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Lampiran 1. Determinasi



UPT-LABORATORIUM

Jl. Letjen Sutoyo, Mojosongo-Solo 57127 Telp. 0271-852518, Fax. 0271-853275

Nomor : 012/DET/UPT-LAB/30.6.2022

Hal : Hasil determinasi tumbuhan

Lamp. : -

Nama Pemesan : Merie Saphira Cahyani

NIM : 21154659A

Prodi : S1 Farmasi, Universitas Setia Budi, Surakarta

Nama Sampel : Nyamplung/*Calophyllum inophyllum* L.

HASIL DETERMINASI TUMBUHAN

Klasifikasi

Kingdom : Plantae

Super Divisi : Spermatophyta

Divisi : Magnoliophyta

Kelas : Magnoliopsida

Ordo : Malpighiales

Famili : Guttiferae/Calophyllaceae

Genus : *Calophyllum*

Species : *Calophyllum inophyllum*

Hasil Determinasi menurut Steenis, C.G.G.J.V, Bloembergen, H, Eyma, P.J. 1992 :

1b – 2b – 3b – 4b – 6b – 7b – 9b – 10b – 11b – 12b – 13b – 14b – 16a. Gol. 10. Daun tunggal, letak berhadapan. 239a - 240a. Familia : Guttiferae – 1b – 2. *Calophyllum inophyllum*.

Deskripsi:

Habitus : Pohon, tinggi 10-20 m, bengkok, besar, mengandung damar.

- Batang : Batang tidak tinggi cenderung sangat pendek, tidak lurus, berkayu keras, cabang rendah dekat tanah, ranting muda gundul, warna coklat atau putih kotor.
- Daun : Daun tunggal, bersilang berhadapan, tangkai daun 1,5-2,5 cm. Helaian daun memanjang sampai ellips bulat telur terbalik, ujung tumpul, membulat atau melekok ke dalam, tepi rata, seperti kulit, mengkilat, 10 – 21 kali 6 -11 cm.
- Bunga : Bunga berkelamin 2, berbau enak, garis tengah 2-3 cm. Bunga majemuk tandan, berbunga 7-13, letaknya di ketiak daun teratas. Anak tangkai bunga putih. Daun kelopak 4, 2 yang terdalam putih bersih. Daun mahkota 4, putih, memanjang, kerap kali tidak beraturan. Benangsari banyak, pangkalnya bersatu menjadi 4-6 berkas. Bakal buah kebanyakan merah. Tangkai putik membengkok saat kuncup, kepala putik bentuk perisai.
- Buah : Buah batu bentuk seperti peluru, ujung meruncing, garis tengah 2,5 – 3 cm. Kulit biji sangat tebal, warna hijau dan berubag kuning kecoklatan saat masak.
- Biji : Biji bulat, ujung meruncing mengandung minyak berwarna kuning.
- Akar : Akar tunggang, coklat, bulat.

Kepala UPT-LAB
Universitas Setia Budi



Asik Gunawan, Amdk

Surakarta, 30 Juni 2022

Penanggung jawab
Determinasi Tumbuhan



Dra. Dewi Sulistyawati. M.Sc.

Lampiran 2. Perhitungan Ekstrak**A. Ekstrak 1 (20%)**

Berat Simplisia	= 10,000 gram
Berat Wadah	= 54,861 gram
Berat Wadah + Ekstrak	= 63,726 gram
Berat Ekstrak	= 8,865 gram
Rendemen	= $\frac{8,865}{10,000} \times 100\% = 88,65\%$

B. Ekstrak II (40%)

Berat Simplisia	= 10,000 gram
Berat Wadah	= 54,878 gram
Berat Wadah + Ekstrak	= 63,382 gram
Berat Ekstrak	= 9,504 gram
Rendemen	= $\frac{9,504}{10,000} \times 100\% = 95,04\%$

C. Ekstrak III (60%)

Berat Simplisia	= 10,000 gram
Berat Wadah	= 54,944 gram
Berat Wadah + Ekstrak	= 64,811 gram
Berat Ekstrak	= 9,876 gram
Rendemen	= $\frac{9,876}{10,000} \times 100\% = 98,76\%$

Lampiran 3. Perhitungan Bobot Jenis

A. Ekstrak I (20%)

$$\text{Piknometer Kosong} = 34,0695 \text{ gram}$$

$$\text{Piknometer + Air} = 83,8199 \text{ gram}$$

$$\text{Piknometer + Zat Uji} = 67,3101 \text{ gram}$$

$$\text{Bobot Jenis} = \frac{67,3101 - 34,0695}{83,8199 - 34,0695} = \frac{33,2406}{49,7504} =$$

