

BAB V

KESIMPULAN DAN SARAN

A. Kesimpulan

Berdasarkan hasil penelitian yang telah dilakukan maka dapat disimpulkan bahwa :

1. Sediaan gel dispersi padat ibuprofen dengan gelling agent HPMC dan Karbopol 940 memiliki mutu fisik yang baik meliputi daya lekat, daya sebar, pH, organoleptis, homogenitas serta stabilitas yang baik.
2. Formula optimum gel dispersi padat ibuprofen diperoleh proporsi HPMC 1,73455% dan karbopol 940 0,26545%.

B. Saran

1. Perlu diadakan penelitian lebih lanjut dalam optimasi formulasi gel dispersi padat ibuprofen – PEG 6000 1:1,5 dengan pemilihan variabel lain.
2. Perlu dilakukan uji penetrasi gel secara *in vitro* agar dapat diketahui jumlah obat yang terpenetrasi dalam pemakaian.

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Lampiran 1. Sertifikat analisis ibuprofen



IOL CHEMICALS AND PHARMACEUTICALS LIMITED

CERTIFICATE OF ANALYSIS

Product Name : **IBUPROFEN BP**

Date of Mfg. : Nov. - 2018

Date of Expiry : Oct. - 2023

Drug Lic No. : 1689-OSP

Dispatch Qty : 1000 Kg

Batch No. : 4001/1201/18/A-4151

Date of Analysis : 27/11/2018

A.R. No. : 4001/1151/1118/A-4151/10667

Batch Qty : 1000 Kg

Packing : 20 X 50 Kg

Sr. No	TEST	SPECIFICATIONS	RESULTS
1.	Appearance	White or almost White Crystalline Powder or Colourless Crystals	White crystalline powder.
2.	Solubility	Practically insoluble in water, freely soluble in Acetone, Methanol, and Methylene chloride. It dissolves in dilute solution of alkali hydroxide and carbonate	Complies
3.	Identification 1 st Identification A,C 2 nd Identification A,B,D	A) Melting Point 75-78°C B) By UV Exhibits 2 absorption maxima, at 264 nm and 272 nm. The ratio of the absorbance measured at the maximum at 264 nm to that measured at the shoulder at 258 nm is 1.20 to 1.30. The ratio of the absorbance measured at the maximum at 272 nm to that measured at the shoulder at 258 nm is 1.00 to 1.10. C) By IR The infra-red absorbance spectrum obtained from the sample should be concordant with spectrum obtained from the standard. D) By TLC The principal spot in the chromatogram obtained with the test solution is similar in position, color and size to the principal spot in the chromatogram obtained with the reference solution.	76°C Omitted as per pharmacopoeia Complies. Omitted as per pharmacopoeia

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Ph. : +91-161-2225531-35 Fax : +91-161-2226929, 2608784 email : contact@iolcp.com Website : iolcp.com

Works : Village Fatehgarh Channa, Mansa Road, District - Barnala, 148101 State - Punjab, INDIA.

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Lampiran 2. Gambar alat penelitian

Neraca analitik



Alat – alat gelas



Water bath



Ice bath



Spektrofotometer UV-Vis



Spektroskopi inframerah



pH meter



Viskometer



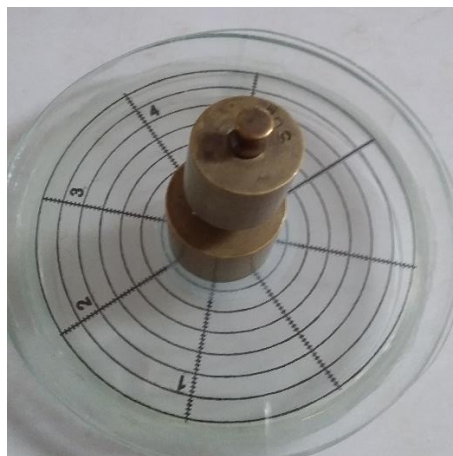
Alat homogenitas



Alat uji daya lekat



Alat uji daya sebar



Oven



Pompa hidrolik



Mortir stamper



Desikator



Lampiran 3. Penimbangan pembuatan dispersi padat dan penetapan kadar dispersi padat

A. Penimbangan pembuatan dispersi padat

Kertas timbang + ibuprofen = 40,236 g

Kertas timbang + sisa = 0,236 g –

Ibuprofen = 40 g

PEG 6000 = 60 g +

Total = 100 g

B. Penimbangan kurva baku

Kertas timbang + ibuprofen = 0,3675 g

Kertas timbang + sisa = 0,2641 g –

Ibuprofen = 0,1034 g = 103,4 mg/100 ml = 1034 ppm

Pengenceran kurva baku

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{150 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 1,5 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{350 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 3,5 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{200 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 2 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{400 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 4 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{250 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 2,5 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{450 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 4,5 \text{ ml}$$

