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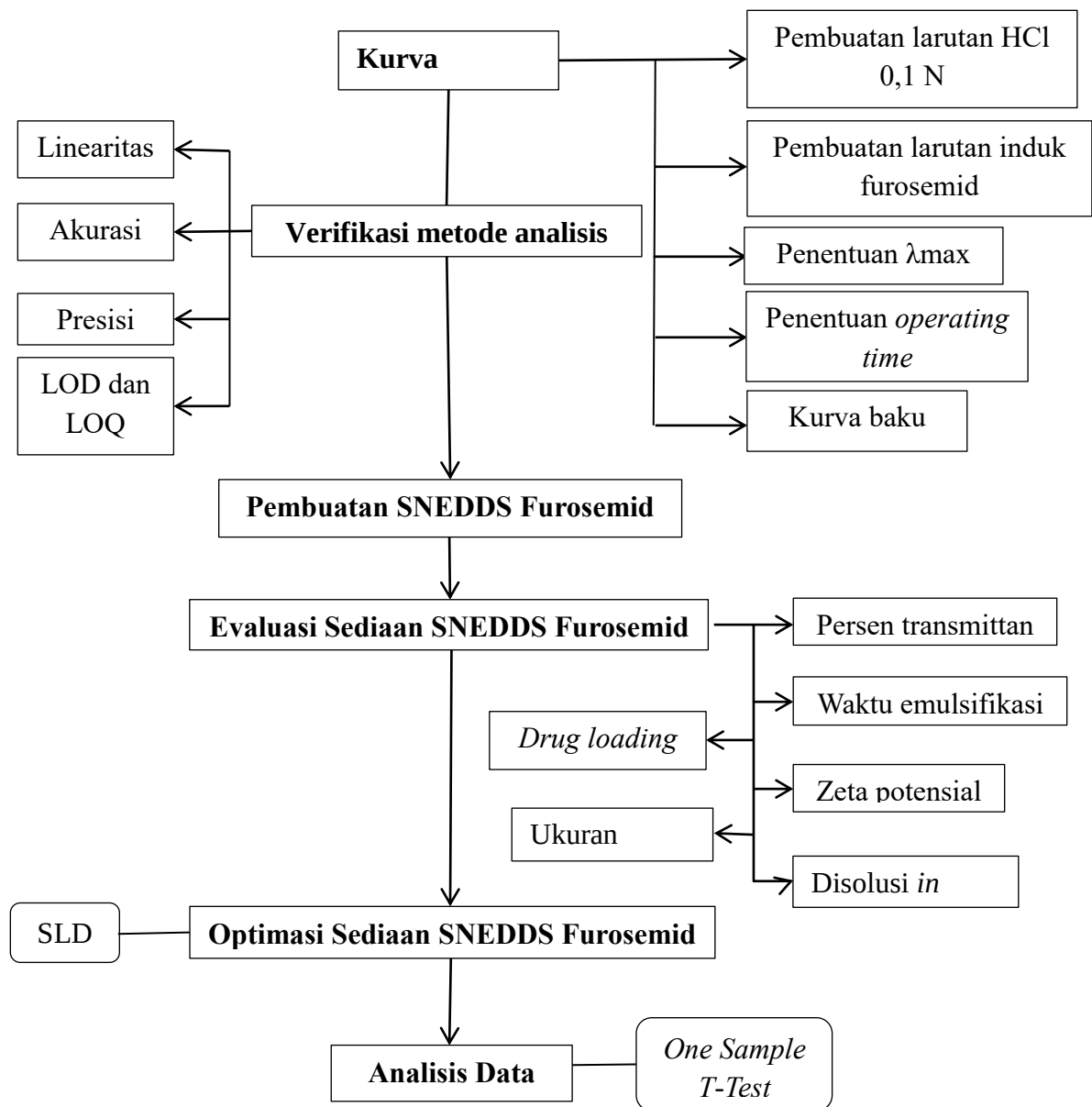
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Lampiran 1. Skema Penelitian



Lampiran 2. Kurva kalibrasi dan verifikasi metode analisis dengan pelarut metanol

a. Pembuatan larutan induk

Furosemid+KS = 0,2766 gram
KS+sisa = 0,2270 gram -
 Berat Furosemid = 0,0496 gram
 Volume metanol pa = 100 ml
 Larutan induk = 0,0496 gram / 100 ml
 = 49,6 mg / 100 ml
 = 496 ppm

Larutan induk furosemid 496 ppm dipipet 1 ml ke 25 ml labu takar, maka diperoleh kadar furosemid :

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 1 \text{ ml} \cdot 496 \text{ ppm} &= 25 \text{ ml} \cdot C_2 \\ C_2 &= 19,84 \text{ ppm} \end{aligned}$$

b. Penetapan panjang gelombang maksimum

Wavelength nm.	RawData ...
251.00	0.1511
252.00	0.1573
253.00	0.1651
254.00	0.1751
255.00	0.1869
256.00	0.2004
257.00	0.2164
258.00	0.2341
259.00	0.2533
260.00	0.2743
261.00	0.2980
262.00	0.3234
263.00	0.3502
264.00	0.3783
265.00	0.4082
266.00	0.4390
267.00	0.4695
268.00	0.4990
269.00	0.5265
270.00	0.5516
271.00	0.5734
272.00	0.5903
273.00	0.6017
274.00	0.6063
275.00	0.6043

276.00	0.5916
277.00	0.5695
278.00	0.5367
279.00	0.4986
280.00	0.4550
281.00	0.4067
282.00	0.3576
283.00	0.3101
284.00	0.2652
285.00	0.2241
286.00	0.1879
287.00	0.1555
288.00	0.1288
289.00	0.1075
290.00	0.0910
291.00	0.0783
292.00	0.0686
293.00	0.0618
294.00	0.0571
295.00	0.0536
296.00	0.0515
297.00	0.0504
298.00	0.0506
299.00	0.0509
300.00	0.0519
301.00	0.0532

Panjang gelombang maksimum larutan furosemid dengan pelarut metanol adalah 274 nm dengan serapan sebesar 0,6063.

c. Penentuan *operating time*

Time (Minute)	RawData ...		
1.000	1.336	23.000	1.800
2.000	1.800	24.000	1.800
3.000	1.800	25.000	1.800
4.000	1.800	26.000	1.800
5.000	1.689	27.000	1.800
6.000	1.590	28.000	1.800
7.000	1.000	29.000	1.800
8.000	1.800	30.000	1.800
9.000	1.800	31.000	1.800
10.000	1.800	32.000	1.800
11.000	1.687	33.000	1.800
12.000	1.800	34.000	1.800
13.000	1.800	35.000	1.686
14.000	1.800	36.000	1.800
15.000	1.800	37.000	1.800
16.000	1.800	38.000	1.800
17.000	1.800	39.000	1.800
18.000	1.800	40.000	1.800
19.000	1.800	41.000	1.800
20.000	1.800	42.000	1.800
21.000	1.800	43.000	1.800
22.000	1.800	44.000	1.800
		45.000	1.800

Scanning operating time menunjukkan bahwa larutan furosemid stabil pada menit ke-12 hingga menit ke-34.

d. Penentuan kurva baku

Konsentrasi (ppm)	Absorbansi
1,984	0,225
2,976	0,277
3,968	0,334
4,96	0,385
5,952	0,432
6,944	0,484
7,936	0,54
8,928	0,601

$$a = 0,1183$$

$$b = 0,0534$$

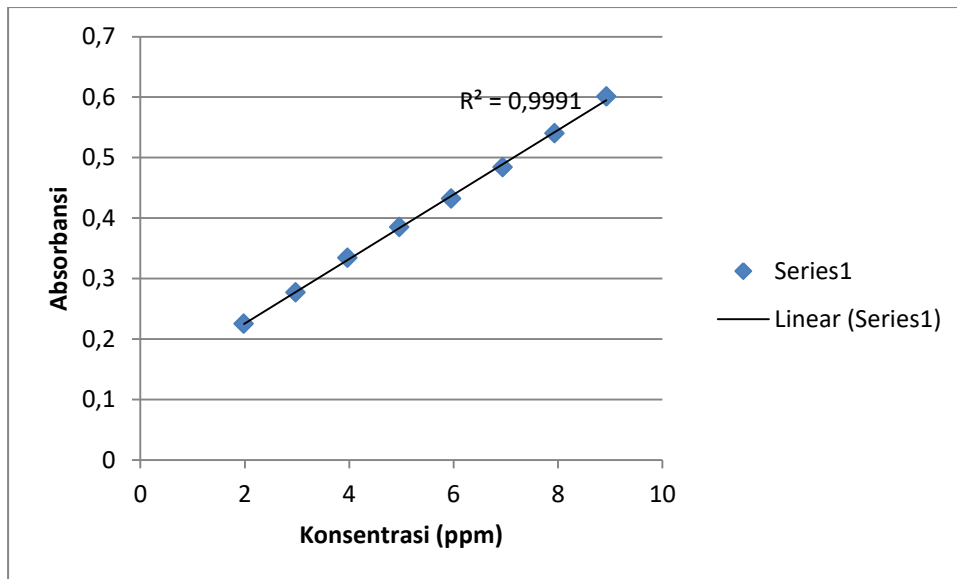
$$r = 0,999$$

Persamaan regresi linear antara konsentrasi (ppm) dan absorbansi adalah

$$y = 0,1183 + 0,0534x$$

dengan y adalah absorbansi dan x adalah konsentrasi dalam ppm.