0,668

B. Ekstrak II (40%)

$$\text{Piknometer Kosong} = 34,2112 \text{ gram}$$

$$\text{Piknometer + Air} = 83,5552 \text{ gram}$$

$$\text{Piknometer + Zat Uji} = 67,1615 \text{ gram}$$

$$\text{Bobot Jenis} = \frac{67,1615 - 34,2112}{83,5552 - 34,2112} = \frac{32,9503}{49,344} =$$

0,667

C. Ekstrak III (60%)

$$\text{Piknometer Kosong} = 34,2102 \text{ gram}$$

$$\text{Piknometer + Air} = 83,5448 \text{ gram}$$

$$\text{Piknometer + Zat Uji} = 67,2814 \text{ gram}$$

$$\text{Bobot Jenis} = \frac{67,2814 - 34,2102}{83,5448 - 34,2102} = \frac{33,0712}{49,3346} =$$

0,670

Lampiran 4. Perhitungan Susut Pengerinan

Berat Kurst Kosong	= 64,921gram
Berat Simplisia	= 2,0009gram
Berat Kurst + Simplisia	= 66,9219gram

Perhitungan

$$1 \quad \frac{66,9219 - 66,490}{66,9219} \times 100 = 0,645\%$$

$$2 \quad \frac{66,9219 - 66,478}{66,9219} \times 100 = 0,663\%$$

$$3 \quad \frac{66,9219 - 66,472}{66,9219} \times 100 = 0,672\%$$

Lampiran 5. Fase Gerak

Fase gerak : Metanol : Kloroform : Eter (5 : 4 : 1)

$$\text{Metanol} = \frac{5}{10} \times 8 = 4 \text{ ml}$$

$$\text{Kloroform} = \frac{4}{10} \times 8 = 3,2 \text{ ml}$$

$$\text{Eter} = \frac{1}{10} \times 8 = 0,8 \text{ ml}$$

Lampiran 6. Perhitungan Rf

Ekstrak I

$$a : 3,7 \quad \text{Nilai Rf} : \frac{3,7}{4,6} = 0,804$$

$$b : 4,6$$

Ekstrak II

$$a : 3,6 \quad \text{Nilai Rf} : \frac{1,6}{4,6} = 0,782$$

$$b : 4,6$$

Ekstrak III

$$a : 3,5 \quad \text{Nilai Rf} : \frac{3,5}{4,6} = 0,760$$

$$b : 4,6$$

Baku Kuersetin

$$a : 3,7 \quad \text{Nilai Rf} : \frac{3,7}{4,6} = 0,804$$

$$b : 4,6$$

Lampiran 7. Pembuatan Baku Induk Kuersetin

$$\text{Kuersetin} = \frac{50\text{mg}}{50\text{ ml}} = 1000\text{ ppm}$$

Lampiran 8. Pembuatan Larutan Kerja Kuersetin

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$5 \cdot 1000 = 10 \cdot N_2$$

$$5000 = 10 N_2$$

$$N_2 = \frac{5000}{10}$$

$$N_2 = 500\text{ ppm}$$

Lampiran 9. Pembuatan Kurva Baku Kuersetin

1. Kuersetin 20 ppm

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$V_1 \cdot .500 = 10 \cdot 20$$

$$500V_1 = 200$$

$$V_1 = \frac{200}{500}$$

$$V_1 = 0,4\text{ mL}$$

2. Kuersetin 40 ppm

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$V_1 \cdot .500 = 10 \cdot 40$$

$$500V_1 = 400$$

$$V_1 = \frac{400}{500}$$

$$V_1 = 0,8\text{ mL}$$

3. Kuersetin 60 ppm

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$V_1 \cdot .500 = 10.60$$

$$500V_1 = 600$$

$$V_1 = \frac{600}{500}$$

$$V_1 = 1,2 \text{ mL}$$

4. Kuersetin 80 ppm

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$V_1 \cdot .500 = 10.80$$

$$500V_1 = 800$$

$$V_1 = \frac{800}{500}$$

$$V_1 = 1,6 \text{ mL}$$

5. Kuersetin 100 ppm

$$V_1 \cdot N_1 = V_2 \cdot N_2$$

$$V_1 \cdot .500 = 10.100$$

$$500V_1 = 1000$$

$$V_1 = \frac{1000}{500}$$

$$V_1 = 2 \text{ mL}$$

Lampiran 10. Penetapan Kadar Flavonoid

1. Persamaan Regresi

X	y	X ²	Y ²	XY
20	0,610	400	0,3721	12,2
40	1,073	1600	1,151329	42,92
60	1,365	3600	1,863225	81,9
80	1,832	6400	3,356224	146,56
100	2,143	10000	4,593449	214,3
Σ 300	Σ 6986	Σ 22.000	Σ 11,335327	Σ 503,88