$$V_1 \times C_1 = V_2 \times C_2$$

$$10 \text{ ml} \times \mathbf{300 \text{ ppm}} = V_2 \times 1000 \text{ ppm}$$

$$V_2 = 3 \text{ ml}$$

Lampiran 4. Foto hasil peleburan, pembekuan, dispersi padat kasar dan halus

Peleburan PEG 6000 dan Ibuprofen



Pembekuan leburan dispersi padat



Dispersi padat kasar



Dispersi padat halus

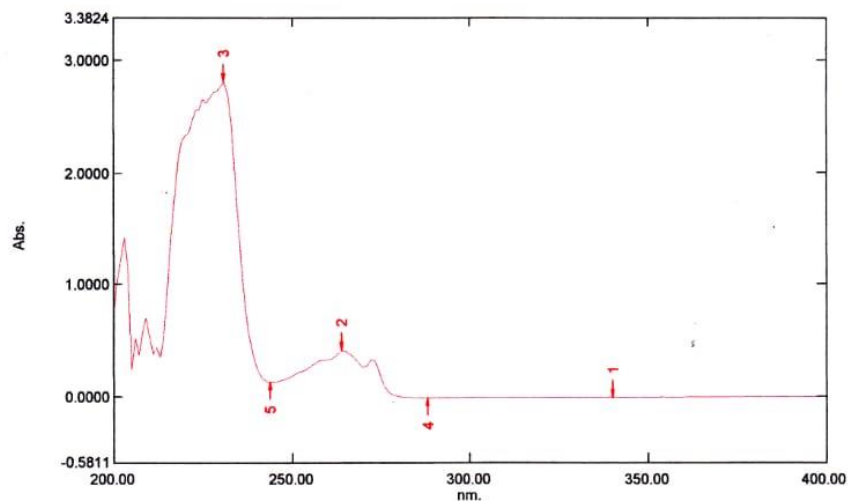


Lampiran 5. Hasil penentuan panjang gelombang maksimum ibuprofen

Spectrum Peak Pick Report

02/01/2020 10:30:15 AM

Data Set: File_200201_102832 - RawData



[Measurement Properties]
Wavelength Range (nm.): 200.00 to 400.00
Scan Speed: Medium
Sampling Interval: 1.0
Auto Sampling Interval: Disabled
Scan Mode: Single

[Instrument Properties]
Instrument Type: UV-1800 Series
Measuring Mode: Absorbance
Slit Width: 1.0 nm
Light Source Change Wavelength: 340.0 nm
S/R Exchange: Normal

[Attachment Properties]
Attachment: None

[Operation]
Threshold: 0.0010000
Points: 4
InterPolate: Disabled
Average: Disabled

[Sample Preparation Properties]
Weight:
Volume:
Dilution:
Path Length:
Additional Information:

No.	P/V	Wavelength	Abs.	Description
1	●	340.00	-0.0081	
2	●	264.00	0.4073	
3	●	231.00	2.8152	
4	●	288.00	-0.0140	
5	●	244.00	0.1251	

Lampiran 6. Hasil penetapan *operating time***Kinetics Data Print Report**

02/03/2020 01:08:45 PM

Time (Minute)	RawData ...
0.000	0.568
1.000	0.568
2.000	0.569
3.000	0.569
4.000	0.569
5.000	0.570
6.000	0.570
7.000	0.569
8.000	0.569
9.000	0.570
10.000	0.569
11.000	0.569
12.000	0.569
13.000	0.568
14.000	0.569
15.000	0.569
16.000	0.569
17.000	0.568
18.000	0.568
19.000	0.568
20.000	0.568
21.000	0.569
22.000	0.569
23.000	0.569
24.000	0.569
25.000	0.569
26.000	0.569
27.000	0.569
28.000	0.569
29.000	0.569
30.000	0.569

Lampiran 7. Perhitungan % *recovery* ibuprofen dalam dispersi padat

$$a = 0,0324$$

$$b = 0,0018$$

$$r = 0,9993$$

$$Y = a + bx$$

$$= 0,0324 + 0,0018x$$

A. Replikasi 1

Dispersi padat ibuprofen yang ditimbang:

$$\text{Kertas timbang} + \text{dispersi} = 0,3077 \text{ g}$$

$$\text{Kertas timbang} + \text{sisas} = 0,2782 \text{ g} -$$

$$\text{Dispersi} = 32,1 \text{ mg}$$

$$\text{Abs} = 0,231$$

Dalam 32,1 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{32,1 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 12,84 \text{ mg ibuprofen}$$

Kadar 12,84 mg ibuprofen yaitu:

$$\frac{12,84 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 128,4 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,231 = 0,0324 + 0,0018x$$

$$x = \frac{0,231 - 0,0324}{0,0018}$$

$$= 110,33 \text{ ppm}$$

$$\% \text{Recovery} = \frac{110,33 \text{ ppm}}{124,4 \text{ ppm}} \times 100\% = 88,6897\%$$

B. Replikasi 2

Dispersi padat ibuprofen yang ditimbang:

Kertas timbang + dispersi = 0,3014 g

Kertas timbang + sisa = 0,2782 g –

Dispersi = 29,4 mg

Abs = 0,222

Dalam 32,1 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{29,4 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 11,76 \text{ mg ibuprofen}$$

Kadar 12,84 mg ibuprofen yaitu:

$$\frac{11,76 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 117,6 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,222 = 0,0324 + 0,0018x$$

$$x = \frac{0,222 - 0,0324}{0,0018}$$

$$= 105,33 \text{ ppm}$$

$$\% \text{Recovery} = \frac{105,33 \text{ ppm}}{124,4 \text{ ppm}} \times 100\% = 84,67\%$$

C. Replikasi 3

Dispersi padat ibuprofen yang ditimbang:

$$\text{Kertas timbang} + \text{dispersi} = 0,3077 \text{ g}$$

$$\text{Kertas timbang} + \text{sisir} = 0,2792 \text{ g} -$$

$$\text{Dispersi} = 30,1 \text{ mg}$$

$$\text{Abs} = 0,224$$

Dalam 32,1 mg dispersi padat mengandung ibuprofen sebanyak:

$$\frac{100 \text{ gram dispersi padat}}{40 \text{ gram ibuprofen}} = \frac{30,1 \text{ mg dispersi padat}}{x \text{ mg ibuprofen}}$$

$$x = 12,04 \text{ mg ibuprofen}$$

Kadar 12,84 mg ibuprofen yaitu:

$$\frac{12,04 \text{ mg ibuprofen}}{100 \text{ ml lar. NaOH } 0,1 \text{ N}} \times 1000 \text{ ppm} = 120,4 \text{ ppm}$$