Berikut adalah grafik dari data kurva baku di atas :



Perhitungan pengenceran seri kurva baku :

- 1 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 1,5 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 1,984 \text{ ppm} \end{aligned}$$
- 1,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 1,5 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 2,976 \text{ ppm} \end{aligned}$$
- 2 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 2 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 3,968 \text{ ppm} \end{aligned}$$
- 2,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 2,5 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 4,96 \text{ ppm} \end{aligned}$$
- 3 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 3 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 5,952 \text{ ppm} \end{aligned}$$
- 3,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 3,5 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 6,944 \text{ ppm} \end{aligned}$$
- 4 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 4 \text{ ml} \cdot 36,64 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 7,936 \text{ ppm} \end{aligned}$$

- 4,5 ml
- $$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 4,5 \text{ ml} \cdot 19,84 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 8,928 \text{ ppm} \end{aligned}$$

e. Verifikasi metode

- Akurasi

No.	Konsentrasi (PPM)	Absorbansi	x	Konsentrasi	Rata-rata	rata rata keseluruhan
1		4,7616	0,373	4,76691	100%	
2	80%	4,7616	0,372	4,74816	100%	100%
3		4,7616	0,37	4,71066	99%	
4		5,952	0,432	5,8732	99%	
5	100%	5,952	0,437	5,96696	100%	100%
6		5,952	0,439	6,00446	101%	
7		7,1424	0,501	7,167	100%	
8	120%	7,1424	0,499	7,1295	100%	100%
9		7,1424	0,5	7,14825	100%	

Akurasi dari metode yang digunakan sebesar 100%.

- Presisi

No.	Konsentrasi (PPM)	Absorbansi	x	x rata-rata	SD	CV
1	5,952	0,429	5,81695			
2	5,952	0,43	5,835701			
3	5,952	0,428	5,7982	5,823201	0,022708	0,4%
4	5,952	0,431	5,854452			
5	5,952	0,43	5,835701			
6	5,952	0,428	5,7982			

$$CV = \frac{SD}{x \text{ rata-rata}} = \frac{0,022708}{5,823201} \times 100\% = 0,4\%$$

- LOD & LOQ

No.	Konsentrasi (ppm)	Absorbansi	Y'	Y-Y'	(Y-Y') ²	x rata rata
1	1,984	0,225	0,22429	0,000714	5,102E-07	
2	2,976	0,277	0,27725	-0,00025	6,25E-08	
3	3,968	0,334	0,33021	0,003786	1,433E-05	
4	4,96	0,385	0,38319	0,001821	3,318E-06	5,456
5	5,952	0,432	0,43614	-0,004143	1,716E-05	
6	6,944	0,484	0,48911	-0,005107	2,608E-05	
7	7,936	0,54	0,54207	-0,002071	4,291E-06	

8	8,928	0,601	0,59504	0,005964	3,557E-05	
				$\sum(Y-Y')^2$	0,0001013	
				Sy/x	0,0045018	
				LOD	0,2529515	
				LOQ	0,8431718	

$$\text{LOD} = 3,3x \frac{\frac{Sy}{\bar{X}}}{b} = 3,3x \frac{0,0045018}{0,0534} = 0,2529515 \text{ ppm}$$

$$\begin{aligned} \text{Absobansi} &= 0,1183 + 0,0534x \\ &= 0,1183 + 0,0534 \cdot 0,253 \\ &= 0,1318 \end{aligned}$$

$$\text{LOQ} = 10x \frac{\frac{Sy}{\bar{X}}}{b} = 10x \frac{0,0045018}{0,0534} = 0,8431718 \text{ ppm}$$

$$\begin{aligned} \text{Absobansi} &= 0,1183 + 0,0534x \\ &= 0,1183 + 0,0534 \cdot 0,843 \\ &= 0,5685 \end{aligned}$$

Lampiran 3. Kurva kalibrasi dan verifikasi metode analisis dengan pelarut HCl 0,1 N

a. Pembuatan larutan induk

Furosemid+KS = 0,3077 gram

KS+sisa = 0,2575 gram -

Berat Furosemid = 0,0502 gram

Volume HCl 0,1 N = 100 ml

Larutan induk = 0,0502 gram / 100 ml

= 50,2 mg / 100 ml

= 502 ppm

Larutan induk furosemid 916 ppm dipipet 1 ml ke 10 ml labu takar, maka diperoleh kadar furosemid :

$$V_1C_1 = V_2C_2$$

$$1 \text{ ml} \cdot 502 \text{ ppm} = 10 \text{ ml} \cdot C_2$$

$$C_2 = 50,2 \text{ ppm}$$

b. Penetapan panjang gelombang maksimum

No.	P/V	Wavelength	Abs.	Description
1		343.00	0.1719	
2		274.00	0.6444	
3		234.00	1.3583	
4		302.00	0.0674	
5		250.00	0.2310	
6		205.00	0.5963	

Panjang gelombang maksimum larutan furosemid dengan pelarut HCl 0,1 N adalah 274 nm dengan serapan sebesar 0,6444.

c. Penentuan *operating time*

Time (Minute)	RawData ...		
1.000	1.891	23.000	1.887
2.000	1.898	24.000	1.888
3.000	1.893	25.000	1.889
4.000	1.899	26.000	1.885
5.000	1.894	27.000	1.885
6.000	1.895	28.000	1.890
7.000	1.896	29.000	1.885
8.000	1.890	30.000	1.880
9.000	1.888	31.000	1.884
10.000	1.889	32.000	1.884
11.000	1.889	33.000	1.885
12.000	1.889	34.000	1.889
13.000	1.887	35.000	1.881
14.000	1.894	36.000	1.884
15.000	1.893	37.000	1.884
16.000	1.895	38.000	1.884
17.000	1.892	39.000	1.884
18.000	1.893	40.000	1.884
19.000	1.887	41.000	1.884
20.000	1.886	42.000	1.884
21.000	1.886	43.000	1.884
22.000	1.886	44.000	1.883
		45.000	1.887

Scanning operating time menunjukkan bahwa larutan furosemid stabil pada menit ke-36 hingga menit ke-43.

d. Penentuan kurva baku

Konsentrasi (ppm)	Absorbansi
7,53	0,225
10,04	0,294
12,55	0,38
15,06	0,462
17,57	0,528
20,08	0,591
22,59	0,672

$$a = 0,00393$$

$$b = 0,02964$$

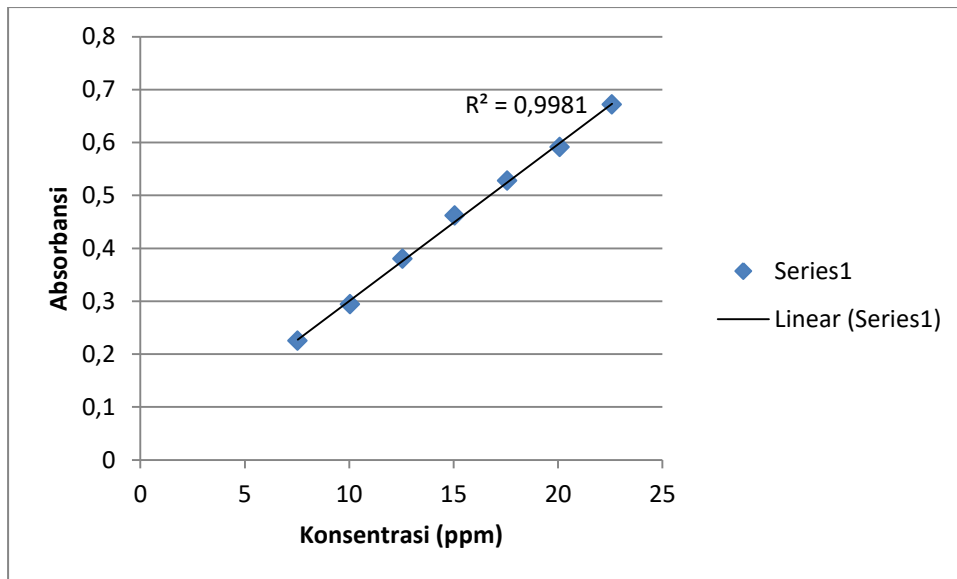
$$r = 0,999$$

Persamaan regresi linear antara konsentrasi (ppm) dan absorbansi adalah

$$y = 0,00393 + 0,02964x$$

dengan y adalah absorbansi dan x adalah konsentrasi dalam ppm.

