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## Lampiran 1. Gambar penelitian

### 1. Gambar bahan penelitian



a. Gambar buah kurma ajwa



b. Gambar buah kurma deglet noor

### 2. Proses maserasi



a. Penimbangan buah deglet noor



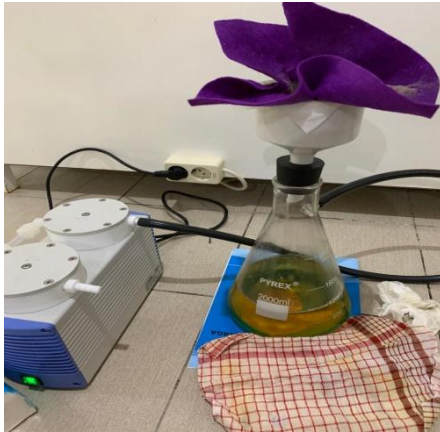
b. penimbangan buah kurma ajwa



c. botol sampel buah kurma ajwa



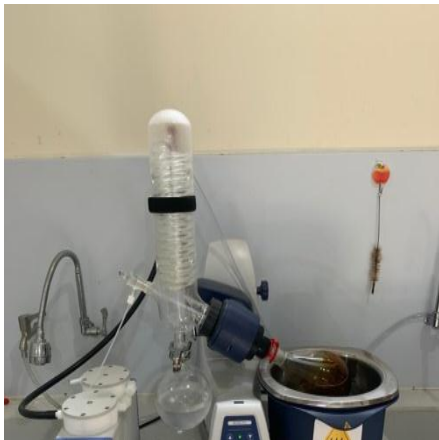
d. Botol sampel buah kurma deglet noor



e. proses penyaringan buah deglet noor



f. Proses penyaringan kurma ajwa



g. Gambar alat evaporator.



h. ekstrak kental kurma ajwa dan deglet noor.

## Lampiran 2. Perhitungan rendemen ekstrak kurma ajwa dan deglet noor.

### 1. Rendemen ekstrak kurma ajwa.

Berat beker gelas = 300 gram

Sampel = 500 gram

Berat sisa = 303 gram

Berat ekstrak kental = 243 gram

- % rendemen =  $\frac{\text{hasil ekstrak yang didapatkan}}{\text{berat kurma ajwa}} \times 100\%$   
 $= \frac{243 \text{ gram}}{500 \text{ gram}} \times 100\%$   
 $= 48,6\%$

### 2. Rendemen ekstrak kurma deglet noor.

Berat beker gelas = 300 gram

Sampel = 500 gram

Berat sisa = 303 gram

Berat ekstrak kental = 160 gram

- % rendemen =  $\frac{\text{hasil ekstrak yang didapatkan}}{\text{berat kurma ajwa}} \times 100\%$   
 $= \frac{160 \text{ gram}}{500 \text{ gram}} \times 100\%$   
 $= 32\%$

**Lampiran 3. Hasil identifikasi kandungan kimia ekstrak kurma ajwa dan deglet noor.**

**1. Uji tabung**



a. uji falvonid kurma ajwa



b. uji falavonoid kurma deglet noor



c. uji fenol kurma ajwa



d. uji fenol kurma deglet noor

## Lampiran 4. Perhitungan DPPH dan perhitungan larutan pengenceran.

### 1. Perhitungan DPPH 0,4mM

$$\text{Molaritas Dpph} = 4 \times 10^{-4}$$

$$\text{Mr Dpph} = 394,32 \text{ gram/mol}$$

$$\text{Volume pembuatan} = 100 \text{ mL}$$

$$\text{Molaritas} = \frac{\text{berat Dpph}}{\text{Mr}} \times \frac{1000}{\text{volume pembuatan}}$$

$$4 \times 10^{-4} = \frac{\text{gr}}{394,32} \times \frac{1000}{100}$$

$$\text{Gram} = \frac{0,0004 \times 394,329/\text{mol}}{10}$$

$$\text{Gram} = 0,01577 \text{ gram}$$

$$\text{Gram} = 15,77 \text{ mg (Dpph yang dipakai)}$$

### 2. Perhitungan larutan pengenceran

Konsentrasi yang digunakan 100 ppm diencerkan menjadi= 10 ppm, 8 ppm, 6 ppm, 4 ppm, 2 ppm.