Kadar percobaan dengan spektrofotometer Uv-Vis:

$$Y = a + bx$$

$$0,224 = 0,0324 + 0,0018x$$

$$x = \frac{0,224 - 0,0324}{0,0018}$$

$$= 106,44 \text{ ppm}$$

$$\% \text{Recovery} = \frac{106,44 \text{ ppm}}{124,4 \text{ ppm}} \times 100\% = 85,562\%$$

$$\text{Rata-rata } \% \text{ recovery} = \frac{88,6897 + 84,67 + 85,562}{3} = 86,307\%$$

$$\begin{aligned} \text{SD} &= \sqrt{\frac{(86,307 - 88,6897)^2 + (86,307 - 84,67)^2 + (86,307 - 85,562)^2}{3-1}} \\ &= \sqrt{\frac{5,677 + 2,6797 + 0,555}{2}} = 2,1106 \end{aligned}$$

$$\text{CV} = \frac{2,1106}{86,307} \times 100\% = 2,44\%$$

Lampiran 8. Perhitungan Penggunaan Ibuprofen dalam Formula Gel

Perbandingan ibuprofen : PEG 6000 = 1 : 1,5

Dalam 100 gram sediaan gel mengandung 2,5 gram dispersi padat ibuprofen-PEG 6000 terdiri atas 1 gram ibuprofen dan 1,5 gram PEG 6000. %Recovery ibuprofen dalam dispersi padat 86,307% sehingga dalam 2,5 gram dispersi padat mengandung 0,86307 gram ibuprofen.

Pada formulasi 100 gram sediaan gel diinginkan mengandung 1 gram ibuprofen.

$$\frac{2,5 \text{ gram dispersi padat}}{x \text{ gram dispersi padat}} = \frac{0,86307 \text{ gram ibuprofen}}{1 \text{ gram ibuprofen}}$$

$$x = 2,896 \text{ gram dispersi padat}$$

Sehingga dapat disimpulkan bahwa dalam formulasi 100 gram sediaan gel mengandung 2,896% dispersi padat ibuprofen yang setara dengan 1% ibuprofen.

Lampiran 9. Foto larutan stok, penetapan % *recovery*, dan akurasi

Larutan stok



Larutan penetapan %*recovery*



Larutan penetapan akurasi



Lampiran 10. Verifikasi metode analisis**A. Linearitas**

$$V_1 \times N_1 = V_2 \times N_2$$

$$1,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 155,1 \text{ ppm}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$2 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 206,8 \text{ ppm}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$2,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 258,5 \text{ ppm}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$3 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 310,2 \text{ ppm}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$3,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 361,9 \text{ ppm}$$

$$V_1 \times N_1 = V_2 \times N_2$$

$$4 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 413,6 \text{ ppm}$$

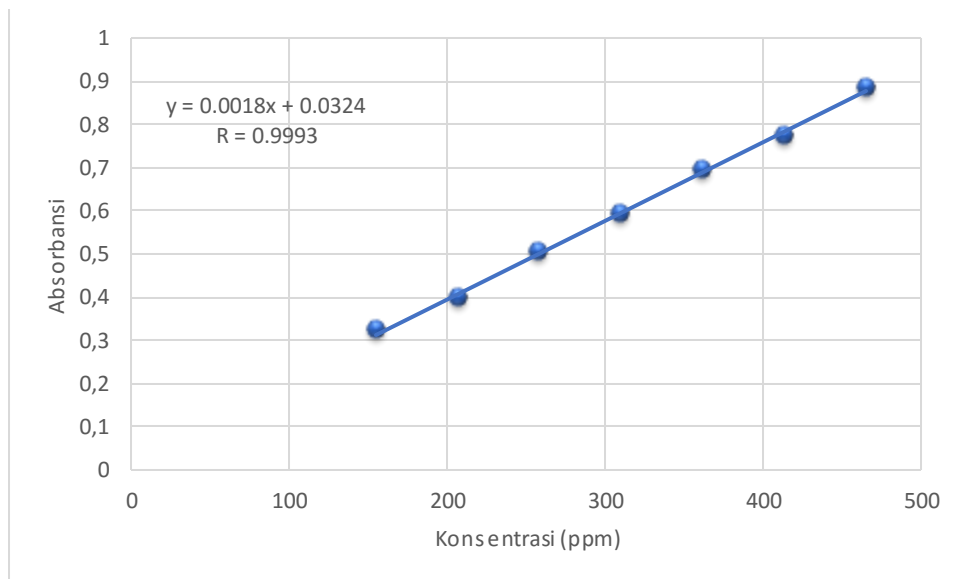
$$V_1 \times N_1 = V_2 \times N_2$$

$$4,5 \text{ ml} \times 1034 \text{ ppm} = 10 \text{ ml} \times N_2$$

$$N_2 = 465,3 \text{ ppm}$$

Konsentrasi	Absorbansi
155.1	0.322
206.8	0.397
258.5	0.504
310.2	0.591
361.9	0.694
413.6	0.773
465.3	0.883

a	0.0324
b	0.0018
r	0.9993



B. Akurasi

Pengambilan larutan:

$$80\% \rightarrow \frac{80}{100} \times 310,2 = 248,16 \text{ ppm} \rightarrow \frac{248,16 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 2,48 \text{ ml} \approx 2,5 \text{ ml}$$

$$100\% \rightarrow \frac{100}{100} \times 310,2 = 310,2 \text{ ppm} \rightarrow \frac{310,2 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 3 \text{ ml}$$

$$120\% \rightarrow \frac{120}{100} \times 310,2 = 372,24 \text{ ppm} \rightarrow \frac{372,24 \text{ ppm} \times 10 \text{ ml}}{1034 \text{ ppm}} = 3,6 \text{ ml} \approx 3,5 \text{ ml}$$

Konsentrasi	Replikasi	Abs	Konsentrasi	Ppm Sebenarnya	%	Rata- Rata	Recovery
80%	1	0.491	252.9262	258.5	98%	99.34%	99.09%
	2	0.497	256.2350	258.5	99%		
	3	0.506	261.1982	258.5	101%		
100%	1	0.589	306.9700	310.2	99%	98.84%	
	2	0.588	306.4185	310.2	99%		
	3	0.588	306.4185	310.2	99%		
120%	1	0.676	354.9476	361.9	98%	99.35%	
	2	0.69	362.6681	361.9	100%		
	3	0.687	361.0137	361.9	100%		