Berikut adalah grafik dari data kurva baku :



Perhitungan pengenceran seri kurva baku :

- 1,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 1,5 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 7,53 \text{ ppm} \end{aligned}$$
- 2 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 2 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 10,04 \text{ ppm} \end{aligned}$$
- 2,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 2,5 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 12,55 \text{ ppm} \end{aligned}$$
- 3 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 3 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 15,06 \text{ ppm} \end{aligned}$$
- 3,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 3,5 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 17,57 \text{ ppm} \end{aligned}$$
- 4 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 4 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \\ C_2 &= 20,08 \text{ ppm} \end{aligned}$$
- 4,5 ml

$$\begin{aligned} V_1 C_1 &= V_2 C_2 \\ 4,5 \text{ ml} \cdot 50,2 \text{ ppm} &= 10 \text{ ml} \cdot C_2 \end{aligned}$$

$$C_2 = 22,59 \text{ ppm}$$

e. Verifikasi metode

- Akurasi

No.	Konsentrasi (ppm)	Absorbansi	x	konsentrasi	Rata-rata	Rata-rata keseluruhan
1	11,81	0,37	12,351	105%		
2	80% 11,81	0,359	11,980	101%	103%	
3	11,81	0,36	12,014	102%		
4	14,76	0,439	14,679	99%		
5	100% 14,76	0,459	15,354	104%	100%	101%
6	14,76	0,43	14,376	97%		
7	17,71	0,531	17,783	100%		
8	120% 17,71	0,525	17,581	99%	99%	
9	17,71	0,515	17,243	97%		

Akurasi dari metode yang digunakan sebesar 101%.

- Presisi

Konsentrasi (PPM)	Absorbansi	x	x rata-rata	SD	CV
14,76 ppm	0,432	14,443			
14,76 ppm	0,433	14,477			
14,76 ppm	0,443	14,814	14,589	0,179	1,2%
14,76 ppm	0,44	14,713			
14,76 ppm	0,44	14,713			
14,76 ppm	0,43	14,376			

$$CV = \frac{SD}{\bar{x} \text{ rata-rata}} = \frac{0,179}{14,589} \times 100\% = 1,2\%$$

- LOD & LOQ

no.	konsentrasi (ppm)	absorbansi	Y	Y-Y'	(Y-Y') ²	x rata-rata
1	7,53	0,225	0,227	-0,0021	4,445E-06	
2	10,04	0,294	0,302	-0,0075	5,625E-05	
3	12,55	0,38	0,376	0,0041	1,687E-05	
4	15,06	0,462	0,450	0,0117	0,0001372	15,06
5	17,57	0,528	0,525	0,0033	1,103E-05	
6	20,08	0,591	0,599	-0,0081	6,515E-05	
7	22,59	0,672	0,673	-0,0015	2,144E-06	
					$\sum(Y-Y')^2$	
					0,00029	
					S_y/X	
					0,0077	

LOD	0,8525
LOQ	2,5833
Vx0	1,7%

$$\text{LOD} = 3,3x \frac{\frac{Sy}{x}}{b} = 3,3x \frac{0,00765}{0,02964} = 0,8525 \text{ ppm}$$

$$\begin{aligned} \text{Absorbansi} &= 0,00393 + 0,02964x \\ &= 0,00393 + 0,02964 \cdot 0,8525 \text{ ppm} \\ &= 0,0292 \end{aligned}$$

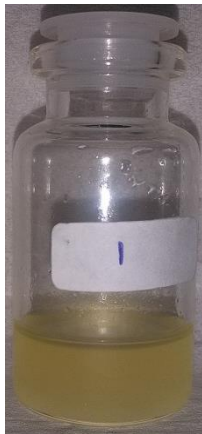
$$\text{LOQ} = 10x \frac{\frac{Sy}{x}}{b} = 10x \frac{0,00765}{0,02964} = 2,5833 \text{ ppm}$$

$$\begin{aligned} \text{Absorbansi} &= 0,00393 + 0,02964x \\ &= 0,00393 + 0,02964 \cdot 2,5833 \text{ ppm} \\ &= 0,0805 \end{aligned}$$

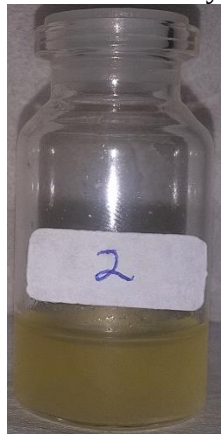
Lampiran 4. Pembuatan formula SNEDDS furosemid

Select	Run	Component 1 A: minyak jarak	Component 2 B: tween 80	Component 3 C: peg 400	Response 1 persen trans...	Response 2 waktu emulsi...	Response 3 drug loading
	1	26,6667	63,3333	10	92,2	150,667	50490,4
	2	20	70	10	99	142,667	23849,1
	3	22,5	67,5	10	96,5	109,333	35397,5
	4	30	60	10	80,8	244,667	38733,7
	5	25	65	10	86,2667	129,333	50491,8
	6	27,5	62,5	10	72,6667	120,333	52540,4
	7	23,3333	66,6667	10	92,9	72	38177,4

Visualisasi Formula SNEDDS furosemid yang dibuat (belum disentrifugasi)



(26,67:63,33:10)



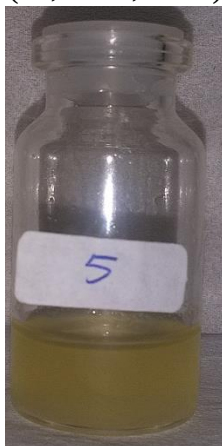
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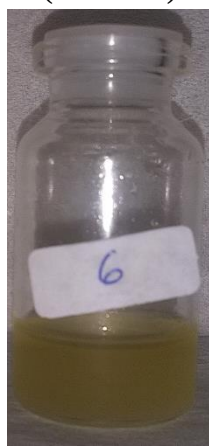
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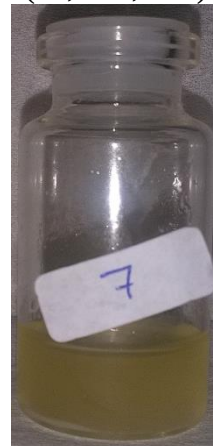
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(25:65:10)



(27,5:62,5:10)



(23,33:66,67:10)

Lampiran 5. Evaluasi sediaan SNEDDS furosemid

a. Waktu emulsifikasi



Bening



Bening



Bening



Bening



Bening



Keruh



Bening

b. Drug loading

No	Absorbansi	FP	Kadar	Rata-rata kadar	SD
1	0,749	4000	47268,67	50490,36	8310,742
	0,758	5000	59929,61		
	0,591	5000	44272,8		
2	0,568	2000	16846,59	23849,07	6068,857
	0,413	5000	27584,7		
	0,408	5000	27115,93		
3	0,613	5000	46335,37	35397,48	11263,42
	0,373	5000	23834,56		
	0,503	5000	36022,5		
4	0,655	4000	40218,41	38733,69	10392,72
	0,634	5000	48304,19		
	0,414	5000	27678,45		
5	0,709	5000	55335,7	50491,77	4195,223
	0,631	5000	48022,93		
	0,632	5000	48116,69		
6	0,716	4000	44793,58	52540,43	6827,317
	0,707	5000	55148,19		
	0,734	5000	57679,53		
7	0,714	4000	44643,57	38177,42	5643,828
	0,499	5000	35647,49		

	0,484	5000	34241,19
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Nilai kadar furosemid didapat dari rumus $y = 0,1183 + 0,0534x$, dengan nilai y adalah absorbansi dan x adalah kadar (ppm).