$$Y = ax + b$$

$$a = \frac{n\Sigma XY - \Sigma x \cdot \Sigma y}{n\Sigma X^2 - (\Sigma X)^2} = -0,0281$$

$$b = \frac{\Sigma y - (a \times \Sigma X)}{n} = 0,02118$$

2. Konsentrasi Sampel

$$y = ax + b$$

$$y = 0,0281x + 0,02118$$

a. Ekstrak i

$$X = \frac{y-b}{a} = \frac{1,660-0,02118}{0,0281} = 58,3 \text{ mg/L}$$

b. Ekstrak ii

$$X = \frac{y-b}{a} = \frac{1,834-0,02118}{0,0281} = 64,5 \text{ mg/L}$$

c. Ekstrak iii

$$X = \frac{y-b}{a} = \frac{2,256-0,02118}{0,0281} = 79,5 \text{ mg/L}$$

3. Kadar Flavonoid Total

$$\text{Kadar b/b} = \frac{\text{Konsentrasi} \times V \times FP}{m}$$

Keterangan : FP: Faktor Pengenceran
 V : Volume Ekstrak (L)
 m : Massa Sampel (gram)

a. Ekstrak i

$$\text{Konsentrasi} = 58,3 \text{ mg/L}$$

$$V = 0,05 \text{ L}$$

$$m = 0,025$$

Perhitungan :

$$\frac{72,5 \frac{\text{mg}}{\text{L}} \times 0,05 \text{ L} \times 10}{0,025} = 1.166 \text{ mg/gram}$$

b. Ekstrak ii

$$\text{Konsentrasi} = 64,5 \text{ mg/L}$$

$$V = 0,05 \text{ L}$$

$$m = 0,025$$

Perhitungan :

$$\frac{64,5 \frac{\text{mg}}{\text{L}} \times 0,05 \text{ L} \times 10}{0,025} = 1.290 \text{ mg/gram}$$

c. Ekstrak iii

$$\text{Konsentrasi} = 79,5 \text{ mg/L}$$




$$V = 0,05 \text{ L}$$


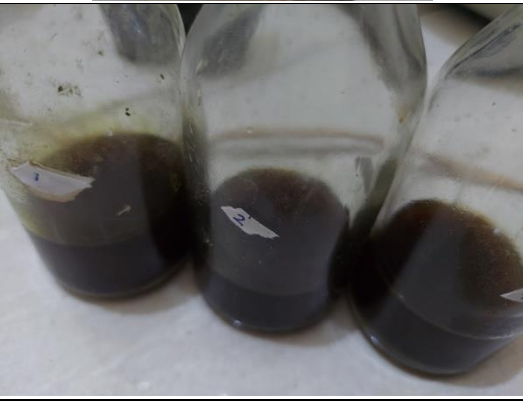
$$m = 0,025$$





Perhitungan :



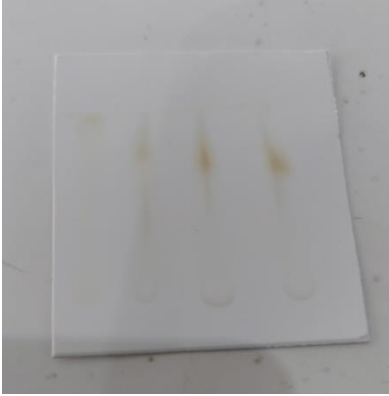

$$\frac{79,5 \frac{\text{mg}}{\text{L}} \times 0,05 \text{ L} \times 10}{0,025} = 1.590 \text{ mg/gram}$$