C. Presisi

REPLIKASI	ABSORBANSI	KONSENTRASI
1	0.267	129.3977
2	0.272	132.1550
3	0.266	128.8462
4	0.271	131.6036
5	0.273	132.7065
6	0.268	129.9492
7	0.27	131.0521
8	0.274	133.2580
9	0.267	129.3977
10	0.272	132.1550

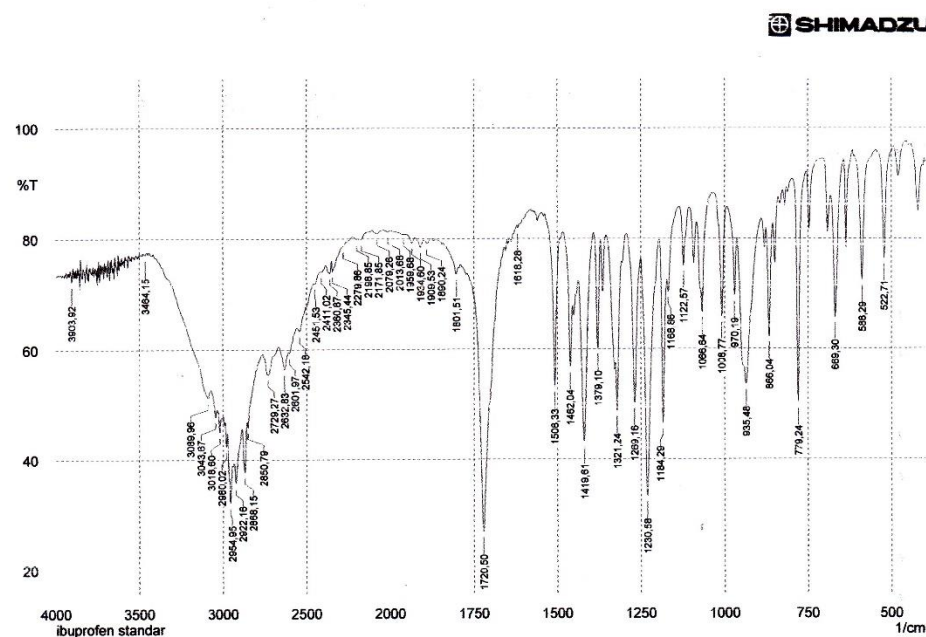
SD 1.559783

RATA-RATA 131.0521

CV 1.190201

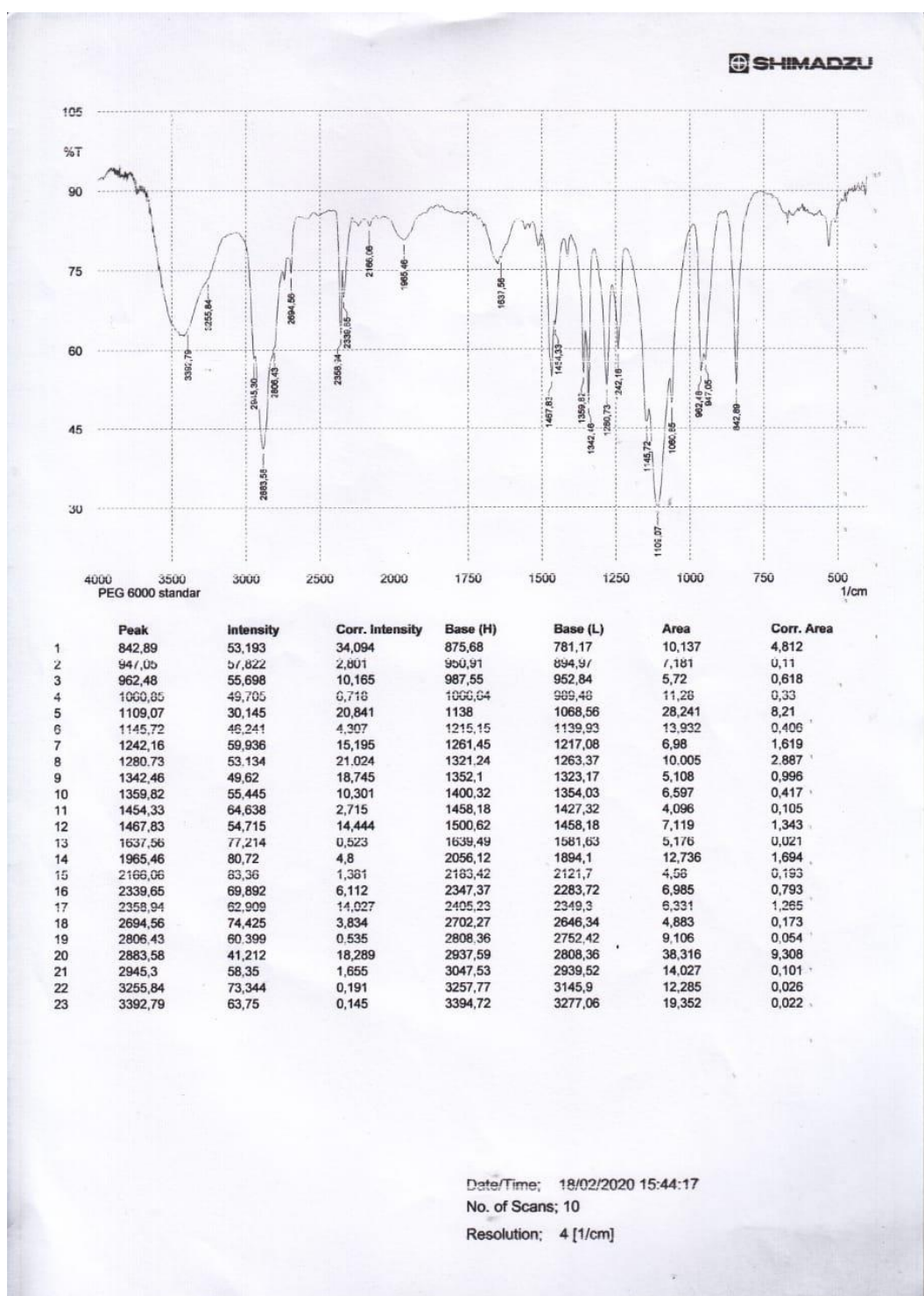
Lampiran 11. Hasil spektra IR

a. Ibuprofen

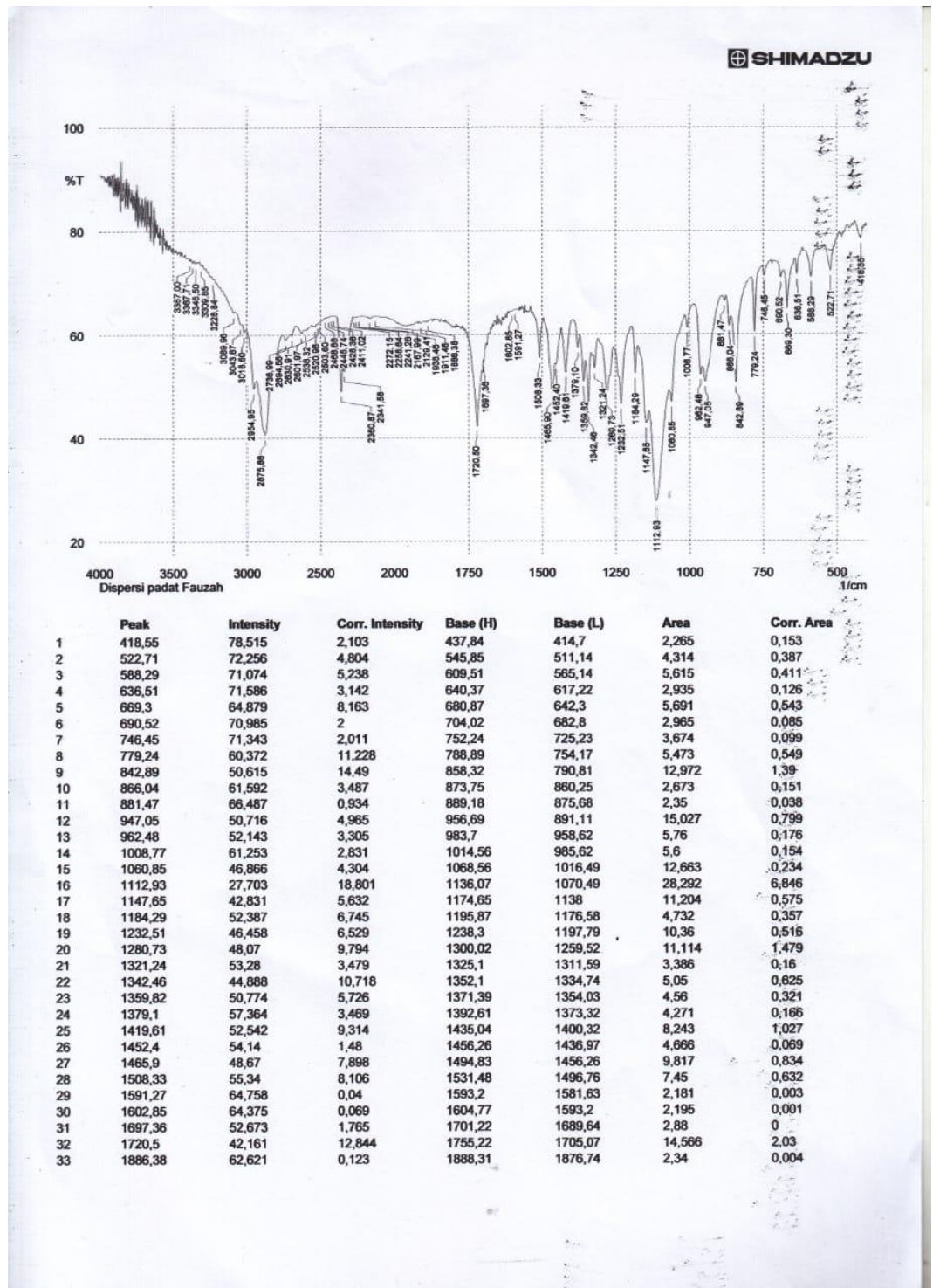


	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	522.71	76,352	19,952	545,85	493,78	2,072	1,231
2	588,29	72,492	23,266	615,29	547,78	3,011	1,757
3	669,3	65,632	24,135	680,87	644,22	3,168	1,685
4	779,24	50,13	41,196	796,6	756,1	3,829	2,257
5	866,04	62,106	19,629	873,75	856,39	2,447	0,94
6	935,48	53,583	5,99	941,26	889,18	8,15	0,41
7	970,19	69,662	11,443	991,41	964,41	2,698	0,45
8	1008,77	65,887	20,865	1031,92	993,34	3,665	1,321
9	1066,64	66,635	5,788	1070,49	1033,85	3,129	0,145
10	1122,57	75,089	10,604	1134,14	1105,21	2,529	0,588
11	1168,86	70,298	2,202	1170,79	1136,07	3,216	0,063
12	1184,29	46,441	29,934	1193,94	1172,72	4,282	1,712