- Formula 1

- $y = 0,1183 + 0,0534x$
 $0,749 = 0,1183 + 0,0534x$
 $x = 11,81717 \cdot FP$
 $= 11,81717 \cdot 4000$
 $= 47268,67 \text{ ppm}$
- $y = 0,1183 + 0,0534x$
 $0,758 = 0,1183 + 0,0534x$
 $x = 11,98592 \cdot FP$
 $= 11,98592 \cdot 5000$
 $= 59929,61 \text{ ppm}$
- $y = 0,1183 + 0,0534x$
 $0,591 = 0,1183 + 0,0534x$
 $x = 8,85456 \cdot FP$
 $= 8,85456 \cdot 5000$
 $= 44272,8 \text{ ppm}$

- Formula 2

- $y = 0,1183 + 0,0534x$
 $0,568 = 0,1183 + 0,0534x$
 $x = 8,423294 \cdot FP$
 $= 8,423294 \cdot 2000$
 $= 16846,59 \text{ ppm}$
- $y = 0,1183 + 0,0534x$
 $0,413 = 0,1183 + 0,0534x$
 $x = 5,51694 \cdot FP$
 $= 5,51694 \cdot 5000$
 $= 27584,7 \text{ ppm}$
- $y = 0,1183 + 0,0534x$
 $0,408 = 0,1183 + 0,0534x$
 $x = 5,42319 \cdot FP$
 $= 5,42319 \cdot 5000$
 $= 27115,93 \text{ ppm}$

- Formula 3

- $y = 0,1183 + 0,0534x$
 $0,613 = 0,1183 + 0,0534x$
 $x = 9,267075 \cdot FP$
 $= 9,267075 \cdot 5000$
 $= 46335,37 \text{ ppm}$
- $y = 0,1183 + 0,0534x$

$$\begin{aligned}
 0,373 &= 0,1183 + 0,0534x \\
 x &= 4,766913 \cdot \text{FP} \\
 &= 4,766913 \cdot 5000 \\
 &= 23834,56 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,503 &= 0,1183 + 0,0534x \\
 x &= 7,2045 \cdot \text{FP} \\
 &= 7,2045 \cdot 5000 \\
 &= 36022,5 \text{ ppm}
 \end{aligned}$$

- Formula 4

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,655 &= 0,1183 + 0,0534x \\
 x &= 10,0546 \cdot \text{FP} \\
 &= 10,0546 \cdot 4000 \\
 &= 40218,41 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,634 &= 0,1183 + 0,0534x \\
 x &= 9,660839 \cdot \text{FP} \\
 &= 9,660839 \cdot 5000 \\
 &= 48304,19 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,414 &= 0,1183 + 0,0534x \\
 x &= 5,53569 \cdot \text{FP} \\
 &= 5,53569 \cdot 5000 \\
 &= 27678,45 \text{ ppm}
 \end{aligned}$$

- Formula 5

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,709 &= 0,1183 + 0,0534x \\
 x &= 11,06714 \cdot \text{FP} \\
 &= 11,06714 \cdot 5000 \\
 &= 55335,7 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,631 &= 0,1183 + 0,0534x \\
 x &= 9,604587 \cdot \text{FP} \\
 &= 9,604587 \cdot 5000 \\
 &= 48022,93 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,632 &= 0,1183 + 0,0534x \\
 x &= 9,623338 \cdot \text{FP} \\
 &= 9,623338 \cdot 5000 \\
 &= 48116,69 \text{ ppm}
 \end{aligned}$$

- Formula 6

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,716 &= 0,1183 + 0,0534x
 \end{aligned}$$

$$\begin{aligned}
 x &= 11,19839 \cdot \text{FP} \\
 &= 11,19839 \cdot 4000 \\
 &= 44793,58 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,707 &= 0,1183 + 0,0534x \\
 x &= 11,02964 \cdot \text{FP} \\
 &= 11,02964 \cdot 5000 \\
 &= 55148,19 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,734 &= 0,1183 + 0,0534x \\
 x &= 11,53591 \cdot \text{FP} \\
 &= 11,53591 \cdot 5000 \\
 &= 57679,53 \text{ ppm}
 \end{aligned}$$

- Formula 7

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,714 &= 0,1183 + 0,0534x \\
 x &= 11,16089 \cdot \text{FP} \\
 &= 11,16089 \cdot 4000 \\
 &= 44643,57 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,499 &= 0,1183 + 0,0534x \\
 x &= 7,129498 \cdot \text{FP} \\
 &= 7,129498 \cdot 5000 \\
 &= 35647,49 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{➤ } y &= 0,1183 + 0,0534x \\
 0,484 &= 0,1183 + 0,0534x \\
 x &= 6,848238 \cdot \text{FP} \\
 &= 6,848238 \cdot 5000 \\
 &= 34241,19 \text{ ppm}
 \end{aligned}$$

Lampiran 6. Simplex Lattice Design

a. Persen transmittan

Transformasi persen transmittan

Transformation:

- ☒ None
- ☐ Square Root
- ☐ Natural Log
- ☐ Base 10 Log
- ☐ Inverse Square Root
- ☐ Inverse
- ☐ Power
- ☐ Logit
- ☐ ArcSine Square Root

Equation

Example Residuals vs. Predicted

Studentized Residuals

Predicted

No Transform ($\lambda = 1$)

$y' = y$

Use with a typical response.

Response ranges from 72,6667 to 99.
Ratio of max to min is 1,36238

A ratio greater than 10 usually indicates a transformation is required. For ratios less than 3 the power transforms have little effect.

Penentuan model persen transmittan

Mixture order: Linear ▼ Add Term Auto Select...

Model type: Scheffe ▼

A-minyak jarak

B-tween 80

AB

AB(A-B)

AB(A-B)²

Terms with red tilde icons (~) were either aliased or not estimated in the Fit Summary calculations.

ANOVA

Response 1 persen transmittan						
ANOVA for Linear Mixture model						
*** Mixture Component Coding is L_Pseudo. ***						
Analysis of variance table [Partial sum of squares - Type III]						
Source	Sum of Squares	df	Mean Square	F Value	p-value	
Model	338,37	1	338,37	9,20	0,0290	significant
1 Linear Mixtu	338,37	1	338,37	9,20	0,0290	
Residual	183,80	5	36,76			
Cor Total	522,17	6				

1 Inference for linear mixtures uses Type I sums of squares.

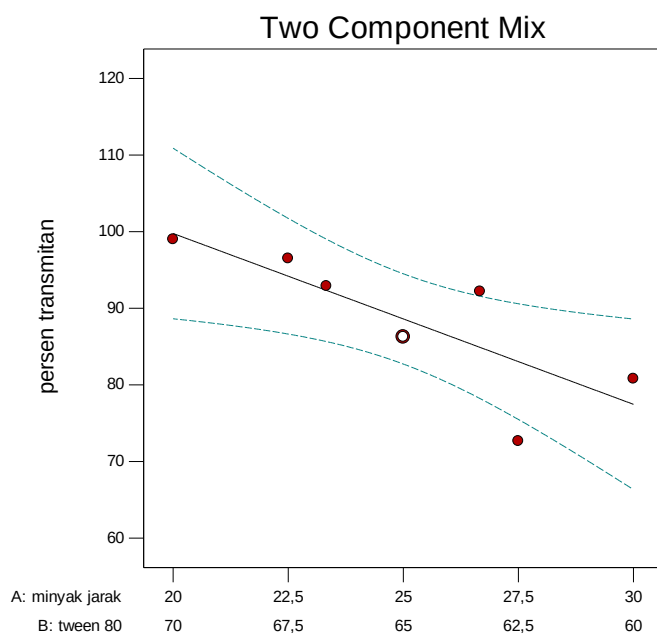
Std. Dev.	6,06	R-Squared	0,6480
Mean	88,62	Adj R-Squared	0,5776
C.V. %	6,84	Pred R-Square	0,3689
PRESS	329,57	Adeq Precisor	6,880
-2 Log Likeliho	42,74	BIC	44,69
		AICc	45,54

Component	Coefficient	df	Standard	95% CI		VIF
	Estimate		Error	Low	High	
A-minyak jarak	77,47	1	4,33	66,34	88,60	1,24
B-tween 80	99,77	1	4,33	88,64	110,90	1,24

Final Equation in Terms of L_Pseudo Components:

$$\begin{aligned} \text{persen transmitan} = & \\ & +77,47 * A \\ & +99,77 * B \end{aligned}$$

Normal plot



b. Waktu emulsifikasi

Transformasi waktu emulsifikasi

Transformation:

- ☒ None
- ☐ Square Root
- ☐ Natural Log
- ☐ Base 10 Log
- ☐ Inverse Square Root
- ☐ Inverse
- ☐ Power
- ☐ Logit
- ☐ ArcSine Square Root

Equation

Example Residuals vs. Predicted

No Transform ($\lambda = 1$)

$y' = y$

Use with a typical response.

Response ranges from 72 to 244,667.
Ratio of max to min is 3,39815

A ratio greater than 10 usually indicates a transformation is required. For ratios less than 3 the power transforms have little effect.

