605

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

Viskositas

Levene Statistic	df1	df2	Sig.
,289	2	33	,751

ANOVA

Viskositas

	Sum of Squares	Df	Mean Square	F	Sig.
Between Groups	146805,556	2	73402,778	2,000	,151
Within Groups	1211250,000	33	36704,545		
Total	1358055,556	35			

b. pH**Descriptive Statistics**

	N	Mean	Std. Deviation	Minimum	Maximum
Hari	36	2.5000	1.13389	1.00	4.00
Formula	36	2.0000	.82808	1.00	3.00
pH	36	6.1628	.13199	5.96	6.36

One-Sample Kolmogorov-Smirnov Test

		Hari	Formula	pH
N		36	36	36
Normal Parameters ^{a,b}	Mean	2.5000	2.0000	6.1628
	Std. Deviation	1.13389	.82808	.13199
	Absolute	.170	.220	.140
Most Extreme Differences	Positive	.170	.220	.099
	Negative	-.170	-.220	-.140
Kolmogorov-Smirnov Z		1.022	1.318	.839
Asymp. Sig. (2-tailed)		.247	.062	.481

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

pH

Levene Statistic	df1	df2	Sig.
1.738	2	33	.192

ANOVA

pH

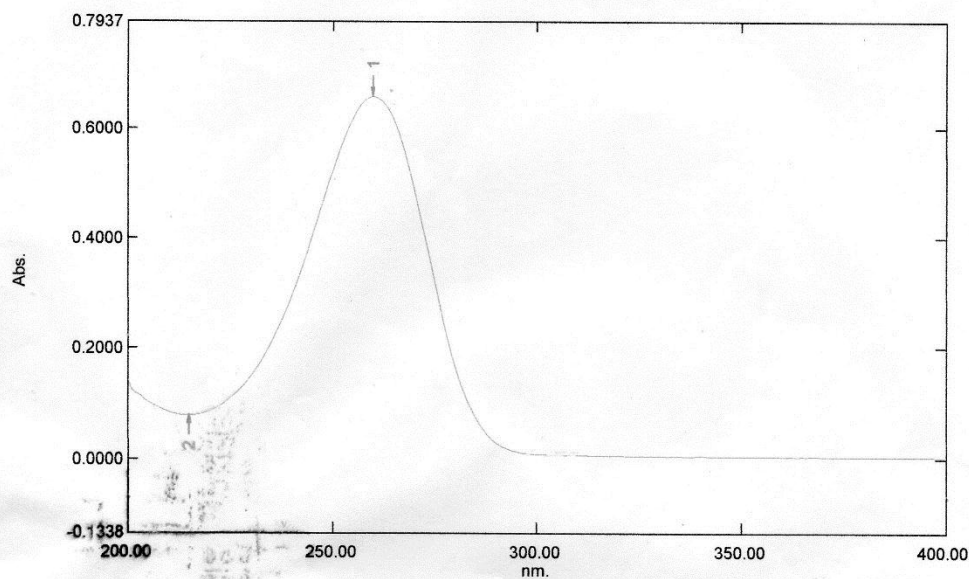
	Sum of Squares	Df	Mean Square	F	Sig.
Between Groups	.463	2	.232	52.164	.000
Within Groups	.147	33	.004		
Total	.610	35			

Lampiran 8. Hasil penentuan panjang gelombang maksimum *retinyl palmitate*

Spectrum Peak Pick Report

03/12/2020 02:33:36 PM

Data Set: File_200312_142730 - RawData



[Measurement Properties]
 Wavelength Range (nm): 200.00 to 400.00
 Scan Speed: Medium
 Sampling Interval: 1.0
 Auto Sampling Interval: Disabled
 Scan Mode: Single

No.	P/V	Wavelength	Abs.	Description
1	①	260.00	0.6576	
2	②	215.00	0.0782	

[Instrument Properties]
 Instrument Type: UV-1800 Series
 Measuring Mode: Absorbance
 Slit Width: 1.0 nm
 Light Source Change Wavelength: 340.0 nm
 S/R Exchange: Normal

[Attachment Properties]
 Attachment: None

[Operation]
 Threshold: 0.0010000
 Points: 4
 InterPolate: Disabled
 Average: Disabled

[Sample Preparation Properties]
 Weight:
 Volume:
 Dilution:
 Path Length:
 Additional Information:

Lampiran 9. Hasil penetapan *operating time***Kinetics Data Print Report**

03/13/2020 09:07:44 AM

Time (Minute)	RawData ...
0.000	0.663
1.000	0.662
2.000	0.663
3.000	0.663
4.000	0.662
5.000	0.663
6.000	0.662
7.000	0.662
8.000	0.662
9.000	0.663
10.000	0.663
11.000	0.662
12.000	0.663
13.000	0.663
14.000	0.663
15.000	0.662
16.000	0.662
17.000	0.662
18.000	0.661
19.000	0.661
20.000	0.662
21.000	0.662
22.000	0.661
23.000	0.662
24.000	0.661
25.000	0.661
26.000	0.662
27.000	0.661
28.000	0.661
29.000	0.661
30.000	0.661

Lampiran 10. Pembuatan kurva kalibrasi dan validasi metode analisis

a. Kurva kalibrasi (Linieritas)

- Membuat larutan induk sebesar 103 ppm dengan menimbang 10,3 mg *retinyl palmitate* ditambahkan etanol *p.a* sampai 100 ml.

Penimbangan *retinyl palmitate* :

Kertas kosong + isi : 0,2864 g

Kertas + sisa : 0,2761 g _

Serbuk *retinyl palmitate* : 0,0103g

- Perhitungan seri konsentrasi

- $V_1 \times C_1 = V_2 \times C_2$
 $2,1 \text{ mL} \times 103 \text{ ppm} = 10 \text{ mL} \times \text{ppm}$
 $C_2 = 21,63 \text{ ppm}$
- $V_1 \times C_1 = V_2 \times C_2$
 $3,1 \text{ mL} \times 103 \text{ ppm} = 10 \text{ mL} \times \text{ppm}$
 $C_2 = 31,93 \text{ ppm}$
- $V_1 \times C_1 = V_2 \times C_2$
 $4,1 \text{ mL} \times 103 \text{ ppm} = 10 \text{ mL} \times C_2$
 $C_2 = 42,23 \text{ ppm}$
- $V_1 \times C_1 = V_2 \times C_2$
 $5,1 \text{ mL} \times 103 \text{ ppm} = 10 \text{ mL} \times C_2$
 $C_2 = 52,53 \text{ ppm}$
- $V_1 \times C_1 = V_2 \times C_2$
 $6,1 \text{ mL} \times 103 \text{ ppm} = 10 \text{ mL} \times \text{ppm}$
 $C_2 = 62,83 \text{ ppm}$

Kadar (ppm)	Absorbansi
21,63	0,234
31,93	0,374
42,23	0,492
52,53	0,621
62,83	0,761

Persamaan regresi linier antara konsentrasi (ppm) dan serapan diperoleh nilai :

$$a = -0,0314$$

$$b = 0,0125$$

$$r = 0,9993$$

$$y = a + bx$$

$$y = -0,0314 + 0,0125x$$

keterangan :

x = konsentrasi ($\mu\text{g/ml}$)

y = serapan

Hasil linearitas diperoleh $r = 0,9993$, sehingga dapat disimpulkan bahwa data linear baik mendekati 1.

b. Akurasi

Konsentrasi	Replikasi	Abs	Konsentrasi (ppm)	ppm sebenarnya	%	Rata-rata	Recovery
80%	1	0,489	41,5513	42,23	98%	98,08%	99,53%
	2	0,487	41,3916	42,23	98%		
	3	0,486	41,3118	42,23	98%		
100%	1	0,620	52,0110	52,53	99%	99,57%	
	2	0,626	52,4901	52,53	100%		
	3	0,625	52,4102	52,53	100%		
120%	1	0,766	63,6684	62,83	101%	100,95%	
	2	0,763	63,4288	62,83	101%		
	3	0,760	62,1893	62,83	101%		

$$\text{Rumus konsentrasi ppm} : \frac{[Abs-(a)]}{b}$$

❖ Konsentrasi 80 %

$$\text{➤ Konsentrasi ppm} : \frac{[0,489-(-0,0314)]}{0,0125} = 41,5513 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,487 - (-0,0314)]}{0,0125} = 41,3916 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,486 - (-0,0314)]}{0,0125} = 41,3118 \text{ ppm}$$

❖ **Konsentrasi 100 %**

$$\text{➤ Konsentrasi ppm : } \frac{[0,620 - (-0,0314)]}{0,0125} = 52,0110 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,626 - (-0,0314)]}{0,0125} = 52,4901 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,625 - (-0,0314)]}{0,0125} = 52,4102 \text{ ppm}$$