- 10 ppm
 
$$(V1 \times N1) = (V2 \times N2)$$

$$X \times 100 = 10 \text{ mL} \times 10 \text{ ppm}$$

$$X = \frac{600}{100}$$

$$X = 1 \text{ mL}$$
- 8 ppm
 
$$(V1 \times N1) = (V2 \times N2)$$

$$X \times 100 = 10 \text{ mL} \times 8 \text{ ppm}$$

$$X = \frac{80}{100}$$

$$X = 0,8 \text{ mL}$$
- 6 ppm
 
$$(V1 \times N1) = (V2 \times N2)$$

$$X \times 100 = 10 \text{ mL} \times 6 \text{ ppm}$$

$$X = \frac{60}{100}$$

$$X = 0,6 \text{ mL}$$
- 4 ppm
 
$$(V1 \times N1) = (V2 \times N2)$$

$$X \times 100 = 10 \text{ mL} \times 4 \text{ ppm}$$

$$X = \frac{40}{100}$$

$$X = 0,4 \text{ mL}$$
- 2 ppm
 
$$(V1 \times N1) = (V2 \times N2)$$

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$$X \times 100 = 10 \text{ mL} \times 2 \text{ ppm}$$

$$X = \frac{20}{100}$$

$$X = 0,2 \text{ mL}$$

## Lampiran 5. Proses pengujian antioksidan

### 1. Proses pengujian antioksidan



a. larutan stok ekstrak kurma ajwa.



b. larutan stok ekstrak kurma deglet noor.



c. Pengeceeran ekstrak kurma ajwa



d. Pengeceeran ekstrak kurma deglet noor



e. Larutan stok DPPH



f. perubahan warna DPPH setelah ditambahkan ekstrak etanol kurma ajwa.

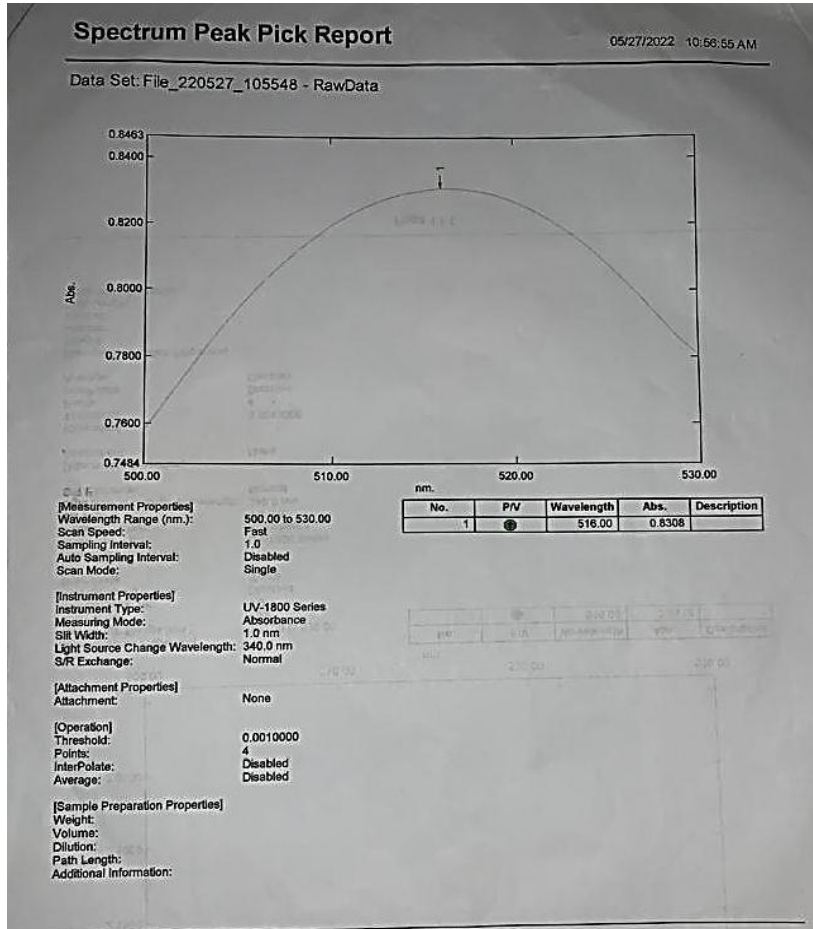


g. perubahan warna DPPH setelah ditambahkan ekstrak etanol kurma deglet noor.