Penentuan model waktu emulsifikasi

Mixture order: Quadratic

Model type: Scheffe

A-minyak jarak

B-tween 80

AB

AB(A-B)

AB(A-B)²

Terms with red tide icons (~) were either aliased or not estimated in the Fit Summary calculations.

ANOVA

Response 2 waktu emulsifikasi

ANOVA for Quadratic Mixture model

*** Mixture Component Coding is L_Pseudo. ***

Analysis of variance table [Partial sum of squares - Type III]

Source	Sum of Squares	df	Mean Square	F Value	p-value	
Model	14437,80	2	7218,90	10,75	0,0246	significant
1 Linear Mixtu	6568,77	1	6568,77	9,78	0,0353	
AB	7869,02	1	7869,02	11,72	0,0267	
Residual	2686,03	4	671,51			
Cor Total	17123,83	6				

1 Inference for linear mixtures uses Type I sums of squares.

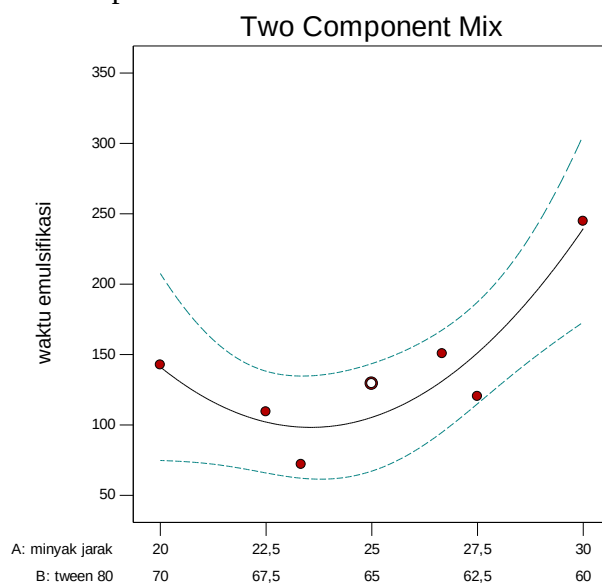
Std. Dev.	25,91	R-Squared	0,8431
Mean	138,43	Adj R-Squared	0,7647
C.V. %	18,72	Pred R-Square	0,6381
PRESS	6197,33	Adeq Precisor	8,311
-2 Log Likeliho	61,51	BIC	65,41
		AICc	68,51

	Coefficient		Standard	95% CI	95% CI	
Component	Estimate	df	Error	Low	High	VIF
A-minyak jarak	239,45	1	23,93	173,02	305,88	2,07
B-tween 80	141,21	1	23,93	74,78	207,63	2,07
AB	-339,70	1	99,23	-615,22	-64,18	3,40

Final Equation in Terms of L_Pseudo Components:

$$\begin{aligned} \text{waktu emulsifikasi} = & \\ & +239,45 * A \\ & +141,21 * B \\ & -339,70 * AB \end{aligned}$$

Normal plot



c. Drug loading

Transformasi drug loading

Transformation:

- ☒ None
- ☐ Square Root
- ☐ Natural Log
- ☐ Base 10 Log
- ☐ Inverse Square Root
- ☐ Inverse
- ☐ Power
- ☐ Logit
- ☐ ArcSine Square Root

Equation

Example Residuals vs. Predicted

No Transform ($\lambda = 1$)
$$y' = y$$

Use with a typical response.

Response ranges from 23849,1 to 52540,4.
Ratio of max to min is 2,20304

A ratio greater than 10 usually indicates a transformation is required. For ratios less than 3 the power transforms have little effect.

Coding for Analysis: Pseudo

Penentuan model drug loading

Mixture order: Cubic

Add Term

Auto Select...

Model type: Scheffe

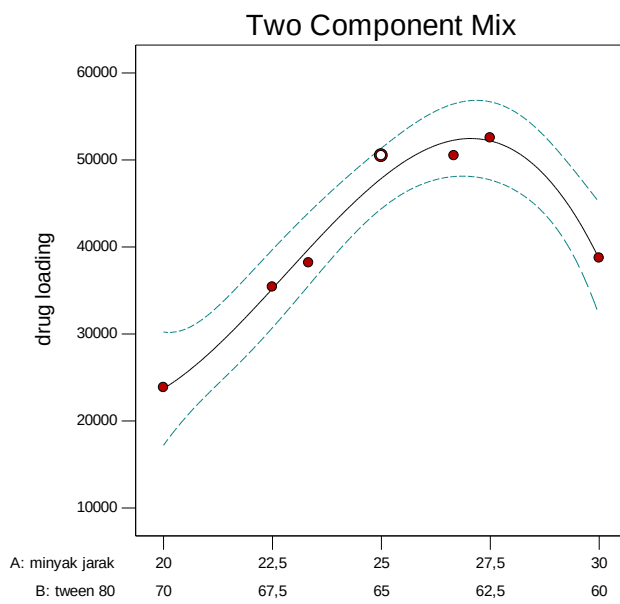
A-minyak jarak
B-tween 80
AB
AB(A-B)
AB(A-B)²

Terms with red tide icons (~) were either aliased or not estimated in the Fit Summary calculations.

ANOVA

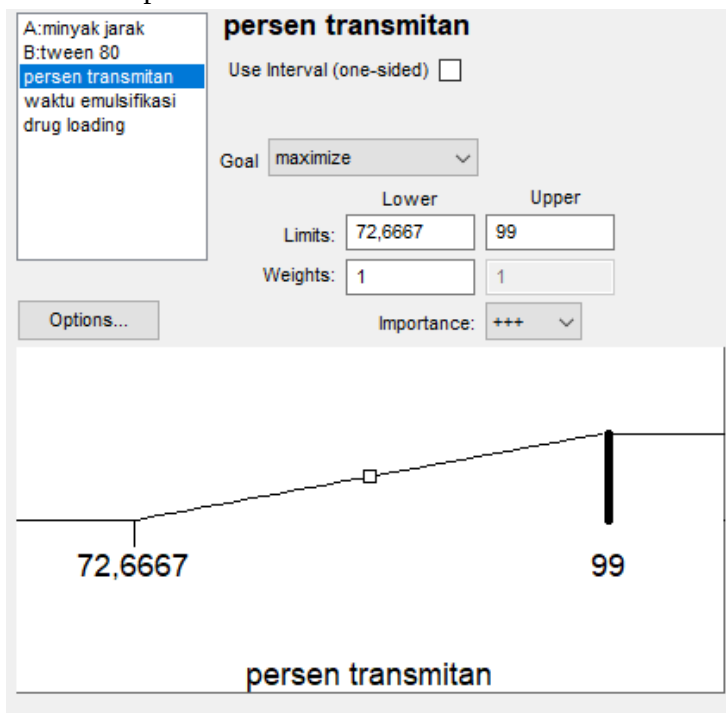
Response 3 drug loading						
ANOVA for Cubic Mixture model						
*** Mixture Component Coding is L_Pseudo. ***						
Analysis of variance table [Partial sum of squares - Type III]						
	Sum of		Mean	F	p-value	
Source	Squares	df	Square	Value	Prob > F	
Model	6,382E+008	3	2,127E+008	50,13	0,0046	significant
1 Linear Mixtu	2,790E+008	1	2,790E+008	65,75	0,0039	
AB	3,047E+008	1	3,047E+008	71,81	0,0035	
Residual	1,273E+007	3	4,244E+006			
Cor Total	6,510E+008	6				
1 Inference for linear mixtures uses Type I sums of squares.						
Std. Dev.	2060,05		R-Squared	0,9804		
Mean	41382,89		Adj R-Squared	0,9609		
C.V. %	4,98		Pred R-Square	0,7762		
PRESS	1,457E+008		Adeq Precisor	18,338		
-2 Log Likeliho	120,76		BIC	126,60		
			AICc	134,76		
	Coefficient		Standard	95% CI	95% CI	
Component	Estimate	df	Error	Low	High	VIF
A-minyak jarak	38617,92	1	2043,48	32114,65	45121,18	2,39
B-tween 80	23721,42	1	2043,48	17218,16	30224,69	2,39
AB	66850,16	1	7888,88	41744,23	91956,10	3,40
Final Equation in Terms of L_Pseudo Components:						
drug loading =						
+38617,92 * A						
+23721,42 * B						
+66850,16 * AB						