❖ **Konsentrasi 120%**

$$\text{➤ Konsentrasi ppm : } \frac{[0,766 - (-0,0314)]}{0,0125} = 63,6684 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,763 - (-0,0314)]}{0,0125} = 63,4288 \text{ ppm}$$

$$\text{➤ Konsentrasi ppm : } \frac{[0,760 - (-0,0314)]}{0,0125} = 63,1893 \text{ ppm}$$

$$\text{Rumus recovery : } \% \text{ Recovery} = \frac{\text{konsentrasi terukur}}{\text{konsentrasi sebenarnya}} \times 100\%$$

➤ **Konsentrasi 42,23 ppm**

$$\bullet \% \text{ Recovery} = \frac{41,5513}{42,23} \times 100\% = 98\%$$

$$\bullet \% \text{ Recovery} = \frac{41,3916}{42,23} \times 100\% = 98\%$$

$$\bullet \% \text{ Recovery} = \frac{41,3118}{42,23} \times 100\% = 98\%$$

➤ **Konsentrasi 52,53 ppm**

$$\bullet \% \text{ Recovery} = \frac{52,0110}{52,53} \times 100\% = 99\%$$

$$\bullet \% \text{ Recovery} = \frac{52,4901}{52,53} \times 100\% = 100\%$$

$$\bullet \% \text{ Recovery} = \frac{51,4102}{52,53} \times 100\% = 100\%$$

➤ **Konsentrasi 62,83 ppm**

$$\bullet \% \text{ Recovery} = \frac{63,6684}{62,83} \times 100\% = 101\%$$

$$\bullet \% \text{ Recovery} = \frac{63,4288}{63,83} \times 100\% = 101\%$$

$$\bullet \% \text{ Recovery} = \frac{63,1893}{63,83} \times 100\% = 101\%$$

Nilai rata- rata % Recovery diatas adalah 99,53%, hal ini menunjukkan nilai persen perolehan kembali yang baik.

c. Presisi

Konsentrasi	Absorbansi	Konsentrasi (ppm)
21,63	0,254	22,7878
21,63	0,258	23,1071
21,63	0,267	23,8257
21,63	0,264	23,5862
21,63	0,256	22,9474
21,63	0,268	23,9056
21,63	0,258	23,1071
21,63	0,253	22,7079
21,63	0,269	23,9854
21,63	0,268	23,9056
	RATA-RATA	23,3866
	SD	0,505334
	CV	0,021608%

Rumus konsentrasi ppm : $\frac{[Abs-(a)]}{b}$

$$1. \text{Konsentrasi ppm} : \frac{[0,254-(-0,0314)]}{0,0125} = 22,7878 \text{ ppm}$$

$$2. \text{Konsentrasi ppm} : \frac{[0,258-(-0,0314)]}{0,0125} = 23,1071 \text{ ppm}$$

$$3. \text{Konsentrasi ppm} : \frac{[0,267-(-0,0314)]}{0,0125} = 23,8257 \text{ ppm}$$

$$4. \text{Konsentrasi ppm} : \frac{[0,264-(-0,0314)]}{0,0125} = 23,5862 \text{ ppm}$$

$$5. \text{Konsentrasi ppm} : \frac{[0,256-(-0,0314)]}{0,0125} = 22,9474 \text{ ppm}$$

$$6. \text{Konsentrasi ppm} : \frac{[0,268-(-0,0314)]}{0,0125} = 23,9056 \text{ ppm}$$

$$7. \text{Konsentrasi ppm} : \frac{[0,258-(-0,0314)]}{0,0125} = 23,1071 \text{ ppm}$$

$$8. \text{Konsentrasi ppm} : \frac{[0,253-(-0,0314)]}{0,0125} = 22,7079 \text{ ppm}$$

$$9. \text{Konsentrasi ppm} : \frac{[0,269-(-0,0314)]}{0,0125} = 23,9854 \text{ ppm}$$

$$10. \text{Konsentrasi ppm} : \frac{[0,268-(-0,0314)]}{0,0125} = 23,9056 \text{ ppm}$$

Nilai CV dilihat dari data diatas adalah 0,021608%, hasil ini sesuai dengan pernyataan bahwa syarat presisi adalah kurang dari 2%.

Lampiran 11. Hasil data uji stabilitas penyimpanan

Formula	Siklus	Pada suhu dingin (4°C) dan suhu panas (40°C)		
		Bentuk	Warna	Bau
1	1	Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
1	2	Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
1	3	Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
1	4	Gel encer	Putih bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
1	5	Gel encer	Putih bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
1	6	Gel encer	Putih bening	Khas <i>retinyl palmitate</i>
2		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>
3		Gel encer	Kuning bening	Khas <i>retinyl palmitate</i>