13	1230,58	33,377	44,276	1249,87	1195,87	12,387	6,613
14	1269,16	50,184	27,89	1292,31	1251,8	6,815	2,504
15	1321,24	48,719	12,755	1325,1	1309,67	3,332	0,613
16	1379,1	60,118	20,15	1390,68	1371,39	2,816	0,981
17	1419,61	43,251	32,481	1436,97	1392,61	9,099	4
18	1462,04	56,709	13,513	1481,33	1456,26	3,936	0,612
19	1508,33	53,193	28	1537,27	1494,83	5,739	2,063
20	1618,28	81,9	0,643	1620,21	1581,63	2,987	0,044
21	1720,5	26,896	48,728	1768,72	1658,78	26,391	13,24
22	1801,51	73,554	2,616	1840,09	1789,94	5,899	0,232
23	1890,24	79,113	0,708	1899,88	1878,67	2,113	0,039
24	1909,53	78,414	1,443	1924,96	1901,81	2,329	0,075
25	1934,6	79,203	1,171	1953,89	1926,89	2,6	0,055
26	1959,68	80,784	0,086	1984,75	1955,82	2,651	0,01
27	2013,68	81,181	0,359	2038,76	2000,18	3,451	0,036
28	2079,26	80,872	0,65	2106,27	2056,12	4,549	0,099
29	2171,85	79,891	0,102	2173,78	2144,84	2,714	0,008
30	2198,85	79,907	0,156	2223,92	2193,06	2,976	0,011
31	2279,86	78,544	0,154	2283,72	2225,85	5,766	0,009
32	2345,44	74,07	2,047	2349,3	2285,65	7,403	0,289
33	2360,87	73,768	2,033	2385,95	2351,23	4,441	0,231