Normal plot

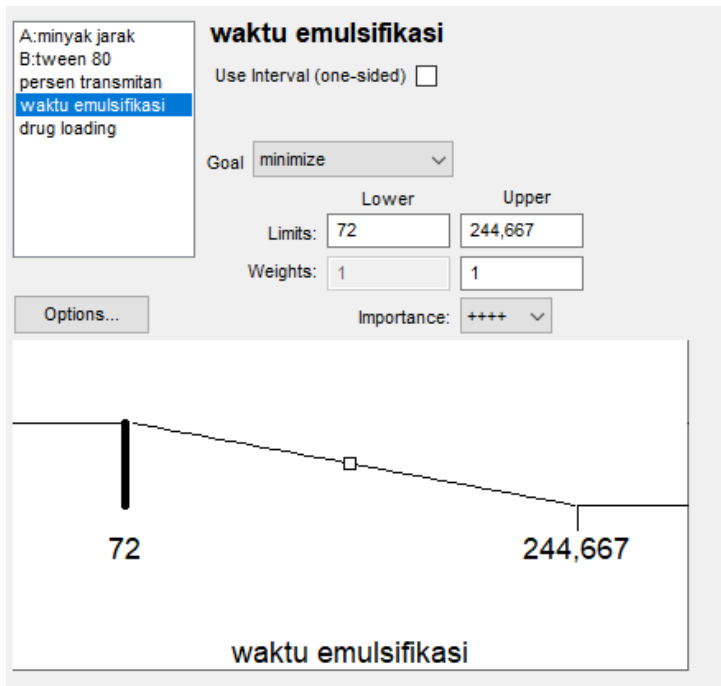


d. Optimasi formula

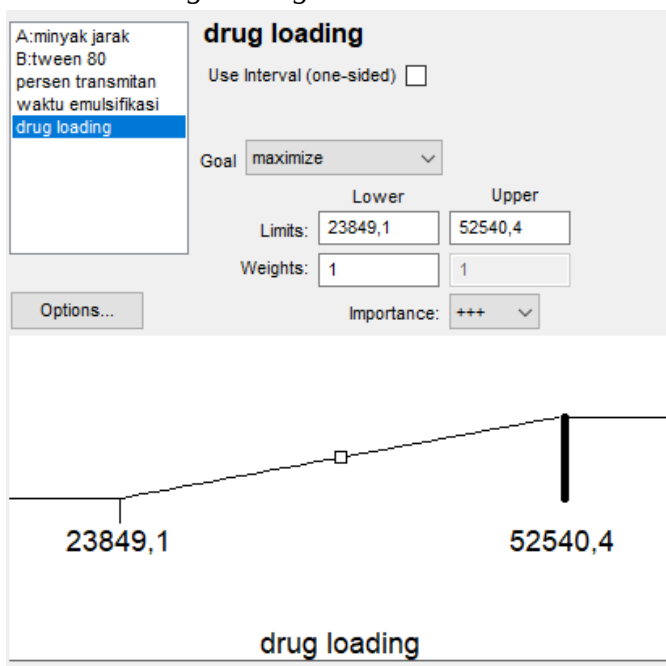
Parameter persen transmitan



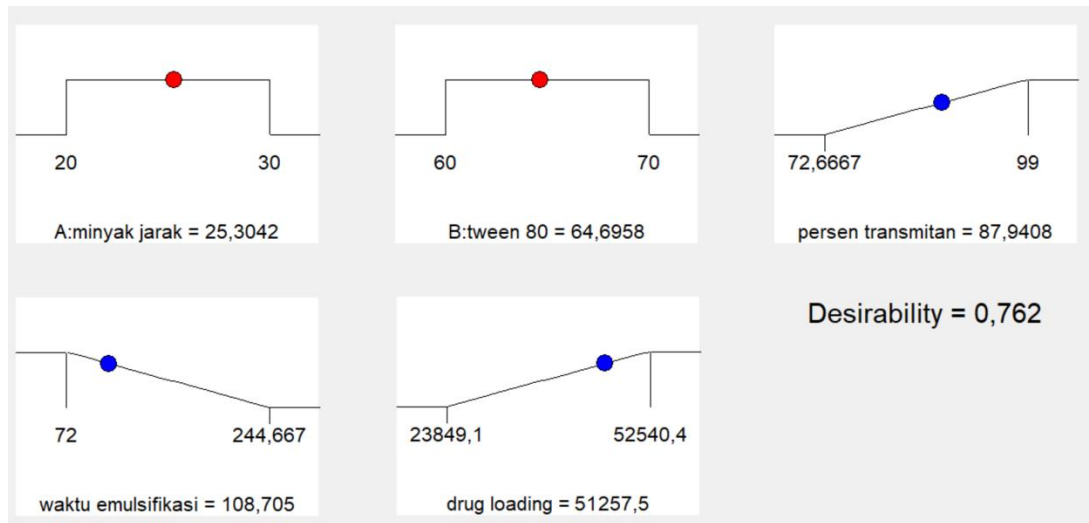
Parameter waktu emulsifikasi



Parameter drug loading



Hasil optimasi



Lampiran 7. Karakterisasi formula optimum SNEDDS furosemid

a. Persen transmitan

Replikasi	Persen transmitan (%)	Rata-rata (%)	SD
I	90,1	87,9	3,72
II	83,6		
III	90		

b. Waktu emulsifikasi

Replikasi	Waktu emulsifikasi (detik)	Rata-rata (detik)	SD
I	111	102,6667	8,02
II	102		
III	95		

c. Drug loading

Replikasi	Absorbansi	FP	Kadar (ppm)	Rata-rata kadar (ppm)	SD
I	0,572	2500	42491,49	51335,55	7660,479
II	0,715	2500	55898,22		
III	0,712	2500	55616,96		
➤ y	$= 0,1183 + 0,0534x$				
0,572	$= 0,1183 + 0,0534x$				
x	$= 16,79498 \cdot FP$				
	$= 16,79498 \cdot 2500$				
	$= 41987,459 \text{ ppm}$				
➤ y	$= 0,1183 + 0,0534x$				
0,715	$= 0,1183 + 0,0534x$				
x	$= 21,37003 \cdot FP$				
	$= 21,37003 \cdot 2500$				
	$= 53425,081 \text{ ppm}$				
➤ y	$= 0,1183 + 0,0534x$				
0,712	$= 0,1183 + 0,0534x$				
x	$= 21,27405 \cdot FP$				
	$= 21,27405 \cdot 2500$				
	$= 53185,13 \text{ ppm}$				

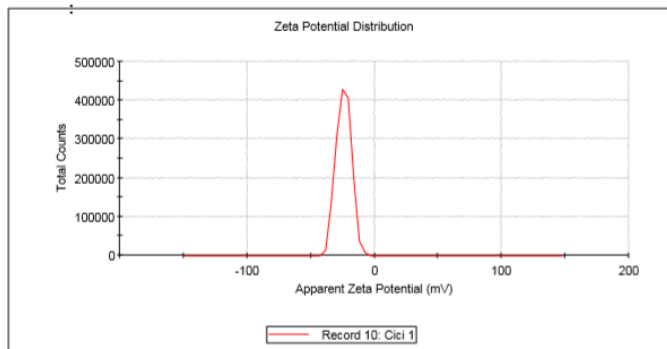
d. Zeta potensial

Replikasi	Zeta potensial (mV)	Rata-rata	SD
I	-24,1	-23,9	0,62
II	-23,2		
III	-24,4		

Replikasi I

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -24,1	Peak 1: -24,1	100,0	5,75
Zeta Deviation (mV): 5,75	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0503	Peak 3: 0,00	0,0	0,00

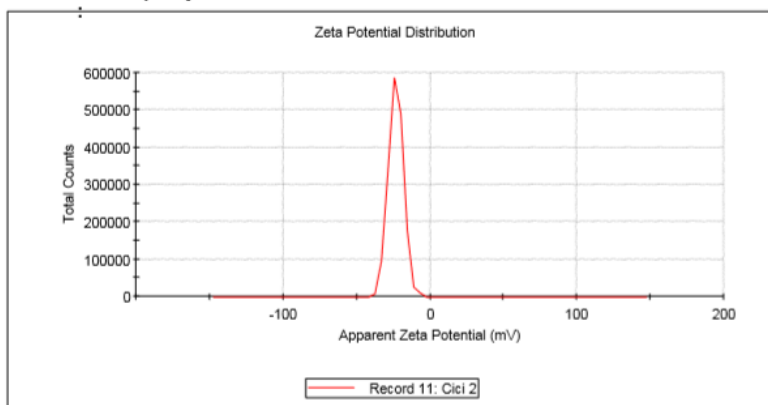
Result quality **Good**



Replikasi II

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -23,2	Peak 1: -23,2	100,0	5,09
Zeta Deviation (mV): 5,09	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0518	Peak 3: 0,00	0,0	0,00