## Lampiran 6. Penentuan panjang gelombang maksimum

### 1. Panjang gelombang maksimum DPPH



## Lampiran 7. Penentuan *operating time*

### 1. *Operating time* ekstrak etanol kurma ajwa

**Kinetics Data Print Report**

05/31/2022 09:30:41 AM

Time ( Minute )	RawData ...
0.000	0.590
1.000	0.584
2.000	0.580
3.000	0.582
4.000	0.582
5.000	0.581
6.000	0.581
7.000	0.581
8.000	0.581
9.000	0.580
10.000	0.580
11.000	0.580
12.000	0.580
13.000	0.580
14.000	0.580
15.000	0.580
16.000	0.580
17.000	0.580
18.000	0.580
19.000	0.580
20.000	0.580
21.000	0.580
22.000	0.580
23.000	0.580
24.000	0.580
25.000	0.580
26.000	0.580
27.000	0.580
28.000	0.580
29.000	0.580
30.000	0.580
31.000	0.580
32.000	0.580
33.000	0.580
34.000	0.580
35.000	0.580
36.000	0.580
37.000	0.580
38.000	0.580
39.000	0.580
40.000	0.580
41.000	0.580
42.000	0.580
43.000	0.580
44.000	0.580
45.000	0.580
46.000	0.580
47.000	0.580
48.000	0.580
49.000	0.580
50.000	0.580

**Kinetics Data Print Report**

05/31/2022 09:30:41 AM

Time ( Minute )	RawData ...
51.000	0.580
52.000	0.580
53.000	0.580
54.000	0.581
55.000	0.580
56.000	0.581
57.000	0.581
58.000	0.581
59.000	0.582
60.000	0.582

2. Operating time ekstrak kurma deglet noor

**Kinetics Data Print Report** 06/07/2022 09:18:26 AM

Time ( Minute )	RawData ...
0.000	0.626
1.000	0.626
2.000	0.626
3.000	0.626
4.000	0.625
5.000	0.626
6.000	0.626
7.000	0.626
8.000	0.627
9.000	0.626
10.000	0.626
11.000	0.626
12.000	0.626
13.000	0.625
14.000	0.625
15.000	0.626
16.000	0.625
17.000	0.625
18.000	0.625
19.000	0.625
20.000	0.626
21.000	0.625
22.000	0.625
23.000	0.626
24.000	0.625
25.000	0.626
26.000	0.626
27.000	0.626
28.000	0.626
29.000	0.625
30.000	0.626
31.000	0.626
32.000	0.626
33.000	0.626
34.000	0.626
35.000	0.626
36.000	0.627
37.000	0.627
38.000	0.627
39.000	0.627
40.000	0.628
41.000	0.628
42.000	0.629
43.000	0.629
44.000	0.630
45.000	0.630
46.000	0.631
47.000	0.631
48.000	0.632
49.000	0.633
50.000	0.632

**Kinetics Data Print Report** 06/07/2022 09:18:27 AM

Time ( Minute )	RawData ...
51.000	0.634
52.000	0.634
53.000	0.635
54.000	0.635
55.000	0.635
56.000	0.636
57.000	0.636
58.000	0.637
59.000	0.637
60.000	0.638

## Lampiran 8. Perhitungan aktivitas antioksidan dan IC<sub>50</sub>

### 1. Perhitungan aktivitas antioksidan kurma ajwa

Aktivitas antioksidan			
Konsentrasi	Replikasi	Absorbansi kontrol	Absorbansi sampel
10	Replikasi 1	0,721	0,671
8			0,664
6			0,621
4			0,618
2			0,589
10	Replikasi 2		0,679
8			0,666
6			0,625
4			0,620
2			0,585
10	Replikasi 3		0,675
8			0,663
6			0,623
4			0,615
2			0,586

$$\text{Inhibisi \%} = \frac{\text{absorbansi balnko} - \text{absorbansi sampel}}{\text{absorbansi balnko}} \times 100\%$$