b. PEG 6000

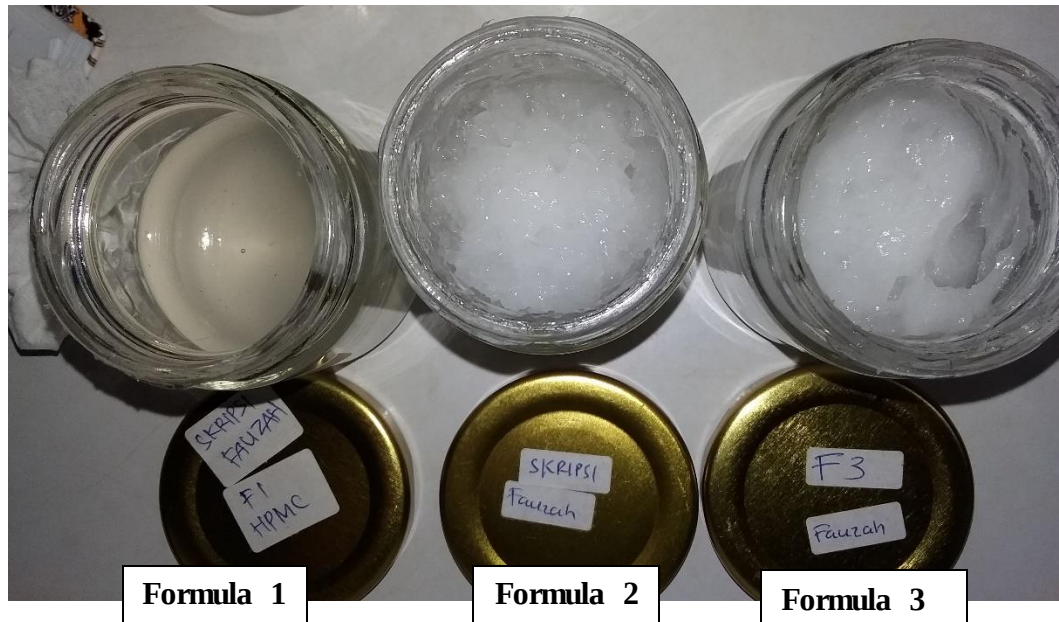


c. Dispersi padat



Lampiran 12. Foto hasil uji mutu fisik gel dispersi padat ibuprofen

a. Pengamatan organoleptis



b. Pengamatan homogenitas



Formula 1



Formula 2



Formula 2

Lampiran 13. Foto uji daya sebar dan daya lekat

Tanpa beban



Beban 50 gram



Beban 100 gram



Beban 150 gram

**Alat uji daya lekat**

Lampiran 14. Uji daya lekat dan daya sebar

Uji	Formula	Replikasi	Hasil	Rata-rata
Viskositas (dpa's)	1	1	110	103,33
		2	110	
		3	100	
	2	1	345	341,67
		2	350	
		3	350	
	3	1	300	306,67
		2	310	
		3	310	
Daya Lekat (detik)	1	1	0,98	0,99
		2	1	
		3	1	
	2	1	1,19	1,18
		2	1,19	
		3	1,18	
	3	1	0,86	0,87
		2	0,89	
		3	0,88	

Daya sebar

Replikasi	Tanpa beban			50 g			100 g			150 g		
Replikasi	F1	F2	F3	F1	F2	F3	F1	F2	F3	F1	F2	F3
1	3,6	3,3	2,9	3,9	3,4,6	3,3,2	3,15,35,13,9	3,93,7	3,55,5,34,2	3,7	3,8	3,7
2	3,6	3,2	2,9	3,8	3,6,4	3,13,25,15,33,9	3,93,5	3,65,7,54,1	3,9	3,7	3,7	3,7
Rata-rata	3,6	3,2	2,8	3,9	3,5	3,1	5,2	3,9	3,55	5,6	4,1	3,8

Replikasi	Tanpa beban			50 g			100 g			150 g		
Replikasi	F1	F2	F3	F1	F2	F3	F1	F2	F3	F1	F2	F3
3	3,5	3,3	2,9	3,9	3,3	3,4	5,5	3,8	3,8	5,7	4,0	4,0
	3,5	3,3	2,9	4,1	3,6	3,1	5,1	3,6	3,5	5,7	4,3	3,9
	3,8	3,4	2,6	3,8	3,7	3,2	5,5	3,8	3,6	5,7	4,2	3,9
	3,6	3,4	2,8	3,9	3,5	3,1	5,4	4	3,5	5,3	3,9	3,8
Rata-rata	3,7	3,35	2,8	3,9	3,5	3,2	5,4	3,8	3,6	5,6	4,1	3,9

Lampiran 15. Data analisis *One Way ANOVA*

a. Daya sebar

One-Sample Kolmogorov-Smirnov Test

		dayasebar
N		9
Normal Parameters ^{a,b}	Mean	4,4889
	Std. Deviation	,79443
Most Extreme	Absolute	,309
Differences	Positive	,309
	Negative	-,232
Kolmogorov-Smirnov Z		,926
Asymp. Sig. (2-tailed)		,358