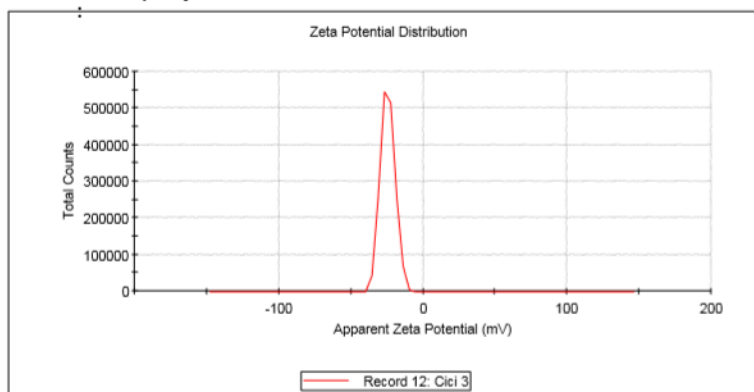
Result quality **Good**



Replikasi III

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -24,4	Peak 1: -24,4	100,0	4,91
Zeta Deviation (mV): 4,91	Peak 2: 0,00	0,0	0,00
Conductivity (mS/cm): 0,0523	Peak 3: 0,00	0,0	0,00

Result quality **Good**

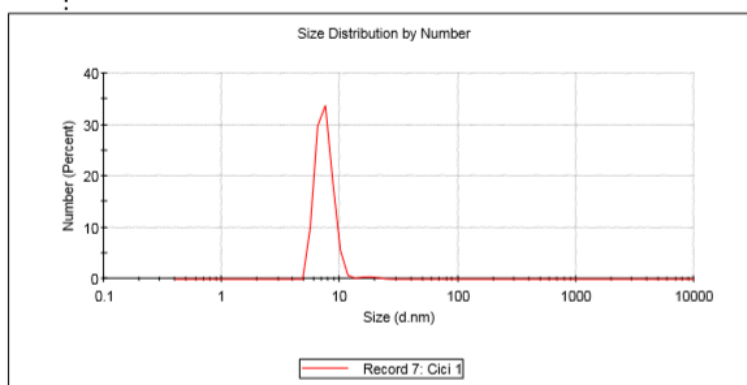


e. Ukuran droplet

Replikasi	Ukuran droplet (nm)	Rata-rata (nm)	SD
I	11,37	11,35	0,03
II	11,37		
III	11,32		

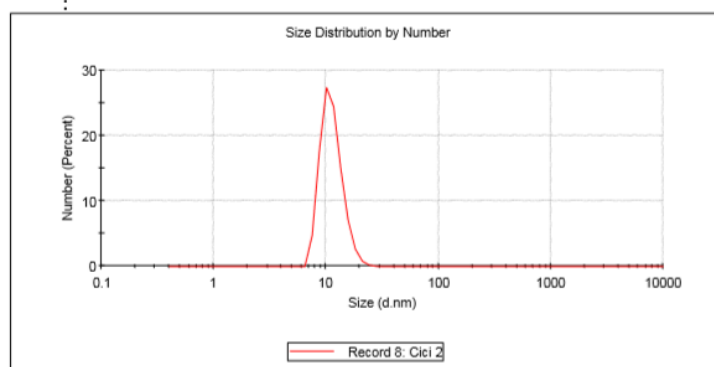
Replikasi I

Z-Average (d.nm): 653,2 **Peak 1:** 18,00 **% Number:** 1,4 **St Dev (d.n...)** 3,085
Pdl: 0,609 **Peak 2:** 7,444 98,6 1,233
Intercept: 1,04 **Peak 3:** 0,000 0,0 0,000
Result quality Refer to quality report
 :

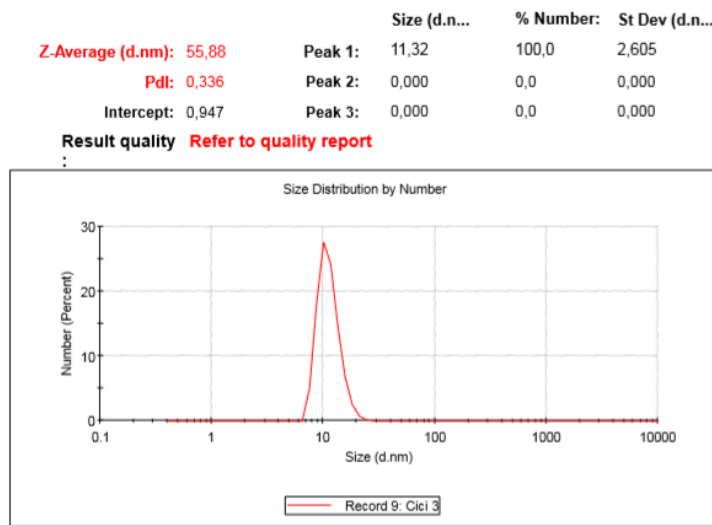


Replikasi II

Z-Average (d.nm): 61,66 **Peak 1:** 11,37 **% Number:** 100,0 **St Dev (d.n...)** 2,624
Pdl: 0,250 **Peak 2:** 0,000 0,0 0,000
Intercept: 0,956 **Peak 3:** 0,000 0,0 0,000
Result quality Refer to quality report
 :



Replikasi III



f. Disolusi *in vitro*

Kadar Furosemid = 51335,55 ppm = 51,33 mg/ml

Jumlah SNEDDS furosemid yang dimasukkan ke dalam kapsul :

$$\text{Volume} = \frac{\text{dosis lazim furosemid}}{\text{kadar furosemid}} = \frac{40 \text{ mg}}{51,33 \text{ mg/ml}} = 0,779 \text{ ml}$$

Waktu	Serapan			Rata-rata	FP	X (ppm)	F koreksi	Kadar koreksi (ppm)	Kadar terkoreksi (mg)	Kadar terkoreksi (%)
	I	II	III							
5	0,321	0,32	0,321	0,321	4	42,747	0	42,747	38,472	96,18
10	0,322	0,325	0,325	0,324	4	43,197	0,237	43,434	39,091	97,73
15	0,441	0,439	0,437	0,439	4	58,717	0,239	58,957	53,061	132,65
20	0,456	0,45	0,454	0,453	4	60,651	0,326	60,978	54,879	137,19
25	0,344	0,347	0,344	0,345	4	46,031	0,337	46,368	41,731	104,33
30	0,329	0,33	0,333	0,331	4	44,096	0,256	44,352	39,917	99,79
35	0,305	0,31	0,307	0,307	4	40,947	0,245	41,192	37,073	92,68
60	0,34	0,343	0,339	0,341	4	45,446	0,227	45,673	41,106	102,77

- 5 menit

$$y = 0,00393 + 0,02964x$$

$$0,321 = 0,00393 + 0,02964x$$

$$x = 10,68668 \cdot \text{FP}$$

$$= 10,68668 \cdot 4$$

$$= 42,747 \text{ ppm}$$

$$\text{Faktor koreksi} = 0$$

$$\text{Kadar terkoreksi (ppm)} = 42,747 \text{ ppm} + 0$$

$$= 42,747 \text{ ppm}$$

$$\text{Kadar terkoreksi (mg)} = \frac{42,747 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml}$$

$$= 38,472 \text{ mg}$$

$$\text{Kadar terkoreksi (\%)} = \frac{38,472 \text{ mg}}{40 \text{ mg}} \times 100\%$$

$$= 96,18\%$$

- 10 menit

$$y = 0,00393 + 0,02964x$$

$$0,324 = 0,00393 + 0,02964x$$

$$x = 10,79915 \cdot FP$$

$$= 10,79915 \cdot 4$$

$$= 43,197 \text{ ppm}$$

$$\text{Faktor koreksi} = \frac{5}{900} \times 42,747 \text{ ppm}$$

$$= 0,237 \text{ ppm}$$

$$\text{Kadar terkoreksi (ppm)} = 43,197 \text{ ppm} + 0,237 \text{ ppm}$$

$$= 43,434$$

$$\text{Kadar terkoreksi (mg)} = \frac{43,434 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml}$$

$$= 39,091 \text{ mg}$$

$$\text{Kadar terkoreksi (\%)} = \frac{39,091 \text{ mg}}{40 \text{ mg}} \times 100\%$$

$$= 97,73\%$$

- 15 menit

$$y = 0,00393 + 0,02964x$$

$$0,439 = 0,00393 + 0,02964x$$

$$x = 14,67922 \cdot FP$$

$$= 14,67922 \cdot 4$$

$$= 58,717 \text{ ppm}$$

$$\text{Faktor koreksi} = \frac{5}{900} \times 43,197 \text{ ppm}$$

$$= 0,239 \text{ ppm}$$

$$\text{Kadar terkoreksi (ppm)} = 58,717 \text{ ppm} + 0,239 \text{ ppm}$$

$$= 58,957 \text{ ppm}$$

$$\text{Kadar terkoreksi (mg)} = \frac{58,957 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml}$$

$$= 53,061 \text{ mg}$$

$$\text{Kadar terkoreksi (\%)} = \frac{53,061 \text{ mg}}{40 \text{ mg}} \times 100\%$$

$$= 132,65\%$$

- 20 menit

$$y = 0,00393 + 0,02964x$$

$$0,453 = 0,00393 + 0,02964x$$

$$x = 11,50768 \cdot FP$$

$$= 11,50768 \cdot 4$$

$$= 46,031 \text{ ppm}$$

$$\text{Faktor koreksi} = \frac{5}{900} \times 58,957 \text{ ppm}$$

$$= 0,337 \text{ ppm}$$

$$\text{Kadar terkoreksi (ppm)} = 46,031 \text{ ppm} + 0,337 \text{ ppm}$$

$$= 46,368 \text{ ppm}$$

$$\text{Kadar terkoreksi (mg)} = \frac{46,368 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml}$$

$$= 41,731 \text{ mg}$$

$$\begin{aligned}\text{Kadar terkoreksi (\%)} &= \frac{41,731 \text{ mg}}{40 \text{ mg}} \times 100\% \\ &= 104,33\%\end{aligned}$$