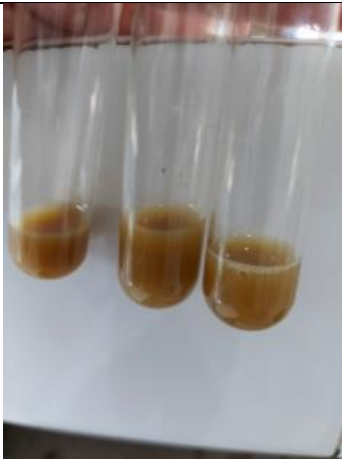
Lampiran 11. Gambar


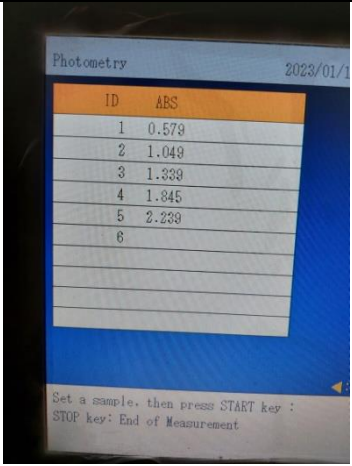
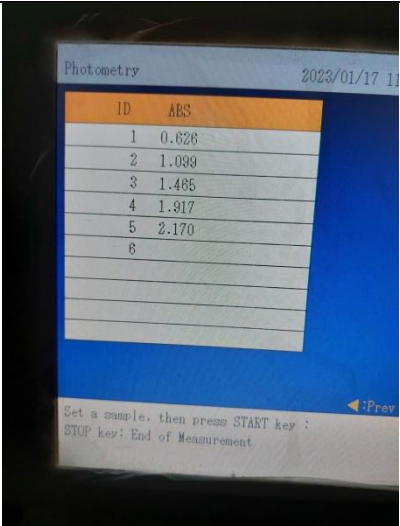
No	Gambar	Keterangan
1		Tanaman Nyamplung
2		Daun Nyamplung
3		Serbuk Simplisia

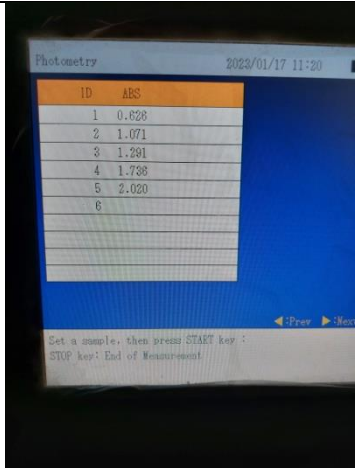
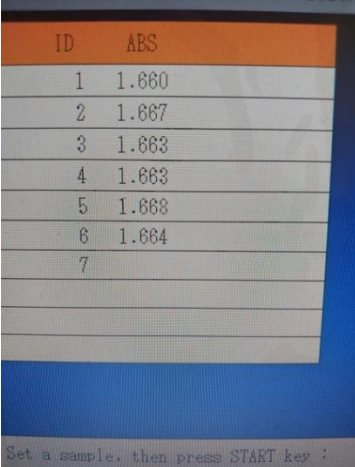
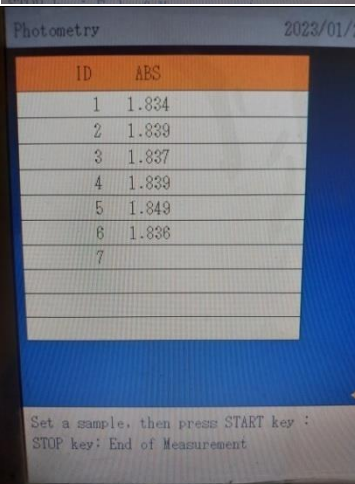
4	 <p>The top image shows the control panel of a QSONICA SONICATORS ultrasonic bath. The digital display shows 'Elapsed Time 01:00:00' and 'Sonication in OFF' with a progress bar. Below the display is a numeric keypad (0-9), a 'POWER' button, a 'TIMER' button, a 'HOLD' button, an 'AMPL' button, a 'CLEAR' button, and a 'REVIEW' button. The bottom image shows a glass beaker containing a dark liquid, with a metal probe inserted into it, placed inside the ultrasonic bath.</p>	Ekstraksi
5	 <p>The image shows three glass bottles containing dark brown liquid extracts. The bottles are arranged in a row on a white surface. The middle bottle has a small white label with the number '2' on it.</p>	Ekstrak

6			Susut Pengerinan i
7			Susut Pengerinan ii
8			Susut Pengerinan iii
9			Bobot Jenis i

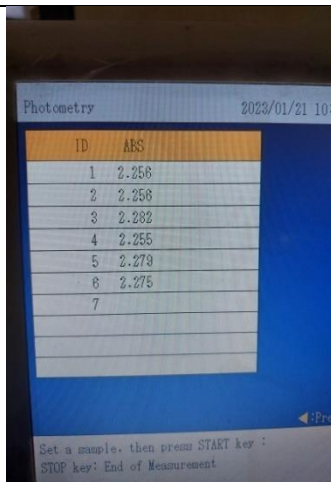
10		Bobot Jenis ii
11		Bobot Jenis iii
12		KLT
13		KLT 254