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

Dayasebar

Levene Statistic	df1	df2	Sig.
,364	2	6	,709

ANOVA

Dayasebar

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	5,016	2	2,508	451,400	,000
Within Groups	,033	6	,006		
Total	5,049	8			

b. Daya lekat

One-Sample Kolmogorov-Smirnov Test

		dayaleka t
N		9
Normal Parameters ^{a,b}	Mean	1,0189
	Std. Deviation	,13596
Most Extreme	Absolute	,222
Differences	Positive	,222
	Negative	-,215
Kolmogorov-Smirnov Z		,666
Asymp. Sig. (2-tailed)		,767

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

Dayalekat

Levene Statistic	df1	df2	Sig.
1,556	2	6	,286

ANOVA

Dayalekat

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	,147	2	,074	551,583	,000
Within Groups	,001	6	,000		
Total	,148	8			

c. Viskositas

One-Sample Kolmogorov-Smirnov Test

		viskositas
N		9
Normal Parameters ^{a,b}	Mean	250,56
	Std. Deviation	111,760
Most Extreme	Absolute	,338
Differences	Positive	,229
	Negative	-,338
Kolmogorov-Smirnov Z		1,013
Asymp. Sig. (2-tailed)		,257

a. Test distribution is Normal.

b. Calculated from data.

Test of Homogeneity of Variances

viskositas

Levene Statistic	df1	df2	Sig.
4,106	2	6	,075

ANOVA

viskositas

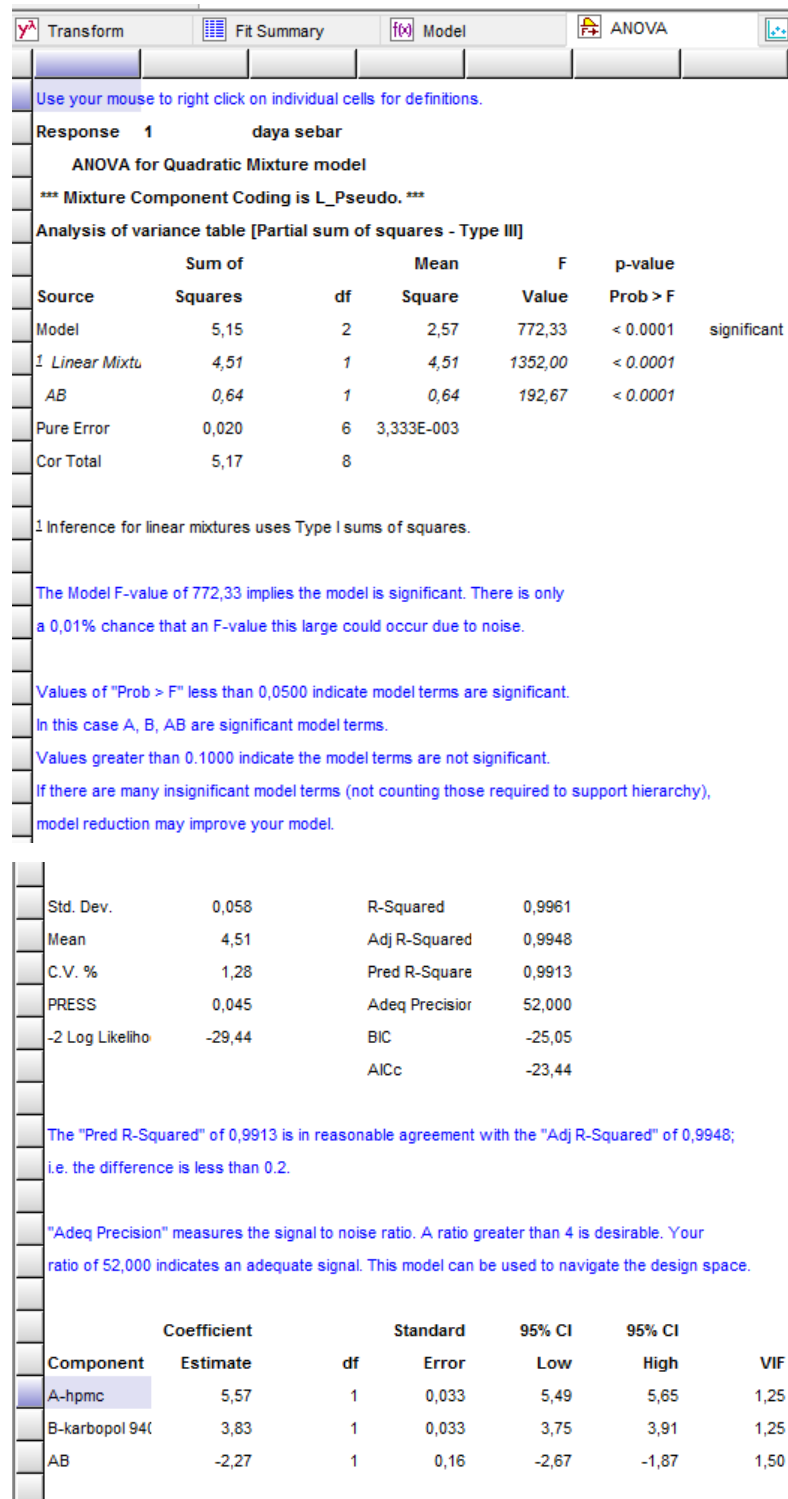
	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	53843,056	2	26921,528	3,505	,098
Within Groups	46079,167	6	7679,861		
Total	99922,222	8			

Lampiran 16. Penentuan formula optimum

Select	Std	Run ▼	Component 1 A:hpmc	Component 2 B:karbopol 940	Response 1 daya sebar cm	Response 2 daya lekat detik	Response 3 viskositas dpa's
	2	1	2	0	5,6	1	100
	5	2	2	0	5,5	0,98	110
	4	3	2	0	5,6	1	100
	1	4	1	1	4,2	1,19	330
	7	5	1	1	4,1	1,19	350
	6	6	1	1	4,1	1,18	345
	3	7	0	2	3,8	0,86	300
	9	8	0	2	3,8	0,89	320
	8	9	0	2	3,9	0,88	300