- 25 menit

$$\begin{aligned}y &= 0,00393 + 0,02964x \\ 0,345 &= 0,00393 + 0,02964x \\ x &= 15,16283 \cdot \text{FP} \\ &= 15,16283 \cdot 4 \\ &= 60,651 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Faktor koreksi} &= \frac{5}{900} \times 58,717 \text{ ppm} \\ &= 0,326 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (ppm)} &= 60,651 \text{ ppm} + 0,326 \text{ ppm} \\ &= 60,978 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (mg)} &= \frac{60,978 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml} \\ &= 54,879 \text{ mg}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (\%)} &= \frac{54,879 \text{ mg}}{40 \text{ mg}} \times 100\% \\ &= 137,19\%\end{aligned}$$

- 30 menit

$$\begin{aligned}y &= 0,00393 + 0,02964x \\ 0,331 &= 0,00393 + 0,02964x \\ x &= 11,02408 \cdot \text{FP} \\ &= 11,02408 \cdot 4 \\ &= 44,096 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Faktor koreksi} &= \frac{5}{900} \times 60,978 \text{ ppm} \\ &= 0,256 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (ppm)} &= 44,096 \text{ ppm} + 0,256 \text{ ppm} \\ &= 44,352 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (mg)} &= \frac{44,352 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml} \\ &= 39,917 \text{ mg}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (\%)} &= \frac{39,917 \text{ mg}}{40 \text{ mg}} \times 100\% \\ &= 99,79\%\end{aligned}$$

- 35 menit

$$\begin{aligned}y &= 0,00393 + 0,02964x \\ 0,307 &= 0,00393 + 0,02964x \\ x &= 10,23682 \cdot \text{FP} \\ &= 10,23682 \cdot 4 \\ &= 40,947 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Faktor koreksi} &= \frac{5}{900} \times 44,352 \text{ ppm} \\ &= 0,245 \text{ ppm}\end{aligned}$$

$$\begin{aligned}\text{Kadar terkoreksi (ppm)} &= 40,947 \text{ ppm} + 0,245 \text{ ppm} \\ &= 41,192 \text{ ppm}\end{aligned}$$

$$\begin{aligned}
 \text{Kadar terkoreksi (mg)} &= \frac{41,192 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml} \\
 &= 37,073 \text{ mg} \\
 \text{Kadar terkoreksi (\%)} &= \frac{37,073 \text{ mg}}{40 \text{ mg}} \times 100\% \\
 &= 92,68\%
 \end{aligned}$$

- 60 menit

$$\begin{aligned}
 y &= 0,00393 + 0,02964x \\
 0,341 &= 0,00393 + 0,02964x \\
 x &= 11,36148 \cdot \text{FP} \\
 &= 11,36148 \cdot 4 \\
 &= 45,446 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{Faktor koreksi} &= \frac{5}{900} \times 41,192 \text{ ppm} \\
 &= 0,227 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{Kadar terkoreksi (ppm)} &= 45,446 \text{ ppm} + 0,227 \text{ ppm} \\
 &= 45,446 \text{ ppm}
 \end{aligned}$$

$$\begin{aligned}
 \text{Kadar terkoreksi (mg)} &= \frac{45,446 \text{ mg}}{1000 \text{ ml}} \times 900 \text{ ml} \\
 &= 41,106 \text{ mg}
 \end{aligned}$$

$$\begin{aligned}
 \text{Kadar terkoreksi (\%)} &= \frac{41,106 \text{ mg}}{40 \text{ mg}} \times 100\% \\
 &= 102,77\%
 \end{aligned}$$

Lampiran 8. Verifikasi formula optimum SNEDDS furosemid

One-Sample Kolmogorov-Smirnov Test

		persen transmitan	waktu emulsifikasi	drug loading
N		3	3	3
Normal Parameters ^{a,b}	Mean	87,9000	102,6667	51335,5567
	Std. Deviation	3,72424	8,02081	7660,47735
	Absolute	,380	,200	,379
Most Extreme Differences	Positive	,277	,200	,276
	Negative	-,380	-,184	-,379
Kolmogorov-Smirnov Z		,659	,346	,656
Asymp. Sig. (2-tailed)		,779	1,000	,783

a. Test distribution is Normal.

b. Calculated from data.

Data uji *One-Sample Kolmogorov-Smirnov* menunjukkan nilai signifikansi $p > 0,05$ (H_0 diterima). Disimpulkan data tersebut terdistribusi normal.

a. Persen transmitan

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
persen transmitan	3	87,9000	3,72424	2,15019

One-Sample Test

	Test Value = 87.9374					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
persen transmitan	-,017	2	,988	-,03740	-9,2889	9,2141

b. Waktu emulsifikasi

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
waktu emulsifikasi	3	102,6667	8,02081	4,63081

One-Sample Test

	Test Value = 108.723					
	t	df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
waktu emulsifikasi	-1,308	2	,321	-6,05633	-25,9811	13,8685

c. Drug loading

One-Sample Statistics

	N	Mean	Std. Deviation	Std. Error Mean
drug loading	3	51335,5567	7660,47735	4422,77866

One-Sample Test

	Test Value = 51257.5					
	t	Df	Sig. (2-tailed)	Mean Difference	95% Confidence Interval of the Difference	
					Lower	Upper
drug loading	,018	2	,988	78,05667	-18951,6240	19107,7373

Lampiran 19. Alat-alat yang digunakan dalam praktikum



Neraca analitik



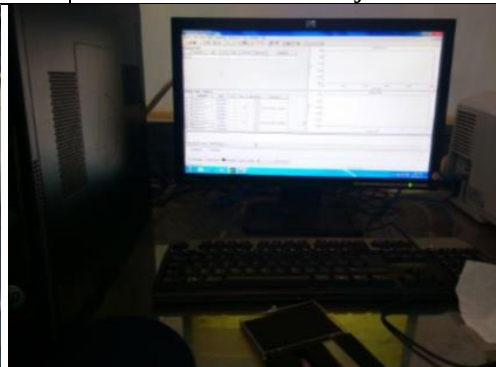
Magnetic stirrer



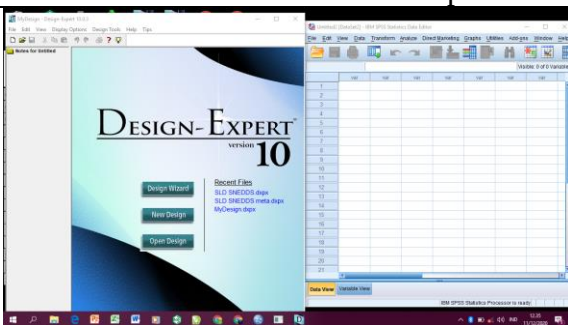
Sentrifugator



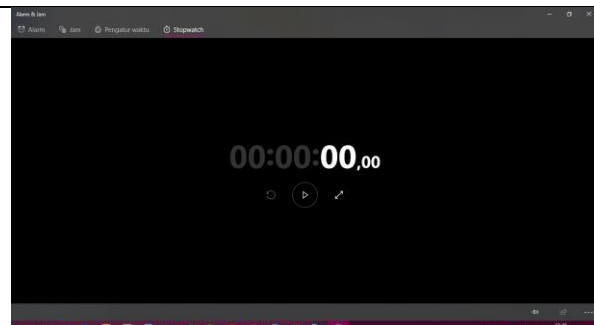
Alat uji disolusi



Spektrofotometer UV-Vis



Perangkat terinstal Design Expert dan SPSS



Stopwatch