14		KLT 366
15		Uji Shinoda
16		Uji NaOH

17		Uji H ₂ SO ₄														
19	 <table border="1" data-bbox="518 681 772 933"> <thead> <tr> <th>ID</th> <th>ABS</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>0.579</td> </tr> <tr> <td>2</td> <td>1.049</td> </tr> <tr> <td>3</td> <td>1.339</td> </tr> <tr> <td>4</td> <td>1.845</td> </tr> <tr> <td>5</td> <td>2.239</td> </tr> <tr> <td>6</td> <td></td> </tr> </tbody> </table>	ID	ABS	1	0.579	2	1.049	3	1.339	4	1.845	5	2.239	6		Kurva Baku Kuersetin i
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1	0.579															
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6																
	 <table border="1" data-bbox="509 1145 762 1396"> <thead> <tr> <th>ID</th> <th>ABS</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>0.626</td> </tr> <tr> <td>2</td> <td>1.089</td> </tr> <tr> <td>3</td> <td>1.465</td> </tr> <tr> <td>4</td> <td>1.917</td> </tr> <tr> <td>5</td> <td>2.170</td> </tr> <tr> <td>6</td> <td></td> </tr> </tbody> </table>	ID	ABS	1	0.626	2	1.089	3	1.465	4	1.917	5	2.170	6		Kurva Baku Kuersetin replikasi ii
ID	ABS															
1	0.626															
2	1.089															
3	1.465															
4	1.917															
5	2.170															
6																

	 <p>Photometry 2023/01/17 11:30</p> <table border="1"><thead><tr><th>ID</th><th>ABS</th></tr></thead><tbody><tr><td>1</td><td>0.628</td></tr><tr><td>2</td><td>1.071</td></tr><tr><td>3</td><td>1.291</td></tr><tr><td>4</td><td>1.738</td></tr><tr><td>5</td><td>2.020</td></tr><tr><td>6</td><td></td></tr></tbody></table> <p>Set a sample, then press START key : STOP key: End of Measurement</p>	ID	ABS	1	0.628	2	1.071	3	1.291	4	1.738	5	2.020	6		<p>Kurva Baku Kuerstin replikasi iii</p>		
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1	0.628																	
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20	 <table border="1"><thead><tr><th>ID</th><th>ABS</th></tr></thead><tbody><tr><td>1</td><td>1.660</td></tr><tr><td>2</td><td>1.667</td></tr><tr><td>3</td><td>1.663</td></tr><tr><td>4</td><td>1.663</td></tr><tr><td>5</td><td>1.668</td></tr><tr><td>6</td><td>1.664</td></tr><tr><td>7</td><td></td></tr></tbody></table> <p>Set a sample, then press START key :</p>	ID	ABS	1	1.660	2	1.667	3	1.663	4	1.663	5	1.668	6	1.664	7		<p>Absorbansi Ekstrak 1</p>
ID	ABS																	
1	1.660																	
2	1.667																	
3	1.663																	
4	1.663																	
5	1.668																	
6	1.664																	
7																		
21	 <p>Photometry 2023/01/2</p> <table border="1"><thead><tr><th>ID</th><th>ABS</th></tr></thead><tbody><tr><td>1</td><td>1.834</td></tr><tr><td>2</td><td>1.839</td></tr><tr><td>3</td><td>1.837</td></tr><tr><td>4</td><td>1.839</td></tr><tr><td>5</td><td>1.849</td></tr><tr><td>6</td><td>1.836</td></tr><tr><td>7</td><td></td></tr></tbody></table> <p>Set a sample, then press START key : STOP key: End of Measurement</p>	ID	ABS	1	1.834	2	1.839	3	1.837	4	1.839	5	1.849	6	1.836	7		<p>Absorbansi Ekstrak 2</p>
ID	ABS																	
1	1.834																	
2	1.839																	
3	1.837																	
4	1.839																	
5	1.849																	
6	1.836																	
7																		

22



Photometry 2023/01/21 10:3

ID	ABS
1	2.258
2	2.256
3	2.282
4	2.255
5	2.279
6	2.275
7	

Set a sample, then press START key :
STOP key: End of Measurement

Absorbansi
Ekstrak 1