#### ➤ Replikasi 1

$$10 \text{ ppm} = \frac{0,721 - 0,671}{0,721} \times 100\% = 6,935\%$$

$$8 \text{ ppm} = \frac{0,721 - 0,664}{0,721} \times 100\% = 7,906\%$$

$$6 \text{ ppm} = \frac{0,721 - 0,621}{0,721} \times 100\% = 13,870\%$$

$$4 \text{ ppm} = \frac{0,721 - 0,618}{0,721} \times 100\% = 14,286\%$$

$$2 \text{ ppm} = \frac{0,721 - 0,589}{0,721} \times 100\% = 18,308\%$$

#### ➤ Replikasi 2

$$10 \text{ ppm} = \frac{0,721 - 0,679}{0,721} \times 100\% = 5,825\%$$

$$8 \text{ ppm} = \frac{0,721 - 0,666}{0,721} \times 100\% = 7,628\%$$

$$6 \text{ ppm} = \frac{0,721 - 0,625}{0,721} \times 100\% = 13,315\%$$

$$4 \text{ ppm} = \frac{0,721 - 0,620}{0,721} \times 100\% = 14,008\%$$

$$2 \text{ ppm} = \frac{0,721-0,585}{0,721} \times 100\% = 18,863\%$$

➤ Replikasi 3

$$10 \text{ ppm} = \frac{0,721-0,675}{0,721} \times 100\% = 6,380\%$$

$$8 \text{ ppm} = \frac{0,721-0,663}{0,721} \times 100\% = 8,044\%$$

$$6 \text{ ppm} = \frac{0,721-0,623}{0,721} \times 100\% = 13,592\%$$

$$4 \text{ ppm} = \frac{0,721-0,615}{0,721} \times 100\% = 14,702\%$$

$$2 \text{ ppm} = \frac{0,721-0,586}{0,721} \times 100\% = 18,724\%$$

$$IC_{50} = \frac{50-a}{b}$$

$$Y = bx + a$$

➤ Replikasi 1

$$Y = 1,456x + 3,523$$

$$50 = 1,456x + 3,523$$

$$\frac{50-3,523}{1,456} = x$$

$$31,914 = x$$

$$IC_{50} = 31,914$$

➤ Replikasi 2

$$Y = 1,623x + 2,191$$

$$50 = 1,623x + 2,191$$

$$\frac{50-2,191}{1,623} = x$$

$$29,462 = x$$

$$IC_{50} = 29,462$$

➤ Replikasi 3

$$Y = 1,567x + 2,885$$

$$50 = 1,567x + 2,885$$

$$\frac{50-2,885}{1,567} = x$$

$$30,067 = x$$

$$IC_{50} = 30,067$$

➤ Rata-rata  $IC_{50}$  yang didapatkan =  $\frac{31,914+29,462+30,067}{3}$   
= 30,433 ppm.

## 2. Hasil perhitungan regresi linier antara konsentrasi dan %inhibisi pada ekstrak kurma ajwa

Konsentrasi	Replikasi	% inhibisi	Regresiliner	IC <sub>50</sub>	Rata-rata
10	Replikasi 1	6,935	a =3,523 b =1,456 r =0,968	31,914	30,433 ppm
8		7,906			
6		13,870			
4		14,286			
2		18,308			
10	Replikasi 2	5,825	a =2,191 b =1,623 r =0,978	29,462	
8		7,628			
6		13,315			
4		14,008			
2		18,863			
10	Replikasi 3	6,380	a =2,885 b =1,567 r =0,982	30,062	
8		8,044			
6		13,592			
4		14,702			
2		18,724			

## 3. Perhitungan aktivitas antioksidan kurma deglet noor

Aktivitas antioksidan			
Konsentrasi	Replikasi	Absorbansi kontrol	Absorbansi sampel
10	Replikasi 1	0,778	0,745
8			0,715
6			0,705
4			0,696
2			0,682
10	Replikasi 2		0,742
8			0,711
6			0,696
4			0,694
2			0,679
10	Replikasi 3		0,744
8			0,701
6			0,698
4			0,692
2			0,681

$$\text{Inhibisi \%} = \frac{\text{absorbansi balnko} - \text{absorbansi sampel}}{\text{absorbansi balnko}} \times 100\%$$