Lampiran 17. Hasil ANOVA Simplex Lattice Design

a. Daya sebar



b. Daya lekat

Transform Fit Summary Model ANOVA						
Use your mouse to right click on individual cells for definitions.						
Response 2 daya lekat						
ANOVA for Quadratic Mixture model						
*** Mixture Component Coding is L_Pseudo. ***						
Analysis of variance table [Partial sum of squares - Type III]						
	Sum of		Mean	F	p-value	
Source	Squares	df	Square	Value	Prob > F	
Model	0,15	2	0,074	551,58	< 0.0001	significant
1 Linear Mixtu	0,020	1	0,020	153,12	< 0.0001	
AB	0,13	1	0,13	950,04	< 0.0001	
Pure Error	8,000E-004	6	1,333E-004			
Cor Total	0,15	8				
1 Inference for linear mixtures uses Type I sums of squares.						
The Model F-value of 551,58 implies the model is significant. There is only a 0,01% chance that an F-value this large could occur due to noise.						
Values of "Prob > F" less than 0,0500 indicate model terms are significant.						
In this case A, B, AB are significant model terms.						
Values greater than 0.1000 indicate the model terms are not significant.						
If there are many insignificant model terms (not counting those required to support hierarchy), model reduction may improve your model.						
Std. Dev.	0,012	R-Squared	0,9946			
Mean	1,02	Adj R-Squared	0,9928			
C.V. %	1,13	Pred R-Square	0,9878			
PRESS	1,800E-003	Adeq Precisor	46,500			
-2 Log Likeliho	-58,41	BIC	-54,02			
		AICc	-52,41			
The "Pred R-Squared" of 0,9878 is in reasonable agreement with the "Adj R-Squared" of 0,9928; i.e. the difference is less than 0.2.						
"Adeq Precision" measures the signal to noise ratio. A ratio greater than 4 is desirable. Your ratio of 46,500 indicates an adequate signal. This model can be used to navigate the design space.						
	Coefficient		Standard	95% CI	95% CI	
Component	Estimate	df	Error	Low	High	VIF
A-hpmc	0,99	1	6,667E-003	0,98	1,01	1,25
B-karbopol 94	0,88	1	6,667E-003	0,86	0,89	1,25
AB	1,01	1	0,033	0,93	1,09	1,50

c. Viskositas

Transform Fit Summary Model ANOVA						
Use your mouse to right click on individual cells for definitions.						
Response 3 viskositas						
ANOVA for Quadratic Mixture model						
*** Mixture Component Coding is L_Pseudo. ***						
Analysis of variance table [Partial sum of squares - Type III]						
	Sum of		Mean	F	p-value	
Source	Squares	df	Square	Value	Prob > F	
Model	99372,22	2	49686,11	542,03	< 0.0001	significant
1 Linear Mixtu	62016,67	1	62016,67	676,55	< 0.0001	
AB	37355,56	1	37355,56	407,52	< 0.0001	
Pure Error	550,00	6	91,67			
Cor Total	99922,22	8				
1 Inference for linear mixtures uses Type I sums of squares.						
The Model F-value of 542,03 implies the model is significant. There is only a 0,01% chance that an F-value this large could occur due to noise.						
Values of "Prob > F" less than 0,0500 indicate model terms are significant.						
In this case A, B, AB are significant model terms.						
Values greater than 0.1000 indicate the model terms are not significant.						
If there are many insignificant model terms (not counting those required to support hierarchy), model reduction may improve your model.						
Std. Dev.	9,57		R-Squared	0,9945		
Mean	250,56		Adj R-Squared	0,9927		
C.V. %	3,82		Pred R-Square	0,9876		
PRESS	1237,50		Adeq Precisor	43,116		
-2 Log Likeliho	62,56		BIC	66,95		
			AICc	68,56		
The "Pred R-Squared" of 0,9876 is in reasonable agreement with the "Adj R-Squared" of 0,9927; i.e. the difference is less than 0.2.						
"Adeq Precision" measures the signal to noise ratio. A ratio greater than 4 is desirable. Your ratio of 43,116 indicates an adequate signal. This model can be used to navigate the design space.						
	Coefficient		Standard	95% CI	95% CI	
Component	Estimate	df	Error	Low	High	VIF
A-hpmc	103,33	1	5,53	89,81	116,86	1,25
B-karbopol 940	306,67	1	5,53	293,14	320,19	1,25
AB	546,67	1	27,08	480,40	612,93	1,50

Lampiran 18. Foto hasil gel dispersi padat dari formula optimum



Lampiran 19. Data uji viskositas, daya lekat, dan daya sebar formula optimum

Uji	Replikasi	Hasil	Rata-rata
Viskositas (dpa's)	1	300	301,67
	2	305	
	3	300	
Daya	1	1,46	1,316
Lekat	2	1,35	
(detik)	3	1,14	

Daya sebar

Beban	Replikasi 1 (cm)	Replikasi 2 (cm)	Replikasi 3 (cm)	Rata-rata
Tanpa beban	2,3	2,4	2,4	2,4
50 g	2,5	2,5	2,7	2,6
100 g	2,9	2,8	3	2,9
150 g	3,2	3,3	3,2	3,2

Lampiran 20. Data analisis *one sample t-test*

a. Daya sebar

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean		
One-Sample Test						
	Test Value = 5.07					
	t	df	Sig. (2- tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
dayasebar	1,900	2	,198	,06333	-,0801	,2068

b. Daya lekat

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean		
One-Sample Test						
	Test Value = 1.09					
	t	df	Sig. (2- tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
dayaleka t	2,082	2	,173	,13000	-,1387	,3987

c. Viskositas

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
viskositas	3	203,3333	5,77350	3,33333

One-Sample Test

	Test Value = 193.286					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
viskositas	3,014	2	,095	10,04733	-4,2948	24,3895

Lampiran 21. Foto hasil stabilitas dari formula optimum

Siklus 1



Siklus 2



Siklus 3



Siklus 4



Siklus 5



Siklus 6