## ➤ Replikasi 1

$$10 \text{ ppm} = \frac{0,778 - 0,742}{0,778} \times 100\% = 4,627\%$$

$$8 \text{ ppm} = \frac{0,778 - 0,711}{0,778} \times 100\% = 8,612\%$$

$$6 \text{ ppm} = \frac{0,778 - 0,696}{0,778} \times 100\% = 10,540\%$$

$$4 \text{ ppm} = \frac{0,778 - 0,694}{0,778} \times 100\% = 10,797\%$$

$$2 \text{ ppm} = \frac{0,778 - 0,679}{0,778} \times 100\% = 12,725\%$$

## ➤ Replikasi 2

$$10 \text{ ppm} = \frac{0,778 - 0,745}{0,778} \times 100\% = 4,242\%$$

$$8 \text{ ppm} = \frac{0,778 - 0,715}{0,778} \times 100\% = 8,098\%$$

$$6 \text{ ppm} = \frac{0,778 - 0,705}{0,778} \times 100\% = 9,383\%$$

$$4 \text{ ppm} = \frac{0,778 - 0,696}{0,778} \times 100\% = 10,540\%$$

$$2 \text{ ppm} = \frac{0,778 - 0,682}{0,778} \times 100\% = 12,339\%$$

## ➤ Replikasi 3

$$10 \text{ ppm} = \frac{0,778 - 0,744}{0,778} \times 100\% = 4,370\%$$

$$8 \text{ ppm} = \frac{0,778 - 0,701}{0,778} \times 100\% = 9,897\%$$

$$6 \text{ ppm} = \frac{0,778 - 0,698}{0,778} \times 100\% = 10,283\%$$

$$4 \text{ ppm} = \frac{0,778 - 0,692}{0,778} \times 100\% = 11,054\%$$

$$2 \text{ ppm} = \frac{0,778 - 0,681}{0,778} \times 100\% = 12,468\%$$

$$\text{IC}_{50} = \frac{50 - a}{b}$$

$$Y = bx + a$$

## ➤ Replikasi 1

$$Y = 0,919x + 3,946$$

$$50 = 0,919x + 3,946$$

$$\frac{50 - 0,919}{3,946} = x$$

$$50,112 = x$$

$$\text{IC}_{50} = 50,112 \text{ ppm}$$

## ➤ Replikasi 2

$$Y = 0,932x + 3,329$$

$$50 = 0,932x + 3,329$$

$$\frac{50 - 3,329}{0,932} = x$$

$$50,083 = x$$

$$\mathbf{IC_{50} = 50,083 \text{ ppm}}$$

➤ Replikasi 3

$$Y = 0,868x + 4,409$$

$$50 = 0,868x + 4,409$$

$$\frac{50 - 4,409}{0,868} = x$$

$$52,548 = x$$

$$\mathbf{IC_{50} = 52,548 \text{ ppm}}$$

➤ Rata-rata  $IC_{50}$  yang didapatkan =  $\frac{50,112 + 50,083 + 52,548}{3}$   
 $= 50,879 \text{ ppm}$

**4. Hasil perhitungan regresi linier antara konsentrasi dan %inhibisi ekstrak kurma deglet noor.**

Konsentrasi	Replikasi	% inhibisi	Regresilinier	$IC_{50}$	Rata-rata
10	Replikasi 1	4,627	a =3,946 b =0,919 r =0,947	50,112	50,879
8		8,612			
6		10,540			
4		10,797			
2		12,725			
10	Replikasi 2	4,242	a =3,329 b =0,932 r =0,968	50,083	
8		8,098			
6		9,383			
4		10,540			
2		12,339			
10	Replikasi 3	4,370	a =4,409 b =0,868 r =0,887	52,548	
8		9,897			
6		10,283			
4		11,054			
2		12,468			



## Lampiran 9. Perbandingan IC<sub>50</sub> menggunakan SPSS uji *t test*

### 1. Hasil uji normalitas dan homogenitas

Tests of Normality						
	Kolmogorov-Smirnov <sup>a</sup>			Shapiro-Wilk		
	Statistic	df	Sig.	Statistic	df	Sig.
IC50	.295	3	.	.920	3	.452
IC50	.220	3	.	.986	3	.777

a. Lilliefors Significance Correction

### Oneway

Test of Homogeneity of Variances					
		Levene Statistic	df1	df2	Sig.
IC50	Based on Mean	.038	1	4	.856
	Based on Median	.000	1	4	.994
	Based on Median and with adjusted df	.000	1	3.794	.994
	Based on trimmed mean	.032	1	4	.866

### ANOVA

C50					
	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	646.986	1	646.986	412.413	.000
Within Groups	6.275	4	1.569		
Total	653.261	5			

## 2. Hasil uji *t test*

### T-Test

[DataSet 0]

#### Group Statistics

	Sampel	N	Mean	Std. Deviation	Std. Error Mean
Hasil Nilai IC50	Kurma ajwa	3	30.4793	1.27816	.73795
	Kurma Deglet nor	3	50.9117	1.41025	.81421

#### Independent Samples Test

		Levene's Test for Equality of Variances		t-Test for Equality of Means					95% Confidence Interval of the Difference	
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference	Lower	Upper
Hasil Nilai IC50	Equal variances assumed	.104	.764	-18.594	4	.000	-20.43233	1.09887	-23.48327	-17.38139
	Equal variances not assumed			-18.594	3.962	.000	-20.43233	1.09887	-23.49487	-17.36980

